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A Simple Synthesis of a Pillar[*n*]arene Building Block – 1,4-bis(4-Bromobenzyl)benzene[†]

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Recent years have seen a veritable explosion of activity in the synthesis of “pillar[*n*]arenes”¹ (previously known as “[*n*]-paracyclophanes”) and their application in the construction of host-guest molecules, supramolecular nanostructures, molecular machines, biological catalysts, and other novel materials. The field has been extensively reviewed,^{2–10} and recent work has expanded this exciting new area of supramolecular chemistry.^{11–14}

We now describe a two-step synthesis of 1,4-bis(4-bromobenzyl)benzene (**1**) from readily available starting materials (Scheme 1). We previously prepared and utilized this building block in the inaugural synthesis of [1₅]- and [1₆]-paracyclophanes, but without experimental details,¹⁵ and it is the purpose of this paper to provide such details. Monolithiation¹⁶ of 1,4-dibromobenzene (**2**) with *n*-BuLi in THF at –78 °C generated 4-lithiobromobenzene (**3**), which was treated with terephthalaldehyde (**4**) to give diol **5**, presumably as a mixture of diastereomers. Treatment of diol **5** with NaBH₄/trifluoroacetic acid¹⁷ gave **1** in high yield.

We believe our method to be superior in terms of yield, simplicity, and cost of starting materials to the procedures of Galun¹⁸ and Stephan,¹⁹ which are shown in Schemes 2 and 3, respectively. Moreover, *ortho* isomers are reported in Stephan’s synthesis and are likely to occur in both of Galun’s syntheses, but are precluded by our method.

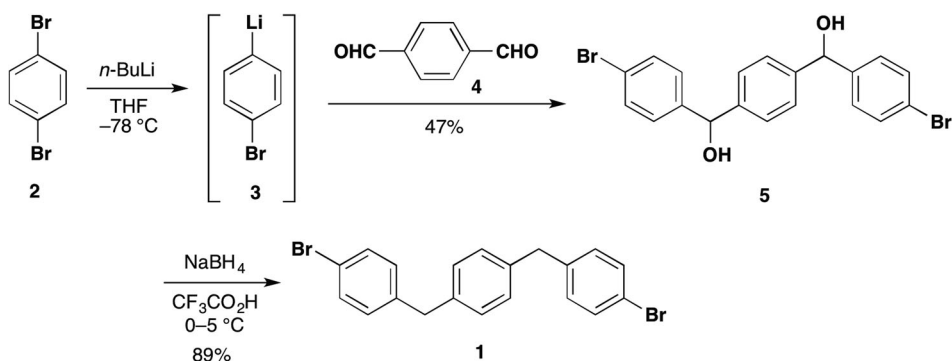
Experimental section

Melting points were obtained on a Thomas-Hoover Mel-Temp capillary apparatus and are uncorrected. Infrared spectra (IR) were recorded on a Perkin-Elmer Model 599 spectrophotometer. Liquids were measured on NaCl plates (neat), while KBr pellets were used for solids. Nuclear Magnetic Resonance spectra (NMR) were recorded on a Perkin-Elmer R-24 or a Varian XL300 spectrometer using TMS as an internal standard. Low resolution mass spectra were obtained with a Finnigan EI-CI gas chromatograph-mass spectrometer, and high-resolution mass spectra were obtained at the NIH Regional Facility at the Massachusetts Institute of Technology. Thin Layer Chromatography (TLC) was carried out on precoated (0.2 mm) Silica Gel 60 F₂₅₄ S

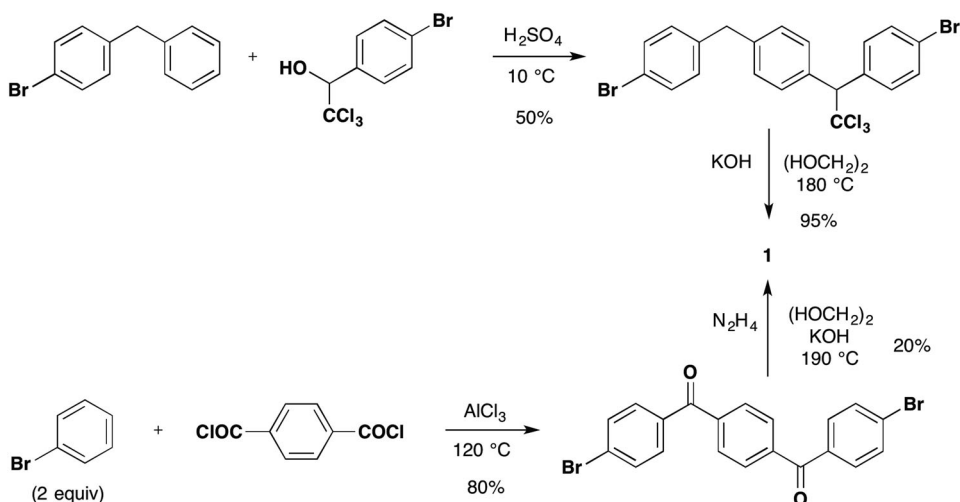
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[†]Dedicated to the memory of Max L. Deinzer – climbing partner extraordinaire and a consummate chemist.

