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EPC Synthesis of 5-Substituted 2-Oxo-cyclopentanecarboxylates via Conjugate Addition of Cuprates to Asymmetric Shielded 2-Oxo-cyclopentenecarboxylates

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Abstract: Asymmetric shielded 2-oxo-cyclopentenecarboxylates 6n and 6x were prepared by transesterification of 2-oxo-cyclopentanecarboxylate 2 with camphor derived concave alcohols 1n and 1xand by subsequent introduction of a double bond via phenylselenides. Diastereoselective conjugate addition of equimolar amounts of mixed cuprates at -78 °C and deprotection by ethanolysis gave enantiomerically pure 5-substituted 2-oxo-cyclopentanecarboxylates 13-18 and *ent*-13-18, valuable as chiral building blocks in natural product synthesis.

Conjugate additions of cuprates to enoates have been widely employed for asymmetric carbon-carbonbond formations in syntheses of chiral building blocks.¹ Particularly successful was the conjugate addition of organocopper compounds to enoates of camphor derived chiral auxiliaries.¹⁻³ Best results were obtained on additions to enoates derived from concave alcohols **1n** and **1x**, which generally proceeded with an extremely high diastereoselectivity (>99%) and in excellent yields (>90%).³ Recently we extended these studies to cyclic enoates and demonstrated the usefulness of asymmetric protected 2-oxo-cyclohexenecarboxylates for the EPC synthesis of 6-substituted 2-oxo-cyclohexanecarboxylates.⁴ In conjunction with a project aiming at the EPC synthesis of (-)-chokol A⁵ we have investigated conjugate additions to asymmetric protected 2-oxocyclopentenecarboxylates derived from **1n** and **1x** and are now able to present an EPC synthesis of 5-substituted 2-oxo-cyclopentanecarboxylates.



Asymmetric shielded 2-oxo-cyclopentanecarboxylates were readily available from auxiliaries 1n or 1x and racemic 2-oxo-cyclopentanecarboxylate 2. We obtained the well crystallizing esters 3n (98%) and 3x (99%) in excellent yields using a DMAP⁶ mediated transesterification reaction⁷ first reported by Taber.⁸ Phenylselenylation⁹ of 3n and 3x gave mixtures of diastereometric selenides (4n:5n = 4x:5x = 80:20) which were separated by chromatography. Oxidative deselenylation of 4n or 5n and 4x or 5x afforded enoates 6n and 6x in good yields (81-90%), respectively.





Preparative scale synthesis of enoates 6n (75%) and 6x (78%) was manufactured by a one pot procedure¹⁰ starting from 3n and 3x without isolation of the intermediate selenides. In contrast to an anticipation¹¹ the enoates 6n and 6x were stable and could be stored without precautions.

Next we studied the addition of Lipshutz cuprates¹² to **6n** and **6x** (see Table 1) and obtained 5-substituted 2-oxo-cyclopentanecarboxylates **7n-12n** and **7x-12x** in good yields $(56-80\%)^{13,14}$ and with an extremely high level of diastereoselection. HPLC analysis of unpurified cuprate addition products outlined excellent selectivities (purity of **7x** > 99.5%; **12n** > 98%).¹⁵ Purification of the crude products by flash chromatography and crystallization yielded pure diastereomers according to HPLC and NMR.

Enone	R	Cuprate Precursor	Cuprate Adduct	Yield (%)	Ethyl Ester	Yield (%)	[α] ²⁰ _D
бn	nBu	R-Li	8n	80	14	88	+81.58
бn	tBu	R-Li	9n	59	15	93	+94.61
бn	H ₂ C=CH	R-MgBr	10n	68	16	82	+85.05
бn	$H_2C=C(CH_3)$	R-Li	11n	63	17	81	+69.88
бn	Ph	R-Li	12n	56	18	79	+18.55
6 x	CH ₃	R-Li	7 x	74	ent-13	75	-95.70
6 x	nBu	R-Li	8 x	69	ent-14	85	-81.26
6 x	tBu	R-Li	9 x	75	ent-15	90	-95.69
6 x	H ₂ C=CH	R-MgBr	10x	80	ent-16	82	-85.61
6 x	H ₂ C=C(CH ₃)	R-Li	11x	75	ent-17	79	-70.21
6 x	Ph	R-Li	12x	71	ent-18	76	-18.30

Table 1. Cuprate Addition to Asymmetric Shielded Enoates and Deprotection by Ethanolysis.



