

STRUCTURE AND SYNTHESIS OF ATALAPHYLLINE AND RELATED ALKALOIDS

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Abstract—The structure of atalaphylline, (1; 1,3,5-trihydroxy-4-(γ,γ -dimethylallyl)acridone an alkaloid of *Atalantia monophylla*, was confirmed by oxidative degradation and by total synthesis. The synthesis involved the preparation of 1,3,5-trihydroxy-9-acridone followed by direct prenylation to give atalaphylline and a monoprenylated product.

INTRODUCTION

Atalantia monophylla Correâ (Rutaceae) is a shrub with brown bark and thorny branches growing wild in the mountainous regions of India, Sri Lanka and Burma. From the petrol extract of its root bark, several acridone alkaloids [1-7] have been isolated and characterized. The structures of atalaphylline and *N*-methylatalaphylline were reported by Govindachari *et al.* [1]. In this report we discuss the oxidative degradation of atalaphylline and its synthesis.

RESULTS AND DISCUSSION

Compound 1 was methylated with methyl iodide and anhydrous potassium carbonate in dry acetone. The methylated product (2) was oxidized with alkaline potassium permanganate to an acid (3) which was converted to its methyl ester (4). Compound 4, $C_{21}H_{21}NO_8$, showed the characteristics of an ester in its IR spectrum 1730 (>C=O of ester), 1650 (>C=O of acridone ring) and 1290 cm^{-1} (C-O-Me). The NMR spectrum indicated the presence of two carbomethoxy groups at δ 3.70 (6H) and three methoxy groups at δ 3.92 (3H) and 4.00 (6H). The NMe protons appeared at δ 3.84 (3H).

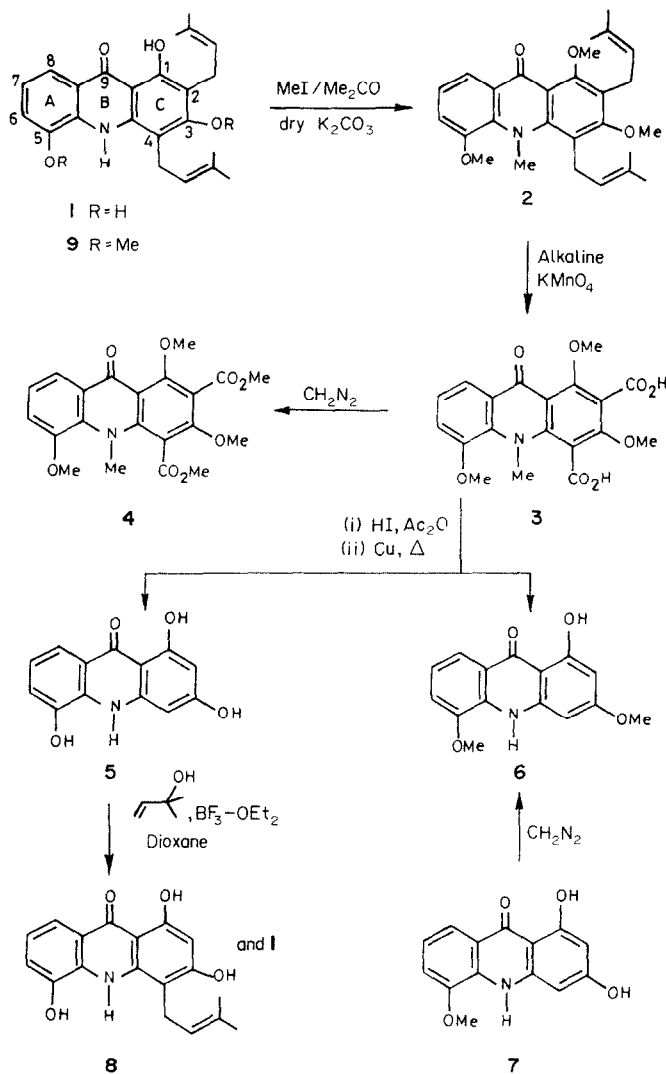
The acid (3) on demethylation followed by decarboxylation gave a mixture which on chromatography gave two pure products (5) and (6).

