

## Enantioselective Synthesis of *ansa*-Zirconocenes

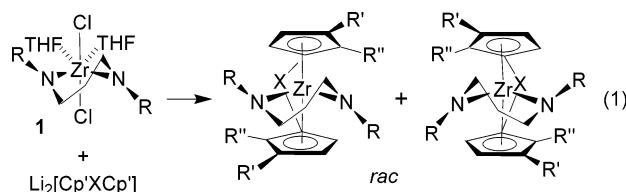
Matthew D. LoCoco and Richard F. Jordan\*

*Department of Chemistry, The University of Chicago, 5735 South Ellis Avenue, Chicago, Illinois 60637*

Received August 6, 2004; E-mail: rfjordan@uchicago.edu

Chiral “Brintzinger type” group 4 metal *ansa*-metallocene complexes exhibit high stereodirecting ability in many reactions.<sup>1,2</sup> However, *ansa*-metallocenes have been under-utilized in enantioselective catalysis, due in large part to the limited availability of enantiomerically pure catalysts.<sup>3,4</sup> Only a few *ansa*-metallocenes have been resolved, and most catalytic studies have used the original (EBTHI)ZrX<sub>2</sub> and (EBTHI)TiX<sub>2</sub> systems.<sup>5</sup> Here we describe a simple, efficient, and potentially general enantioselective synthesis of *ansa*-zirconocenes.

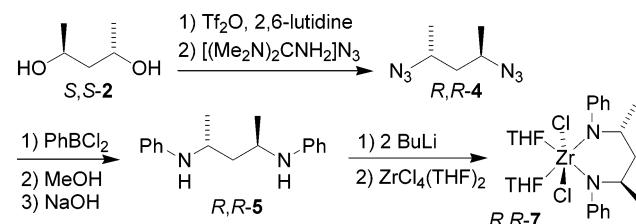
We recently reported a diastereoselective “chelate-controlled” synthesis of racemic *ansa*-zirconocenes, which is based on the reaction of Li<sub>2</sub>[Cp'Xcp'] salts with chelated bis-amide compounds Zr{RN(CH<sub>2</sub>)<sub>3</sub>NR}Cl<sub>2</sub>(THF)<sub>2</sub> (**1**; R = Ph, SiMe<sub>3</sub>), as shown in eq 1.<sup>6</sup> The twist conformation of the Zr{RN(CH<sub>2</sub>)<sub>3</sub>NR} ring in **1**



positions the N–R substituents on opposite sides of the N–Zr–N plane. This conformation is matched with the *rac*-metallocene structure but is incompatible with the *meso* isomer due to steric crowding between the N–R and Cp' groups, which leads to complete selectivity for the *rac* product.

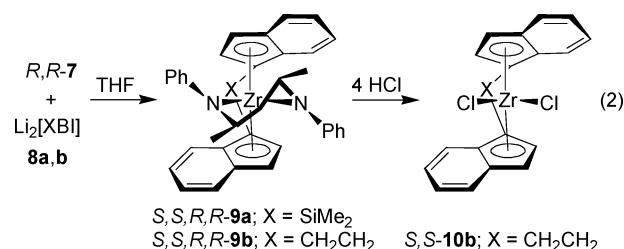
We reasoned that incorporation of substituents at C1 and C3 of the bis-amide ligand of **1** in a stereo-defined manner would favor one enantiomer of the twist conformation in the resulting metallocene, and possibly provide a means of controlling enantioselectivity. To test this concept, we prepared Zr{(2*R*,4*R*)-PhNCHMeCH<sub>2</sub>CHMeNPh}Cl<sub>2</sub>(THF)<sub>2</sub> (*R,R*-**7**; Scheme 1), a chiral analogue of **1**.

**Scheme 1**



Conversion of (2*S*,4*S*)-pentanediol (*S,S*-**2**) to the ditriflate (*S,S*-**3**), followed by reaction with [(Me<sub>2</sub>N)<sub>2</sub>CNH<sub>2</sub>]N<sub>3</sub>, yields (2*R*,4*R*)-diazidopentane (*R,R*-**4**) quantitatively.<sup>7</sup> Reaction of (*R,R*-**4**) with PhBCl<sub>2</sub> followed by methanol workup and neutralization yields *N,N'*-diphenyl-(2*R*,4*R*)-pentanediamine (*R,R*-**5**, 76%, >99.5% ee).<sup>8</sup> Deprotonation of *R,R*-**5** with 2 equiv of *n*-BuLi yields Li<sub>2</sub>[(2*R*,4*R*)-PhNCHMeCH<sub>2</sub>CHMeNPh] (*R,R*-**6**) (99%). Reaction of ZrCl<sub>4</sub>(THF)<sub>2</sub> and 1 equiv of *R,R*-**6** affords *R,R*-**7** (84% isolated).

