

Pergamon

0040-4039(95)01379-2

HETEROCYCLIC KETONES AS PRE-FORMED BUILDING BLOCKS. SYNTHESIS OF (-)-SLAFRAMINE.

Peter Szeto, * David C. Lathbury^b and Timothy Gallagher** * School of Chemistry, University of Bristol, Bristol BS8 1TS U.K. * SmithKline Beecham Pharmaceuticals, Leigh, Tonbridge TN11 9AN U.K.

Summary. A synthesis of lactam 2, which constitutes a formal total synthesis of (-)-slaframine 1, is described with key steps being (a) intramolecular aldol reaction of ketoaldehyde 4 to establish the indolizidine framework and (b) diastereoselective reduction of enone 5 using the Corey oxazaborolidine 7.

(-)-Slaframine 1 is a fungal metabolite produced by *Rhizoctonia leguminicola* that is known to undergo oxidative activation *in vivo* to produce a potent and neurotoxic muscarinic agent.¹ Since its isolation in 1965,² both the biosynthetic origins and metabolism of slaframine have been investigated³ and, especially in recent years, this indolizidine alkaloid has been the subject of extensive synthetic studies.⁴ Our interest in this area focused on use of this molecule as a vehicle for the development of new methodology for the construction and elaboration of functionalised indolizidines and in this paper we describe an enantiospecific synthesis of lactam 2^{4m} which constitutes a formal total synthesis of (-)-slaframine 1.



The indolizidine nucleus of slaframine was assembled using the intramolecular aldol sequence shown in **Scheme 1**. Protection of (S)-2-amino-4-pentenoic acid followed by N-acylation of 3, 3-dimethoxypyrrolidine gave amide 3 (82 % overall). Oxidative alkene cleavage followed by ketal hydrolysis then gave ketoaldehyde 4 (63 % from 3) and the key intramolecular aldol reaction of 4 was carried out in the presence of piperidine (in THF at r.t.) to give the indolizidine enone 5 in 31 % yield. This aldol cyclisation process, which requires enolisation at C(2) of the pyrrolidine ring (a selectivity that is generally disfavoured⁵), has only been achieved using a secondary amine - a variety of other, more conventional aldol reaction conditions failed. The major side reaction observed in the conversion of 4 to 5 involved competitive enolisation of the aldehyde component and elimination of CbzNH₂.⁶



Scheme 1. Reagents: i, (S)-N-(Cbz)-2-amino-4-pentenoic acid, EDC, CH_2Cl_2 ; ii, OsO₄ (cat.), NaIO₄, aq. THF; iii, 2M HCl, aq. THF; iv, piperidine, THF, r.t. 24 h., then H_3O^+ , 1 h.

With the indolizidine framework available, methods were examined for introduction of the two remaining stereocentres (at C(1) and C(8a)) required for slaftarmine with the intention that the amino residue at C(6) could provide a diastereofacial directing role. Initial reduction of the C=C bond of 5 was unsuccessful because of (a) the level of selectivity observed and (b) complications encountered in the subsequent ketone manipulation. 1, 2-Reduction of the enone moiety of 5 gave allylic alcohols 6a/b but with poor selectivity (0-30 %d.e.) for 6a, regardless of the nature of the hydride source used (Scheme 2). A possible explanation for this observation may be found within the structure of 5. With five adjacent sp²-centres (at C(1), N(4), C(5), C(8), and C(8a)) the indolizidine ring system is severely flattened. In addition, the C(6)-N-Cbz group most likely adopts a quasi-equatorial position thereby minimising its directing ability. Given this, indolizidine 5 is perhaps better viewed as a "prochiral" substrate and we have sought to establish the stereochemistry of the C(1)-OH of 6a via reagent rather than substrate control.

Of a series of asymmetric reducing agents examined, the Corey oxazaborolidine^{7,8} was the most effective. Reduction of 5 using a stoichiometric amount of 7 (from (R)-proline) gave the desired (1S)-allylic alcohol 6a (60 %; \geq 95 %d.e. as judged by ¹H NMR). A similar level of selectivity was observed for the formation of the (1R)-isomer 6b using *ent*-7 and the effective lack of any significant matched/mismatched relationship between reagent and substrate served to reinforce our notion of the nominally "prochiral" nature of indolizidine 5.⁹



Completion of the synthetic sequence was straightforward and was carried out as shown in Scheme 2. Silylation of 6a followed by hydrogenation and re-protection gave lactam 2,¹⁰ in 86 % overall from 6a and the synthesis of lactam 2 and its conversion (in 3 steps) to (-)-slaframine 1 has been reported by Hua.^{4m} A problem associated with the hydrogenation step must be highlighted: long reaction times led to formation (up to 15 %) of the C(6) epimer of 2 - the precursor to 6-epislaframine.



Scheme 2. Reagents: i, e.g. NaBH₄, LiAlH₄, *i*-Bu₂AlH, LS-Selectride, ii, 7, CH₂Cl₂, -20 °C; iii, TBDMSCl, imidazole, DMF; iv, H₂, 10 % Pd/C, EtOH then Boc₂O, CH₂Cl₂.

In summary, the synthesis of lactarn 2, which constitutes a formal total synthesis of (-)-slaframine 1, has been achieved. This chemistry illustrates a new approach to indolizidine synthesis based on an intramolecular aldol reaction, together with the application of the Corey oxazaborolidine towards solving a problem of diastereoselectivity in the 1,2-reduction of an enone.