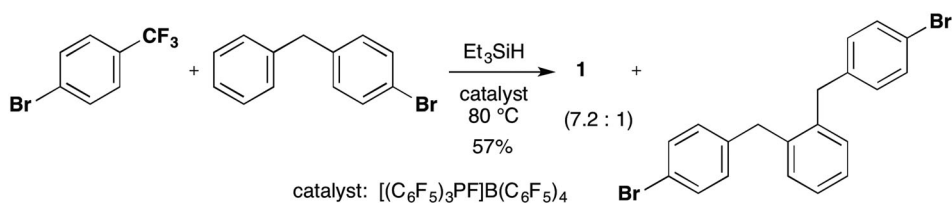
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Scheme 1. Present synthesis of 1,4-bis(4-bromobenzyl)benzene.



Scheme 2. Two syntheses of 1 by Galun and co-workers.



Scheme 3. Synthesis of 1 by Stephan and co-workers.

plastic sheets. Microanalysis was performed by Atlantic Microlabs, Atlanta, Georgia. Sodium borohydride was obtained from Morton Thiokol and was oven dried at 110 °C before use. Trifluoroacetic acid was provided by Halocarbon, Hackensack, New Jersey, and was distilled before use. Tetrahydrofuran was dried over and distilled from sodium/benzophenone. The *n*-BuLi was purchased from Aldrich and standardized by titration against 2,5-dimethoxybenzyl alcohol.

1,4-Phenylenebis(4-bromophenylmethanol) (5)

To a magnetically stirred solution of *p*-dibromobenzene (**2**) (1.03 g, 4.36 mmol) in dry THF (15 mL) at -78°C was added quickly via syringe a solution of *n*-BuLi/hexane (1.52 M, 2.90 mL, 4.41 mmol). The resulting solution was stirred at -78°C for 5 min, then a solution of terephthalaldehyde (**4**) (0.29 g, 2.2 mmol) in dry THF (10 mL) was added quickly via syringe. The resulting mixture was allowed to warm to 25°C , stirred at 25°C for 24 h, then poured into H_2O (75 mL), diluted with brine (100 mL), and extracted with Et_2O (3 x 100 mL). The combined extracts were dried (Na_2SO_4), filtered, and concentrated *in vacuo* to afford a yellow oil (0.84 g). Flash chromatography (2:1 hexane/ Et_2O) gave **5** (0.46 g, 47%) as a white solid (presumed mixture of diastereomers) which was recrystallized from hexane/ Et_2O : mp $141\text{--}142^{\circ}\text{C}$; ^1H NMR (acetone- d_6) δ 7.4–7.3 (m, 12H), 5.7 (broad, 2H); ^{13}C NMR (DMSO- d_6) δ 145.0, 143.8, 130.8, 128.3, 126.0, 119.6, 73.3; IR (CHCl_3): 3180 cm^{-1} (broad).

Anal. Calcd for $\text{C}_{20}\text{H}_{16}\text{Br}_2\text{O}_2$: C, 53.60; H, 3.60; Br, 35.66. Found: C, 54.48; H, 3.92; Br, 36.12. HRMS Calcd for $\text{C}_{20}\text{H}_{16}\text{Br}_2\text{O}_2$: 445.9517. Found: 445.9486.

1,4-bis(4-Bromobenzyl)benzene (1)

To trifluoroacetic acid (100 mL) under N_2 , magnetically stirred at $0\text{--}5^{\circ}\text{C}$ was added NaBH_4 (15 pellets, 4.5 g, 120 mmol) over 10 min, and the resulting mixture was allowed to warm to 25°C and stirred at 25°C for 45 min. To this was added in portions over 2 h a suspension of α,α' -bis(4-bromophenyl)-1,4-benzenedimethanol (**5**) (3.10 g, 6.92 mmol) in CH_2Cl_2 (25 mL). The resulting mixture was stirred at 25°C for 24 h, then carefully poured into 25% aqueous NaOH (100 mL)/ice chips to make strongly alkaline (pH 11), diluted with brine (200 mL), and extracted with Et_2O (3 x 100 mL). The combined extracts were dried (Na_2SO_4), filtered, and concentrated *in vacuo* to afford **1** (2.55 g, 89%) as a white solid, which was recrystallized from hexane/ Et_2O to give colorless needles; mp $121\text{--}122.5^{\circ}\text{C}$; (lit.¹⁸ mp 127°C): ^1H NMR (CDCl_3) δ 7.3–7.0 (m, 12H), 3.8 (s, 4H); ^{13}C NMR (CDCl_3) δ 140.0, 138.3, 131.4, 130.5, 128.9, 119.8, 40.8; MS *m/e* (relative intensity) 418 (M^+ , 17), 416 (M^+ , 33), 414 (M^+ , 16), 337 (12), 335 (12), 247 (100), 245 (99), 166 (66); IR (CHCl_3): 2945, 1605, 1085, 1025 cm^{-1} . A reduction of **4** on a 9.5-gram scale gave **1** in 91% yield.

