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## Aryl *gem*-Difluorovinyl Pinacolboronates: Synthesis and Utility for Suzuki-Miyaura Coupling Reaction

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# Aryl *gem*-Difluorovinyl Pinacolboronates: Synthesis and Utility for Suzuki-Miyaura Coupling Reaction

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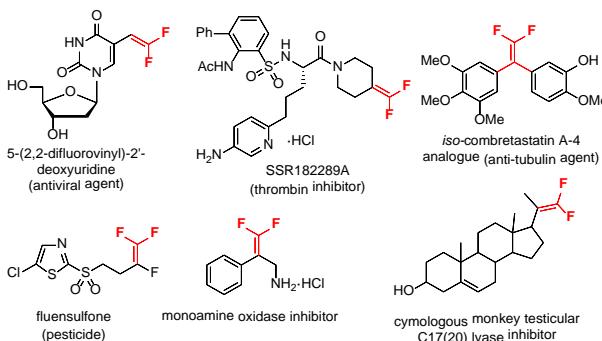
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**1** The synthesis of unstable aryl difluorovinyl pinacolboronates was achieved by the dehydrofluorination of *α*-trifluoromethyl arylmethyl pinacolboronates with LDA. These aryl difluorovinyl pinacolboronates can be used for Suzuki-Miyaura coupling with various aryl halides under Pd-catalysis to furnish diaryl *gem*-difluorovinyl compounds.

**7** **Keywords:** *gem*-difluorovinyl | pinacolboronates |  
**8** Suzuki-Miyaura coupling

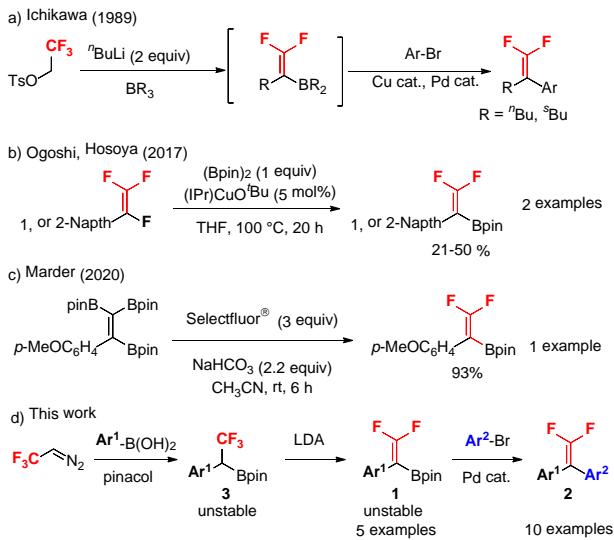
**9** In recent decades, organofluorine compounds have been  
**10** the subject of intense research activity because of their  
**11** pharmaceutical and agrochemical applications.<sup>1</sup> The efficacy  
**12** of such molecules in these fields of study/research is  
**13** rigorously related to the remarkable physical and chemical  
**14** properties created by fluoro-functional groups such as  
**15** trifluoromethyl ( $\text{CF}_3$ ), difluoromethyl ( $\text{CF}_2\text{H}$ ),  
**16** trifluoromethylthio ( $\text{SCF}_3$ ), pentafluorosulfanyl ( $\text{SF}_5$ ) and  
**17** others in their structures.<sup>1,2</sup> Among distinctly varying fluoro-  
**18** functional groups, the *gem*-difluorovinyl group ( $\text{C}=\text{CF}_2$ ) is  
**19** particularly attractive synthetically<sup>3-7</sup> since the *gem*-  
**20** difluorovinyl moiety is exceptionally reactive towards  
**21** nucleophiles, allowing it to be efficiently transformed into  
**22** other fluorinated functional groups such as  $\text{CF}_3$ ,<sup>4</sup> *gem*-  
**23** difluoromethylene ( $\text{R}-\text{CF}_2\text{R}'$ ),<sup>5</sup> monofluoroalkene ( $\text{C}=\text{CFR}$ ,  
**24**  $\text{C}=\text{CFAr}$ ),<sup>6</sup> and even non-fluorinated functional groups of  
**25** carboxylic acids and esters.<sup>7</sup> The *gem*-difluorovinyl moiety is  
**26** also a known bioisostere for a carbonyl group.<sup>8</sup> Moreover, the  
**27** high electrophilicity of the  $\text{C}=\text{CF}_2$  group is attractive for the  
**28** design of biologically active molecules, including enzyme  
**29** inhibitors,<sup>9</sup> pesticides,<sup>10</sup> and others<sup>11</sup> (Figure 1).



