## Highly Diastereoselective 1,2-Asymmetric Addition of Dialkylzincs to Chiral 2-Phenylpropanal Catalyzed by Amino Alcohol

Kenso Soai,\* Seiji Niwa, and Toshihiro Hatanaka Department of Applied Chemistry, Faculty of Science, Science University of Tokyo, Shinjuku-ku, Tokyo 162 (Received February 20, 1990)

**Synopsis.** Diastereoselective addition of dialkylzinc reagents to 2-phenylpropanal using amino alcohol as catalyst afforded *erythro* alcohols (Cram-selectivity) in high diastereomeric excess (up to 88% d.e.).

Many efforts have been devoted to understanding 1,2-asymmetric alkylation of chiral carbonyl compounds, and useful models for predicting the relative stereochemistry have been provided.<sup>1)</sup> Among chiral carbonyl compounds, 2-phenylpropanal (1) has been one of the most widely used fundamental indicators of the diastereoselectivity of 1,2-asymmetric induction (Cram's open-chain<sup>2)</sup> and Felkin's<sup>3)</sup> models) in addition reactions of organometallic reagents. However, from the standpoint of organic synthesis, diastereomeric excesses (d.e.'s) of Cram addition to 1 have been low to

$$\begin{array}{c} \text{Ph} \\ \text{Me} \\ \text{CHO} \\ \text{(1)} \\ \text{(2)} \\ \\ \text{(3)} \\ \text{(3)} \\ \text{(3)} \\ \text{(3)} \\ \text{(3)} \\ \text{(4)} \\ \text{(4)} \\ \text{(5)} \\ \text{(5)} \\ \text{(5)} \\ \text{(5)} \\ \text{(5)} \\ \text{(6)} \\ \text{(2)} \\ \text{(2)} \\ \text{(3)} \\ \text{(4)} \\ \text{(2)} \\ \text{(5)} \\ \text{(5)} \\ \text{(5)} \\ \text{(6)} \\ \text{(2)} \\ \text{(2)} \\ \text{(3)} \\ \text{(4)} \\ \text{(5)} \\ \text{(5)} \\ \text{(5)} \\ \text{(5)} \\ \text{(2)} \\ \text{(2)} \\ \text{(3)} \\ \text{(4)} \\ \text{(5)} \\ \text{(5)} \\ \text{(5)} \\ \text{(5)} \\ \text{(5)} \\ \text{(6)} \\ \text{(7)} \\ \text{(7)} \\ \text{(7)} \\ \text{(8)} \\ \text{(7)} \\ \text{(8)} \\ \text{(8)} \\ \text{(8)} \\ \text{(1)} \\ \text{(1)} \\ \text{(2)} \\ \text{(2)} \\ \text{(3)} \\ \text{(3)} \\ \text{(4)} \\ \text{(4)} \\ \text{(5)} \\ \text{(5)} \\ \text{(5)} \\ \text{(5)} \\ \text{(5)} \\ \text{(6)} \\ \text{(7)} \\ \text{(7)} \\ \text{(8)} \\ \text{(8)} \\ \text{(1)} \\ \text{(8)} \\ \text{(1)} \\ \text{(1)} \\ \text{(1)} \\ \text{(2)} \\ \text{(2)} \\ \text{(3)} \\ \text{(4)} \\ \text{(4)} \\ \text{(5)} \\ \text{(5)} \\ \text{(5)} \\ \text{(5)} \\ \text{(6)} \\ \text{(7)} \\ \text{(7)} \\ \text{(8)} \\ \text{(7)} \\ \text{(8)} \\ \text{($$

of auxiliaries such as TiCl<sub>4</sub><sup>10</sup> and crown ether.

We report a highly diastereoselective addition to **1** with an organozinc reagent using a catalytic amount of auxiliary (catalyst). During our continuing study on the enantioselective addition of dialkylzincs to aldehydes,<sup>11)</sup> we found that **1** is alkylated diastereoselectively in high d.e.'s with dialkylzincs (**2**) using amino alcohols (**5**)<sup>12)</sup> as catalysts. When **1** was treated with Et<sub>2</sub>Zn using 10 mol% of 2-dimethylaminoethanol

(5c) in hexane at room temperature, 2-phenylpentan-3-ols (3b+4b) were obtained in a combined 64% isolated yield. 13) Erythro-3b (Cram selectivity) was formed predominatly in 86% d.e. (determined by GLC analysis) (Table 1, Entry 4). When the reaction was

run at 0 °C, d.e. of **3b** increased to 88% d.e. (Entry 5). Tratment of **1** with dibutylzinc afforded Cram-**3c** in