Compound 5, mp 319-320° (d), $C_{13}H_9NO_4$ (M^+ 243), gave a positive ferric chloride test indicating the presence of a phenolic hydroxyl group. The UV spectrum showed the characteristic absorptions of 9-acridones. The IR spectrum showed bands assignable to nonbonded and bonded hydroxyl (3500, 3350 cm^{-1}), >NH (3275 cm^{-1}) and carbonyl of the acridone (1640 cm^{-1}). The NMR spectrum indicated the presence of two hydroxyl groups at δ 14.39

(strongly deshielded OH-1 by O-9) and 11.11 (OH-5 in the vicinity of acridone >NH), δ 8.28 (NH). These signals were exchangeable with D_2O . The aromatic region of the spectrum (integrating for 4H) showed the presence of an ABX system [8]: C-8 proton at δ 7.66 (*dd*, $J = 7.5$ and 2.5 Hz), C-6 and C-7 at δ 7.10 (*m*). The other proton in this region which was exchangeable with D_2O was assigned to a hydroxyl group probably at C-3. The signals at δ 6.73 (1H, *s*) and 6.05 (1H, *s*) were expected to be in ring C only. As these are singlets, they cannot be *ortho* to each other. These two signals were absent in the NMR spectrum of atalaphylline (1) and in 4 and were generated after decarboxylation of the acid (3).

Compound 5 was synthesized for the first time in an excellent yield by condensing 3-hydroxyanthranilic acid with phloroglucinol in butanol-benzene (2:1) in the presence of anhydrous zinc chloride under reflux. Hence, the NMR signals at δ 6.73 and 6.05 could be assigned to C-4 and C-2 protons. This also proved that the two prenyl groups present in atalaphylline could be at C-4 and C-2. This synthesis also supports the placement of the third hydroxyl group at C-3.

Compound 6 was found to be identical with the product obtained on diazomethane treatment of compound 5, mp 215-218°, $C_{15}H_{13}NO_4$ (M^+ 271), showing IR bands at 1250 and 1065 cm^{-1} (=C-O-C) and NMR signals at δ 4.09 (3H, *s*) and 3.93 (3H, *s*) for two methoxyl groups. This compound also responded to the ferric chloride test indicating the presence of a phenolic hydroxyl group suggesting the methylation of OH-3 and OH-5 groups. The hydroxyl group at C-1 could not be methylated under such conditions because of strong hydrogen bonding with the C-9 carbonyl group. Compound 6 was found to be identical with a sample obtained by diazomethane treatment of 1,3-dihydroxy-5-methoxyacridone [9] (7, M^+ 257).



Thus, from the structure of the degradation products, it could be concluded that the hydroxyl groups in atalaphylline are at positions C-1, C-3 and C-5 and the prenyl groups at C-2 and C-4.

Atalaphylline was synthesized from trihydroxy-acridone (5) by treatment [10] with 2-methyl-3-butene-2-ol in dioxane in the presence of BF_3 -etherate to give 1, mp 242–245°, $\text{C}_{23}\text{H}_{25}\text{NO}_4$ (M^+ 379) and 8. Compound 1 was found to be identical with atalaphylline in all respects.

Compound 8, mp 248–249° (d), $\text{C}_{18}\text{H}_{17}\text{NO}_4$ (M^+ 311) had the characteristic UV and IR absorptions of a hydroxy acridone. The NMR spectrum showed the presence of a γ,γ -dimethylallyl group, δ 1.76 (3H, s, signal for the other methyl of the gem-dimethyl overlapping with the solvent impurity of D_3CCOCD_3); benzylic methylene at δ 3.62 (2H, d, $J = 7$ Hz) and vinylic at δ 5.20 (1H, t, $J = 7$ Hz). It also had signals for a chelated hydroxyl at δ 14.08, a phenolic hydroxyl at δ 9.60, >NH at δ 9.00, and another hydroxyl in the region δ 7.00–7.30 in addition to the expected ABX system for C-6, C-7 and C-8 protons.

Furthermore, the spectrum showed a signal at δ 6.24 (1H, s) which could be that of the C-2 proton if the prenyl group is placed at C-4 or vice-versa. From the study of NMR spectra of several synthetic and natural acridone derivatives [7], it was found that the C-2 proton resonated between δ 6.00 and 6.20 whereas the C-4 proton resonated between δ 6.40 and 6.70. Therefore, the signal at δ 6.24 could be due to the C-2 proton. Hence 8 could be named as 1,3,5-trihydroxy-4-(γ,γ -dimethylallyl)acridin-9-one.

Atalaphylline-3,5-dimethyl ether (9), which was isolated [6] from *A. manophylla*, was prepared from atalaphylline by treatment with diazomethane.

EXPERIMENTAL

Mps are uncorr. NMR spectra were recorded at 100 MHz using TMS as int. standard.

