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A One-pot Aza-Payne Rearrangement-Epoxide Ring Opening Reaction of 2-Aziridinemethanols: A Regio- and Stereoselective Synthetic Route to Diastereomerically Pure 1,2-Amino Alcohols

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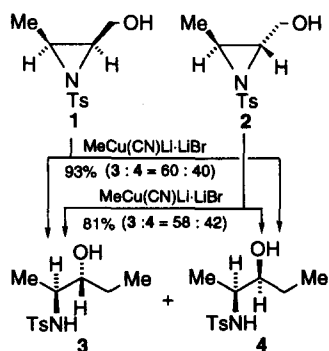
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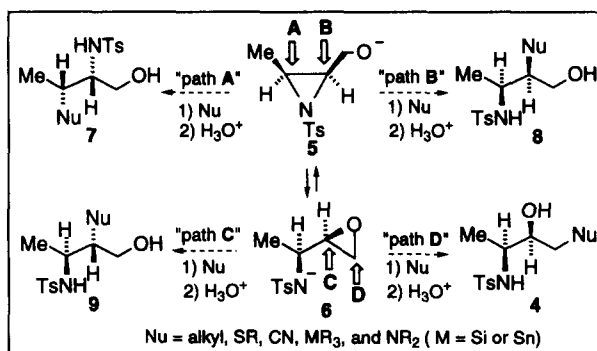
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Abstract: A regio- and stereoselective synthetic route to diastereomerically pure 1,2-amino alcohols via a one-pot aza-Payne rearrangement - epoxide ring opening reaction of 2-aziridinemethanols is reported. Satisfactory yields are obtained in excellent diastereoisomeric excesses by successive exposure of 2-aziridinemethanols to potassium hydride and nucleophilic reagents in a one-pot manner.

N-Activated aziridines have been utilized as substrates in a number of important synthetic transformations.¹ On account of the ring strain present in aziridines,² the potential utility of *N*-activated aziridines as building blocks for synthesis of biologically active compounds reflects their ability to undergo regioselective ring-opening reactions with a wide range of nucleophilic reagents.³ One very important aspect of aziridine ring-opening reactions is that they are usually stereospecific, producing secondary amines with inversion of configuration at the site of the ring opening via an S_N2 mechanism.⁴ We have been interested in the synthetically useful ring-opening reaction of aziridine-ring bearing compounds with organocopper reagents in connection with synthetic studies on bioactive compounds of stereochemically well defined structure.⁵ Unexpectedly, when both 2-aziridinemethanols **1** and **2** were treated with $\text{MeCu}(\text{CN})\text{Li}\cdot\text{LiBr}$, 60:40 and 58:42 mixtures of protected 1,2-amino alcohols **3** and **4** were obtained as shown in Scheme 1.^{5a,c,e} Although these transformations with a lower order cuprate were interesting from the viewpoint of the reaction mechanism, the low stereoselectivity precludes their use for synthetic applications.



