

Synthesis of Optically Active Selenium-Containing Isotryptophan, Homoisotryptophan, and Homotryptophan

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Selenoisotryptophan and its higher homologues were synthesized by Sonogashira coupling of iodophenyl methyl selenide and alkynyloxazolidines followed by iodocyclization as

phene with ethynyloxazolidine allowed the synthesis of selenohomotryptophan.

the key step. Sonogashira coupling of 3-iodobenzoseleno-

Introduction

Selenium-containing unnatural amino acids are well known to reduce phase problems in protein crystallography.^[1a,1b] In the last decade, selenium analogues of tryptophan (i.e., **1a** and **1b**; Figure 1) are found to be used in this direction.^[1c] Besides crystallography, amino acids bearing radioactive selenium are used for imaging purpose. In this area, different selenium-substituted aromatic or heterocyclic α -amino acids (e.g., **1d**, **1e**)^[2a] have been synthesized as model compounds. Radioactive Se⁷⁵ labeled tryptophan **1c** is in clinical trials as a pancreatic scanning agent.^[2b]



Figure 1. Important selenotryptophan analogues.

As part of our ongoing research toward the synthesis of tryptophan analogues, we decided to prepare new Se analogues in optically active form. Herein we report the synthesis of selenium-containing isotryptophan **5a**, homoisotryptophan **5b** and homotryptophan **9** by using a chiral pool approach.

Results and Discussion

Our retrosynthetic route is depicted in Scheme 1; suitably substituted alkynyloxazolidines were used as building

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blocks for the introduction of a chiral center to the amino acids. We anticipated that selenoisotryptophan **5a** and its higher homologue **5b** could possibly be synthesized by iodocyclization^[3] of intermediate **4**, which could be obtained by Sonogashira coupling^[4] of iodophenyl methyl selenide (**2**)^[3a,3b] with alkynyloxazolidines **3a** and **3b**^[5a-5d] (retrosynthesis A, Scheme 1). The synthesis of selenohomotryptophan **9** could be achieved from intermediate **8**, which in turn could be synthesized by Sonogashira coupling of 3-iodobenzoselenophene (**6**)^[3a] and ethynyloxazolidine **7**^[6] (retrosynthesis B, Scheme 1).



Scheme 1. Retrosynthetic pathway.

The synthesis of tryptophan analogues was previously reported by Rutjes et al.^[7] by using Sonogashira coupling as the key step; however, a similar synthesis of the Se analogues has not been attempted. Moreover, the starting acetylene-containing α -amino acids were prepared by a chemoenzymatic method, as these compounds are very expensive commercially. Recently, we developed a strategy to prepare indole/benzofuran-containing amino acids through the palladium-catalyzed coupling of alkynyloxazolid-

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ines.^[5d,5e] At the onset of the synthesis, we prepared iodophenyl methyl selenide $2^{[3a,3b]}$ from commercially available 2-iodoaniline. This was then coupled with acetylene 3a to form adduct 4a. Upon treatment with iodine in dichloromethane at room temperature, compound 4a gave desired 10a in 60% yield along with amino alcohol 11a as a result of the Lewis acidic behavior of molecular iodine. Deiodination^[8] of **10a** by using tributyltin hydride (TBTH) in the presence of 2,2'-azobisisobutyronitrile (AIBN) afforded 12a, which after acetonide removal under acidic conditions provided deiodinated amino alcohol 13a. Sequential oxidation^[9] and esterification afforded selenoisotryptophan 5a from 13a. Repetition of the same sequence with 11a led to the formation of 3-iodoselenoisotryptophan derivative 14a. Similarly, higher homologue 5b and iodo derivative 14b were prepared from butynyloxazolidine 3b and selenide 2 by

