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Studies toward the total synthesis of azaspiracids: synthesis of the FGHI ring domain

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Abstract—Synthesis of the FGHI ring domain of azaspiracids, the causative agents for a new type of shellfish poisoning, azaspiracid poisoning (AZP), has been achieved. The synthesis features dithiane anion–epoxide coupling for convergent fragment assembly.

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Azaspiracid poisoning (AZP) is a new type of shellfish poisoning prevailing in Europe since 1995.¹ Azaspiracid (1) and its congeners (2-5) were isolated as the causative toxins from the contaminated Irish mussels, Mytilus edulis, and their structures were elucidated by extensive NMR analysis and FAB collision-induced dissociation MS/MS experiments by Yasumoto and co-workers (Fig. 1).² These toxins are characterized by a bisspiroketal structure fused to a tetrahydrofuran ring and an unusual azaspiro ring fused with a 2,9-dioxabicyclo[3.3.1]nonane ring system. However, the relative configuration between C25 and C28, and the absolute stereochemistry remained to be determined. The unprecedented molecular architecture and intriguing biological activity of these natural toxins have attracted considerable interest among synthetic chemists.³ As part of our efforts toward the total synthesis of azaspiracids, we describe herein the synthesis of the FGHI ring domain 6a. Although the synthesis of ent-6a has been recently reported by the Nicolaou^{3f} and Forsyth groups,^{3g} our approach features dithiane-epoxide coupling for the C35–C36 bond construction.⁴

Our synthetic plan for the assembly of the C21–C40 domain **6b** of azaspiracids involved convergent coupling of the HI ring aldehyde **7** with the E ring allyl-stannane **8** followed by closure of the 2,9-bicyclo[3.3.1]nonane ring (Scheme 1).⁵ The azaspiro ring system of **7** would, in turn, be accessible from an

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open-chain precursor 9, which was envisioned to be obtained from coupling of the C36–C40 dithiane 10 and the C28–C35 epoxide 11. The viability of the strategy was investigated by the synthesis of a more simple system 6a.^{3f,g}

The synthesis of dithiane **10** is summarized in Scheme 2. Desymmetrization of the known *meso*-diol **12**⁷ by treatment with vinyl acetate and lipase AK provided the corresponding monoacetate **13**, $[\alpha]_{D}^{23} + 10.8$ (*c* 1.00, CHCl₃).⁸ Protection as its *tert*-butyldimethylsilyl (TBS) ether followed by deacetylation gave alcohol **14**. Oxidation⁹ to the aldehyde and subsequent treatment



5 azaspiracid-5: $R_1 = R_2 = R_3 = H$, $R_4 = OH$

Figure 1. Structures of azaspiracid (1) and its analogues.

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Scheme 1. Retrosynthetic analysis of the C21–C40 domain 6.



Scheme 2. Reagents and conditions: (a) CH_2 =CHOAc, lipase AK, THF, 0°C, 47%; (b) TBDMSCl, imidazole, CH_2Cl_2 , rt; (c) K_2CO_3 , MeOH, rt, 86% (two steps); (d) IBX, DMSO, rt, 90%; (e) HS(CH₂)₃SH, BF₃·OEt₂, CH₂Cl₂, 0°C, 96%; (f) TIPSCl, imidazole, CH₂Cl₂, rt, 89%.

with propanedithiol and $BF_3 \cdot OEt_2$ afforded dithiane alcohol 15, which was reprotected as the TIPS ether to give 10.

The synthesis of epoxide 11 began with lactone 16 ($[\alpha]_D^{25}$ -33.7 (c 0.89, CHCl₃)),¹⁰ which was subjected to DIBALH reduction and Wittig methylenation to give, after benzylation, olefin 17 (Scheme 3). Hydroborationoxidation followed by benzylation of the derived alcohol and removal of the trityl ether provided alcohol 18, which was then oxidized to the aldehyde and homologated to enoate 19. Dihydroxylation of 19 with $OsO_4/$ NMO produced an approximately 3:1 diastereomeric mixture of diols (77% yield), with the desired 20 predominating as predicted by the empirical rule developed by Cha et al.¹¹ On the contrary, Sharpless asymmetric dihyroxylation with AD mix- β^{12} afforded **20** with high selectivity (>20:1). Diol 20 was then converted to triol 21 by a three-step sequence. Selective sulforylation of the primary hydroxyl group with 2,4,6-trimethylbenzenesulfonyl chloride followed by base treatment led to epoxide 23, which was then protected as the MOM ether to provide 11.