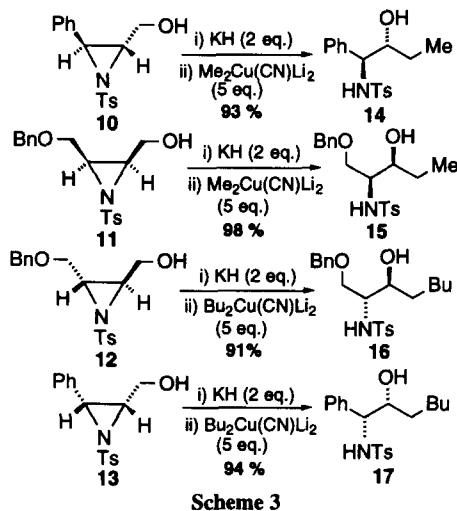
Scheme 1



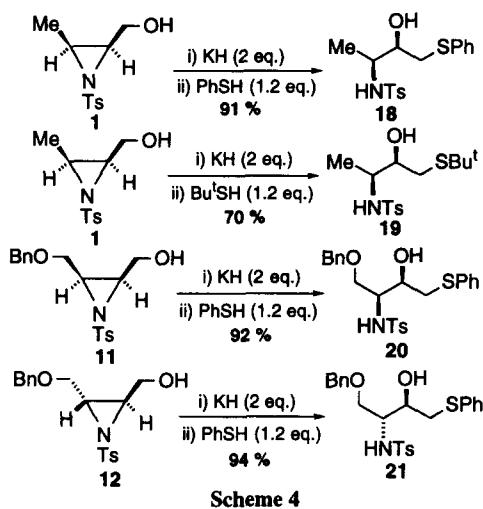
Scheme 2

It was our expectation to be able to synthesize stereochemically pure amino alcohols in a stereo- and regioselective manner in a one-pot sequence from readily available 2-aziridinemethanols by successive treatment with potassium hydride and various nucleophilic reagents as shown in Scheme 2. In principle, reaction of 2-aziridinemethanol **1** with bases such as KH followed by nucleophilic reagents in a one-pot manner could afford

one or a mixture of four amino alcohols **4**, **7**, **8**, and **9** via the anionic intermediates **5** and **6**. Thus, it is not an easy matter to predict whether A, B, C, or D is the major reaction pathway. In order to make such reactions practically useful, it is inevitable to know the favored reaction pathway before an evaluation regarding the relative merits of aza-Payne rearrangement - epoxide ring opening reaction. In a preliminary experiment, the rearrangement-epoxide ring opening reaction scenario does in fact lead to the stereochemically pure amino alcohols via the path D in Scheme 2. We now report that this approach is a viable and potentially useful route to stereodefined 1,2-amino alcohols from readily available chiral 2-aziridinemethanols.



Scheme 3



Scheme 4

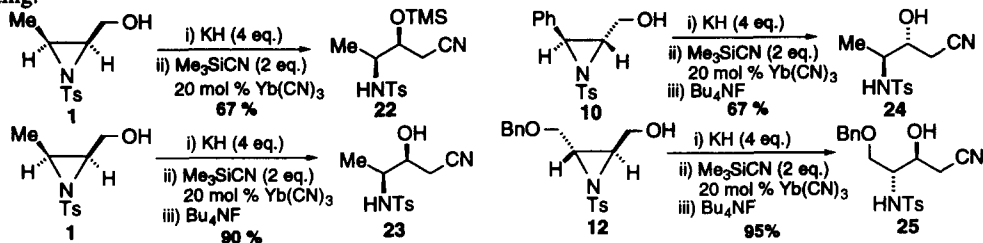
Exposure of **10**^{5c} and **11**^{5c} to potassium hydride (2 equiv.) in THF at 0 °C for 1 h, followed by the addition of Me₂Cu(CN)Li₂·2LiBr (5 equiv.) by syringe, and the whole mixture was stirred for 1 h at 0 °C to give the single products **14** and **15** in 93% and 98% yield, respectively (Scheme 3). This contrasted with our previous work ^{5a,e} where the reaction of **1**^{5c} and **2**^{5c} with MeCu(CN)Li·LiBr gave a mixture of 1,2-amino alcohols **3** and **4**. In a series of bases, KH^{5e} gave the best results in combination with 1.2–5 equivalents of nucleophilic reagents. It should be clearly noted that the use of MeMgBr instead of MeCu(CN)Li·LiBr or Me₂Cu(CN)Li₂·2LiBr was not successful for a clean transformation. In a similar manner, reaction of the anionic equilibrium mixture derived from **12**^{5c} with Bu₂Cu(CN)Li₂ afforded the single product **16** in 91% isolated yield. We were unable to detect any regio- or stereoisomeric compounds in these reactions.

