BASIC STEROID SAPONINS FROM AERIAL PARTS OF FRITILLARIA THUNBERGII*

JUNICHI KITAJIMA, TETSUYA KOMORI, TOSHIO KAWASAKI† and HANS-ROLF SCHULTEN‡

Faculty of Pharmaceutical Sciences, Kyushu University, Maedashi 3-1-1, Higashi-ku, 812 Fukuoka, Japan; ‡ Institute of Physical Chemistry, University of Bonn, Wegelerstr. 12, 5300 Bonn, West Germany

(Received 23 April 1981)

Key Word Index—*Fritillaria thunbergii*; Liliaceae; structural elucidation; field desorption mass spectrometry; basic steroid saponins; solanidine- and hapepunine-oligoglycosides.

Abstract—From the aerial parts of *Fritillaria thunbergii*, three glycosidal *Solanum* alkaloids (basic steroid saponins) were isolated together with minor amounts of the two ceveratrum alkaloids verticine and verticinone. The saponins were identified as solanidine $3-O-\alpha$ -L-rhamnopyranosyl- $(1 \rightarrow 2)$ - β -D-glucopyranoside (β_1 -chaconine), solanidine $3-O-\alpha$ -L-rhamnopyranosyl- $(1 \rightarrow 2)$ - β -D-glucopyranoside and hapepunine $3-O-\alpha$ -L-rhamnopyranosyl- $(1 \rightarrow 2)$ - β -D-glucopyranoside and hapepunine $3-O-\alpha$ -L-rhamnopyranosyl- $(1 \rightarrow 2)$ - β -D-glucopyranoside. The latter two are newly discovered natural compounds. Field desorption MS was used to provide information on the purity and MW of the complex oligoglycosides, and, to some extent, the nature and sequence of their sugar moieties. The direct examination of yet unknown polar glycosides prior to derivatization and application of conventional techniques proves to be of great diagnostic value and aids the structural elucidation considerably.

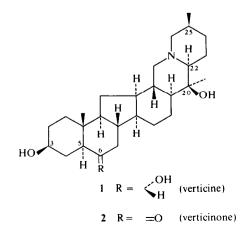
INTRODUCTION

The processed bulbs of Fritillaria thunbergii Miq. [F. verticillata Willd. var. thunbergii (Miq.) Baker have long been known as one of the principal Chinese crude drugs [1]. The alkaloids of unprocessed fresh bulbs of the plant cultivated in Japan were studied by Fukuda [2], Morimoto et al. [3], Itô et al. [4, 5], and Kaneko et al. [6], and three ceveratrum alkaloids, verticine (1), verticinone (2) and isoverticine were isolated in a free state and characterized. A study on the alkaloids [7] and other constituents of the crude drug in comparison with those of fresh bulbs has been carried out in this laboratory. With regard to the constituents of the fresh aerial parts of this plant, Kaneko et al. have surveyed the alkamine obtained by acid hydrolysis of the extracts in the course of biogenetic studies on Fritillaria alkaloids. They obtained, in addition to solanidine (3) [8,9] and happenine (4)[10], five new steroid alkamines and identified them as 5α cevanine- $3\alpha, 6\beta, 20\beta$ -triol (epimeric at C-3 and C-6 to 1) (baimonidine) [6], the 3-epimer of 1 (isobaimonidine) [11], (22S, 25S)-cev-5-enine-3 α , 20 β -diol (fritillarizine) [12] and (22S, 25R)-5 α -cevanine-3 α ,6 α -diol (petilinine) and its 6β -epimer [13].

We now report on the isolation and characterization from unhydrolysed MeOH extracts of the aerial parts of F. *thunbergii* of three glycosidal *Solanum* alkaloids (basic steroid saponins) [14], **5**, **6** and **7**, along with small amounts of the cervatum alkaloids **1** and **2**. In view of the established capacity of field desorption (FD) mass spectrometry (MS) for the investigation of micrograms of free oligoglycosides [15–19], in order to know the purity, to determine the MW and to evaluate the structural features of each glycoside, FD mass spectra were taken directly after chromatographic separation and purification. The foreknowledge at this early stage of the study facilitated the subsequent unequivocal identification made on the basis of the conventional chemical and spectral data.