Pd(OAc)₂, PPh NBoo Cul, Et₃N, 65 °C SeCH₃ 6 h **4a**: *n* = 1, 58 % 3a: n = 1 **4b**: *n* = 2, 52 % 3b: n = 2I2, CH2CI2 r.t., 10 min Boch BocHN ÒН **10a**: n = 1, 60 % 11a: n = 1, 37 % 10b: n = 2, 38 % 11b: n = 2, 47 % (a) Dess-Martin TBTH, AIBN periodinane toluene, 90 °C CH₂Cl₂, r.t., 30 min 4 h (b) NaClO₂, NaH₂PO₄ tBuOH, H₂O r.t., 1 h (c) CH₂N₂, ether, 0 °C Boch 12a: n = 1, 61 % CO₂Me 12b: n = 2, 57 % BocHN **14a**: *n* = 1, 48 % 14b: n = 2, 46 % PTSA, MeOH r.t., 2 h (a) Dess-Martin periodinane. CH2Cl2, r.t., 30 min (b) NaClO₂, NaH₂PO₄ CO₂Me BocHN BocHN *t*BuOH, H₂O, r.t., 1 h ÒН 5a: n = 1, 46 % (c) CH₂N₂, ether, 0 °C 13a: n = 1 75 % **5b**: n = 2.48 %13b: n = 2, 80 %

Scheme 2. Synthesis of selenoisotryptophan and selenohomoisotryptophan derivatives (PTSA = p-toluenesulfonic acid). following the reaction sequence shown in Scheme 2. Chiral acetylenes 3a and 3b were synthesized from naturally occurring L-aspartic acid and L-glutamic acid according to a literature procedure.^[5a-5d]

Synthesis of selenohomotryptophan **9** commenced with the preparation of 3-iodobenzoselenophene $(6)^{[3a]}$ from iodophenyl methyl selenide (**2**). Compound **6** was then coupled with ethynyloxazolidine **7**^[6] under Sonogashira^[4] conditions to obtain coupling product **8**. Ethynyloxazolidine **7** was made from Garner's aldehyde,^[6] which in turn was prepared from naturally occurring L-serine. Application of standard hydrogenation^[10] conditions to reduce the triple bond of **8** resulted in a mixture of half-saturated product **17**, completely saturated product **18**, and starting material **8**, in which **17** was the major species (Scheme 3). Hydrogenation at high (>80 psi) pressure^[10] provided a parallel outcome (Scheme 3).



Scheme 3. Synthesis of selenohomotryptophan derivative from iodophenyl methyl selenide (TBAF = tetrabutylammonium fluoride, DIPA = diisopropylamine, PTSA = p-toluenesulfonic acid).

After purification, semisaturated compound **17** was subjected to hydrogenation under several conditions^[10] (Table 1, entries 1–6) but none of these provided **18** in more than 7% yield.

However, overnight hydrogenation in a Parr hydrogenator (>80 psi) afforded desired product **18** in moderate yield (Table 1, entry 7). Unreacted starting material **17** was recovered in every case (Scheme 3). At this juncture, acetonide deprotection of **18** resulted in amino alcohol **19**, which upon oxidation^[9] and esterification afforded selenohomotryptophan derivative **9** (Scheme 3).



Table 1. Conditions for the reduction of the double bond to a single bond. $^{[a]}$

	NBoo Se	catalyst, so	olvent Se BocN-		
	17 reductant		nt	18	
Entry	Catalyst	Amount	Reductant	Time [h]	Yield ^[b] [%]
1	10 % Pd/C	15 wt%	Et ₃ SiH ^[c]	2	5
2	10 % Pd/C	15 wt%	H_2 (1 atm)	12	7
3	NiCl ₂	50 mol-%	NaBH4 ^[d]	4	_[e]
4	Pd(OH) ₂	10 wt%	H_2 (1 atm)	12	6
5	RhCl(PPh ₃) ₃	10 mol-%	H_2 (1 atm)	12	6
6	PtO ₂	10 mol-%	H_2 (1 atm)	5	_[e]
7	10 % Pd/C	20 wt-%	H ₂ (>80 psi)	24	40

[a] Reagents and conditions: Reactions were performed with 17 (0.05 mmol) in methanol (1 mL, entries 1 to 5) and AcOH (1 mL, entry 6). Reaction was performed with 17 (0.26 mmol) in methanol (20 mL, entry 7). 1 psi = 6.89 kPa; 1 atm = 101.32 kPa. [b] Isolated yield. [c] Et₃SiH (10 equiv.). [d] NaBH₄ (1.5 equiv.). [e] No reaction.