This one-pot regio- and stereoselective method would be truly useful if it could be successfully extended to other nucleophiles. As shown in Scheme 4, the reaction of **1** with KH followed by PhSH and ^tBuSH gave the diastereomerically pure phenylthio- and *tert*-butylthio amino alcohols **18** and **19** in 91% and 70% yields, respectively.⁷ Similarly, 2-aziridinemethanols **11**^{5c} and **12**^{5c} could easily be converted into the phenylthio amino alcohols **20** and **21**, respectively, in high yields (Scheme 4).

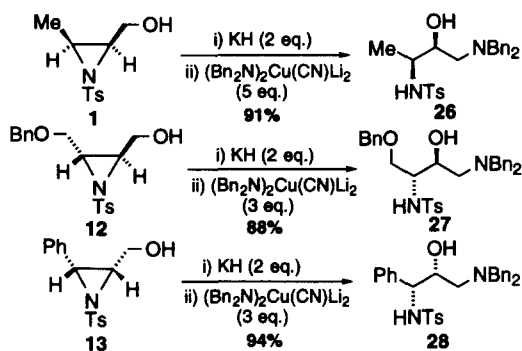
Likewise, the reaction of **1** with KH followed by the sequential addition of Me₃SiCN and Yb(CN)₃⁸ yielded the nitrile **22** in 67% yield after flash chromatographic purification. The low yield of **22** may be attributed to the rather low stability of the siloxy group in **22**. In the event, sequential treatment of **1** with KH, Me₃SiCN (3 equiv.) in the presence of Yb(CN)₃ (0.2 equiv.), and Bu₄NF (1.5 equiv.) yielded the cyano amino alcohol **23** in 90% isolated yield. By a procedure identical with that described for the reaction of **1**, 2-aziridinemethanols **10** and **12** afforded cyano amino alcohols **24** and **25**, respectively, in high yields (Scheme 5).

Attempts to use either dibenzylamine or lithium dibenzylamide to effect the epoxy ring opening of anionic intermediates derived from 2-aziridinemethanols **1**, **12**, and **13** were unsuccessful. Fortunately, use of a higher

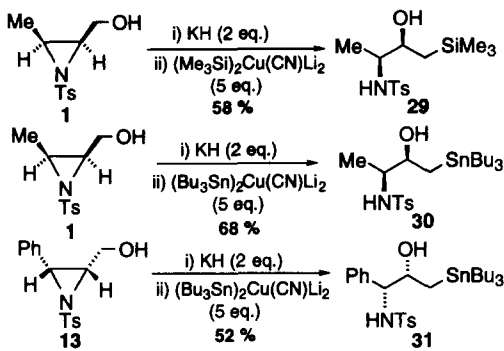
order amide cuprate,⁹ $(\text{Bn}_2\text{N})_2\text{Cu}(\text{CN})\text{Li}_2$,^{9a} had overcome the problem to yield diastereomerically pure diamino alcohols **26**, **27**, and **28** in high yields as shown in Scheme 6 and had placed the method on a synthetic footing.¹⁰



Scheme 5

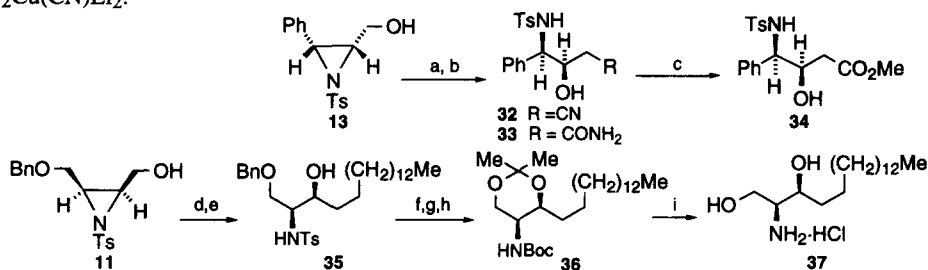


Scheme 6



Scheme 7

As can be seen from Scheme 7, acceptable chemical yields of compounds **29**, **30**, and **31** were obtained from the treatment of 2-aziridinemethanols **1** and **13** with KH followed by $(\text{Me}_3\text{Si})_2\text{Cu}(\text{CN})\text{Li}_2$ or $(\text{Bu}_3\text{Sn})_2\text{Cu}(\text{CN})\text{Li}_2$.



