

## PRUNETIN 5-O- $\beta$ -D-GLUCOPYRANOSIDE, AN ISOFLAVONE FROM THE PEDUNCLE OF *PRUNUS AVIUM* AND *P. CERASUS*

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**Key word Index**—*Prunus avium*; *P. cerasus*; Rosaceae; cherry; peduncle; prunetin 5-glucoside; isoflavone.

**Abstract**—Prunetin 5-glucoside was isolated from cherry peduncle *Prunus avium* and *P. cerasus*. The structure was fully assigned on the basis of NMR evidence. It is the second 5-O-glycosylated isoflavonoid to be reported.

Prunetin-4'-O-glucoside has been reported previously from three species of *Prunus*, whereas the parent aglycone, prunetin, has been identified from eight species of this genus [1]. Unlike the isomeric flavones, O-glycosylation among the naturally occurring isoflavonoids seems to be largely restricted to the 7- and 4'-positions [1]. To date there is only one single isoflavone mentioned in the literature being 5-O-glycosylated [2]. This communication represents the second report of such glycosylation pattern. For this compound (1) we propose the trivial name prunetinoside.

The acetone extract of cherry peduncle *Prunus avium* and *P. cerasus* [3] (commercially available as Nicardin®) was subjected to chromatography to yield prunetinoside (1) which analysed for C<sub>22</sub>H<sub>21</sub>O<sub>10</sub>. Its UV spectrum showed a prominent maximum at 259 nm indicative of an isoflavone nucleus. This was supported by a <sup>13</sup>C chemical shift of  $\delta$  153.3 for a methine carbon which corresponds to C-2 of an isoflavone and excluded the isomeric flavone structure [4]. Moreover, the H-2 chemical shift value of  $\delta$  8.15 substantiated its isoflavone nature.

The glycosidic nature of 1 and its sugar moiety was proved by hydrolysis followed by paper chromatography. The aglycone was identified as prunetin and the sugar as glucose (see Experimental). The presence of a distinct bathochromic shift (12 nm) on addition of AlCl<sub>3</sub>-HCl to its aglycone and its absence in the UV spectrum of the respective glycoside 1 indicated that the sugar was attached to C-5-OH. The <sup>1</sup>H and <sup>13</sup>C chemical shifts as well as <sup>1</sup>H-<sup>1</sup>H coupling constants confirmed that the sugar was glucose and it has the  $\beta$ -configuration [ $J_{1'',2''} \approx 7.6$  Hz].

The sites of linkages of the two substituents, Me and  $\beta$ -D-glucosyl, were ascertained by two NOE difference experiments. On irradiation of the anomeric proton (H-1'',  $\delta$  4.91) signal enhancements are observed for one aromatic proton only (H-6) proving that the sugar is attached to C-5-OH. The appearance of H-3''/5'' responses provides a further argument for the  $\beta$ -configuration of the glucosyl grouping. Irradiation of the methoxy protons, however, induces NOE of both the H-6 and H-8 signals; thus, this substituent is positioned between them, i.e. at C-7-OH. The NOE on H-6 is clearly smaller than

that on H-8. Therefore, the methyl group is directed preferably towards the H-8 atom due to its steric interference with the bulky sugar moiety.

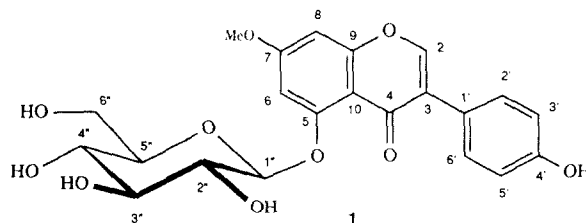
### EXPERIMENTAL

The Me<sub>2</sub>CO extract of the peduncle was concd under red. pres. and the resulting semi-solid (50 g) was chromatographed over a polyamide column (580 g; MN polyamide SC6) yielding six fractions containing mainly genistin [5]. The mother liquor was further chromatographed over polyamide. The 40% MeOH elute provided a crude prunetinoside which was further subjected to Sephadex LH 20 to yield pure 1 (28.1 mg) as crystalline colourless plates (mp 237-239°). UV  $\lambda_{\max}$  (MeOH) nm: 259; (+ AlCl<sub>3</sub>) 259; (+ AlCl<sub>3</sub>-HCl) 259; (+ NaOAc) 259; (+ NaOAc-H<sub>3</sub>BO<sub>3</sub>) 259; (+ NaOMe) 280.

**Hydrolysis.** Compound 1 (10 mg) was dissolved in 6% aq. HCl (0.5 ml) and heated at 100° for 45 min. The mixture was worked-up according to standard methods to give a pale yellow solid which yielded silky needles on crystallization. UV  $\lambda_{\max}^{\text{MeOH}}$  nm: 263, 327 (sh); (+ AlCl<sub>3</sub>) 275, 310 (sh); (+ AlCl<sub>3</sub>/HCl) 275, 310 (sh); (+ NaOMe) 275, 325 (sh). It was identified as prunetin (mmp, co-TLC). In the aqueous filtrate glucose was identified by PC.

