

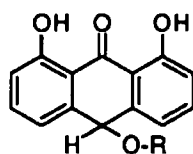
Electron-Impact Induced and Thermal Decomposition of Dithranol Derivatives, III ¹⁾:

Fragmentations of 10-Alkoxy- and 10-Benzyloxydithranol Radical Cations*)

Elektronenstoß-induzierter und thermischer Zerfall von Dithranol-Derivaten, 3. Mitt. ¹⁾:
Fragmentierungen der Molekülionen von 10-Alkoxy- und 10-Benzyloxy-dithranolHsu-Shan Huang^{a)+}, Klaus K. Mayer^{b)}, and Wolfgang Wiegreb^{a)}Institut für Pharmazie^{a)} and Zentrale Analytik^{b)}, NWF IV, Universität Regensburg, Universitätsstr. 31, D-93040 Regensburg, Germany

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Derivatives of dithranol [anthralin; 1,8-dihydroxy-9(10*H*)-anthracenone] with substituents at C-10 are of interest in the search for antipsoriatic remedies ²⁾. While a wealth of compounds with a C-10 carbon bond was tested ³⁾, only little is known about substituted anthralins with side-chains bound by hetero-atoms directly to C-10 ⁴⁻⁷⁾. Our studies on 10-alkylthio- and 10-arylthio-1,8-dihydroxy-9(10*H*)-anthracenones ^{1, 6, 8)} caused us to synthesize *i.a.* the dithranol ethers 1–3 and their deuterium-labelled analogues 1a and 3a–3c.



R=	1	CH ₃
	1a	CD ₃
	2	C ₂ H ₅
	3	CH ₂ C ₆ H ₅
	3a	CD ₂ C ₆ H ₅
	3b	CH ₂ C ₆ D ₅
	3c	CD ₂ C ₆ D ₅

Scheme 1

Under standard electron-impact ionization conditions (70/12 eV), *ipso*-cleavage ⁹⁾ processes dominate dissociations in the ion source, in the case of 1 – 3 giving rise to strong ions at *m/z* 241 (*M* – [•]R) and *m/z* 225 (*M* – [•]OR), while in particular molecular ions of 3 additionally decompose to rearrangement ions at *m/z* 240, 226, and 92 (Table 1).

In the spectra of 1, 1a, and 2 *m/z* 240 ions arise by loss of H[•] from *m/z* 241; *m/z* 226 ions come directly from the molecular ions by ejecting CH₂O (1a: CD₂O, → *m/z* 227) or CH₃CHO, respectively, as confirmed for metastable *M*^{•+} (*B/E* = const. linked scans). α-Cleavage (*M* – [•]CH₃) and elimination of C₂H₄, preponderant features of ionized ethyl ethers ⁹⁾, could not be detected. Metastable *M*^{•+} of 3 decompose to *m/z* 331 (–H[•]), *m/z* 241 (–[•]C₇H₇), *m/z* 240 (–C₇H₈), *m/z* 226 (–C₆H₅CHO), and *m/z* 92 (C₇H₈^{•+}) (Table 2).

Table 2: MIMS of 3 (*m/z* 332) (*m/z*; % rel. int.)^{a)}.

	331 (100)	254 (2)	241 (80)	240 (5)	226 (65)	92 (3)
70 eV						
12 eV	331 (100)	254 (3)	241 (95)	240 (2)	226 (70)	92 (5)

^{a)} average of 10 runsTable 1: Partial EIMS (70/12 eV; % rel. int.) of 1 – 3 (¹³C corr.)^{a)}.

<i>m/z</i> No.	332	331	270	269	256	255	242	241	240	226	225	197	92	91
1 (256)	–	–	–	–	66/ 100	2/4	–	49/ 11	8/5	15/ 12	100/ 36	49/-	–	–
2 (270)	–	–	36/ 100	1/1	–	1/-	2/3	65/ 12	20/ 14	51/ 52	100/ 23	30/-	–	–
3 (332)	3/4	<1/ <1	–	–	–	–	–	47/ 27	100/ 66	98/ 100	49/ 7	34/-	25/ 18	34/ 10

^{a)} average of 20 runs.^{*)} On leave from the Institute of Pharmacy, National Defense Medical Center, Taipei, Taiwan, Republic of China.^{*)} Dedicated to Prof. Dr. Dr.h.c. F. Eiden, München, on the occasion of his 70th birthday.

The data of the specifically deuterated ethers 3a – 3c support the concept of multiple H-migrations in their molecular ions before or in the course of fragmentation (Table 3):

m/z 226 ions are shifted mainly to *m/z* 227 in the MIMS of 3a and 3c, they keep their position in the case of 3b. However,

Table 3: MIMS (70/12 eV) of **3**, **3a**, **3b**, and **3c**; % $\Sigma(m/z$ 226 to m/z 229) and % $\Sigma(m/z$ 240 to m/z 243)^{a)}.

