

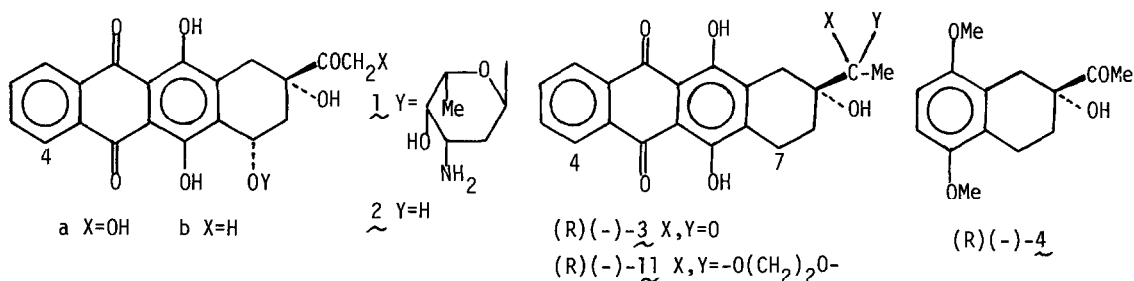
(+)- AND (-)-1-O, 4-O-BIS(p-CHLOROBENZYL)THREITOL AS NOVEL RESOLVING AGENTS
 FOR OPTICALLY ACTIVE ANTHRACYCLINONE SYNTHESIS AN EFFICIENT
 SYNTHESIS OF OPTICALLY PURE 4-DEMETHOXYDAUNOMYCINONE

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Summary (±)-7-Deoxy-4-demethoxydaunomycinone((±)-3) was cleanly resolved by forming a mixture of the diastereomeric acetals((-)-9 and (+)-10 or (+)-9 and (-)-10) with the title vicinal-diol((+)- or (-)-5) to give optically pure (R)(-)-3. The method for racemizing the undesired enantiomer((S)(+)-3) was also explored. Optically pure (+)-4-demethoxydaunomycinone((+)-2b) was elaborated from (R)(-)-3 according to the reported reaction scheme.

The 4-demethoxyanthracyclines, 4-demethoxyadriamycin(1a) and 4-demethoxydaunorubicin(1b), attract much attention since improved therapeutic indexes can be expected for these modified antibiotics which exhibit antineoplastic activity ca. 10 times higher than that of the natural anthracyclines.¹⁾ While most of the reported synthetic works on anthracyclines, the aglycones of anthracycline antibiotics, have been focused on synthesis of 4-demethoxyanthracyclines such as 4-demethoxyadriamycinone(2a) and 4-demethoxydaunomycinone(2b),²⁾ a limited number of methods is only available for producing optically active 2a,b.²⁾

Optically pure (R)(-)-7-deoxy-4-demethoxydaunomycinone((R)(-)-3), from which optically active 2a,b can be readily elaborated,³⁻⁵⁾ is anticipated to hold a pivotal position in the synthesis of optically active 2a,b, and has been synthesized from the optically pure α-hydroxyketone((R)(-)-4) produced by optical resolution^{3,6)} or by asymmetric synthesis.^{5,7)} However, this synthetic route seems to reduce its practical value since we have found that, being different from the reported results,^{3,4)} the simultaneous inter- and intramolecular Friedel-Crafts acylation of (R)(-)-4 always accompanies a slight racemization to afford (R)(-)-3 being ca. 70-75% ee.⁵⁾



We wish to report here that preparation of optically pure (R)(-)-3 can be readily achieved by the optical resolution in which (+)- or (-)-1,0-, 4,0-bis(p-chlorobenzyl)threitol ((+)- or (-)-5) is utilized as a resolving agent to form a mixture of the diastereomeric acetals((-)-9 and (+)-10 or (+)-9 and (-)-10) from the racemic α -hydroxy ketone((\pm)-3)^{5,8,9}