RESULTS AND DISCUSSION

The isolation and separation of compounds 1, 2, 5, 6 and 7, were carried out by combination of CC on Amberlite XAD-2, Sephadex LH-20, Al_2O_3 and Si gel, and droplet countercurrent chromatography (DCCC) [20] followed by crystallization. Compounds 1 and 2 were identified as verticine (1) and verticinone (2) [4, 5].



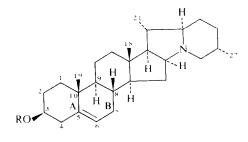
^{*} Part 9 in the series 'Field Desorption Mass Spectrometry of Natural Products'. For Part 8 see Komori, T., Maetani, I., Okamura, N., Kawasaki, T., Nohara, T. and Schulten, H.-R. (1981) Justus Liebigs Ann. Chem. 683.

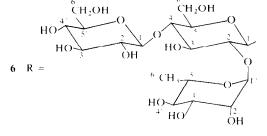
[†] To whom correspondence should be addressed.

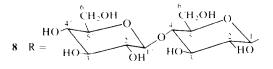
Compound 5 was first subjected to FDMS. The spectrum showed $[M + Na]^+$, $[M + H]^+$ and the fragment ions $[(M + Na) - 146]^+$, $[(M + H) - 146]^+$, $[(M + Na) - 308]^+$ and $[(M + H) - 308]^+$, suggesting that 5 was a homogeneous oligoglycoside having a methylpentosyl-hexose moiety attached to an aglycone of MW 397. Based on these results and elemental analytical data 5 was given the molecular formula $C_{39}H_{63}NO_{10}$. On treatment with crude hesperidinase 5 gave an aglycone and two sugars, which were identified as solanidine (3) [8,9], D-glucose and L-rhamnose, respectively. The ¹³C NMR spectrum [21, 22] of 5 exhibited signals due to the carbon atoms of the sugar moiety and of the A and B rings of the aglycone (Table 1), the δ values of which were in good agreement with those of diosgenin $3-O-\alpha-1$. rhamnopyranosyl- $(1 \rightarrow 2)$ - β -D-glucopyranoside [23]. Permethylation of 5 followed by methanolysis afforded the methyl pyranosides of 2,3,4-tri-O-methyl-x-Lrhamnose and 3,4,6-tri-O-methyl-x-D-glucose. Two anomeric proton signals in the ¹H NMR spectrum of permethyl 5 and the coupling constant (J = 170 Hz, at)101.8 ppm) of $C_1 - H_1$ of the rhamnose unit observed in the ¹³C NMR spectrum of **5** indicated a β -configuration for the D-glucose unit and an α configuration for the Lrhamnose. From this, it followed that 5 was solanidine 3-O- α -L-rhamnopyranosyl- $(1 \rightarrow 2)$ - β -D-glucopyranoside $(\beta_1$ -chaconine) [24].

The FDMS of compound 6 (Fig. 1) showed a cationized cluster ion $[M + Na]^-$ and the fragment ion peaks $[(M + Na) - 146]^+$, $[(M + Na) - 162]^-$ and $[(M + Na) - 308]^+$. These data suggested that 6 was in a