Scheme 2

Removal of the chiral auxiliary from the sterically highly crowded 2-oxo-cyclopentanecarboxylates 7n-12n and 7x-12x was accomplished using a Ti(OEt)₄ mediated transesterification reaction¹⁶ which allowed recycling of 1n and 1x (78-89%) and gave the enantiomerically pure ethyl esters 13-18 and *ent*-13-18 in good yields (Table 1). Comparison of the optical rotation of 13^{17} allowed the determination of absolute configuration at C-5', because 13 was previously prepared from (*R*)-pulegone.¹⁸

The steric course of cuprate additions to **6n** and **6x** can be rationalized by an attack of the organocopper nucleophile from the less hindered halfspace of the *s*-trans enoate reactive species (Scheme 2), as it was found for simpler acyclic enoates³ and 2-oxo-cyclohexenecarboxylates.⁴

In conclusion, the present conjugate addition approach to 5-substituted 2-oxo-cyclopentanecarboxylates enable a short and extremely stereoselective route to valuable intermediates in natural product synthesis.

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REFERENCES AND NOTES

- 1. For a recent review see: Rossiter, B. E.; Swingle, N. M. Chem. Rev. 1992, 92, 771-806.
- 2. Oppolzer, W.; Dudfield, P.; Stevenson, T.; Godel, T. Helv. Chim. Acta 1985, 68, 212-215.
- 3. Helmchen, G.; Wegner, G. Tetrahedron Lett. 1985, 26, 6051-6054.
- 4. Urban, E.; Riehs, G.; Knühl, G. Tetrahedron Lett. 1995, 36, 4773-4776.
- 5. Urban, E.; Knühl, G.; Helmchen, G. Sci. Pharm. 1994, 62, 195.
- 6. Steglich, W.; Höfle, G. Angew. Chem. Int. Ed. Engl. 1969, 8, 981.
- 7. Decicco, C. P.; Buckle, R. N. J. Org. Chem. 1992, 57, 1005-1008.
- 8. Taber, D. F.; Amedio, J. C.; Patel, Y. K. J. Org. Chem. 1985, 50, 3618-3619.
- 9. Liotta, D.; Barnum, C.; Puleo, R.; Zima, G.; Bayer, C.; Kezar, H. S. J. Org. Chem. 1981, 46, 2920-2923.
- 10. Typical procedure: A solution of PhSeBr (4.72 g, 20 mmol) in CH₂Cl₂ (100 ml) was cooled to 0 °C, pyridine (1.61 g, 20.4 mmol) and then a solution of 3n (10.47 g, 20 mmol) in CH₂Cl₂ were added. The mixture was stirred for 2 h at 20 °C, extracted with 1 M HCl and dried (Na₂SO₄). Then a solution of 3-chloroperbenzoic acid (3.97 g, 23 mmol) in CH₂Cl₂ (100 ml) was added. The mixture was stirred for 2 h at 20 °C, extracted NaHCO₃ and NaHSO₃ solution, the organic layer was dried (Na₂SO₄) and the solvent was removed at reduced pressure. Crystallization of the residue from ether gave 6n (8.14 g, 78%, mp 171 °C).¹⁹
- Nugent, W. A.; Hobbs, F. W.; Wustrow, D. J.; Kende, A. S. Org. Synth. 1988, 66, 52-59; Marx, J. N.; Cox, J. H.; Norman, L. R. J. Org. Chem. 1972, 37, 4489-4491.
- 12. Lipshutz, B. H.; Koerner, M.; Parker, D. A. Tetrahedron Lett. 1987, 28, 945-948.
- 13. Typical procedure: A stirred solution of lithium 2-thienylcyanocuprate¹² (8 ml, 0.25 M in THF, 2 mmol) was cooled to -78 °C and treated dropwise with a solution of methyllithium (1.3 ml, 1.55 M in ether/hexane, 2 mmol). The mixture was stirred at -78 °C for 1 h, then a solution of 6n (1.04 g, 2 mmol) in THF (15 ml) was added and stirring was continued at -78 °C for 2 h. Then a saturated NH₄Cl solution was added, the mixture was stirred at 20 °C for 1 h and extracted with CH₂Cl₂. The organic layer was dried (Na₂SO₄) and the solvent was evaporated at reduced pressure. The residue was purified by flash chromatography (silica gel, hexane/EtOAc = 9:1); recrystallization from 2-propanol yielded 7n (785 mg, 73%, mp 140 °C).¹⁹
- 14. ¹H and ¹³C NMR spectroscopic studies of the keto-enol equilibrium revealed that compounds with a bulky group showed mainly signals of the keto form, while compounds with a planar substituent provided both signals of the keto and the enol form (ratio strongly depending on concentration and choice of the solvent).
- 15. Diastereomers of 7x and 12n with opposite configuration at C-5' have been prepared for HPLC purposes by a sequence which will be published later in a full paper.
- Seebach, D.; Hungerbühler, E.; Naef, R.; Schnurrenberger, P.; Weidmann, B.; Züger, M. Synthesis 1982, 138-141.
- 17. The optical rotation of 13 ($[\alpha]_{D}^{25} = +68.01$ {neat, $d_{4}^{25} = 1.032$ }) agreed with a previous determination.¹⁸
- 18. Marx, J. N.; Norman, L. R. J. Org. Chem. 1975, 40, 1602-1606.
- 19. All new compounds gave satisfactory spectroscopic data and elementary analyses which will be published later in a full paper.

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