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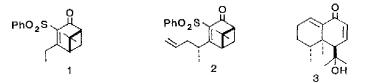
A HIGHLY STEREOCONTROLLED SYNTHESIS OF (-)-KANSHONE A

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Abstract: Described herein is the first and stereoselective synthesis of (-)-kanshone A (3), a nardosinan sesquiterpere, starting with enore 2.

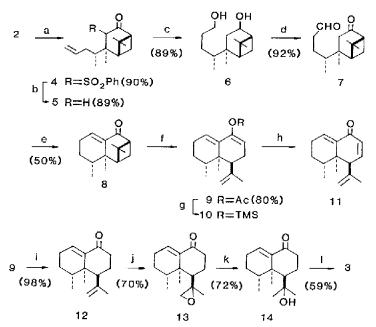
In the preceding paper,¹ we reported that allylation of enone 1, derived from (+)-nopinone, with allyl bromide proceeded in an extracyclic stereocontrolled fashion to give enone 2 in synthetically satisfied yield. In order to study use of 2 in natural product synthesis, (-)-kanshone A (3), a nardosinan sesquiterpene isolated from <u>Nardostachys chinensis</u>,² was chosen as the first target molecule. We wish to describe here the highly stereocontrolled and first total synthesis of 3.



Treatment of 2 with Me₂CuLi led to stereoselective Michael addition to give the adduct 4, which on desulfurization provided 4,4-disubstituted nopinone 5 (Scheme 1). Hydroboration-oxidation of 5 followed by Swern oxidation of the resulting diol 6 afforded keto-aldehyde 7, and cyclization of the latter with KOBu^t gave the tricyclic enone 8 in fair yield.

 $BF_3 \cdot OEt_2$ -promoted cyclobutane cleavage³ in 8 gave high yield of enol acetate 9 which possesses the nardosinan carbon skeleton with correct stereochemistry. Conversion of 9 to enol silyl ether 10 followed by palladium catalyzed dehydrosilylation⁴ provided cross-conjugated dicnone 11 whose physical data including the sign of the specific rotation were identical with those of authentic 11 derived from natural $3.^2$

Regioselective epoxidation of dienone 12 prepared from hydrolysis of 9 was performed by treatment with mCPBA (1 equiv) to give epoxide 13, whose lithium aluminum hydride reduction in refluxing THF followed by PDC oxidation gave dihydrokanshone A (14). Finally, transformation of 14 to 3 was accomplished by phenylselenenylation and subsequent selenoxide elimination. TR and ¹H NMR data of the synthetic 3 (crystals, mp 96-98 °C), $[\alpha]^{20}_{D}$ -245.1°(<u>c</u> 0.39, CHCl₃)⁵ were in good accordance with those of natural 3 (oil), $[\alpha]_D = 147.8^\circ$ (c = 0.35, CHCl₃)⁵, in all respects.^{2,6}



a) Me₂CuLi, THF; b) Na(Eg), MeOE; c) B_2E_6 , THF, then 30% H_2O_2 , aq NaOH; d) DMSO, (COCI), CH₂Cl₂, then Et₂N; e) <u>t</u>-BuOK, CE₂Cl₂, rt, 20 min; f) BF; OEt, Zn(OAc), Ac,O; g) Mell, TMSCl, HMPA, Et,O; h) Pd(OAc), MeCN; i) K₂CO₃, MCOE; j) mCPBA, CH₂Cl₂; k) (i) LiAlH₂, THF, reflux, (ii) PDC, CH₂Cl₂; 1) (i) (TMS)₂NLi, PhSeCl, HMPA, THF, (ii) 30% H₂O₂, Py, THF,

Scheme 1

Acknowledgment: We are grateful to Dr. Y. Oshima for providing us $^{1}\mathrm{H}$ NMR and IR spectra of (-)-kanshone A.

References and Notes

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- Kato, M.; Watanabe, M.; Awen, B. Z.; Vogler, B.; Yoshikoshi, A. <u>Tetrahedron Lett.</u> 1991, <u>Bagchi, A.; Oshima, Y.; Hikino, H. Phytochemistry, 1988, 27. 1199.</u> Kato, M.; Kamat, V. P.; Tooyama, Y.; Yoshikoshi, A. <u>J. Org. Chem.</u> 1989, 54, 1536. Ito, Y.; Hirano, T.; Saegusa, T. <u>J. Org. Chem.</u> 1978, 43, 1011. The difference of specific rotation between natural and synthetic 3 arises from impurities contaminated in natural 3 as a result of difficulty of purification (private communication). Attempts to synthesize 3 by reductive cleavage of the oxirane in epoxy dienone i derived from regioselective epoxidation of 11 failed because the oxirane was unreactive under the conditions shown below. 6