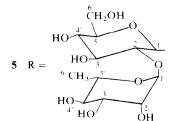
pure state and had a branched-chain trisaccharide, methylpentosyl (hexosyl)-hexose, combined with a compound of MW 397. From the elemental analytical data, the molecular formula of 6 was $C_{45}H_{73}NO_{15}$. The fact that the relative abundance $[31; [M + Na]^+ (100)]$ of the fragment ion $[(M + Na) - 146]^+$ was larger than that (12) of $[(M + Na) - 162]^+$ in the FDMS of **6** suggested, according to an established analogy between fragmentation in FDMS and sugar cleavage in acidic solvolysis [19], that on mild acid hydrolysis, removal of a methylpentose unit would be easier than that of the hexose unit. Subsequently, 6 was treated with 1 N H₂SO₄ to yield a dihexoside (8). $C_{39}H_{63}NO_{11}$ (FDMS: m/z 722 $[M - H]^+$). When 6 and 8 were hydrolysed with hesperidinase, 6 gave the same products, 3, D-glucose and L-rhamnose, as 5, while 8 provided 3 and D-glucose. The signals due to sugar carbons (Table 1) in the ¹³C NMR spectrum [21, 22] of 8 were in good agreement with those methyl β -D-glucopyranosyl- $(1 \rightarrow 4)$ - β -Dof glucopyranoside (β -cellobioside) [15]. Taking the coexistence of 5 and 6 into account, 6 was presumed to be the 3-O-x-L-rhamnopyranosyl- $(1 \rightarrow 2)$ -[β -Dglucopyranosyl- $(1 \rightarrow 4)$]- β -D-glucopyranoside of **3**. Unequivocal evidence for this structure was provided by the methanolysis of permethyl 6, yielding methyl pyranosides of 2,3,4-tri-O-methyl-x-L-rhamnose, 2,3,4,6tetra-O-methyl- and 3,6-di-O-methyl-x-D-glucoses, by the ¹HNMR spectrum of permethyl 6 showing three anomeric proton signals, and by the molecular rotation difference, -233.5° , between 6 and 11 ([M]_D: methyl α -Lrhamnopyranoside, -111° ; β -anomer, $+170^{\circ}$) [25].







3 R = H (solanidine)

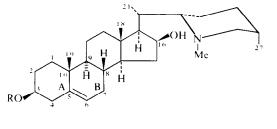


The FDMS of compound 7 (Fig. 2) showed ion peaks of $[M + Na]^+$, $[M + H]^+$, $[(M + Na) - 146]^+,$ $[(M + H) - 146]^+$, and $[(M + H) - 308]^+$, indicating that 7 was homogeneous and had a methylpentosylhexose moiety linked to an aglycone of MW 429. Elemental analysis and the above data gave the molecular formula C₄₀H₆₇NO₁₁. Incubation of 7 with hesperidinase afforded D-glucose and L-rhamnose, together with an aglycone, which was identified as hapepunine (4) [10]. Therefore, 7 was a rhamnosyl-glucoside of 4. The ¹³C NMR spectrum of 7 was the same as that of 5 with respect to the signals due to carbon atoms of the sugar moiety and of the A and B rings of the aglycone (Table 1). Consequently, 7 was characterized as the 3-O- α -Lrhamnopyranosyl- $(1 \rightarrow 2)$ - β -D-glucopyranoside of **4**.

is noteworthy that the glycosidal *Solanum* alkaloids of the aerial parts of *F. thunbergii* are different from those (predominantly ceveratrum and jerveratrum alkaloids in a free state) in the bulbs [6, 7, 12]. **5**, **6** and **7** correspond to the parent glycosides of **3** and **4**, respectively, reported by Kaneko *et al.* [6].

FDMS, requiring a very minute amount (in the order of micrograms or less) of underivatized sample, has proved to be of great diagnostic value prior to conventional chemical and spectral studies in purity control, MW determination, and structure clucidation of unknown oligoglycosides. It should be noted that the essential processes of ion formation are due to field-promoted ion-molecule reactions such as proton transfer and cationization. In particular, the analogy between acidic

4 R = H (hapepunine)



7 $R = \begin{array}{c} & \stackrel{6}{\text{CH}_2\text{OH}} \\ & \stackrel{4}{\text{HO}_3} \\ & \stackrel{6}{\text{CH}_3} \\ & \stackrel{6}{\text{CH}_3} \\ & \stackrel{6}{\text{OH}_3} \\ & \stackrel{6}{\text{HO}_3} \\ & \stackrel{6}{\text{HO}_4} \\ & \stackrel{1}{\text{HO}_4} \\ & \stackrel{6}{\text{OH}_3} \end{array}$

5 is one of the glycosidal *Solanum* alkaloids (basic steroid saponins) [14] previously isolated from the leaves of *Solanum chacoense* Bitt. (Solanaceae) and characterized by Kuhn *et al.* [24], and might be regarded as a prosapogenin of **6** with which it coexists in *F. thunbergii.* It

solvolysis and FD fragmentation [19] is a very helpful heuristic principle and can be taken as a guideline for straightforward interpretation of the FDMS and for the choice and design of appropriate analytical techniques for more subtle structural work.