Scheme 8

Reagents: a) i. KH (2 eq.), ii. TMSCN - 20 mol % $\text{Yb}(\text{CN})_3$, iii. Bu_4NF (85%); b) 30% H_2O_2 - K_2CO_3 - MeOH (54%); c) i. 2N-KOH - MeOH , ii. ethereal CH_2N_2 (80%); d) KH (2 eq.); e) $[\text{Me}(\text{CH}_2)_{13}\text{Cu}(\text{CN})(\text{MgCl})_2]$ (5 eq.) (89.5%); f) H_2 - PtO_2 - MeOH (91%); g) 2,2-dimethoxypropane - Me_2CO (1:2), $\text{BF}_3 \cdot \text{Et}_2\text{O}$ (94.2%); h) i. Na - NH_3 , ii. $(\text{Boc})_2\text{O}$ (75.5%); i) 3N-HCl in THF (98.7%)

Having established useful reaction conditions for the synthesis of various amino alcohols with very high optical purity, the utility of aziridinemethanols **13** and **11**^{5c} was briefly investigated. As shown in Scheme 8, the nitrile **32**, derived from **13** in 85% yield by the same procedure described above, was treated with an alkaline hydrogen peroxide to yield an amide **33**, which upon successive treatment with 2N-KOH in MeOH , 5% hydrochloric acid, and ethereal diazomethane yielded an amino acid methyl ester **34**, m.p. 106°C , $[\alpha]_D^{20}$ -33° (in CHCl_3), in 80% yield. *L-threo*-C18-dihydrosphingosine **37** {hydrochloride: m.p. $75\sim77^\circ\text{C}$, $[\alpha]_D^{20}$

- 10.4° (in MeOH)} can be readily synthesized from 2-aziridinemethanol **11** as shown in Scheme 8 via a sequence of reactions.¹¹

In conclusion, an attractive one-pot regio- and stereoselective synthetic route to 1,2-amino alcohols from readily available 2-aziridinemethanols has been developed. Isolation or purification of the intermediate resulting from the aza-Payne rearrangement is not necessary. This methodology leads to a series of useful diastereomerically pure amino alcohols to be utilized in the synthesis of more complex molecules.