Conclusions

In summary, the synthesis of optically active new selenotryptophan analogues was reported from alkynyloxazolidines by using a chiral pool approach. Though the yield in the reduction step was moderate, the synthesis of selenohomotryptophan is reported for the first time. During their synthesis, we prepared 3-iodosusbtituted analogues that can be further derivatized by standard coupling reactions to afford several other analogues.

Supporting Information (see footnote on the first page of this article): Experimental procedures and characterization data, ¹H NMR spectra of all new compounds and ¹³C NMR spectra of compounds **3a**, **3b**, **4a**, **4b**, **5a**, **5b**, **8**, **9**, **12a**, **12b**, **13a**, **13b**, **14a**, **14b**, **17**, **18**, and **19**.

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- a) W. A. Hendrickson, A. Pahler, J. L. Smith, Y. Satow, E. A. Merritt, R. P. Phizackerley, *Proc. Natl. Acad. Sci. USA* 1989, 86, 2190–2194; b) N. Budisa, C. Minks, S. Alefelder, W. Wenger, F. Dong, L. Moroder, R. Huber, *FASEB J.* 1999, 13, 41–51; c) J. O. Boles, J. Henderson, D. Hatch, L. A. P. Silks, *Biochem. Biophys. Res. Commun.* 2002, 298, 257–261.
- [2] a) T. Sadeh, M. A. Davis, R. Gil, U. Zoller, J. Heterocycl. Chem. 1981, 18, 1605–1607; b) J. E. Agnew, M. Maze, British J. Radiology 1978, 51, 206–209.
- [3] a) T. Kesharwani, S. A. Worlikar, R. C. Larock, J. Org. Chem. 2006, 71, 2307–2312; b) S. Mehta, J. P. Waldo, R. C. Larock, J. Org. Chem. 2009, 74, 1141–1147.
- [4] a) K. Sonogashira, Y. Tohda, N. Hagihara, *Tetrahedron Lett.* 1975, 16, 4467–4470; b) R. Chinchilla, C. Nájera, *Chem. Rev.* 2007, 107, 874–922.
- [5] a) A. Dondoni, P. P. Giovannini, A. Massi, Org. Lett. 2004, 6, 2929–2932; b) A. Dondoni, P. P. Giovannini, D. Perrone, J. Org. Chem. 2005, 70, 5508–5518; c) G. Mahler, G. Serra, E. Manta, Synth. Commun. 2005, 35, 1481–1492; d) K. Goswami, S. Paul, S. T. Bugde, S. Sinha, Tetrahedron 2012, 68, 280–286; e) K. Goswami, I. Duttagupta, S. Sinha, J. Org. Chem. 2012, 77, 7081–7085.
- [6] G. T. Crisp, Y. L. Jiang, P. J. Pullman, C. D. Savi, *Tetrahedron* 1997, 53, 17489–17500, and references cited therein.
- [7] B. C. J. van Esseveldt, F. L. van Delft, J. M. M. Smits, R. de Gelder, H. E. Schoemaker, F. P. J. T. Rutjes, *Adv. Synth. Catal.* 2004, 346, 823–834.
- [8] A. Dobbs, J. Org. Chem. 2001, 66, 638-641.
- [9] a) D. B. Dess, J. C. Martin, J. Org. Chem. 1983, 48, 4155–4156;
 b) B. S. Bal, W. E. Childers Jr., H. W. Pinnick, Tetrahedron 1981, 37, 2091–2096; c) K. J. Hale, S. Manaviazar, J. H. George, M. A. Walters, S. M. Dalby, Org. Lett. 2009, 11, 733–736.
- [10] a) P. K. Mandal, J. S. McMurray, J. Org. Chem. 2007, 72, 6599–6601; b) J. F. Sauvage, R. H. Baker, A. S. Hussey, J. Am. Chem. Soc. 1960, 82, 6090–6095; c) A. Jourdant, E. Gonzalez-Zamora, J. Zhu, J. Org. Chem. 2002, 67, 3163–3164. Received: March 8, 2013

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