All NMR spectra were recorded at 400.1 MHz <sup>1</sup>H and 100.6 MHz in a 1:1 mixture of deuterated methanol and DMSO. The MeOH peaks were used as internal reference:  $\delta$ (<sup>1</sup>H) = 3.30 and  $\delta$ (<sup>13</sup>C) = 49.0.

**NMR spectral data.** <sup>1</sup>H chemical shifts ( $\delta$ ) 8.15 (H-2, s), 7.38 (H-2'/6', m), 6.99 (H-6, d), 6.87 (H-3'/5', m), 6.85 (H-8, d), 4.91 (H-



1", d), 3.95 (OMe, s), 3.92 (H-6", dd), 3.70 (H-6", dd), 3.56 (H-2", dd), 3.51 (H-5", m), 3.48 (H-3", dd), 3.38 (H-4", dd);  $^1\text{H}$ - $^1\text{H}$  coupling constants (Hz)  $J$  (6, 8)=2.4,  $J$  (1", 2")=7.6,  $J$  (2", 3")=9.0,  $J$  (3", 4")=9.0,  $J$  (4", 5")=9.5,  $J$  (5", 6")=2.4 and 6.2,  $J$  (6", 6'")=12.0;  $^{13}\text{C}$  chemical shifts: 153.3 (C-2), 126.9 (C-3), 177.6 (C-4), 160.6/160.3 (C-5/C-9), 97.2 (C-6), 165.6 (C-7), 105.2 (C-8), 111.3 (C-10), 124.0 (C-1'), 131.8 (C-2'/6'), 116.2 (C-3'/5'), 158.7 (C-4'), 104.1 (C-1''), 74.8 (C-2''), 77.2 (C-3''), 71.3 (C-4''), 78.9 (C-5''), 62.5 (C-6''), 56.9 (OMe).

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## REFERENCES

- Ingham, J. L. (1983) in *Progress in the Chemistry of Organic Natural Products* (Herz, W., Grisebach, H. and Kirby, G. W. eds), **43**, p. 1. Springer, Wien.
- Mitra, J., Das, A. and Joshi, T. (1983) *Phytochemistry* **22**, 1063.
- Chem. Abstr.* **56**: 1176g.
- Markham, K. R., Chari, V. M. and Mabry, T. J. (1982) in *The Flavanoids, Advances in Research*, (Harborne, J. B. and Mabry, T. J., eds), p. 19. Chapman & Hall, London.
- Gellért, M., Szendrei, K., Dinya, Z. and Répási, J. (1985) *Flavanoids and Bioflavanoids* (Farkas, L., Gábor, M. and Kállay, F., eds). Akademia Kiado, Budapest.

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## DELBIDINE, AN ALKALOID FROM A HYBRID POPULATION OF *DELPHINIUM OCCIDENTALE* AND *DELPHINIUM BARBEYI*

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**Key Word Index**—*Delphinium occidentale/barbeyi*; Ranunculaceae;  $\text{C}_{20}$ -diterpenoid alkaloid, delbidine.

**Abstract**—A new  $\text{C}_{20}$ -diterpenoid alkaloid delbidine has been isolated from a hybrid population of *Delphinium occidentale* (S. Wats.) S. Wats. and *Delphinium barbeyi* (Huth) Huth and its structure elucidated by spectroscopic methods. The structure was confirmed by its correlation with geyeridine.

### INTRODUCTION

The diterpenoid alkaloids occur mostly in several of the *Aconitum* and *Delphinium* species which belong to the Ranunculaceae. These alkaloids have long been of interest because of their usefulness in medicine and intensely poisonous property. *Delphinium barbeyi* Huth is a herbaceous perennial plant native to the Rocky Mountain region of the United States. It grows above 7000 ft in the mountains of Utah, Wyoming and Colorado [1].

The occurrence of the  $\text{C}_{19}$ -diterpenoid alkaloids antranoyllycoctonine [2], delpheline [3], deltaline [3] and lycoctonine [2, 3] from *D. barbeyi* Huth (*D. glaucum* S. Wats) has been recorded by earlier investigators. Recently from a hybrid population of *Delphinium occiden-*

*tale* (S. Wats.) S. Wats. and *Delphinium barbeyi* (Huth) Huth delcosine [4], deltamine [4] and dictyocarpine [4] have been reported. The isolation of an unusual alkaloid barbeline containing a  $\text{C}(19)=\text{N}$  azomethine group in the  $\text{C}_{19}$ -diterpenoid alkaloid skeleton has also been reported [5] from this hybrid population. In the present note we wish to report the isolation and structure elucidation of a new  $\text{C}_{20}$ -diterpenoid alkaloid from this hybrid population.

### RESULTS AND DISCUSSION

Ethanol extraction and subsequent isolation of the basic fraction of the *Delphinium* hybrid gave a crude alkaloid which was found by TLC to be a mixture of a number of alkaloids. The mixture was purified by vacuum liquid chromatography [6] on alumina and a polar fraction was collected by elution with acetone-25% methanol. From this fraction, delbidine (3) was obtained

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