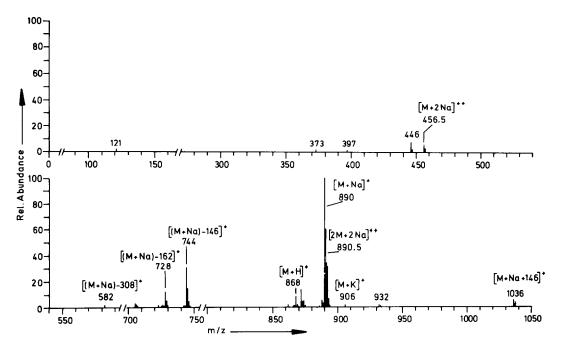


Fig. 1. FDMS of 6 (MW, 867, 498).

100 112 80 60 [M+2 Na]** 40 [M + H+ Na] 20 [(M+H)-308] Rel. Abundance 135 147 380 430 0 100 150 300 350 400 450 500 0 100 80 [M+H]* [M+Na]* 738 . 760 60 [2M+2Na]** [(M+Na)-146] 40 760 5 614 20 [(M+H)-146] 686 592 0 550 600 650 700 750 800 850 1000 m/z

Fig. 2. FDMS of 7 (MW, 737, 471).

EXPERIMENTAL

Table 1. ¹³C NMR chemical shifts of compounds 5-8 (25.05 MHz, C_5D_5N , TMS as int. standard, FT mode)

Carbon	5	8	6	7
1	37.5			37.5
2	30.2			30.2
3	77.6			77.5
4	39.0			38.8
5	140.5			140.5
6	121.6			121.7
7	31.9			32.0
8	31.5			31.8
9	50.4			50.2
10	37.0			36.9
19	19.4			19.4
1'	100.1	102.1	99.7	100.1
2'	79.4	74.6	78.0	79.4
31	77.8	76.6	75.9	77.7
4′	71.6	81.1	81.7	71.5
5′	78.0	76.2	77.4	78.0
6'	62.4	62.3	61.8	62.4
1″	101.8		101.5	101.8
2″	72.6		72.5	72.6
3″	72.3		72.1	72.3
4"	73.9		73.9	73.9
5″	69.3		69.1	69.3
6″	18.6		18.5	18.6
1 ‴		104.7	104.9	
2‴		74.6	74.7	
3‴		78.2	78.0	
4‴		71.3	70.9	
5‴		78.2	78.0	
6‴		62.3	61.8	

All mps were uncorr. Optical rotations were taken at 15-20° using a 1-dm cell. GLC (FID mode): glass column (1.5 m \times 3 mm) packed with 10 $^{\rm o}_{\rm o}$ SE-30, column temp. 150°, injection temp. 170°, carrier gases N_2 (0.8 kg/cm²) and H_2 (0.7 kg/cm²). CC and TLC (Si gel and Avicel); solvent systems: (A) CHCl₃ MeOH-H₂O (14:6:1): (B) upper layer of n-BuOH pyridine H_2O (6:2:3) + equal vol. of pyridine. The FDMS [15-19] were produced on a double-focusing MS (in Bonn) and recorded electrically with scan-times between 4 and 8 sec, decade and at a mass resolution of better than 3000 (10%, valley definition). All spectra were produced at ion source potentials of 8 kV for the field anode and - 3 kV for the slotted cathode plate, an ion source pressure of $ca = 10^{-1}$ Torr and an ion source temp, between 50 and 60°. The samples were desorbed by direct heating using the supplied emitter heating current (0-70 mA) and by controlling the emission of FD ions roughly by the total ion monitor. MeOH was used as solvent and an estimated amount of ca 1 µg was transferred to the emitter via the syringe technique.