No./ m/z	226	227	228	229	240	241	242	243
3 (332)	100/100	—	—	—	5/7	95/93	—	—
3a (334)	—	95/94	5/6	—	11/9	85/80	3/8	1/3
3b (337)	81/76	15/18	4/6	—	10/6	50/45	35/41	5/8
3c (339)	2/3	87/79	8/13	3/5	8/7	54/50	33/36	5/7

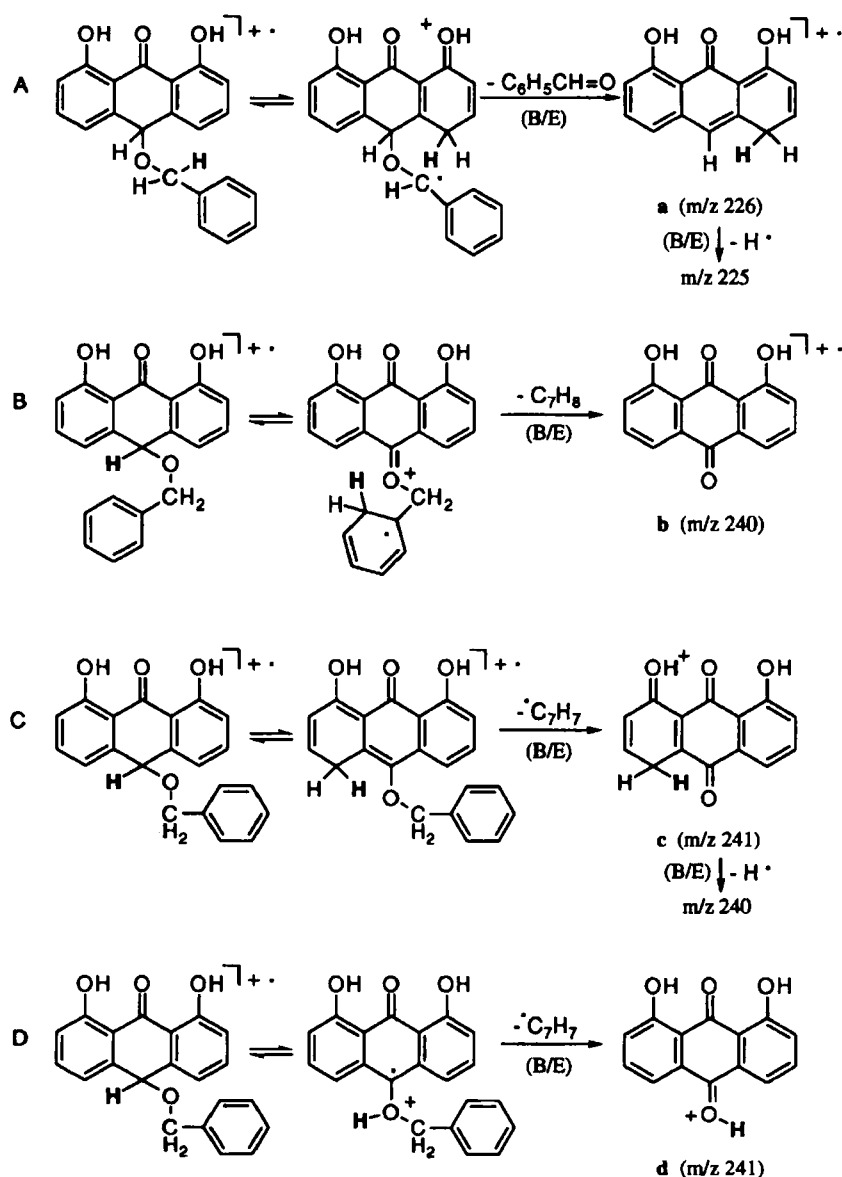
^{a)} average of 10 runs

signals of minor to medium intensity at m/z 226 (**3c**), m/z 227 (**3b**), and m/z 228 and 229 (**3b**, **3c**) substantiate partial exchange of H-atoms of the 'dithranol' moiety and the benzyloxy group.

The ions at m/z 241 are partly shifted to m/z 242 and 243 (**3a–3c**) with a strong increase of % Σ -values (m/z 242) in the case of **3b** and **3c** as compared with **3a** due to D-atom

migration from the benzylic CD₂-group (**3a**, **3c**) and to a higher extent from the aromatic nucleus (**3b**, **3c**) into the 'dithranol' part.

A mechanistic rationalization is given by equations A – D in Scheme 2, where reversible H-exchange between the positions C - 4/5 and C - 10, the benzylic methylene and the phenyl group is depicted.