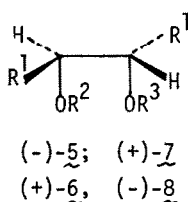
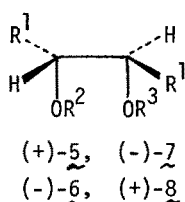
After several unsuccessful attempts,¹³⁾ (+)- and (-)-5 were chosen as the most suitable vicinal-diols for the optical resolution of (\pm)-3 due to excellent crystallization and separation properties of the diastereomeric acetals, and were readily prepared from unnatural D-(-)- and natural L-(+)-tartaric acid((-)- and (+)-6) Thus, alkylation (NaH-p-ClC₆H₄CH₂Cl in THF, 50°C, 3 hr, 86%) of the oily (-)-1,4-diol((-)-7),^{14a)} $[\alpha]_D^{20}$ -4 2°(c=5.31, CHCl₃), prepared from (-)-6 according to the reported procedure,¹⁵⁾ followed by acidic hydrolysis(conc HCl in THF-dioxane, reflux, 6 hr, 93%) of the formed (+)-acetal((+)-8),^{14b)} $[\alpha]_D^{20}$ +8 1°(c=6.17, CHCl₃), gave (+)-5^{14b,c)} as a colorless solid, mp 76-78°C, $[\alpha]_D^{20}$ +6 3°(c=3.08, CHCl₃) Levo-rotatory vicinal-diol((-)-5),^{14b,c)} mp 74-77°C, $[\alpha]_D^{20}$ -6 5°(c=3.01, CHCl₃), was similarly prepared from (+)-6 by way of (+)-7,^{14a,16)} $[\alpha]_D^{20}$ +4 5°(c=5.06, CHCl₃), and (-)-8,^{14b)} $[\alpha]_D^{20}$ -8 2°(c=6.14, CHCl₃)

Whereas unnatural (-)-6 is fairly expensive for a large scale preparation, (-)-7 could be also obtained from D-(-)-mannitol according to the reported procedure with a little modification.¹⁷⁾ Thus, triacetal formation(Me₂CO-conc H₂SO₄, 70%) of D-(-)-mannitol, followed by partial deacetalization of the triacetal(AcOH-H₂O, 80%), oxidative cleavage(NaIO₄ in MeOH-H₂O) of the two terminal 1,2-diols, and reduction of the dialdehyde(NaBH₄ in MeOH-H₂O, 2 steps, 69%), successfully gave (-)-7,^{14a)} $[\alpha]_D^{20}$ -4 4°(c=5.11, CHCl₃)(vide supra)

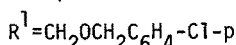
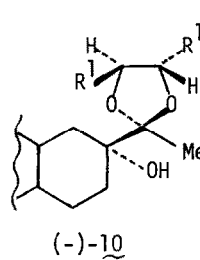
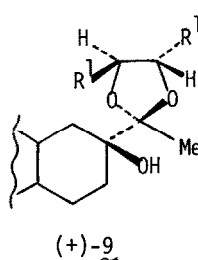
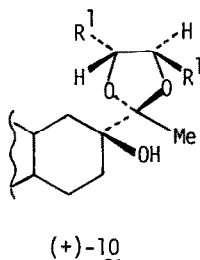
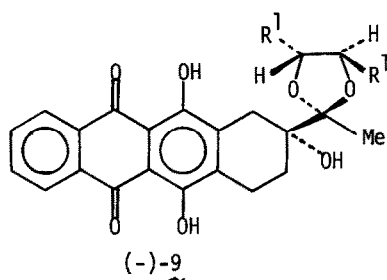
Acetalization(p-TsOH(0.06 eq) in C₆H₆, reflux, 13 hr, 100%) of (\pm)-3,^{5,8)} mp 214-216°C, with (+)-5(1.1 eq) gave an oily mixture of the diastereomeric acetals((-)-9 and (+)-10), $[\alpha]_D^{20}$ +4 8°(c=0.65, CHCl₃). This was triturated in ether at room temperature for 15 hr, giving crude (-)-9((-)-9 (+)-10 85.15¹⁸⁾)(47%), mp 132-136°C Evaporation of the mother liquor in vacuo afforded crude (+)-10((-)-9 (+)-10 20.80¹⁸⁾)(49%) as a red foam Two recrystallizations of crude (-)-9 from acetonitrile gave pure (-)-9^{14b,c)} (35%), mp 141-142°C, $[\alpha]_D^{20}$ -53 6°(c=0.50, CHCl₃). On the other hand, three recrystallizations of crude (+)-10 from ether afforded pure (+)-10^{14b,c)} (18%), mp 120-121°C, $[\alpha]_D^{20}$ +66 8°(c=0.51, CHCl₃)