Plant material. The aerual parts (excluding flowers) were collected in April from plants cultivated in the Nara Prefecture and in the Medicinal Botanical Garden of Kyushu University, Fukuoka Prefecture, Japan.

Isolation of constituents. Cut material (8.7 kg) was extracted with MeOH (251.) and the solvent removed in vacuo. The extract (610g) was shaken with Et₂O and H₂O, then the H₂O layer was extracted with *n*-BuOH and the extracts evapd in vacuo to dryness. The residue (67g) dissolved in 5°_{\circ} HOAc (11.) was shaken with *n*-BuOH, the aq. layer was then adjusted to pH9 with aq. 28° o NH₃ (200 ml) and extracted with *n*-BuOH. The extracts (21g) were passed through an Amberlite XAD-2 (11.) column using H₂O and 80° o MeOH as eluants. The latter eluate (9g) was chromatographed on Si gel (500g) and elution with solvent A gave fraction (Fr.) 1- Fr. 6. Further chromatography of Fr. 2 on Al₂O₃ using CHCl₃ MeOH (9:1 to 4:1) and then on Si gel (CHCl₃ MeOH H₂O. 7:12:8, lower layer) afforded 1 (13.1 mg) and **2** (6.7 mg). Fr. 4 was separated on a Si gel column (solvent A) into two fractions each of which was passed through a Sephadex LH-20 column (eluant, MeOH) to give 7 (50 mg) and 5 (74 mg) respectively. Fr. 5 was purified in the same way as above to give 5 (40 mg). Fr. 6 was subjected to DCCC [400 glass tubes (60 cm \times 1.65 mm); stationary phase, upper layer of CHCl₃-MeOH-aq. 1% NH₃ (7:12:8); moving phase, bottom layer of the above mixt.], and the first eluate was chromatographed on Si gel (EtOAc-CHCl₃-MeOH-H₂O, 2:4:4:1) to give 6 (580 mg). Compound 1, needles (from EtOAc-hexane), mp 220-222°, and 2, needles (from EtOAc-hexane), mp 212-214°, were identified [mmp, co-TLC (Si gel, solvent A), IR] as verticine and verticinone, respectively.

Compound 5. Needles (from di. MeOH), mp 287-292° (dec.), $[\alpha]_{\rm D}$ -52.5° (pyridine; c 0.9). FDMS [m/z (rel. abundance (cf. Figs. 1 and 2)]: 744 $[M + K]^+$ (7), 728 $[M + Na]^+$ (100), 706 $[M + H]^{+}$ 582 $[(M + Na) - 146]^+$ (86), (30), 560 $[(M + H) - 146]^+$ (8), 420 $[(M + Na) - 308]^+$ (4), 398 $[(M + H) - 308]^+$ (12), 375.5 $[M + 2Na]^{2+}$ (1), 364.5 $[M + Na + H]^{2+}$ (16); ¹HNMR (100 MHz, C₅D₅N. TMS as int. standard): δ 0.84 (3 H, d, J = 7 Hz, H-27), 0.95 (3 H, s, H-18), 0.97 (3 H, d, J = 7 Hz, H-21), 1.07 (3 H, s, H-19), 1.78 (3 H, d, J = 6 Hz,H-6''), 5.04 (1 H, d, J = 7 Hz, H-1'), 5.36 (1 H, m, H-6), 6.37 (1 H, s, H-1"). IR v_{max}^{KBr} cm⁻¹: 3380 (OH). ¹³C NMR: Table 1. (Found: C, 63.3; H, 8.9; N, 1.9. Calc. for C₃₉H₆₃NO₁₀. 2H₂O:C, 63.1; H 9.1; N, 1.9%).