**Scheme 2**

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Experimental Part

EIMS (70/12 eV) and MIMS: Finnigan MAT 95 double-focusing instrument. Samples were introduced by the direct insertion probe (quartz crucibles) at $T = 100^\circ\text{C}$; ion source temp. $100\text{--}120^\circ\text{C}$. High resolution measurements were performed with $m/\Delta m = 15\,000$.—Melting points: Büchi 510 melting point apparatus, uncorrected.— ^1H -NMR spectra: Varian EM 390 (90 MHz) or Bruker Spectrospin WM 250 spectrometer (250 MHz), TMS as int. standard.—Fourier-transform IR spectra (KBr): Nicolet 510M FT-IR spectrometer.—The D-content of the samples was measured by MS.

$[\alpha, \alpha\text{-D}_2]$ -Benzyl alcohol ¹⁾

A solution of benzoyl chloride (11.2 g, 80 mmole) in 20 ml of anhydrous ether was added dropwise with good stirring over a period of 30 min to a suspension of LiAlD_4 (1.89 g, 45 mmole) in 50 ml of anhydrous ether. After stirring for 2 h, water was added to destroy excess reagent followed by 10 % H_2SO_4 (10 ml). The solution was extracted twice with ether (2×100 ml). The extracts were dried (Na_2SO_4) and then distilled. Yield 8.1 g (92 %).—MS: m/z 110 (100 %); 99 % d_2 , 1 % d_1 .

$[2,3,4,5,6\text{-D}_5]$ -Benzyl alcohol

A solution of $[2,3,4,5,6\text{-D}_5]$ -benzoic acid (Aldrich) (1 g, 7.9 mmole) in 20 ml of anhydrous ether was added dropwise with good stirring over a period of 30 min to a suspension of LiAlH_4 (0.5 g, 13 mmole) in 20 ml of anhydrous ether. After stirring for 4 h, water was added, followed by 10 % H_2SO_4 (10 ml).—Yield 0.8 g (90 %).—MS: m/z 113 (100 %); 98 % d_5 , 2 % d_4 .

$[\alpha, \alpha, 2, 3, 4, 5, 6\text{-D}_7]$ -Benzyl alcohol

A solution of $[2,3,4,5,6\text{-D}_5]$ -benzoic acid (Aldrich) (1 g, 7.9 mmole) in 20 ml of anhydrous ether was added dropwise with good stirring over a period of 30 min to a suspension of LiAlD_4 (0.5 g, 12 mmole) in 20 ml of anhydrous ether. After stirring for 4 h, water was added, followed by 10 % H_2SO_4 (10 ml).—Yield 0.8 g (88 %).—MS: m/z 115 (57 %); 97 % d_7 , 2 % d_6 , 1 % d_5 .

10-Alkoxy and 10-Benzyloxy-1,8-dihydroxy-9-(10H)-anthracenone (1, 2, 3, 3a, 3b, 3c); General Procedure

To a solution of 10-bromo-1,8-dihydroxy-9-(10H)-anthracenone ¹⁰⁾ (305 mg, 1.0 mmole) and 0.1 ml of trifluoroacetic acid in dry CH_2Cl_2 (20 ml) was added methanol (20 ml), ethanol (20 ml) or benzyl alcohol (1 g, 1.5 mmole) under N_2 . The reaction mixture was stirred at room temp. for 4 h. Removal of the solvent and recrystallization of the residue from CH_2Cl_2 gave solid products.

10-Methoxy-1,8-dihydroxy-9-(10H)-anthracenone (1)

Mp. $92\text{--}93^\circ\text{C}$; lit.: $77\text{--}78^\circ\text{C}$ ^{7b)}.—FT-IR (KBr): $1636 (\text{CO}\cdots\text{HO}) \text{ cm}^{-1}$.—EIMS (m/z ; % rel. int.): 256 (66/100), 241 (49/11), 240 (8/5), 226 (15/12), 225 (100/36), 197 (49/-).—MIMS (B/E, m/z 256 (M^{++})): 241 (90), 226 (35), 225 (100).

10-[D₃]-Methoxy-1,8-dihydroxy-9-(10H)-anthracenone (1a)

Mp. $92\text{--}93^\circ\text{C}$.—FT-IR (KBr): $1635 (\text{CO}\cdots\text{HO}) \text{ cm}^{-1}$.—EIMS (m/z ; % rel. int.): 259 (68/100), 241 (60/10), 240 (9/4), 227 (37/21), 225 (100/25), 197 (74/-).—MIMS (B/E, m/z 259 (M^{++})): 241 (100), 227 (20), 225 (75).