The same acetalization of (\pm)-3 with (-)-5 as that described above, gave a crude mixture of (+)-9 and (-)-10(100%), $[\alpha]_D^{20}$ -5 4°(c=0.61, CHCl₃), from which crude (+)-9((+)-9 (-)-10 85.15¹⁸⁾)(45%), mp 133-138°C, and (-)-10((+)-9 (-)-10 17.83¹⁸⁾)(50%) were obtained by the trituration with ether Similar sequential recrystallizations of crude (+)-9 and (-)-10 yielded pure (+)-9^{14b,c)} (39%), mp 141-142°C, $[\alpha]_D^{20}$ +53 8°(c=0.55, CHCl₃), and pure (-)-10^{14b,c)} (21%), mp 120-121°C, $[\alpha]_D^{20}$ -66 4°(c=0.53, CHCl₃), respectively

Treatment of pure (-)-9 under the condition for transacetalization(BF₃ Et₂O(10 eq) in Me₂CO, reflux, 13 hr) or hydrolysis(conc HCl in dioxane-H₂O, reflux, 2 hr), followed by filtration through a short silica gel column(C₆H₆-CH₂Cl₂ 3:1), gave optically pure (R)(-)-3^{14a)} (98% or 97%), mp 218-219°C, $[\alpha]_D^{20}$ -89 9°(c=0.102, CHCl₃), or mp 218-219°C, $[\alpha]_D^{20}$ -90 3°(c=0.106, CHCl₃)(lit,^{4a)} mp 228-230°C, $[\alpha]_D^{20}$ -87°(c=0.1, CHCl₃), lit,⁵⁾ mp 218-220°C, $[\alpha]_D^{20}$ -87 0°(c=0.115, CHCl₃) Similar transacetalization or hydrolysis of crude (-)-9, mp 132-136°C (vide supra), gave partially optically active(R)(-)-3,^{14a)} mp 195-200°C, $[\alpha]_D^{20}$ -64 2°(c=0.108,



5. $R^1 = \text{CH}_2\text{OCH}_2\text{C}_6\text{H}_4\text{-Cl-p}$, $R^2 = R^3 = \text{H}$
 6. $R^1 = \text{COOH}$, $R^2 = R^3 = \text{H}$
 7. $R^1 = \text{CH}_2\text{OH}$, $R^2, R^3 = \text{CMe}_2$
 8. $R^1 = \text{CH}_2\text{OCH}_2\text{C}_6\text{H}_4\text{-Cl-p}$, $R^2, R^3 = \text{CMe}_2$



CHCl_3), 71% ee, or mp 190–197°C, $[\alpha]_D^{20} -62.5^\circ$ ($c=0.110$, CHCl_3), 69% ee. When pure and crude (+)-10 (vide supra) were similarly subjected to transacetalization, there could be obtained optically pure (S)(+)-3^{14a} (97%), mp 219–220°C, $[\alpha]_D^{20} +89.5^\circ$ ($c=0.102$, CHCl_3), and partially optically active (S)(+)-3^{14a} (90%), mp 190–196°C, $[\alpha]_D^{20} +55.5^\circ$ ($c=0.110$, CHCl_3), 61% ee. In completely the same manner, optically pure and partially optically active (S)(+)-3 (or (R)(-)-3) could be obtained from pure and crude (+)-9 (or pure and crude (-)-10), respectively. Recovery of the resolving agents ((+)- and (-)-5) might be simply accomplished since (+)-8 (70%), $[\alpha]_D^{20} +7.8^\circ$ ($c=0.97$, CHCl_3), and (+)-5 (75%), mp 76–78°C, $[\alpha]_D^{20} +5.7^\circ$ ($c=2.98$, CHCl_3), could be readily separated by the short silica gel column which was utilized for isolating (R)(-)-3 produced by transacetalization and hydrolysis (vide supra).