Enzymatic hydrolysis of 5. A mixture of 5 (25 mg), a crude hesperidinase (5 mg) and aq. 1 % HOAc (5 ml) was left to stand at 37° for 24 hr. The mixture was evaporated *in vacuo* to dryness and the residue was chromatographed on Si gel (10g) using CHCl₃-MeOH-H₂O (40:10:1) and solvent A has eluants to give an aglycone, needles (from MeOH), mp 217-218° (dec.), $[\alpha]_D$ - 29.0° (CHCl₃; c 0.9), and two sugars: sugar 1, syrup, $[\alpha]_D$ + 54.0° (MeOH; c, 0.3), Avicel TLC (R_f 0.25, solvent B); sugar 2, syrup, $[\alpha]_D$ + 10.0° (MeOH; c 0.3), Avicel TLC (R_f 0.50, solvent B). The aglycone was identified (mmp, co-TLC, ¹H NMR) as solanidine, while sugars 1 and 2 were identified (Avicel TLC) as D-glucose and L-rhamnose, respectively.

Methylation of 5 [26]. NaH (100 mg) in DMF (3 ml) was stirred for 10 min, then 5 (25 mg) and MeI (1 ml) were added, the mixture was stirred for 5 hr and then poured into ice–H₂O. Extraction with EtOAc and the usual work-up gave a crude product, which was chromatographed on Si gel (5g) (CHCl₃–MeOH–H₂O (95:5:0.5)) to provide permethyl 5 as an oil (23 mg). IR: no OH; ¹H NMR (100 MHz, CDCl₃ TMS as int. standard): δ 4.26 (1 H, d, J = 7 Hz, H-1'), 5.21 (1 H, d, J = 1.5 Hz, H-1").

Methanolysis of permethyl 5. A soln of permethyl 5 (20 mg) in 8 % HCl-MeOH (5 ml) was refluxed for 2.5 hr. After cooling the mixture was neutralized with NaHCO₃, filtered, and the filtrate was passed through Sephadex LH-20 (eluant, MeOH) and ionexchange resin IR-4B (eluent, MeOH) columns. The methylated sugars in the eluate were identified as the methyl pyranosides of 2,3,4-tri-O-methyl- α -L-rhamnose and 3,4,6-tri-O-methyl- α -Dglucose by direct comparison [GLC, Si gel TLC (EtOAc-MeOH, 49:1)] with authentic samples [27].

Compound **6**. Needles (from dil. MeOH), mp 278–283° (dec.), $[\alpha]_{D} - 58.4^{\circ}$ (pyridine; *c* 1.0). MS: Fig. 1; ¹H NMR (100 MHz, C₅D₅N, TMS as int. standard): δ 0.84, 0.95 (each 3 H, *d*, *J* = 7 Hz, H-27, H-21), 0.93, 1.04 (each 3 H, s, H-18, H-19), 1.75 (3 H, *d*, *J* = 6 Hz, H-6"), 5.34 (1 H, *m*, H-6), 6.21 (1 H, *d*, *J* = 1.5 Hz, H-1"); IR v^{MBx}_{Max} cm⁻¹: 3400 (OH). ¹³C NMR: Table 1. (Found: C, 59.6; H, 8.4; N, 1.6. C_{4.5}H_{7.3}NO_{1.5} · 2 H₂O requires: C, 59.8; H, 8.6; N, 1.6%.)

Partial hydrolysis of 6. A soln of 6 (70 mg) in 1 N H₂SO₄ (10 ml) was heated on a boiling water-bath for 2 hr. The mixture was made alkaline with 10% aq. NH₃, extracted with *n*-BuOH, and the organic layer was evapd *in vacuo* to dryness. The residue was

passed through Sephadex LH-20 column (eluent, MeOH) and the eluate was further chromatographed on Si gel (solvent A) to afford a crude product. It was crystd from dil. MeOH to give a glycoside (8) as needles (from dil. MeOH) (38 mg), mp 260-261° (dec.), $[\alpha]_D - 38.0^\circ$ (Py; c 1.2). FDMS: m/z 722 $[M(C_{39}H_{63}NO_{11}) + H]^+$; ¹³C NMR: Table 1.

Enzymatic hydrolysis of 6 and 8. 6 (150 mg) and a crude hesperidinase (20 mg) in aq. 1 % HOAc (10 ml) were left to stand at 37° for 24 hr and worked up in the same way as for 5 to yield 3, D-glucose and L-rhamnose, which were identified by direct comparison with authentic samples. 8 (15 mg) treated in the same manner gave 3 and D-glucose.