The reaction of *R,R*-**7** with Li<sub>2</sub>[SBI](Et<sub>2</sub>O) (**8a**)<sup>5</sup> or Li<sub>2</sub>[EBI](Et<sub>2</sub>O) (**8b**)<sup>5</sup> in THF affords the corresponding metallocenes *S,S*-(SBI)Zr{(2*R*,4*R*)-PhNCHMeCH<sub>2</sub>CHMeNPh} (*S,S,R,R*-**9a**) or *S,S*-(EBI)Zr{(2*R*,4*R*)-PhNCHMeCH<sub>2</sub>CHMeNPh} (*S,S,R,R*-**9b**) in >95% isolated yield (eq 2). The <sup>1</sup>H NMR spectra of isolated **9a,b** display,

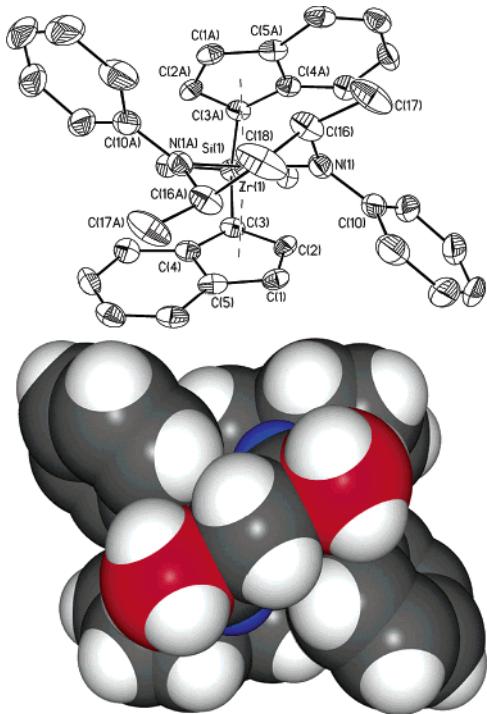


in each case, a single set of resonances and show that a single diastereomer is present. These results establish that the metallocene units of **9a,b** are formed in >99% ee. No significant side products were detected in these reactions. The use of THF as the solvent is important, as reactions of *R,R*-**7** with **8a,b** in Et<sub>2</sub>O yielded mixtures of products. Compound **9b** was converted to the corresponding enantiomerically pure dichloride *S,S*-**10b** (91% isolated, >99% ee) by reaction with HCl in Et<sub>2</sub>O. The chiral diamine *R,R*-**5** was recovered.

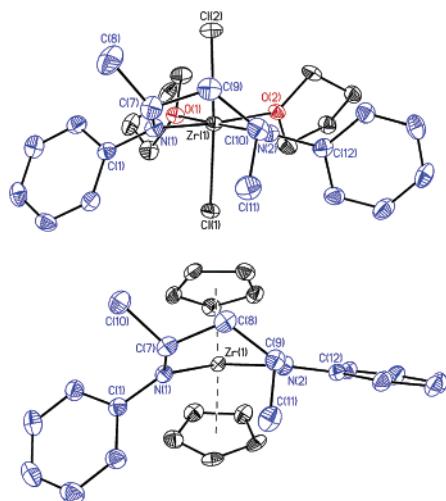
The configurations of *S,S,R,R*-**9a** and *S,S*-**10b** were established by X-ray crystallography. Additionally, *S,S*-**10b** was hydrogenated to the known *S,S*-(EBTHI)ZrCl<sub>2</sub> (*S,S*-**11b**), and the configuration and optical purity of *S,S*-**11b** were established from the optical rotation of this species and NMR analysis of its bis-(*R*)-*O*-acetyl-mandelate derivative.<sup>3,5,9</sup> These results confirm the configurations and optical purity of *S,S*-**10b** and *S,S,R,R*-**9b**.

