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Synthesis of Oligonucleotides Labelled with a Novel Type of Chemically Stable Acridine Dye

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Abstract: Alkoxy derivatives of 6-chloro-2-methoxy-9-methylacridine 1 have been attached to oligonucleotides 6 - 10. The coupling step may be performed either with phosphoramidite-, *H*-phosphonate or phosphodiester reagents prepared from 2. This dye, in contrast to 9-aminoacridines, is completely stable under the conditions of oligonucleotide deprotection (NH₃ or CH₃NH₂). The dye conjugates tightly bind to complementary strands and are useful tools in the study of nonenzymatic oligomerizations of ribonucleotides according to L. Orgel.

In our continuing search for the catalysis of transphosphorylation reactions ¹) we are interested in the template directed but nonenzymatic oligomerization of ribonucleotides ²). In order to monitor the proceeding of chain elongation we planned to apply short oligonucleotide primers labelled with a fluorescent dye. Especially, intercalating dyes should be advantageous in stabilizing the complexes between primers and templates ^{3,4}). Derivatives of 9-aminoacridines are well established for this purpose ^{3c-e)}. However, their C9-N-bond is prone to hydrolysis which complicates preparation and use of such oligonucleotide conjugates. Here we present a novel type of chemically stable acridine dye ⁴) attached to oligonucleotides via side chains of variable length.

Starting from 6-chloro-2-methoxy-9-methylacridine 1^{5} the methyl ether was cleaved by use of hydrobromic acid. The crude phenole was then alkylated with the THP protected side chain building blocks ⁶) (91 - 98% overall yield). After acidolysis the acridine alcohols **2a** and **b** were obtained as readily crystallizing yellow compounds (93 - 96%).

Scheme 1: Attachment of side chains to 6-chloro-2-methoxy-9-methylacridine



Reagents and conditions: **a**, HBr, glacial acetic acid, reflux, 7 h; **b**, 1.5 eq. Br-(CH₂)_n-OTHP, 5 eq. Cs₂CO₃, DMF, 110 °C, 10 h, 91 - 98%; **c**, dil. HCl, MeOH, reflux, 6 h, 93 - 96%.

2a and **b** could be transformed to all major types of phosphorylated derivatives used in oligonucleotide synthesis. Treatment of **2a** with bis(diisopropylamino)cyanoethoxyphosphine gave phosphoramidite **3a** in 69% yield ⁷). The *H*-phosphonates **4a** and **b** were obtained by reaction of **2a** and **b** with phosphorus trichloride, 1H-1,2,4-triazole and *N*-methylmorpholine in 65% yield ⁸). The phosphodiester **5a** was pre-

pared from 2a and 2-chlorophenyldichlorophosphate for application in the phosphotriester technique ⁹). Alternatively conjugation of 2a with oligonucleotides was possible in one step according to van Boom's in situ phosphorylation/coupling method ¹⁰).

Scheme 2: Preparation of phosphorylated acridines 3 - 5¹⁶



Reagents and conditions: **a**, 2.2 eq. bis(diisopropylamino)cyanoethoxyphosphine, 1.1 eq. diisopropylammonium-tetrazolide, pyridine, rt, 5 h, 69%; **b**, 40 eq. 1*H*-1,2,4-triazole, 50 eq. *N*-methylmorpholine, 5 eq. PCl₃, THF, pyridine, rt, 1.5 h, 65%; **c**, 1) 2 eq. 2-chlorophenyldichlorophosphate, pyridine, -10 °C, 10 min, 2) H₂O, 65%.

In order to obtain large amounts of labelled oligonucleotides (20 - 100 mg) we used both solution chemistry and manual solid phase preparation with highly charged supports ¹¹). Our acridine dye proved to be stable under the conditions of oligonucleotide synthesis, especially during base deprotection with ammonia or methylamine ¹²). Results are summarized in table 1.

Table 1: Preparation of Acr-O-(CH₂)_n-O-PO₂-O-(oligonucleotides) 6 - 10¹⁵

acridine building block	nucleotide moiety	product	method
H-phosphonate 4a (n = 6)	dG-dC-dA-dC-G dG-dG-G dA-dA-dA-dA	6 7 8	a)H-phosphonate, solid phase
phosphoramidite 3a (n = 6)	dG-dG-G	7	b) phosphoramidite, solid phase
acridine alcohol 2b (n = 7)	dA-dA-A	9	c) phosphotriester, solution (v. Boom)
phosphodiester 5a (n = 6)	dT-dT-U	10	d) phosphotriester, solution