N-Methyl-tri-O-methylatalaphylline (2). A soln of atalaphylline (1) (100 mg) in dry Me_2CO was refluxed with MeI (10 ml) and dry K_2CO_3 (5 g) for 96 hr (TLC monitoring). The reaction mixture, after work-up, was chromatographed on Si gel. Elution with petrol (bp 60–80°)–EtOAc (9:1) yielded 2 (27 mg) as a pale yellow gum.

1,3,5 - Trimethoxy - 10 - methylacridone - 2,4 - dicarboxylic acid (3). A soln of *N* - methyl - tri - *O* - methylatalaphylline (2) (2 g) in Me₂CO (200 ml) was refluxed with a mixture of NaOH (5 g) and KMnO₄ (excess) for 48 hr. The mixture was cooled, filtered and the solvent removed under vacuum. The gum so obtained was extracted with CHCl₃ to remove unreacted material. It was then acidified with HCl, extracted with CHCl₃ and the solvent removed to obtain crude 3 (750 mg).

1,3,5 - Trimethoxy - 10 - methylacridone - 2,4 - dimethylcarboxylate (4). An Et₂O soln of 1,3,5 - trimethoxy - 10 - methylacridone - 2,4 - dicarboxylic acid (3, 750 mg) was treated with an excess of CH₂N₂-Et₂O. After usual work-up, a product was obtained which was chromatographed on Si gel, and elution with C₆H₆-EtOAc (3:2) gave 4 (100 mg) as a gum. IR $\nu_{\text{max}}^{\text{Nujol}}$ cm⁻¹: 1730, 1650, 1600, 1590, 1290, 1170 and 1050. (Found: C, 60.48; H, 5.04. C₂₁H₂₁NO₈ requires: C, 60.72; H, 5.06%.) NMR (CDCl₃): δ 7.30 (3H, s), 4.00 (6H, s), 3.92 (3H, s), 3.84 (3H, s) and 3.70 (6H, s).

1,3,5 - Trihydroxyacridone (5). 1,3,5 - Trimethoxy - 10 - methyl - 2,4 - dicarboxylic acid (3, 500 mg) in HI (10 ml) and Ac₂O (10 ml) was boiled for 30 min. Usual work-up afforded a crude product (200 mg) which gave a positive FeCl₃ test. This product was then heated with a Cu catalyst (50 mg) at 250° for 30 min. Et₂O extraction of the resultant tar gave a yellow gum which was chromatographed on Si gel. Fractions collected with C₆H₆-EtOAc (1:1) afforded 5 (80 mg), mp 319–320° (d). UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm: 205, 228, 255, 269, 278, 299, 315 and 380 (log ϵ 4.09, 3.92, 4.47, 4.17, 4.16, 3.94, 3.81 and 3.46). IR $\nu_{\text{max}}^{\text{Nujol}}$ cm⁻¹: 3500, 3350, 3275, 1640, 1600, 1365, 1150, 1100 and 800. (Found: C, 64.43; H, 3.95. C₁₃H₉NO₄ requires: C, 64.20; H, 3.70%.) NMR (DMSO-*d*₆): δ 14.39 (1H, s), 11.11 (1H, s), 8.28 (1H, s), 7.66 (1H, *dd*, *J* = 7.5 and 2.5 Hz), 7.00–7.20 (3H, *m*), 6.73 (1H, s) and 6.05 (1H, s). MS *m/z* (rel. int.): 243 [M]⁺ (100), 242 (13), 214 (21), 176 (36), 163 (46), 156 (16), 154 (15), 102 (45) and 77 (25).

1-Hydroxy-3,5-dimethoxyacridone (6). Fractions collected with petrol-EtOAc (17:3) from the above column afforded crystalline 6, mp 215–218° as the major component. UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm: 204, 229, 256, 268, 277, 298, 312 and 388 (log ϵ 3.91, 3.44, 4.01, 3.79, 3.81, 3.46, 3.33 and 3.09). IR $\nu_{\text{max}}^{\text{Nujol}}$ cm⁻¹: 3370, 1665, 1610, 1575, 1250, 1150 and 800. (Found: C, 66.30; H, 4.70. C₁₅H₁₃NO₄ requires: C, 66.41; H, 4.79%.) NMR (CDCl₃): δ 8.26 (1H, s), 7.94 (1H, *dd*, *J* = 7.5 and 2.5 Hz), 7.14–7.30 (2H, *m*), 6.30 (1H, s), 6.26 (1H, s), 4.09 (3H, s) and 3.93 (3H, s). MS *m/z* (rel. int.): 271 [M]⁺ (100), 256 (28), 242 (7), 228 (3), 213 (5), 200 (4), 185 (8), 157 (2) and 135 (3).