Methylation of 6. This was carried out in the same way as for 5 (NaH, 100 mg; DMF, 4 ml; 6, 40 mg; MeI, 2 ml) to provide permethyl 6 as an oil (37 mg). IR : no OH; ¹H (100 MHz, C₅D₅N, TMS as int. standard) NMR : δ 4.26, 4.36 (each 1 H, d, J = 7 Hz, H-1', H-1'''), 5.25 (1 H, d, J = 2 Hz, H-1'').

Methanolysis of permethyl 6. This was carried out in the same way as for permethyl 5 (permethyl 6, 37 mg; 8% HCl-MeOH, 5 ml). The methylated sugars were identified as methyl pyranosides of 2,3,4-tri-O-methyl- α -L-rhamnose, 2,3,4,6-tetra-Omethyl- and 3,6-di-O-methyl- α -D-glucoses by comparison (GLC, TLC) with authentic samples [27] run in parallel.

Compound 7. Needles (from dil. MeOH). mp 269–274° (dec.), $[\alpha]_D - 67.2^\circ$ (Py; *c* 1.5). FDMS: Fig. 2; ¹H NMR (100 MHz, C₅D₅N, TMS as int. standard): δ 1.07 (6 H, *s*, H-18, H-19), 1.07 (6 H, *d*, *J* = 7 Hz, H-21, H-27), 1.85 (3 H, *d*, *J* = 6 Hz, H-6"), 2.21 (3 H, *s*, N-Me), 5.00 (1 H, *d*, *J* = 7 Hz, H-1'), 5.25 (1 H, *m*. H-6), 6.32 (1 H, *s*, H-1"); ¹³C NMR: Table 1. (Found: C, 64.1; H, 9.3; N, 1.6. C₄₀H₆₇NO₁₁ · H₂O requires: C, 63.6; H, 9.2; N, 1.6 %.)

Enzymatic hydrolysis of 7. This was carried out in the same way as for 5 (7, 40 mg; hesperidinase, 5 mg; aq. 1% HOAc, 5 ml) providing D-glucose and L-rhamnose together with an aglycone, needles (from Me₂CO-hexane), mp 201-202°, $[\alpha]_D - 71.2°$ (Py; c 1.0). IR $v_{\text{Mar}}^{\text{Kar}}$ cm⁻¹: 3450 (OH); ¹H NMR [100 MHz, CDCl₃-CD₃OD (9:1), TMS as int. standard]: δ 0.94, 1.01 (each 3 H, s, H-18, H-19), 1.07 (6 H, d, J = 7 Hz, H-21, H-27), 2.40 (3 H, s, N-Me), 3.52 (1 H, m, H-3\alpha), 4.45 (1 H, m, H-16\alpha), 5.35 (1 H, m, H-6). (Found: C, 76.0; H, 11.0; N, 3.0. Calc. for C₂₈H₄₇NO₂·1/2 H₂O: C, 76.7; H, 11.0: N, 3.2%.) It was identified as hapepunine by direct comparison (mmp, TLC, ¹H NMR) with an authentic sample.

Acknowledgements -- The authors are grateful to Prof. I. Nishioka of Kyushu University for the collection and identification of the plant, to Prof. H. Mitsuhashi and Prof. K. Kaneko, Hokkaido University, for the authentic samples of solanidine and hapepunine, to Prof. S. Itô, Tohoku University, for verticine and verticinone, and to Prof. O. Tanaka, Hiroshima University, for a crude hesperidinase. Thanks are also due to Mr. R. Müller, University of Bonn, Mr. A. Tanaka, Miss K. Soeda and the members of the Central Analytical Department of Kyushu University for FDMS, ¹³C NMR, ¹H NMR and elemental analytical data. This work was supported by a Grantin-Aid for scientific research from the Ministry of Education, Science and Culture, Japan, a grant from the Japan Society for the Promotion of Science, the Deutsche Forschungsgemeinschaft (Schu 416/1-3), the Ministerium für Wissenschaft und Forschung des Landes Nordrhein-Westfalen, and the Umweltbundesamt, Berlin, all of which are gratefully acknowledged.