**32** **Figure 1.** Examples of biologically active molecules with a *gem*-  
**33** difluorovinyl unit.

**35** The *gem*-difluorovinyl compounds have been traditionally  
**36** synthesized from carbonyl compounds by the Wittig

**37** reaction,<sup>12</sup> Julia-Kocienski cross-coupling,<sup>13</sup> Horner-  
**38** Wadsworth-Emmons reactions,<sup>14</sup> and others.<sup>15</sup> Nowadays,  
**39** vast synthetic approaches for *gem*-difluorovinyl compounds  
**40** have emerged.<sup>16</sup> Despite the tremendous methodologies for  
**41** synthesizing *gem*-difluorovinyl compounds,<sup>3,12-16</sup> we noticed  
**42** that the Suzuki-Miyaura cross-coupling reaction of aryl *gem*-  
**43** difluorovinyl boronates ( $\text{R}_2\text{B}(\text{Ar})\text{C}=\text{CF}_2$ ) with aryl halides  
**44** has never appeared in the literature, except for Ichikawa's  
**45** seminal work in 1989.<sup>17</sup> Ichikawa and co-workers reported  
**46** the synthesis of 2-alkyl-2-aryl-substituted *gem*-difluorovinyl  
**47** compounds by the coupling reaction of aryl bromides and  
**48** *gem*-difluorovinyl alkyl boranes ( $\text{R}_2\text{B}(\text{R})\text{C}=\text{CF}_2$ ,  $\text{R} = {}^n\text{Bu}$ ,  
**49**  ${}^t\text{Bu}$ ) generated *in situ* from 2,2,2-trifluoromethyl *p*-toluene  
**50** sulfonate with trialkyl boranes *via* a one-pot transformation  
**51** (Scheme 1a). Although this method is efficient, the *gem*-  
**52** difluorovinyl boranes are limited to alkyl-aryl-substituted  
**53** molecules ( $\text{Ar}(\text{R})\text{C}=\text{CF}_2$ ). Thus, diaryl *gem*-difluorovinyl  
**54** molecules ( $\text{Ar}'(\text{Ar})\text{C}=\text{CF}_2$ ) cannot be accessed.



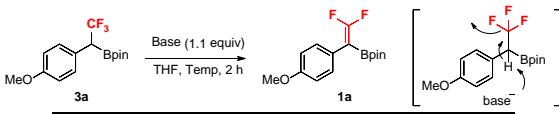
**55** **Scheme 1.** *gem*-Difluorovinyl boronates and their reaction. a) Coupling reaction of alkyl *gem*-difluorovinyl boronates with aryl bromides. b) First report of aryl *gem*-difluorovinyl boronates. c) Second report of aryl *gem*-difluorovinyl boronates. d) The preparation of aryl *gem*-difluorovinyl boronates and their application for coupling reaction (this work).

**62** We envisaged that if the aryl *gem*-difluorovinyl boronates  
**63** ( $\text{R}_2\text{B}(\text{Ar})\text{C}=\text{CF}_2$ ) could be synthesized, then Ichikawa's  
**64** methodology would become a standard for the synthesis of  
**65** both alkyl-aryl and diaryl *gem*-difluorovinyl compounds by