84% d.e. (Entry 8). The Cram-selectivity obtained here

is much higher than those of the existing organozinc

moderate, i.e., MeMgI,<sup>2)</sup> EtMgBr,<sup>2)</sup> MeZnX,<sup>4)</sup> Et<sub>2</sub>Zn-

TiCl<sub>4</sub>,<sup>5)</sup> MeLi.<sup>6)</sup> It is only recently that more

diasetereoselective methods using organometallic reagents

with certain auxiliaries, i.e., Bu<sub>2</sub>CuLi-Me<sub>3</sub>SiCl-crown ether,<sup>7)</sup> MeLi-TiCl<sub>4</sub>,<sup>6)</sup> Et<sub>4</sub>Pb-TiCl<sub>4</sub>,<sup>8)</sup> have been re-

ported.<sup>9)</sup> These methods require stoichiometric amounts

methods [MeZnX (ca. 20% d.e.),4) Et<sub>2</sub>Zn-TiCl<sub>4</sub> (54% d.e.)<sup>5)</sup>].

D.e. was also high when a lesser amount (5 mol%) of the catalyst (5c) was used (Entry 3). Both *primary*-amino (5a), *secondary*-amino (5b), and *tertiary*-amino (5c, 5d, 5e) alcohols act as catalysts to afford Cram-3b

in high d.e.'s (Entries 1, 2, 4, 9, and 10).15) The high

Table 1. Diastereoselective Alkylation of 1 with Dialkylzincs (2)
Using Amino Alcohols (5) as Catalysts<sup>a)</sup>

Entry	$R^1$ in $2$	Catalyst (5)		Alcohols (3+4)	
				Yield (3+4),b)/%	D.e.[Cram- <b>3</b> ],0/%
1	Et	a	b	55	83
2	Et	b	b	45	83
3 <sup>d)</sup>	Et	c	b	53	85
4	Et	c	b	64	86
5 <sup>e)</sup>	Et	c	b	60	88
$6^{f)}$	Et	c	b	65	87
7	Me	С	a	55	74
8	n-Bu	С	c	37	84
9	Et	d	b	53	85
10	Et	e	b	45	84

a) Unless otherwise noted reactions were run in hexane at room temperature for 45—51 h. Molar ratio. 1:2:5=1.0:2.0:0.1. b) Isolated yields. c) Determined by GLC analyses. Conditions: FID detector, OV-1, 50 m capillary column, column temp. 120°C. Retention time 15.7 min for 4a (minor), 16.3 min for 3a (major). Column temp. 105°C. Retention time 46.0 min for 4b (minor), 48.2 min for 3b (major). PEG-20M 25 m capillary column, column temp. 125°C. Retention time 16.5 min for 4c (minor), 19.9 min for 3c (major). d) Molar ratio. 1:2:5=1.0:2.0:0.05. e) Reaction was run at 0°C for 69 h. f) Reaction was run at 0°C in a mixed solvent of hexane and toluene (2:1, v/v) for 44 h.

diastereoselectivity of the present R<sub>2</sub>Zn with amino alcohol catalyst may results from the steric bulk of the activated organozinc reagent formed in situ by the complex formation of amino alcohol with R<sub>2</sub>Zn. These bulky alkylating reagents are more diastereoselective probably because of their more effective steric interaction with the large Ph group substituent of 1.

## **Experimental**

GLC analyses were performed on a Shimadzu GC-4C gas chromatograph with Chromatopac C-R6A data processor. 2-Phenylpropanal (1) and amino alcohols (5a—e) were purchased from Tokyo Kasei, Inc. Diethylzinc (hexane solution) was purchased from Kanto Chemical Co. Dibutylzinc was prepared according to the literature procedure. 14)

Typical procedure (Table 1, Entry 5): To a solution of (5c)(13.5 mg, 0.15 mmol) in hexane (1.5 ml) was added (1) (0.2 ml, 202 mg, 1.5 mmol). The mixture was stirred at room temperature for 20 min. Then, Et<sub>2</sub>Zn (3 mmol, 3.0 ml of 1 M hexane solution, M=mol dm<sup>-3</sup>) was added at 0 °C. The reaction mixture was stirred at 0 °C for 69 h and was quenched with 1 M HCl. The mixture was extracted with dichloromethane, and the extract was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, and evaporated under reduced pressure. The residue was purified by silica-gel TLC [hexane-AcOEt=4:1 (v/v) as developing solvent]. Alcohols (3b and 4b) were obtained in 60% total yield (148 mg). NMR and IR spectra were identical with those of authentic samples. Conditions of GLC analyses are described in a footnote to Table 1.