Reagents and conditions: **a**, 1) 5 eq. **4a**, 15 eq. pivaloyl chloride, oligonucleotide attached to Tentagel[®] (ca. 0.2 mmol/g), acetonitrile/pyridine 1 : 1, rt, 2 min: 2) 0.1 M I₂ in pyridine/*N*-methylmorpholine/H₂O/THF 5 : 1 : 5 : 90 followed by 0.1 M I₂ in NEt₃/H₂O/THF 5 : 5 : 90; 3) methylamine/ammonia 1 : 1, rt, 90 min or ammonia, 55 °C, 20 h; **b**, 1) 30 eq. **3a**, 60 eq. tetrazole, oligonucleotide attached to CPG-support (1 µmol), acetonitrile, rt, 2 h; 2) 0.1 M I₂ in THF/H₂O/lutidine 2 : 2 : 1; 3) methylamine/ammonia 1 : 1, rt, 4 h; **c**, 1) 1.1 eq. 2-chlorophenyldichlorophosphate, 2.2 eq. CF₃HOBt ^{10c}), 2.2 eq. pyridine; 2) 1 eq. $N_{6,2}$ ',3'-benzoyl-protected dAdArA; 3) 4 eq. **2b**, THF, rt, 6h; 4) 25 eq. pyridine-2-carbaldoxime, 25 eq. tetramethylguanidine, THF, rt, 23 h; 5) ammonia, 55 °C, 72 h; **d**, 1) 1 eq. 2',3'-isopropylidene-protected dTdTU, 1.2 eq. **5a**, 11 eq. MSNT, pyridine, rt, 6 h; 2) 25 eq. pyridine-2-carbaldoxime, 25 eq. tetramethylguanidine, THF, rt, 12 h; 3) 80% acetic acid, reflux, 40 min. The interactions of **6** - **10** with complementary oligonucleotides were investigated by CD- and UVmeasurements. A strong induced Cotton effect in the visible band of the dye was observed in the CD spectrum of **9** when mixed with poly-U¹³ (figure 1a). Together with a pronounced hypochromicity at 354 nm this observation proves a strong interaction between the dye and the bases which can be explained most likely by intercalation. No effect was visible either with **9** alone or in the presence of the noncomplementary poly-A. To quantify the stability of the aggregate, the melting point was determined to 42°C as shown in figure 1b¹⁴. However, the exact nature of these aggregates (duplex versus triplex) is still unknown. An unequivocal case of duplex formation (T_M = 49°C) was observed with the labelled oligonucleotide **6** Acr-d(GCAC)rG and d(CCCCCGTGCG), also proven by UV- and CD-measurements.



Template assisted elongation experiments of **6** and **9** according to Orgel ^{2a)} were carried out with guanosine-5'-phosphoro(2-methylimidazolide) and the adenosine derivative, respectively. The reactions could be successfully monitored by HPLC, due to the colour, fluorescence and lipophilic properties of the acridine moiety. So, the new intercalating agent is well suited for our purposes. Furthermore some applications in antisense strategy may be envisaged.

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- CD: 0.18 mM 9, 2 mM poly-U or poly-A (calculated as triplets), H₂O, 10 mM Tris-buffer pH 8.3, 1 M NaCl; the effects changed only slightly between pH 5.6 and pH 10.3 and disappeared at pH 10.9; UV: 0.1 mM 9, 0.5 mM poly-U, H₂O, 100 mM Tris-buffer pH 5.2, 1 M NaCl; the same value of T_M was observed at pH 7.0.
- 15. Satisfactory ¹H NMR and ESI MS data have been obtained.
- 16. Selected physical data: **2a**: mp. 162°C (ethyl acetate); UV/VIS: (ethanol); λ_{max} (ε) = 263 (142500), 319 (2670), 335 (5000), 352 (7360), 382 (6500), 401 nm (6190); **2b**: mp. 154 155°C (ethyl acetate); UV/VIS: like spectrum of 2a; ¹H NMR: (270 MHz, [D₆]DMSO): δ = 1.33 1.53 (m, 8 H, methylene-H), 1.82 (m, 2 H, methylene-H), 3.02 (s, 3 H, CH₃), 3.39 (m, 2 H, methylene-H), 4.18 (t, *J* = 6.5 Hz, 2 H, methylene-H), 4.34 (t, *J* = 5.1 Hz, 1 H, OH), 7.44 (d, *J* = 2.6 Hz, 1 H, 1-H), 7.52 (dd, *J* = 9.4 and 2.5 Hz, 1 H, 3-H), 7.56 (dd, *J* = 9.2 and 2.1 Hz, 1 H, 7-H), 7.99 (d, *J* = 9.4 Hz, 1 H, 4-H), 8.09 (d, *J* = 2.1 Hz, 1 H, 5-H), 8.36 (d, *J* = 9.4 Hz, 1 H, 8-H); **3a**: ³¹P NMR: (162 MHz, C₆D₆): δ = 148.5 (m); **4a** / **4b**: ³¹P NMR: (162 MHz, [D₆]-DMSO): δ = 2.55 (dt, *J*_d = 589 Hz, *J*_t = 8.1 Hz); **5a**: ³¹P NMR: (162 MHz, [D₆]DMSO): δ = -5.50 (t, *J* = 7.1 Hz).

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