The molecular structure of *S,S,R,R*-**9a** is shown in Figure 1. The Zr{PhNCHMeCH<sub>2</sub>CHMeNPh} ring adopts a twist  $\lambda$  conformation,<sup>10</sup> which is matched to the *S,S*-metallocene unit. The C1 and C3 methyl groups (C17 and C17A) occupy pseudoequatorial positions and point away from the metallocene unit, while, as illustrated in the space-filling view, the pseudoaxial C1–H and C3–H hydrogens point directly toward the indenyl rings. To form the *R,R*-metallocene without severe N–Ph/indenyl steric crowding, the bis-amide chelate ring would have to invert to the  $\delta$  configuration. However, in this case, the C1 and C3 methyl groups would occupy pseudoaxial positions, and severe methyl/indenyl steric crowding would result.

While these structural data and steric arguments suggest that the matched metallocenes *S,S,R,R*-**9a** and *S,S,R,R*-**9b** may be more stable than the other stereoisomers of these complexes, the operative stereocontrol mechanism in eq 2 is unknown. X-ray crystallographic analyses show that the chelate rings of *R,R*-**7** and Cp<sub>2</sub>Zr{(2*R*,4*R*)-PhNCHMeCH<sub>2</sub>CHMeNPh} (*R,R*-**12**) adopt twist-boat conformations (Figure 2), suggesting that this may be the most stable conformation of these rings in the absence of significant steric interactions with other ligands. However, both *R,R*-**7** and *R,R*-**12** exhibit *C*<sub>2</sub> symmetry on the NMR time scale at 25 °C, which



**Figure 1.** Molecular structure and space-filling view of *S,S,R,R-9a*. H atoms are omitted from the ORTEP view. The axial C1–H and C3–H hydrogen atoms are colored blue, and the equatorial methyl groups are colored red in the space-filling view. Bond distances (Å): Zr–N(1) 2.095(4), Zr–cent 2.301. Bond angles (deg): N(1)–Zr(1)–N(1A) 90.2(2), cent(1)–Zr(1)–cent(2) 125.5.



**Figure 2.** Molecular structures of *R,R-7* (top) and *R,R-12* (bottom). H atoms are omitted. Bond distances (Å) and angles (deg): *R,R-7*, Zr(1)–N(1) 2.028(2), Zr(1)–N(2) 2.046(2), Zr(1)–O(1) 2.348(2), Zr(1)–O(2) 2.319(2), Zr(1)–Cl(1) 2.484(2), Zr(1)–Cl(2) 2.504(2), N(1)–Zr(1)–N(2) 92.01(7), Cl(1)–Zr(1)–Cl(2) 161.91(2), O(1)–Zr(1)–O(2) 77.3(5); *R,R-12*, Zr–N(1) 2.087(2), Zr–N(2) 2.173(2), Zr–cent(1) 2.258, Zr–cent(2) 2.287, N(1)–Zr–N(2) 91.62(8), cent(1)–Zr–cent(2) 124.1.

indicates that the chelate rings are flexible. Given the potential complexity of the mechanism of substitution of the chloride and

THF ligands of **7** by the incoming indenyl ligands,<sup>11</sup> more detailed studies will be required to understand the origin of stereocontrol in these reactions.

The methodology described here should be useful for the enantioselective synthesis of other bis-indenyl metallocenes and has the advantage that both enantiomers are available by using either *R,R*- or *S,S*-**7**. Tuning the steric interactions between the chiral bis-amide directing ligand and the metallocene framework by variation of the C1, C3, and N-substituents should enable broad application of this strategy.<sup>6b</sup>

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**Supporting Information Available:** Experimental procedures and characterization data for new compounds (PDF); X-ray crystallographic data for *R,R-7*, *S,S,R,R-9a*, *S,S-10b*, and *R,R-12* (CIF). This material is available free of charge via the Internet at <http://pubs.acs.org>.

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- (5) Abbreviations: EBTHI, 1,2-ethylene-bis(tetrahydroindenyl); Cp', generic cyclopentadienyl or indenyl; X, generic bridge; SBI, Me<sub>2</sub>Si(indenyl)<sub>2</sub>; EBI, 1,2-ethylene-bis(indenyl); cent, Cp' ring centroid.
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