the Suzuki-Miyaura coupling reaction. When we started this work in 2016,<sup>18</sup> aryl *gem*-difluorovinyl boronates ( $R_2B(Ar)C=CF_2$ ) were completely unknown in the SciFinder® database. In 2017, Ogoshi, Hosoya, and co-workers reported the first preparation of *gem*-difluorovinyl pinacolboronates having a naphthyl group ( $\text{pinB}(\text{Naph})C=CF_2$ ) by regioselective monodefluoroborylation of trifluoroalkenes under Cu-catalysis, with 21–50% yield (two examples) (Scheme 1b).<sup>19</sup> In 2020, Marder et al. disclosed a single example of the synthesis of *p*-methoxyphenyl *gem*-difluorovinyl pinacolboronate ( $\text{pinB}(p\text{-MeOC}_6\text{H}_4)C=CF_2$ ) by difluorination of 1,1,2-triborylalkene by Selectfluor® with 93% yield (one example) (Scheme 1c).<sup>20</sup> However, there is no report of their use for any coupling reaction. Moreover, their methods are limited (total 3 examples), and the preparation of the corresponding starting materials requires steps. Thus, the development of a novel synthetic method for the aryl *gem*-difluorovinyl boronates remains a challenge. Herein, we report the concise synthesis of aryl *gem*-difluorovinyl pinacolboronates ( $\text{pinB}(Ar)C=CF_2$ , **1**) and their application for the Suzuki-Miyaura cross-coupling reaction to access the unsymmetrical diaryl *gem*-difluorovinyl compounds ( $Ar'(Ar)C=CF_2$ , **2**). The key to preparing aryl *gem*-difluorovinyl pinacolboronates **1** is the dehydrofluorination of unstable  $\alpha$ -trifluoromethyl- $\alpha$ -arylmethyl pinacolboronates **3** by treatment with LDA. While the aryl *gem*-difluorovinyl pinacolboronates **1** were also found to be unstable, the following Suzuki-Miyaura cross-coupling reaction of **1** with aryl bromides **4** proceeded smoothly to provide the desired diaryl-*gem*-difluorovinyl compounds **2** in moderate overall yields. Our method for the preparation of aryl *gem*-difluorovinyl pinacolboronates **1** is very different from previous two approaches.<sup>19,20</sup> Besides, in our knowledge, this is the first example of Suzuki-Miyaura cross-coupling reaction of aryl *gem*-difluorovinyl pinacolboronates **1**.

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**Table 1.** Optimization of dehydrofluorination reaction conditions of **3a**<sup>a</sup>



Entry	Base	Temp	Yield (%) <sup>a</sup>
1	$^n\text{BuLi}$	rt	N.D.
2	LiHMDS	rt	7
3	KHMDS	rt	9
4	LDA	rt	30
5	LDA	-20 °C	62

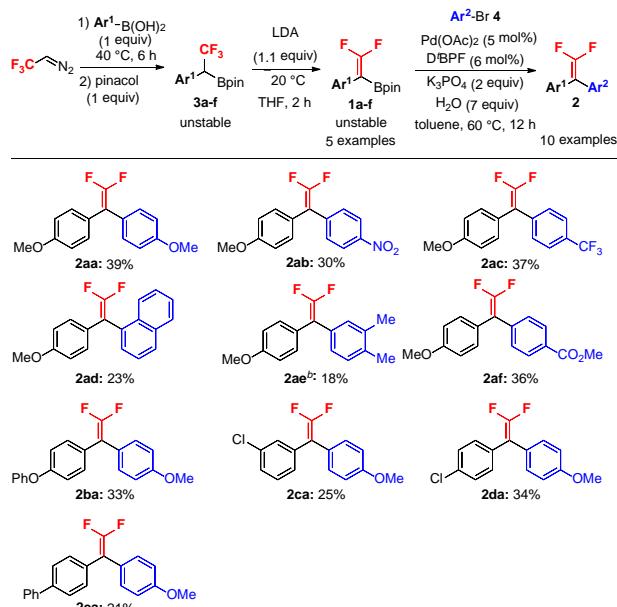
<sup>a</sup> Isolated yield

42

Molander and co-workers reported the synthesis of  $\alpha$ -trifluoromethyl- $\alpha$ -arylmethyl pinacolboronates **3** from 2,2,2-trifluorodiazethane ( $\text{CF}_3\text{CH=N}_2$ ) and aryl boronic acids ( $Ar^1\text{-B(OH)}_2$ ).<sup>21</sup> We imagined that the treatment of

pinacolboronates **3** with a base should induce dehydrofluorination triggered by negative hyperconjugation of the  $\text{CF}_3$  group,<sup>22</sup> resulting in the targeted aryl *gem*-difluorovinyl pinacolboronates **1** (Table 1). Our study commenced by optimizing the dehydrofluorination of  $\alpha$ -trifluoromethyl pinacolboronates **3**. We selected  $\alpha$ -trifluoromethyl-*p*-methoxyphenylmethyl pinacolboronate **3a** as a model substrate. According to Molander's protocol, **3a** was prepared from  $\text{CF}_3\text{CH=N}_2$  and *p*-MeOC<sub>6</sub>H<sub>4</sub>B(OH)<sub>2</sub>.<sup>21</sup> Using  $\alpha$ -CF<sub>3</sub>-pinacolboronate **3a**, we examined the base-mediated dehydrofluorination of **3a** to **1a**. The results of a brief screening of bases are summarized in Table 1. The use of  $^n\text{BuLi}$  at room temperature did not afford the desired dehydrofluorination product **1a** (entry 1). On the other hand, LiHMDS or KHMDS gave the desired dehydrofluorinated product **1a** with low yield (entries 2 and 3). Moreover, the use of LDA at room temperature provided **1a** in 30% yield (entry 4). Yield improved to 62% when the reaction was performed at -20 °C (entry 5). Starting material **3a** is unstable, and a couple of byproducts were observed in the reaction mixture, but we could not identify them. While the dehydrofluorinated product **1a** could be isolated, **1a** slowly decomposed during the purification on silica-gel column chromatography.