Acknowledgements. We thank Dr. David J. Mathre (Merck Research Laboratories, Department of Process Research, Rahway, NJ) for a generous gift of both 7 and *ent*-7, Professor Duy Hua for copies of spectra for lactam 2 and its C(6) epimer, Dr. M.S. Hadley (SB Pharmaceuticals) for his support and SmithKline Beecham Pharmaceuticals and the University of Bristol for financial support.

References.

- 1. Guengerich, F. P.; Broquist, H. P. In *Bioorganic Chemistry*; Van Tamelen, E. E. Ed.; Academic: New York, 1979; Vol. 2, p 97.
- (a) Rainey, D. P.; Smalley, E.B.; Crump, M.H.; Strong, F.M. Nature (London) 1965, 205, 203.
 (b) Aust, S.D.; Broquist, H.P. Nature (London), 1965, 205, 204. (c) Aust, S.D.; Broquist, H.P.;

Rinehart, K.L. J. Am. Chem. S oc. 1966, 88, 2879. (d) Whitlock, B. J.; Rainey, D. P.; Riggs, N. V.;
Strong, F. M. Tetrahedron Lett. 1966, 3819. (e) Gardiner, R. A.; Rinehart, K. L.; Snyder, J. J.;
Broquist, H. P. J. Am. Chem. Soc. 1968, 90, 5639.

- 3. Harris, C. M.; Schneider, M. J.; Ungemach, F. S.; Hill, J.E.; Harris, T. M. J. Am. Chem. Soc. 1988, 110, 940.
- (a) Cartwright, D.; Gardiner, R. A.; Rinehart, K. L. J. Am. Chem. Soc. 1970, 92, 7615. (b) Gensler, W. J.; Hu, M. W. J. Org. Chem. 1973, 38, 3848. (c) Gobao, R. A.; Bremmer, M. L.; Weinreb, S.M. J. Am. Chem. Soc. 1982, 104, 7065. (d) Schneider, M. J; Harris, T. M. J. Org. Chem. 1984, 49, 3681.
 (e) Dartmann, M.; Flitsch, W.; Krebs, B.; Pandl, K.; Westfechtel, A. Liebigs Ann. Chem. 1988, 695.
 (f) Shono, T.; Matsumura, Y.; Katoh, S.; Takeuchi, K.; Sasaki, K.; Kamada, T.; Shimizu, R. J. Am. Chem. Soc. 1990, 112, 2368. (g) Pearson, W. H.; Bergmeier, S. C. J. Org. Chem. 1991, 56, 1976.
 (h) Wasserman, H. H.; Chi, B. V. Tetrahedron Lett. 1994, 35, 9779. For recent syntheses of (-)-slaframine see: (i) Choi, J. -R.; Han, S.; Cha, J. K. Tetrahedron Lett. 1991, 32, 6469. (j) Pearson, W. H.; Bergmeier, S. C.; Williams, J. P. J. Org. Chem. 1992, 57, 3977. (k) Sibi, M. P.; Christensen, J. W.; Li, B.; Renhowe, P. A. J. Org. Chem. 1992, 57, 4329. (l) Knapp, S.; Gibson, F. S. J. Org. Chem. 1992, 57, 4802. (m) Hua, D. H.; Park, J.-G.; Katsuhira, T.; Bharathi, S. N. J. Org. Chem. 1993, 58, 2144. (n) Knight, D. W.; Sibley, A. W. Tetrahedron Lett. 1993, 34, 6607. For the synthesis of 8a-epi and 1,8a-diepislaframine see Gmeiner, P.; Junge, D. Kärtner, A. J. Org. Chem., 1994, 59, 6766.
- For enolisation studies of pyrrolidin-3-ones, see Garst, M. E.; Bonfiglio, J. N.; Grudoski, D. A.; Marks, J. J. Org. Chem. 1980, 45, 2307. For a solution to the problem of C(2) alkylation of pyrrolidin-3-ones, see Giles, M.; Hadley, M. S.; Gallagher, T. J. Chem. Soc., Chem. Commun. 1990, 1047.
- The aldol cyclisation protocol has been successfully applied to a range of other systems but the principle limitation remains competitive enolization (Stockman, R.; Szeto, P; Hadley, M. S.; Lathbury, D. C.; Gallagher, T. unpublished results).
- (a) Corey, E. J.; Bakshi, R. K.; Shibata, S. J. Am. Chem. Soc. 1987, 109, 5551.
 (b) Singh, V. K. Synthesis 1992, 605.
- Mathre, D. J.; Thompson, A. S.; Douglas, A. W.; Hoogsteen, K.; Carroll, J. D.; Corley, E. G.; Grabowski, E. J. J. J. Org. Chem. 1993, 58, 2880 and references therein.
- There are few examples of the asymmetric ketone reduction of prochiral enones of this general type. (a) Iwasaki, G.; Sano, M.; Sodeoka, M.; Yoshida, K.; Shibasaki, M. J. Org. Chem. 1988, 53, 4864. (b) Fronza, G.; Fogliato, G.; Fuganti, C.; Lanati, S.; Rallo, R.; Servi, S. Tetrahedron Lett. 1995, 36, 123.
- 10. Lactam 2 had $[\alpha]_{D}^{22} = +53$ (c 1.0, CHCl₃) {lit., ^{4m} $[\alpha]_{D}^{22} = +53$ (c 1.09, CHCl₃)} and ¹H and ¹³C NMR spectra obtained for 2 matched those provided by Professor Hua. All new compounds gave satisfactory spectral data and were characterised by elemental analysis and/or high resolution mass measurement.

(Received in UK 22 June 1995; revised 17 July 1995; accepted 21 July 1995)