REFERENCES

 Pharmaceutical Institute Chinese Academy of Medical Science and others (1979) Zhong Yao Zhi, Vol. 1, p. 93. Peking.

- 2. Fukuda, M. (1948) Nippon Kagaku Zasshi 69, 165.
- Morimoto, H. and Kimata, S. (1960) Chem. Pharm. Bull. 8, 871.
- 4. Itô, S., Kato, M., Shibata, K. and Nozoe, T. (1963) Chem. Pharm. Bull. 11, 1337.
- 5. Itô, S., Fukazawa, Y., Okuda, T. and Iitaka, Y. (1968) Tetrahedron Letters 5373.
- Kaneko, K., Tanaka, M., Haruki, K., Naruse, N. and Mitsuhashi, H. (1979) *Tetrahedron Letters* 25, 3737.
- Kitajima, J., Noda, N., Ida, Y., Miyahara, K. and Kawasaki. T. (1981) *Heterocycles* 15, 791.
- Sato, Y. (1970) in *Chemistry of the Alkaloids* (Pelletier, S. W., ed.) p. 618. Van Nostrand-Reinhold, New York.
- Schreiber, K. (1968) in *The Alkaloids* (Manske, R. H. F., ed.) Vol. 10, p. 1. Academic Press, New York.
- Kaneko, K., Nakaoka, U., Tanaka, M., Yoshida, N. and Mitsuhashi, H. (1981) *Phytochemistry* 20, 157.
- 11. Kaneko, K., Naruse, N., Haruki, K. and Mitsuhashi, H. (1980) Chem. Pharm. Bull. 28, 1345.
- Kaneko, K., Naruse, N., Tanaka, M., Yoshida, N. and Mitsuhashi, H. (1980) Chem. Pharm. Bull. 28, 3711.
- Kaneko, K., Naruse, N., Yoshida, N. and Mitsuhashi, H. (1980) 27th Annual Meeting of the Japanese Society of Pharmacognosy, Nagoya, Abstracts of Papers, p. 28.
- Kawasaki, T. (1978) in *Methodicum Chimicum* (Korte, F. and Goto, M., eds.) Vol. 11/3, p. 88. Academic Press, New York.

- 15. Schulten, H.-R., Komori, T. and Kawasaki, T. (1977) Tetrahedron 33, 2595.
- Schulten, H.-R., Komori, T., Nohara, T., Higuchi, R. and Kawasaki, T. (1978) *Tetrahedron* 34, 1003.
- Komori, T., Kawamura, M., Miyahara, K., Kawasaki, T., Tanaka, O., Yahara, S. and Schulten, H.-R. (1979) Z. Naturforsch. Teil C 34, 1094.
- 18. Geiger, H. and Schwinger, G. (1980) Phytochemistry 19, 897.
- Komori, T., Maetani, I., Okamura, N., Kawasaki, T., Nohara, T. and Schulten, H.-R. (1981) Justus Liebigs Ann. Chem. 683.
- 20. Hostettmann, K. (1980) Planta Med. 39, 1.
- Yahara, S., Tanaka, O., Komori, T. and Kawasaki, T. (1977) 97th Annual Meeting of the Pharmaceutical Society of Japan, Tokyo, Abstracts of Papers Part 2, p. 221.
- Mahato, S. B., Sahu, N. P., Ganguly, A. N., Kasai, R. and Tanaka, O. (1980) *Phytochemistry* 19, 2017.
- Kawasaki, T. and Yamauchi, T. (1968) *Chem. Pharm. Bull.* 16, 1070.
- 24. Kuhn, R., Löw, I. and Trischmann, H. (1955) *Chem. Ber.* 88, 1690.
- 25. Klyne, W. (1950) Biochem. J. 47, 41.
- Miyahara, K., Kodama, I., Matsumoto, M. and Kawasaki, T. (1978) 98th Annual Meeting of the Pharmaceutical Society of Japan, Okayama, Abstracts of Papers, p. 306.
- Okabe, H., Koshito, N., Tanaka, K. and Kawasaki, T. (1971) Chem. Pharm. Bull. 19, 2394.