10-Ethoxy-1,8-dihydroxy-9-(10H)-anthracenone (2)

Mp. $98\text{--}99^\circ\text{C}$.—FT-IR (KBr): $1630 (\text{CO}\cdots\text{HO}) \text{ cm}^{-1}$.—EIMS (m/z ; % rel. int.): 270 (36/100), 241 (65/12), 240 (21/14), 226 (56/36), 225 (100/25), 197 (51/-).—MIMS (B/E, m/z 270 (M^{++})): 269 (100), 255 (3), 242 (2), 241 (72), 226 (25), 225 (20).

10-Benzyloxy-1,8-dihydroxy-9-(10H)-anthracenone (3)

Mp. 102°C .—FT-IR (KBr): $1630 (\text{CO}\cdots\text{HO}) \text{ cm}^{-1}$.— ^1H -NMR (250 MHz, CDCl_3): δ (ppm) = 4.15 (s, 2H, CH_2), 5.87 (s, 1H, 10-H), 7.00 (d, $J=8.5$ Hz, 2H, 2-H, 7-H), 7.23 (m, $J=8.5$ Hz, 7H, C_6H_5 , 4-H, 5-H), 7.58 (t, $J=8.5$ Hz, 3-H, 6-H), 12.18 (s, 2H, 1-OH, 8-OH).— $\text{C}_{21}\text{H}_{16}\text{O}_4$ (332.3) Calcd. C 75.9 H 4.85 Found C 75.5 H 4.83. —EIMS (m/z ; % rel. int.): 332 (3/4), 254 (1/1), 241 (47/27), 240 (100/66), 226 (98/100), 225 (49/77), 197 (34/-), 92 (25/18), 91 (34/10).—MIMS (B/E, m/z 332 (M^{++})): 331 (100/100), 254 (2/3), 241 (80/95), 240 (5/2), 226 (65/70), 92 (3/5).—(B/E, m/z 241) : 240 (100).

$[\alpha, \alpha\text{-D}_2]$ -10-Benzyloxy-1,8-dihydroxy-9-(10H)-anthracenone (3a)

Mp. 102°C .— $\text{C}_{21}\text{H}_{14}\text{D}_2\text{O}_4$ Calcd. 334.1174 Found 334.1173. —EIMS: (m/z ; % rel. int.): 334 (13/22), 256 (4/6), 241 (98/98), 240 (73/76), 227 (94/100), 226 (39/30), 225 (100/16), 197 (53/-), 94 (22/18), 93 (23/13), 92 (5/1).—MIMS: (B/E, m/z 334 (M^{++})): 333 (30), 256 (2), 242 (4), 241 (100), 240 (15), 228 (2), 227 (40), 94 (1).—(B/E, m/z 241): 240 (100), 239 (1), 213 (1), 212 (2).—(B/E, m/z 227): 226 (100), 199 (1), 198 (1), 197 (2).

$[2,3,4,5,6\text{-D}_5]$ -10-Benzyloxy-1,8-dihydroxy-9-(10H)-anthracenone (3b)

Mp. 102°C .— $\text{C}_{21}\text{H}_{11}\text{D}_5\text{O}_4$ Calcd. 337.1362 Found 337.1364. —EIMS: (m/z ; % rel. int.): 337 (11/3), 254 (2/1), 242 (7/6), 241 (52/24), 240 (65/48), 227 (1/16), 226 (100/100), 225 (56/14), 197 (21/-), 97 (18/9), 96 (26/7), 95 (3/1).—MIMS: (B/E, m/z 337 (M^{++})): 336 (≈ 1), 335 (12), 254 (1), 242 (55), 241 (75), 240 (15), 228 (2), 227 (20), 226 (100), 97 (1).—(B/E, m/z 242): 241 (100), 240 (2), 213 (2), 212 (3).

$[\alpha, \alpha, 2, 3, 4, 5, 6\text{-D}_7]$ -10-Benzyloxy-1,8-dihydroxy-9-(10H)-anthracenone (3c)

Mp. 102°C .— $\text{C}_{21}\text{H}_9\text{D}_7\text{O}_4$ Calcd. 339.1488 Found 339.1487. —EIMS: (m/z ; % rel. int.): 339 (9/19), 256 (4/4), 242 (24/26), 241 (72/53), 240 (38/36), 227 (74/100), 226 (30/19), 225 (100/20), 197 (32/-), 99 (16/20), 98 (17/3), 97 (3/1).—MIMS: (B/E, m/z 339 (M^{++})): 338 (5), 256 (3), 242 (65), 241 (100), 240 (10), 228 (2), 227 (55), 226 (1), 99 (2).—(B/E, m/z 242): 241 (100), 213 (1), 212 (1).—(B/E, m/z 241): 240 (100), 213 (1), 212 (2).

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