

Tetrahedron Letters 42 (2001) 3101-3103

TETRAHEDRON LETTERS

An enantioselective synthetic strategy toward the polyhydroxylated agarofuran

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Received 9 August 2000; revised 1 March 2001; accepted 2 March 2001

Abstract—This paper describes a new approach for the enantioselective syntheses of naturally occurring polyhydroxylated dihydroagarofuran sesquiterpenoids starting from (–)-carvone. Through utilizing asymmetric Robinson annulation and acetoxylation as key steps, the agarofuran skeleton was enantioselectively constructed and an oxyfunctionalized group was introduced into the C-1 position. The important intermediate **23** to dihydroagarofuran was obtained in eleven steps. © 2001 Elsevier Science Ltd. All rights reserved.

In the sesquiterpenic family, a large number of β -dihydroagarofuran polyol esters based on dihydroagarofuran skeleton **1** have been isolated^{1,2} from the species *Celastraceae*. These kind of compounds have been demonstrated to exhibit a wide spectrum of biological properties, including significant cytotoxic,³ immunosuppressive,⁴ anticancer,⁵ insect antifeedant⁶ and potent anti-HIV activity.⁷ Their classic derivatives range from the least structurally complex diol **2**,⁸ and the triol isocelorbicol **3**,⁹ to complex polyols such as **4** (Fig. 1).¹⁰

Their highly oxygenated tricyclic frameworks, comprising a number of contiguous stereocenters, pose a formidable synthetic challenge and have attracted immense interest from synthetic chemists. Several racemic total syntheses of polyhydroxylated agarofurans have been reported since 1984.^{11,12} However, there is no report on the total synthesis of optically active polyhydroxylated agarofuran to our knowledge. In our own quest for an enantioselective avenue toward a bioactive polyhydroxylated agarofuran, we are interested in developing a facile and general approach for the total synthesis of a polyhydroxylated agarofuran. Reported herein is an efficient approach for the enantioselective construction of a carbon skeleton and the introduction of an oxyfunctionalized group at the C-1 position.

In our earlier efforts, research interest was mainly focused on the introduction of an oxygenated group at the C-1 position. Our initial synthetic strategy was to employ (\pm)-9-keto- α -agarofuran **5** as a key intermediate, which was readily prepared from carvone in seven steps.¹³ After considerable endeavor, several intermediates **6–12** have been obtained in several steps (Fig. 2). Unfortunately, attempted allylic oxidation or enone's α' -position oxidation of these compounds failed under various conditions, including PCC, SeO₂, CrO₃–SiO₂, PDC, CrO₃–*t*BuOOH, Pb(OAc)₄ and Mn(OAc)₃. Employing oxidative agents mentioned above only resulted in undesired products, complex mixtures or recovered substrates.



Figure 1.

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Keywords: dihydroagarofuran; sesquiterpenoid; enantioselective synthesis.

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Figure 2.

On the assumption that failure to introduce an oxyfunctionalized group at the C-1 position was due to the steric hindrance of a rigid tetrahydrofuran ring, combined with a consideration of the enantioselective synthesis at the same time, we then turned our attention to an alternative synthetic approach. The detailed pathway of synthesis is summarized in Scheme 1.