We thank Tri Chemical Inc. for a generous gift of Me<sub>2</sub>Zn.

## References

- 1) For reviews, see J. D. Morrison, H. S. Mosher, "Asymmetric Organic Reactions," Prentice-Hall, New York (1971); E. L. Eliel, "Asymmetric Synthesis," ed by J. D. Morrison, Academic Press, New York (1983), Vol. 2A, Chap. 5; P. A. Bartlett, *Tetrahedron*, **36**, 3 (1980).
- 2) D. J. Cram and F. A. Abd Elhafez, J. Am. Chem. Soc., 74, 5828 (1952).
- 3) M. Cherest, H. Felkin, and N. Prudent, *Tetrahedron Lett.*, **1968**, 2199.

- 4) P. R. Jones, E. J. Goller, and W. J. Kauffman, J. Org. Chem., 36, 3311 (1971).
- 5) M. T. Reetz, R. Steinbach, and B. Wenderoth, Synth. Commun., 11, 261 (1981).
- 6) M. T. Reetz, S. H. Kyung, and M. Hullmann, *Tetrahedron*, **42**, 2931 (1986).
- 7) S. Matsuzaka, M. Isaka, E. Nakamura, and I. Kuwajima, *Tetrahedron Lett.*, **30**, 1975 (1989).
- 8) Y. Yamamoto and J. Yamada, J. Am. Chem. Soc., 109, 4395 (1987).
- 9) For an anti-Cram selectivity, see K. Maruoka, T. Itoh, and H. Yamamoto, J. Am. Chem. Soc., 107, 4573 (1985); Y. Yamamoto and K. Maruyama, ibid., 107, 6411 (1985).
- 10) For a diastereoselective allylation of chiral  $\alpha$ -keto amides using TiCl<sub>4</sub>, see K. Soai and M. Ishizaki, *J. Org. Chem.*, **51**, 3290 (1986).
- 11) a) K. Soai, A. Ookawa, K. Ogawa, and T. Kaba, J. Chem. Soc., Chem. Commun., 1987, 467; K. Soai, S. Yokoyama, K. Ebihara, and T. Hayasaka, ibid., 1987, 1690; K. Soai and M. Watanabe, ibid., 1990, 43; K. Soai, S. Niwa, Y. Yamada, and H. Inoue, Tetrahedron Lett., 28, 4841 (1987); K. Soai, S. Niwa, and M. Watanabe, J. Org. Chem., 53, 927 (1988); idem., J. Chem. Soc., Perkin Trans. 1, 1989, 109; K. Soai, S. Yokoyama, T. Hayasaka, K. Ebihara, Chem. Lett., 1988, 843; K. Soai and S. Niwa, ibid., 1989, 481; K. Soai, A. Ookawa, T. Kaba, and K. Ogawa, J. Am. Chem. Soc., 109, 7111 (1987); K. Soai, H. Hori, and S. Niwa, Heterocycles, 29, 2065 (1989); K. Soai, Y. Kawase, and S. Niwa, ibid., 29, 2219 (1989); b) K. Soai, M. Watanabe, and M. Koyano, J. Chem. Soc., Chem. Commun., 1989, 534; For the related reactions, see M. Yoshioka, T. Kawakita, and M. Ohno, Tetrahedron Lett., 30, 1657 (1989); N. N. Joshi, M. Srebnik, and H. C. Brown, ibid., 30, 5551 (1989); A. Oeveren, W. Menge, and B. L. Feringa, ibid., 30, 6427 (1989); N. Oguni, Y. Matsuda, and T. Kaneko, J. Am. Chem. Soc., 110, 7877 (1988); M. Kitamura, S. Okada, S. Suga, R. Noyori, ibid., 111, 4028 (1989); K. Tanaka, H. Ushio, and H. Suzuki, J. Chem. Soc., Chem. Commun., 1989, 1700, and references cited in ref.11b.
- 12) Et<sub>2</sub>Zn adds to benzaldehyde in the presence of amino alcohol. T. Sato, K. Soai, K. Suzuki, and T. Mukaiyama, *Chem. Lett.*, **1978**, 601; T. Mukaiyama, K. Soai, T. Sato, H. Shimizu, and K. Suzuki, *J. Am. Chem. Soc.*, **101**, 1455 (1979).
- 13) To simplify the equation, only one of the enatiomers are shown. The compounds are racemic mixtures.
- 14) C. R. Noller, Org. Synth., Coll. Vol. II, 184 (1966).