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**Scheme 2.** Substrate scope of the Suzuki-Miyaura coupling reaction of aryl *gem*-difluorovinyl pinacolboronates **1** with aryl bromides **4**<sup>a</sup>

<sup>a</sup> Isolated yield was calculated according to aryl boronic acid ( $Ar^1\text{-B(OH)}_2$ ). See the details of experimental procedure for **2** in supporting information (SI). <sup>b</sup> Commercial grade of 4-bromo-*o*-xylene (**4e**) (contaminated with 3-bromo-*o*-xylene) was used.

80

With the desired **1a** in hand, we examined the Suzuki-Miyaura cross-coupling of **1a** with aryl bromides. Since both pinacolboronate **1a** and precursor **3a** are unstable, we decided to examine the reaction from the start without purification in each step (Scheme 2). After the initial screening of reaction conditions focusing on the Suzuki-Miyaura coupling reaction,

we found that  $\text{Pd}(\text{OAc})_2$ , 1,1'-bis(di-tert-butylphosphino)ferrocene (DBPF),  $\text{K}_3\text{PO}_4$ , and  $\text{H}_2\text{O}$  in toluene at 60 °C provided the desired coupling product **2aa** in 39% overall yield. With these optimized reaction conditions, the substrate scope for aryl bromides **4** was investigated. Taking into consideration the instability and purification difficulty of pinacol boronates **1** and **3**, a stoichiometric amount of aryl boronic acid ( $\text{Ar}^1\text{-B(OH)}_2$ ) was used as starting material to proceed with the subsequent reactions to afford coupling products **2**. When using **1a**, electron-donating and -withdrawing groups substituted aryl bromides **4a-c** were well-tolerated in the coupling reaction to provide the corresponding diaryl *gem*-difluorovinyl products **2** in good yields (**2aa-ac**). Besides, 1-bromonaphthalene (**4d**) with **1a** also proceeded with the coupling reaction under optimal reaction conditions to afford the corresponding diaryl-*gem*-difluorovinyl-product (**2ad**: 23%). 4-Bromo-*o*-xylene (**4e**) with **1a** gave the desired product in lower yield (**2ae**: 18%). The aryl bromide with a methyl ester moiety **4f** was also tolerated and accepted for the coupling reaction with **1a** to furnish product **2af** in 36% yield. We further examined the substrate scope on *gem*-difluorovinyl boronates **1**. Electron-donating phenoxy-substituted *gem*-difluorovinyl boronates **1b**, electron-withdrawing chloro-substituted **1c, 1d**, and aryl-substituted **1e** were transformed with *p*-methoxyphenyl bromide (**4a**) into the corresponding unsymmetrical diaryl *gem*-difluorovinyl compounds **2ba, 2ca, 2da** and **2ea** at an acceptable overall yield of 21-34%, independent of the substitution of the  $\text{Ar}^1$  group of **2**.

In summary, we report a feasible method for the synthesis of aryl *gem*-difluorovinyl boronates **1**, and subsequent access to unsymmetrical diaryl *gem*-difluorovinyl products **2** via the Pd-catalyzed Suzuki-Miyaura cross-coupling reaction. While the aryl *gem*-difluorovinyl pinacolboronates **1** were found to be unstable, they can be straightforwardly synthesized from readily available trifluorodiazooethane and aryl boronic acids followed by LDA treatment. The wide synthetic application of aryl *gem*-difluorovinyl boronates **1** for a variety of coupling reactions is expected.

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Supporting Information is available on [http://dx.doi.org/10.1246/cl.\\*\\*\\*\\*\\*](http://dx.doi.org/10.1246/cl.*****).

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