Note



Resolution and Synthesis of (S)-1-(2-Naphthyl)ethanol with Immobilized Pea Protein: As a New Biocatalyst

Hiroyuki Nagaoka*,*** and Hiroshi Kayahara**,†

*The United Graduate School of Agricultural Science, Gufu University, 1-1 Yanagido, Gufu 501-1193, Japan **Department of Bioscience and Biotechnology, Shinshu University, 8304 Minamiminowa, Kamiina, Nagano 399-4598 Japan

***Sanyo Shokuhin Co., Ltd., 555-4 Asakura, Maebashi, Gunma 371-0811, Japan

Received April 28, 1999; Accepted July 14, 1999

(S)-1-(2-Naphthyl)ethanol was yielded by immobilized pea (*Pisum sativum L.*) protein (IPP) from (R,S) 2-naphthyl ethanol (>99% ee, yield; about 50%), in which the (R)-enantiomer was selectively oxidized to 2-acetonaphthone. IPP could be reused consecutively at least three times without any decrease of yield and optical purity.

Key words: immobilized pea protein; selective oxidation; biotransformation; (\pm) 2-naphthyl ethanol; (S)-1-(2-naphthyl)ethanol

Recently, baker's yeast, 10 lipase, 20 cultured plant cells, 30 and bacteria 40 have been found to be useful biocatalysts in synthesizing chiral compounds.

We report here new catalysts, *i.e.* pea (*Pisum sativum. L.*) protein as the method of resolution and synthesis of racemic alcohol more stereoselectively and more economically than the above-mentioned biocatalysts. Additionally, this method may have such merits as the ease in separation the catalyst from the reaction solution and its reuse it without any decrease in yield and optical purity.

Pea protein (PP) powder was obtained from Orugano Co. Ltd., which was prepared by the spray dry method from a water extract of hulled and milled pea seeds under low pressure. The PP was immobilized as follows: the protein powder (20 g) was dissolved in distilled water (200 ml), and 5% aq. sodium arginate (250 ml) was poured into the solution with stirring. When the homogeneous solution was added dropwise to 0.6% aq. calcium chloride solution (2000 ml), calcium arginate gel beads (ϕ =about 4 mm) involving the pea protein were made. After the beads were kept further in the calcium chloride solution for 5 h, the beads were separated from the solution and washed with distilled water.

Then, the immobilized pea protein (IPP) was used for the resolution of (\pm) 2-naphthyl ethanol (1). That is, 200 mg of I was added to distilled water containing IPP at $33 \sim 37$ °C in the absence of any organic solvent, buffer, or surfactant without stirring. After filtration, the separation of (S)-alcohol and 2-acetonaphthone was done with silica gel chromatography (elution solvent; hexane:ethyl acetate=9:1). I was biotransformed into (S)-alcohol with excellent optical yield (>99% ee, yield: about 50%) through the stereoselective oxidation of its

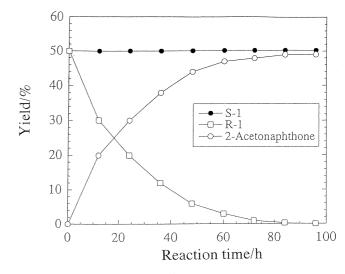


Fig. 1. Biotransformation of (*S*)-1-(2-Naphthyl)ethanol with Immobilized Pea (*Pisum sativum L.*) Protein through Selective Oxidation of Its (*R*)-Enantiomer to 2-Acetonaphthone.

Table 1. Chiral Resolution of Racemic 1-(2-Naphthyl)ethanol Using Immobilized Pea Protein (IPP) and Consecutive Reuse (2nd, 3rd) of IPP

Reuse	Time/h ^a	Resolved Alcohol	OP/% ee ^b	CY/%°
1st	96	S	99	50
2nd	48	S	99	50
3rd	36	S	99	50

- a Reaction time.
- ^b Optical purity measured by HPLC.
- ^c Chemical yield.

(R)-isomer to 2-acetonaphthone in the way depicted in Fig. 1. HPLC analysis was done by using a Chiralcel OB column (elution solvent; hexane:2-propanol=9:1, flow rate; $0.5 \text{ cm}^3 \text{ min}^{-1}$, retention time=14.2 min (99.9%, (S)-alcohol), 18.1 min (0.1%, (R)-alcohol).

As can be seen in Table 1, IPP was found to be reusable at least three times with accelaration in the reaction rate without any decrease of yield and optical purity.

This acceleration may be due to a steric change of the catalyst in favor of fitting the substrate during the process of reaction.

[†] To whom correspondence should be addressed. Hiroshi KAYAHARA, Tel: 81-265-77-1600; Fax: 81-265-77-1629

We suggest that pea protein powder may be very useful in stereoselectively catalytic reactions of the powder as a new biotransformation of chiral compound, in which the reaction condition are mild (35°C, pH=7) and needs no preincubation or bacteria-free operation in contrast to the case of cultured plant cell.

References

- 1) Cruk, T. and Glanzer, B. I., Baker's yeast mediated transformation in organic chemistry. Chem. Rev., 91, 49-97 (1991).
- Santaniello, E., Ferraboschi, P., Grisenti, P., and Manzocchi, A., The biocatalytic approach to the preparation of enantiometrically pure chiral building blocks. *Chem. Rev.*, 92, 1071-1140 (1992).
- a) Naoshima, Y., Akakabe, Y., and Watanabe, F., Biotransformation of acetoacetic esters with immobilized cells of *Nicotiana tabacum*. Agric. Biol. Chem., 53, 545-547 (1989); b) Naoshima, Y. and Watanabe, Y., Biotransformation of some ketoesters through the consecutive reuse of immobilized Nicotiana tabacum cells. J. Org. Chem., 54, 4237-4239 (1989); c) Naoshima, Y. and Akakabe, Y., Biotransformation of aromatic ketones with cell cul-

- tures of carrot, tobacco, and gardenia. *Phytochemistry*, **30**, 3595–3597 (1991); d) Naoshima, Y. and Akakabe, Y., The mechanistic pathway of the biotransformation of acetophenone by immobilized cell culture of gardenia. *Phytochemistry*, **32**, 1189–1191 (1993); e) Naoshima, Y. and Akakabe, Y., Biotransformation of acetophenone with immobilized cell of carrot, tobacco and gardenia. *Phytochemistry*, **35**, 661 (1994).
- Takemoto, M., Moriyasu, Y., and Achiwa, K., Synthesis of optically active α-phenylpyridylmethanols with cell cultures of Nicotiana tabacum. Chem. Pharm. Bull., 43, 1458-1461 (1995).
- a) Seebach, D., Gipovannini, F., and Lamatsch, B., Preparative asymmetric reduction of 3-ketobutyrate and -valerate by suspended cell of thermophilic bacteria (*Thermoanaerobium brockii*) in ordinary laboratory equipment. *Helv. Chim. Acta.*, 68, 958-960 (1985); b) Yamazaki, Y. and Hosono, K., Microbial asymmetric reduction of organometallic ketones: Acetyl ferrocene and (acetophenone)-tricarbonylchromium. *Agric. Biol. Chem.*, 52, 3239-3240 (1988); c) Miya, M., Kawada, M., and Sugiyama, Y., Stereoselective reduction of ethyl 2-methyl-3-oxobutanoate by bacteria. *Biosci. Biotechnol. Biochem.*, 60, 95-98 (1996).
- 6) Yamada, H. and Shimizu, S., Microbial and enzymatic processes for the production of biologically and chemically useful compounds. Angew. Chem. Int. Ed. Engl., 27, 622-642 (1988).