In order to improve efficacy of the exploited optical resolution, racemization of the undesired enantiomer ((S)(+)-3) was next examined. While attempted racemization of (S)(+)-3, 61% ee, under the same condition as that previously reported,⁶ was fruitless (only 30% racemization), the use of trifluoromethanesulfonic acid (70 eq to (S)(+)-3 in $\text{AcOH-H}_2\text{O}$ (7.4), 110°C, 20 hr) in place of p-toluenesulfonic acid was found to afford racemized (S)(+)-3^{14a} $[\alpha]_D^{20} +20.3^\circ$ ($c=0.118$, CHCl_3), 22% ee (62% racemization), from partially optically active (S)(+)-3, $[\alpha]_D^{20} +52.8^\circ$ ($c=0.108$, CHCl_3), 58% ee.

Since the preparation of optically pure (R)(-)-3 was completed, synthesis of (+)-2b was attempted according to the reported procedure.⁴ Successive bromination and substitution ($\text{Br}_2\text{-AIBN}$ in $\text{CHCl}_3\text{-CCl}_4\text{-H}_2\text{O}$) of the C₇-position of (R)(-)-acetal ((R)(-)-11),^{14b,c} mp 224–226°C, $[\alpha]_D^{20} -81.0^\circ$ ($c=0.120$, CHCl_3), prepared ($\text{HOCH}_2\text{CH}_2\text{OH-p-TsOH}$ in C_6H_6 , 98%) from (R)(-)-3, followed by hydrolysis and epimerization (80% CF_3COOH , rt, 15 hr), gave optically pure (+)-2b^{14a,19} (41% based on (R)(-)-11), mp 184–185.5°C, $[\alpha]_D^{20} +157^\circ$ ($c=0.114$, dioxane) (lit.,³) mp 185–187°C, $[\alpha]_D^{20} +165^\circ$ ($c=0.1$, dioxane), lit.,⁵) mp 183.5–184.5°C, $[\alpha]_D^{20} +153^\circ$ ($c=0.09$, dioxane).

Since (+)-2b had been transformed to (+)-2a,^{2,5} it became also possible to obtain optically pure (+)-2a from (R)(-)-3 produced by the optical resolution of (±)-3. Considering

operational simplicity, possible uses of both enantiomers of the resolving agents, and simple racemization of the undesired enantiomer((S)(+)-3), the exploited optical resolution of (\pm)-3 should have wide practical values in the synthesis of optically pure (+)-2a,b.

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References and Notes

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- 13) Prior to selecting (+)- and (-)-5 as acetalizing reagents, acetalization of (\pm)-3 was attempted using L-(+)-dimethyl tartrate, (S)(+)-1-phenyl-1,2-ethanediol, and (-)-1-0, 4-0-bis(benzyl)threitol. However, the acetal formation was not observed for L-(+)-dimethyl tartrate, and a mixture of the four diastereomeric acetals was obtained from (S)(+)-1-phenyl-1,2-ethanediol. Although (-)-1-0, 4-0-bis(benzyl)threitol successfully gave a mixture of the diastereomeric acetals in a quantitative yield, separation of the acetals attempted by column chromatography or by recrystallization turned out to be fruitless
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- 18) Determined by the optical purity of (R)(-)-3(or (S)(+)-3) derived from this sample
- 19) Separation of the C₇-epimer of (+)-2b was not attempted

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