REFERENCES AND NOTES

1. Church, N. J.; Young, D. W. *Tetrahedron Lett.* **1995**, *36*, 151 and references cited.
2. a) Fanta, P. E. In *Heterocyclic Compounds with Three- and Four-membered Rings*; Weissberger, A. Ed.; Part One; Interscience Publishers; New York, **1964**, p. 524. b) Rauk, A.; Allen, L. C.; Mislow, K. *Angew. Chem. Int. Ed. Engl.* **1970**, *9*, 400. c) Gilchrist, T. L. *Heterocyclic Chemistry*, 2nd ed.; Longman, Harlow, **1992**, p. 38.
3. a) Eis, M. J.; Ganem, B. *Tetrahedron Lett.* **1985**, *26*, 1153. b) Legters, J.; Thijs, L.; Zwanenburg, B. *Tetrahedron Lett.* **1989**, *30*, 4881. c) Hudlicky, T.; Luna, H.; Price, J. D.; Rulin, F. *J. Org. Chem.* **1990**, *55*, 4683. d) Pearson, W. H.; Bergmeier, S. C.; Degan, S.; Lin, K.-C.; Poon, Y.-F.; Schkeryantz, J. M.; Williams, J. P. *J. Org. Chem.* **1990**, *55*, 5719. e) Baldwin, J. E.; Spivey, A. C.; Schofield, C. J.; Sweeney, J. B. *Tetrahedron* **1993**, *49*, 6309. f) Lygo, B. *Synlett* **1993**, 764. g) Osborn, H. M. I.; Sweeney, J. B.; Howson, B. *Synlett* **1993**, 675. h) Evans, D. A.; Faul, M. M.; Bilodeau, M. T.; Anderson, B. A.; Barnes, D. M. *J. Am. Chem. Soc.* **1993**, *115*, 5328. i) Davis, F. A.; Zhou, P.; Reddy, G. V. *J. Org. Chem.* **1994**, *59*, 3243. j) Toshimitsu, A.; Abe, H.; Hirosawa, C.; Tamao, K. *J. Chem. Soc., Perkin Trans. 1*, **1994**, 3465. k) Gmeiner, P.; Orecher, F.; Thomas, C.; Weber, K. *Tetrahedron Lett.* **1995**, *36*, 381.
4. a) Dubois, L.; Mehta, A.; Tourette, E.; Dodd, R. H. *J. Org. Chem.* **1994**, *59*, 434 and references cited. b) Tanner, D.; He, H. M. *Tetrahedron* **1992**, *48*, 6079.
5. a) Ibuka, T.; Nakai, K.; Habashita, H.; Fujii, N.; Garrido, F.; Mann, A.; Chounan, Y.; Yamamoto, Y. *Tetrahedron Lett.* **1993**, *34*, 7421. b) Ibuka, T.; Nakai, K.; Habashita, H.; Hotta, Y.; Fujii, N.; Mimura, N.; Miwa, Y.; Taga, T.; Yamamoto, Y. *Angew. Chem. Int. Ed. Engl.* **1994**, *33*, 652. c) Fujii, N.; Nakai, K.; Habashita, H.; Hotta, Y.; Tamamura, H.; Otaka, A.; Ibuka, T. *Chem. Pharm. Bull.* **1994**, *42*, 2241. d) Wada, M.; Doi, R.; Hosotani, R.; Ibuka, T.; Habashita, H.; Nakai, K.; Fujii, N.; Imamura, M. *Pancreas* **1995**, *10*, 301. e) Ibuka, T.; Nakai, K.; Habashita, H.; Hotta, Y.; Otaka, A.; Tamamura, H.; Fujii, N.; Mimura, N.; Miwa, Y.; Taga, T.; Chounan, Y.; Yamamoto, Y. *J. Org. Chem.* **1995**, *60*, 2044.
6. Recently, we reported that the aza-anionic energy minimum of type **6** was predicted to be 18.6 kcal mol⁻¹ lower than the oxa-anionic minimum of type **5** at the RHF/3-21+G* level.^{5c}
7. Payne rearrangement - ring opening of 2,3-epoxy alcohols usually yields a 2:1~20:1 mixture of products. see Behrens, C. H.; Ko, S. Y.; Sharpless, K. B.; Walker, F. J. *J. Org. Chem.* **1985**, *50*, 5687.
8. a) Matsubara, S.; Kodama, T.; Utimoto, K. *Tetrahedron Lett.* **1990**, *31*, 6379 and references cited. b) Meguro, M.; Asao, N.; Yamamoto, Y. *Tetrahedron Lett.* **1994**, *35*, 7395.
9. Yamamoto, Y.; Asao, N.; Meguro, M.; Tsukada, N.; Nemoto, H.; Sadayori, N.; Wilson, J. G.; Nakamura, H. *J. Chem. Soc., Chem. Commun.* **1993**, 1201. see also Meguro, M.; Asao, N.; Yamamoto, Y. *J. Chem. Soc., Perkin Trans. 1*, **1994**, 2597.
10. The *N,N*-dibenzylamino group has been used for the synthesis of many important compounds, see Reetz, M. T.; Kayser, F.; Harms, K. *Tetrahedron Lett.* **1992**, *33*, 3453 and references cited.
11. Shibuya, H.; Kawashima, K.; Naritas, N.; Ikeda, M.; Kitagawa, I. *Chem. Pharm. Bull.* **1992**, *40*, 1154 and references cited.