With the requisite fragments 10 and 11 in hand, we next focused on their coupling (Scheme 4). After some

experiments, the best result was realized by generation of the anion of 10 with n-BuLi in THF at room temperature for 5 min followed by reaction with 11 at 0°C for 20 min. Thus, the desired coupling product 24 was obtained in 92% yield. Desilylation and tosylation gave 25, which upon displacement with NaN₃ afforded azide 9. After removal of the MOM group, treatment with $Hg(ClO_4)_2$ in the presence of CaCO₃ in THF/ methanol gave a mixture of methyl ketal 26 and the corresponding hemiketal (ca. 1.3:1), which without separation was treated with PPTS in methanol to afford 26 in 74% yield from 9. Reduction of the azide and protection of the resulting amine as the *tert*-butylcarbamate gave N-Boc derivative 27. Subsequent spiroaminal formation was carried out according to the procedure described by Forsyth et al.^{3g} Thus, treatment of 27 with Yb(OTf)₃ (0.01 equiv.) in CH₃CN at -20°C gave the desired spiroaminal 29 in 57% yield (73% based on recovered 27). In this reaction, hemiketal 28 was obtained in 13% yield, but this could be readily converted into 27 by treatment with PPTS in methanol (77% yield) and recycled. Reductive debenzylation with lithium di-tert-butylbiphenilide (LiDBB)¹³ and selective acetylation of the primary hydroxyl group gave monoacetate 30. Protection of the diol as the TES ethers was followed by deacetylation and oxidation of the derived alcohol to give aldehyde 7. Reaction with methallyltributylstannane 31 in the presence of MgBr₂·OEt₂ and oxidation of the resulting secondary alcohol produced ketone 32. Finally, treatment of 32 with HF pyridine in pyridine/THF at room temperature effected formation of the 2,9-dioxabicy-



Scheme 3. Reagents and conditions: (a) DIBALH, CH_2Cl_2 , $-78^{\circ}C$; (b) $Ph_3P^+CH_3Br^-$, NaHMDS, THF, 0°C, 53% (two steps); (c) NaH, BnBr, DMF, 0°C \rightarrow rt; (d) 9-BBN, THF, rt, then NaBO₃·4H₂O; (e) NaH, BnBr, DMF, 0°C \rightarrow rt, 60% (three steps); (f) *p*-TsOH, MeOH/CH₂Cl₂ (1:1), rt, 70%; (g) IBX, DMSO, rt; (h) (*i*-PrO)₂P(O)CH₂CO₂Et, KO*t*-Bu, THF, 0°C, 90% (two steps); (i) AD mix- β , CH₃SO₂NH₂, OsO₄ (0.05 equiv.), *t*-BuOH/H₂O (1:1), rt, 83% (>20:1); (j) PhCH(OMe)₂, CSA, CH₂Cl₂, rt; (k) LiAIH₄, THF, 0°C; (l) *p*-TsOH, MeOH, rt, 76% (three steps); (m) 2,4,6-Me₃PhSO₂Cl, Et₃N, DMAP, CH₂Cl₂, 0°C \rightarrow rt; (n) K₂CO₃, EtOH, rt, 59% (two steps); (o) MOMCl, *i*-Pr₂NEt, CH₂Cl₂, rt, 67%.



Scheme 4. Reagents and conditions: (a) 10, n-BuLi, THF, rt, then 11, 0°C, 92%; (b) TBAF, THF, rt, 81%; (c) *p*-TsCl, Et₃N, DMAP, CH₂Cl₂, rt; (d) NaN₃, DMF, 50°C, 94% (two steps); (e) BF₃·OEt₂, Me₂S, CH₂Cl₂, 0°C, 87%; (f) Hg(ClO₄)₂, CaCO₃, THF/MeOH (4:1), 0°C; (g) PPTS, MeOH, rt, 85% (two steps); (h) Ph₃P, THF, then H₂O, rt; (i) Boc₂O, Et₃N, CH₂Cl₂, rt, quant. (two steps); (j) Yb(OTf)₃, CH₃CN, -20°C, 57%; (k) LiDBB, THF, -78°C, 92%; (l) AcCl, 2,4,6-collidine, CH₂Cl₂, -40°C, 79%; (m) TESOTf, 2,6-lutidine, CH₂Cl₂, -40°C, 86%; (n) DIBALH, CH₂Cl₂, -78°C, 88%; (o) TPAP, NMO, 4 Å MS, CH₂Cl₂, rt, 91%; (p) **31**, MgBr₂·OEt₂, cH₂Cl₂, -20°C→rt, 78%; (q) TPAP, NMO, 4 Å MS, CH₂Cl₂, rt, 73%; (r) HF·pyr, pyr, THF, 0°C→rt, 30%.

clo[3.3.1]nonane ring system to complete the synthesis of the FGHI ring domain **6a**, $[\alpha]_{D}^{20}$ -3.7 (*c* 0.22, CHCl₃), in 30% yield (not optimized). The spectral data for **6a** were fully consistent with those reported previously.^{3f}

In conclusion, we have developed a convergent synthetic route to the FGHI ring system 6a of azaspiracids based on dithiane anion–epoxide coupling for the C35– C36 bond construction. Further studies toward the total synthesis of azaspiracids are currently underway and will be reported in due course.

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- Because the sign of optical rotation ([α]_D²⁵ -59 (c 0.016, MeOH))⁶ of acid i derived from natural I was opposite to that of synthetic *ent*-6a ([α]_D²⁰ +4.6 (c 0.22, CH₂Cl₂)^{3f}; [α]_D²³ +5 (c 0.26, CHCl₃)^{3g}), we determined to synthesize compound 6a in the present study.



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