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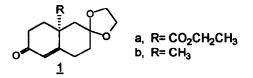
## ENANTIOSELECTIVE SYNTHESIS OF SUBSTITUTED OCTALONES

Lúcia C. Sequeira<sup>a,b</sup>; Paulo R. R. Costa<sup>b\*</sup>; Alexandre Neves<sup>b</sup> and Pierre Esteves<sup>a</sup>

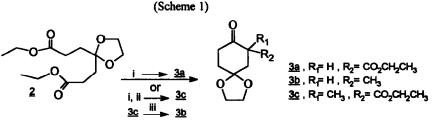
a) Instituto de Química b) Núcleo de Pesquisas de Produtos Naturais - Universidade Federal do Rio de Janeiro 21941, Cidade Universitária (CCS - Bloco H), Rio de Janeiro, R. J., Brasil

**Abstract:** The octalones (R)-(-)-<u>1a</u> and (S)-(-)-<u>1b</u> were prepared from the Michael adducts <u>7a</u> and <u>7b</u> respectively. These intermediates were obtained by the "deracemizing alkylation" of cyclanones <u>3a</u> and <u>3b</u>.

As part of a program aimed at the synthesis of terpenoids isolated from Brazilian plants, we needed to prepare the octalones  $\underline{1a-b}$  in high enantiomeric purity. In this paper, we describe the enantioselective synthesis of these octalones, using the "deracemizing alkylation of chiral imines" as the key step to introduce the chirality. Both the operational simplicity and the recovery of the chiral auxiliary make this method one of the most attractive for the enantioselective construction of quaternary stereogenic centers.

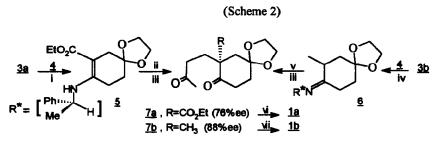


The starting materials, the cyclohexanones <u>3a-b</u> were prepared from available diethyl  $\gamma$ ethylenedioxypimelate <u>2</u>, obtained from furfural<sup>2</sup>. It has been demonstrated that the generation of cyclic  $\beta$ -ketoesters by Dieckmann condensations is promoted by the use of potassium in toluene under sonication<sup>3</sup>. Using Luche's conditions<sup>3</sup>, <u>2</u> was transformed into <u>3a</u> in 30-40 min. at room temperature in 80-85% yield. In addition, the trapping of the K-enolate intermediate with methyl iodide (in a one pot procedure) also led to <u>3c</u> in good yield (80-85%), giving, after descarboxylation, 2-methylcyclanone <u>3b</u> (75-80%) (Scheme 1).



(i: K/toluene - ultrasound ; ii: CH3 ; iii: KOH 10% (aq.)/EtOH- refl. (16h))

Compounds <u>3a</u> and <u>3b</u> were allowed to react with (S)-(-)-phenylethylamine <u>4</u> (89% ee), respectively, leading to the enamine <u>5</u><sup>4</sup> and imine <u>6</u><sup>1</sup>. These chiral intermediates were alkylated with methylvinylketone (MVK), furnishing, after hydrolytic work-up, the corresponding Michael adducts <u>7a</u><sup>5</sup> and <u>7b</u><sup>6</sup> (Scheme 2).



(i: benz. refl. 14h ; ii: MVK / ZnCl<sub>2</sub> / THF ; iii: 10%AcOH ; iv. toluene (TsOH cat.) ;refl 6h ; v. MVK / THF ; vi: EtONa / EtOH ; vi: MeONa / MeOH )