Anal. Calcd for $\text{C}_{20}\text{H}_{16}\text{Br}_2$: C, 57.72; H, 3.88; Br, 38.40. Found: C, 57.48; H, 3.91; Br, 38.48.

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References

1. T. Ogoshi, S. Kanai, S. Fujinami, T. A. Yamagishi, and Y. Nakamoto, *J. Am. Chem. Soc.*, **130**, 5022 (2008). doi:[10.1021/ja711260m](https://doi.org/10.1021/ja711260m)
2. H. Takemura, *Curr. Org. Chem.*, **13**, 1633 (2009). doi:[10.2174/138527209789578117](https://doi.org/10.2174/138527209789578117)
3. P. J. Cragg and K. Sharma, *Chem. Soc. Rev.*, **41**, 597 (2012). doi:[10.1039/c1cs15164a](https://doi.org/10.1039/c1cs15164a)
4. T. Ogoshi, *J. Incl. Phenom. Macrocycl. Chem.*, **72**, 247 (2012). doi:[10.1007/s10847-011-0027-2](https://doi.org/10.1007/s10847-011-0027-2)
5. M. Xue, Y. Yang, X. Chi, Z. Zhang, and F. Huang, *Acc. Chem. Res.*, **45**, 1294 (2012). doi:[10.1021/ar2003418](https://doi.org/10.1021/ar2003418)
6. D. Cao and H. Meier, *Asian J. Org. Chem.*, **3**, 244 (2014). doi:[10.1002/ajoc.201300224](https://doi.org/10.1002/ajoc.201300224)
7. D. Cao and H. Meier, *Synthesis*, **47**, 1041 (2015). doi:[10.1055/s-0034-1378688](https://doi.org/10.1055/s-0034-1378688)
8. T. Ogoshi, T. Yamagishi, and Y. Nakamoto, *Chem. Rev.*, **116**, 7937 (2016). doi:[10.1021/acs.chemrev.5b00765](https://doi.org/10.1021/acs.chemrev.5b00765)
9. K. Du and A. C.-H. Sue, *Synlett*, **30**, 2209 (2019). doi:[10.1055/s-0037-1611921](https://doi.org/10.1055/s-0037-1611921)
10. J.-D. Ding, W.-J. Jin, Z. Pei, and Y. Pei, *Chem. Comm.*, **56**, 10113 (2020). doi:[10.1039/d0cc03682j](https://doi.org/10.1039/d0cc03682j)
11. X. Wang, R.-X. Chen, A. C.-H. Sue, H. Zuilhof, A. J. A. Aquino, and H. Lischka, *Compt. Theor. Chem.*, **1161**, 1 (2019). doi:[10.1016/j.comptc.2019.05.011](https://doi.org/10.1016/j.comptc.2019.05.011)
12. P. Liu, Q. Li, H. Zeng, B. Shi, J. Liu, and F. Huang, *Org. Chem. Front.*, **6**, 309 (2019). doi:[10.1039/C8QO01218K](https://doi.org/10.1039/C8QO01218K)
13. W. Yang, K. Sumanta, X. Wan, T. U. Thikekar, Y. Chao, S. Li, K. Du, J. Xu, Y. Gao, H. Zuilhof, and A. C.-H. Sue, *Angew. Chem. Int. Ed.*, **59**, 3994 (2020). doi:[10.1002/anie.201913055](https://doi.org/10.1002/anie.201913055)
14. X. Xu, V. V. Jerca, and R. Hoogenboom, *Angew. Chem. Int. Ed.*, **59**, 6314 (2020). doi:[10.1002/anie.202002467](https://doi.org/10.1002/anie.202002467)
15. G. W. Gribble and C. F. Nutaitis, *Tetrahedron Lett.*, **26**, 6023 (1985). doi:[10.1016/S0040-4039\(00\)95115-3](https://doi.org/10.1016/S0040-4039(00)95115-3)
16. S. Caron and N. M. Do, *Synlett*, 1440 (2004). doi:[10.1055/s-2004-825611](https://doi.org/10.1055/s-2004-825611)
17. G. W. Gribble, W. T. Kelly, and S. E. Emery, *Synthesis*, 763 (1978). doi:[10.1055/s-1978-24884](https://doi.org/10.1055/s-1978-24884)
18. A. B. Galun, A. Kaluszyner, and E. D. Bergmann, *J. Org. Chem.*, **27**, 2373 (1962). doi:[10.1021/jo01054a020](https://doi.org/10.1021/jo01054a020)
19. J. Zhu, M. Perez, C. B. Caputo, and D. W. Stephan, *Angew. Chem. Int. Ed.*, **55**, 1417 (2016). doi:[10.1002/anie.201510494](https://doi.org/10.1002/anie.201510494)