By the published method,¹⁴ an inseparable trione mixture **15** could be readily prepared from commercially available (–)-carvone **13** by epoxidation, cleavage of epoxide and condensation with ethyl vinyl ketone. D-(+)-Phenylalanine-catalyzed asymmetric Robinson annulation¹⁵ of trione **15** afforded the (–)-9-oxo-10-*epi*- α -cyperone **16** (54% e.e.; $[\alpha]_D^{10}$ –25.5, *c* 2.54, CHCl₃) in 51% yield by Hagiwara's method. Selective reduction of the keto group of **16** by sodium borohydride in MeOH provided the hydroxyl enone **17**. The enantiomeric excess of **16** was determined by esterification of **17** with (R)-O-acetylmandelic acid.¹⁶ Tosyl hydrazone formation followed by treatment with excess n-BuLi (6 equiv.) in anhydrous THF gave the conjugated diene 19 in 86% total yield.¹⁷ Diels-Alder cycloaddition of diene¹⁸ 19 to singlet oxygen generated from the adducts between triphenylphosphite and ozone¹⁹ afforded the desired endo-peroxide 20 as a single product in 81% yield, in which the peroxide bridge was assigned as the β -configuration based on the consideration that the steric hindrance of the angular methyl group restricts the approach of oxygen from the site undergoing reaction. Rearrangement of peroxide 20 on K₂CO₃ in THF gave the hydroxyl ketone 21 in 78% yield. The β configuration of the C-5 hydroxyl group in 21 was further confirmed by the latter's easy cyclization. Regioselective acetoxylation with Mn(OAc)₃ at the C-1 of enone 21 ultimately furnished the less polar acetate 22 in 53% yield.²⁰ The stereochemistry of the acetate 22 was confirmed via an NOE experiment. According to



Scheme 1. (a) 30% H₂O₂, KOH, MeOH–H₂O, 0°C, 2.5 h; (b) 1N NaOH, H₂O, reflux, 1 h; (c) EVK, KOH, MeOH, reflux, 6 h; (d) D-(+)-phenylalanine, D-(+)-CSA, $30-70^{\circ}$ C, 6 days; (e) NaBH₄, MeOH, 0°C, 3 h; (f) TsNHNH₂, BF₃·Et₂O, rt, 2 h; (g) *n*-BuLi, THF, -78° C to 0°C, 10 h; (h) O₃, P(OPh)₃, CH₂Cl₂, -78° C to 0°C, 2.5 h; (i) K₂CO₃, THF, 40°C, 24 h; (j) Mn(OAc)₃, benzene, reflux, 10 h; (k) TfOH, THF, rt, 5 min.

the literature method,^{11a} treatment of **22** with trifluoromethanesulfonic acid smoothly afforded pure agarofuran **23** (60% e.e.;¹⁶ $[\alpha]_{D}^{21}$ –12.9, *c* 0.78, CHCl₃) after chromatographic purification of the crude product on silica gel, which has shown identical spectral data with its structure.²¹ The downfield ¹³C NMR signals at δ 83.62 and 83.43 for two oxygen-linked quaternary carbons indicated the successful cyclization of the tetrahydrofuran ring.

In summary, we have presented a new enantioselective synthetic strategy to construct the skeleton and introduce an oxyfunctionalized group into the C-1 position as anticipated, which allows us to obtain the optically active important intermediate 23 to polyhydroxylated agarofuran starting from (–)-carvone in eleven steps. Application of the established method to the further syntheses of a number of naturally occurring polyhydroxylated agarofurans is under active investigation.

Acknowledgements

We are grateful for the financial support from the National Natural Science Foundation of China (Grant No. 29732060), the National Outstanding Youth Fund (No. 29925204) and the Foundation for University Key Teacher by the Ministry of Education of China.

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- 21. Spectral data for **23**: ¹H NMR (400 MHz, CDCl₃) δ 0.90 (s, 3H), 1.23 (s, 3H), 1.33 (s, 3H), 2.00 (s, 3H), 2.10 (s, 3H), 2.77 (d, *J*=4.4 Hz, 1H), 3.98 (m, 1H), 4.88 (s, 1H), 6.01 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 192.6, 172.3, 158.5, 127.7, 83.6, 83.4, 75.5, 65.2, 46.3, 42.7, 34.2, 30.7, 29.9, 22.5, 21.0, 20.2, 17.5. EIMS (70 eV) *m/z* 308 (M⁺, 3%), 293 (15%), 291 (50%), 290 (5%), 240 (13%), 149, 75 (100); IR (film): 3486, 2976, 2934, 1729, 1680 cm⁻¹; FAB-HRMS: found 331.1510, C₁₇H₂₂O₅+Na⁺ requires: 331.1521.