The enantiomer of  $\underline{7a}$  was previously prepared by Hermann and Wynberg<sup>7</sup> in 20% ee and transformed by these authors into its octalone. We achieved  $\underline{7a}$  using the conditions described by Guingant<sup>4</sup> in 76% ee, estimated from <sup>1</sup>H NMR, run in the presence of [Eu(hfc)<sub>3</sub>] as chiral shift reagent. In the same way, Michael adduct  $\underline{7b}$  was prepared in 88% ee. While we were developing our work, a synthesis of the enantiomer of  $\underline{7b}$ , using the same methodology as ours, was reported<sup>8</sup>. Treatment of (R)-(-)- $\underline{7a}$  and (S)-(+)- $\underline{7b}$  with sodium alkoxide led to the desired octalones (R)-(-)- $\underline{1a}^9$  and (S)-(-)- $\underline{1b}^{10}$ , already used in racemic forms as starting materials for the preparation of racemic natural products. The synthesis of terpenes isolated from Brazilian plants using these octalones are under investigation in our laboratory.

## **References and Notes :**

- 1) d'Angelo, J.; Desmaele, D.; Dumas, F. and Guingant, A. Tetrahedron : Asymmetry, 1992, 3, 459
- 2) Narang, S. A. and Dutta, P.C. J. Chem. Soc. , 1960, 2842
- 3) (a)Einhorn, C.; Einhorn, J. and Luche, J. L. Synthesis, 1989, 782 (b) note: Ultrasonic irradiation were realized using a common cleaning bath (Thornton T-14)
- 4) Guingant, A. and Hammami, H. Tetrahedron : Asymmetry, 1991, 2, 411
- 5) (R)-(-)-<u>7a</u> (77% yield) colorless oil(bulb to bulb distillation 120° / 2.25 mm); [α]<sub>D</sub> = 36 (c=3; CCl<sub>4</sub>)<sup>7</sup> (76% ee, by NMR with [Eu(hfc)<sub>3</sub>]); <sup>1</sup>H NMR (200Mz, CDCl<sub>3</sub>) δ (ppm): 4.35-4.09(m,2H) 4.09-3.90(m,4H) 3.15-1.70(several absorptions,10H) 2.13(s,3H) 1.29(t,3H,J=7.4Hz).
- 6) (S)-(+)-<u>7</u>b (71%) pale yellow oil (bulb to bulb distillation  $120^{\circ} / 0.6 \text{ mm}$ ); [ $\alpha$ ]<sub>D</sub> = +4.6 (c=2.8,CCl<sub>4</sub>) (88%ee, by NMR with [Eu(hfc)<sub>3</sub>]); <sup>1</sup>H NMR(200Mz, CDCl<sub>3</sub>)  $\delta$ (ppm): 4.0(broad unresolved multiplet,4H) 2.70-1.60(several absorptions, 10H) 2.15(s,3H) 1.13(s,3H).
- 7) Hermann, K. and Wynberg, H. J. Org. Chem., 1979, 44, 2238
- 8) Pfau, M.; Jabin, I. and Revial, G. J. Chem. Soc. Perkin Trans I, 1993, 1935
- 9) (R)-(-)-<u>1a</u> colorless oil ( bulb to bulb dist. 145<sup>o</sup> / 1.0 mm ); [α]<sub>D</sub> = -108 (c=0.5,CCl<sub>4</sub>) <sup>1</sup>H
  NMR (200Mz,CDCl<sub>3</sub>) δ(ppm): 6.01(d,1H,J=2Hz) 4.35-4.10(m,2H) 4.10-3.85(m,4H) 3.05-1.49(several absorptions, 10H) 1.28(t,3H,J= 6.3 Hz).
- 10) (S)-(-)-<u>1b</u> colorless oil (bulb to bulb dist.  $122^{\circ} / 0.8 \text{ mm}$ ) [ $\alpha$ ]<sub>D</sub> = -171 (c=0.9, CCl<sub>4</sub>) <sup>1</sup>H NMR (200Mz, CDCl<sub>3</sub>)  $\delta$ (ppm): 5.80(br. s, 1H) 4.10-3.90(m,4H) 2.80-1.50(several absor.,10H) 1.35(s, 3H).

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