1,3,5-Trihydroxyacridone (5) was prepared by condensing phloroglucinol with 3-hydroxyanthranilic acid according to the method described in ref. [11].

3-Hydroxyanthranilic acid was prepared from 3-amino-benzoic acid following the procedure of ref. [12].

1,3 - Dihydroxy - 5 - methoxyacridone (7) was prepared from 3-methoxyanthranilic acid and phloroglucinol according to the method of ref. [9].

Atalaphylline (1). To a stirred soln of 1,3,5-trihydroxyacridone (5, 500 mg) in dioxane (100 ml) was gradually added 2 - methyl - 3 - butene - 2 - ol (1.25 ml) in dioxane (50 ml) and then BF₃-etherate (10 drops). The reaction mixture was stirred for 90 hr. It was then diluted with H₂O (200 ml), extracted with EtOAc, dried and the solvent removed to give a gum. The gum was chromatographed on Si gel (40 g) and 50 ml fractions were collected; the separation was monitored by TLC. Fractions collected with C₆H₆ afforded atalaphylline 1 (30 mg), mp 242–245° (hexane-EtOAc).

(Found: C, 72.82; H, 6.50. C₂₃H₂₅NO₄ requires: C, 72.82; H, 6.59%.)

Chromatography of the mixture gave 1 and 1,3,5 - trihydroxy - 4 - (γ,γ - dimethylallyl)acridone (8) in pure crystalline form. TLC of the mixture showed the presence of some other minor products which could not be obtained in a pure state.

1,3,5 - Trihydroxy - 4 - (γ,γ - dimethylallyl)acridone (8). Fractions collected with petrol-EtOAc (9:1) from the above column afforded crystalline 8 (30 mg), mp 248–249° (d). UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm: 206, 230, 254, 280, 304, 330 and 396 (log ϵ 4.17, 4.09, 4.59, 4.33, 4.13, 3.72 and 3.71). IR $\nu_{\text{max}}^{\text{Nujol}}$ cm⁻¹: 3350, 1640, 1615, 1500 and 800. (Found: C, 69.34; H, 5.35. C₁₈H₁₇NO₄ requires: C, 69.45; H, 5.46%.) NMR (D₃CCOCD₃): δ 14.08 (1H, s) 9.60 (1H, unresolved), 9.00 (1H, *br* s), 7.77 (1H, *dd*, *J* = 7.5 and 2.5 Hz), 7.00–7.30 (3H, *m*), 6.24 (1H, s), 5.20 (1H, *t*, *J* = 7 Hz), 3.62 (2H, *d*, *J* = 7 Hz) and 1.76 (3H, s). MS *m/z* (rel. int.): 311 [M]⁺ (73), 296 (16), 294 (15), 268 (19), 256 (51), 255 (40), 243 (100), 227 (10), 190 (6), 170 (6), 147 (8), 121 (9) and 107 (11).

Atalaphylline - 3,5 - dimethyl ether (9). A soln of 1 (25 mg) in MeOH (5 ml) was treated with excess CH₂N₂-Et₂O in the cold and left overnight at room temp. Evaporation of the solvent followed by crystallization from petrol-Et₂O gave 9 (20 mg) as yellow needles, mp 145–147°. UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm: 215, 262, 280, 310, 323 and 405 (log ϵ 4.10, 4.58, 4.42, 3.94, 3.99 and 3.43). IR $\nu_{\text{max}}^{\text{Nujol}}$ cm⁻¹: 3350, 1640, 1605, 1590, 1570, 1375, 1325, 1270, 1245, 1225, 1195, 1170, 1120, 1080, 1030, 975, 955, 895, 850 and 725. (Found: C, 73.90; H, 7.46. C₂₅H₂₅NO₄ requires: C, 73.71; H, 7.13%.) NMR (CDCl₃): δ 14.20 (1H, s), 8.78 (1H, *br* s), 7.75 (1H, *dd*, *J* = 7.5 and 3 Hz), 7.20–6.80 (2H, *m*), 5.14 (2H, *br*), 3.93 (3H, s), 3.80 (3H, s), 3.40 (4H, *br*), 1.92 (3H, s), 1.76 (6H, s) and 1.64 (3H, s). MS *m/z* (rel. int.): 407 [M]⁺ (78), 392 (25), 364 (84), 352 (100), 336 (51), 308 (62), 296 (29) and 282 (70).

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