

Stereoselective Synthesis of 2,3-Dihydropyrroles from Terminal Alkynes, Azides, and $\alpha.\beta$ -Unsaturated Aldehydes via N-Sulfonyl-1,2,3triazoles

Tomova Miura.* Takamasa Tanaka, Kentaro Hiraga, Scott G. Stewart.† and Masahiro Murakami*

Department of Synthetic Chemistry and Biological Chemistry, Kyoto University, Katsura, Kyoto 615-8510, Japan

Supporting Information

ABSTRACT: A stereoselective method for synthesis of trans-2,3-disubstituted 2,3-dihydropyrroles is reported. N-Sulfonyl-1,2,3-triazoles prepared from terminal alkynes generate α -imino rhodium carbene complexes, which when combined with α,β -unsaturated aldehydes produce trans-2,3-disubstituted dihydropyrroles. The method can be successfully applied to a one-pot process starting from terminal alkynes.

he 2,3-dihydropyrrole ring system is a valuable structural motif found in a number of biologically active compounds. 1 In addition, 2,3-dihydropyrroles have been widely employed as important intermediates in the synthesis of natural products² and other complex molecules.³ Thus, the development of efficient methods for their synthesis from readily accessible starting materials is highly desired.⁴ Now, we report a sequential procedure for the diastereoselective synthesis of trans-2,3disubstituted 2,3-dihydropyrroles from terminal alkynes, Nsulfonyl azides, and α,β -unsaturated aldehydes (Figure 1).

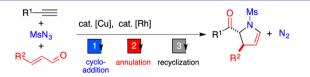


Figure 1. Construction of 2,3-dihydropyrroles starting from terminal alkynes, N-sulfonyl azides, and $\alpha_1\beta$ -unsaturated aldehydes.

N-Sulfonyl-1,2,3-triazoles, which can be easily prepared from terminal alkynes by copper-catalyzed 1,3-dipolar cycloaddition with N-sulfonyl azides, have recently received much attention as precursors of α -imino metal carbenes.⁶ The generated metal carbene species are electrophilic in nature to accept various nucleophiles, including those having acidic protons like water⁷ and allylic alcohols. In our previous report, the addition of allylic alcohols was immediately followed by a Claisen rearrangement to afford α -allyl- α -amino-ketones. On the other hand, the nitrogen atom of the α -imino group exhibits a higher nucleophilic character than the related α -oxo metal carbenes. This nucleophilicity, when combined with the electrophilic character of the carbene carbon, enables the compound to incorporate unsaturated compounds such as nitriles, alkynes, allenes, isocyanates, furans, and indoles to produce the corresponding N-heterocycles. For example, the reaction with aldehydes leads

Table 1. Denitrogenative Reaction of Triazole 1a with (E)-Crotonaldehyde (4a): Screening of Rhodium(II) Catalysts^a

		yield (%) ^b	
entry	$Rh_2(L)_4$	5aa	6aa
1	$Rh_2(OCOC_7H_{15})_4$	32	34
2	$Rh_2(OCO^tBu)_4$	49	18
3	$Rh_2(OCO-1-Ad)_4$	67	19
4	$Rh_2[(S)-DOSP]_4$	46	48
5	$Rh_2[(S)-PTPA]_4$	40	16
6 $\operatorname{Rh}_{2}[(S)\operatorname{-NTTL}]_{4}$		74 (80)	3

^aConditions: **1a** (0.20 mmol), **4a** (0.22 mmol), and 4 Å MS (40 mg) were heated in toluene (1 mL) at 120 °C for 1 h in the presence of $Rh_2(L)_4$ (2.0 μ mol). ^{b1}H NMR yield using CHBr $_2$ CHBr $_2$ as an internal standard, with isolated yield in parentheses.

to the production of 4-oxazolines, as recently reported by Fokin et al. ¹⁰ As a continuation of our previous studies on the utilization of N-sulfonyl-1,2,3-triazoles for synthetic purposes, we next examined the use of α,β -unsaturated aldehydes as the reaction partner. Thus, we initially prepared 1-methanesulfonyl-4-phenyl-1,2,3-triazole (1a) from phenylacetylene (2a) and methanesulfonyl azide (3a) according to the procedure using copper(I) thiophene-2-carboxylate (CuTC). Triazole 1a (0.20 mmol) was reacted with (E)-crotonaldehyde (4a, 0.22 mmol) in the presence of Rh₂(OCOC₇H₁₅)₄ (1.0 mol %) and 4 Å molecular sieves (MS) in refluxing toluene (Table 1, entry 1). Triazole 1a was completely consumed in 1 h, giving a mixture of 2,3dihydropyrrole (5aa, 32% NMR yield) and 4,5-dihydro-1,4oxazepine (6aa, 34% NMR yield). The relative stereochemistry at the 2,3-positions of 5aa was unambiguously assigned as trans by a single-crystal X-ray analysis. Next, a variety of ligands on rhodium(II) (Figure 2) were examined in terms of product selectivity, which showed a significant dependence on the ligands (entries 2-6). In particular, when the sterically bulky chiral ligand (S)-NTTL¹¹ was employed, formation of **6aa** was suppressed (3% NMR yield) and trans-5aa was obtained in 80% isolated yield as a racemic mixture. Production of the fivemembered-ring compound 5aa is in sharp contrast to the

Received: July 13, 2013

Figure 2. Chiral Rh(II) catalysts examined in the optimization studies.

reaction of rhodium α -oxo-carbene intermediates with α , β -unsaturated aldehydes, which furnished alkenyl-substituted epoxides without participation of the α -oxo moieties.¹²

A pair of (*E*)- and (*Z*)-isomers of cinnamaldehyde (**4b**) was then used to compare the stereochemical outcomes. The two isomers were independently subjected to the reaction conditions ($Rh_2[(S)-NTTL]_4$, 120 °C, 1 h), and both isomers furnished exclusively the *trans*-2,3-disubstituted dihydropyrrole **5ab** in high yield as a racemic mixture (eq 1).

Ph (1.1 equiv)
$$\frac{\text{Rh}_{2}[(S)-\text{NTTL}]_{4}}{\text{toluene, MS 4Å}}$$
 $\frac{\text{Ph}}{\text{Sab}}$ (1) $\frac{\text{Rh}_{2}[(S)-\text{NTTL}]_{4}}{\text{toluene, MS 4Å}}$ $\frac{\text{Ph}}{\text{Sab}}$ (1) $\frac{\text{E:}Z}{\text{Sab}}$ $\frac{\text{E:}Z}{\text{Sab}}$ $\frac{\text{Sab}}{\text{Sab}}$ \frac

When the reaction of triazole 1a with (E)- and (Z)-4b was carried out for 1 h at room temperature in chloroform, the corresponding (E)- and (Z)-2-styryl-4-oxazolines 7ab, which retained their original double-bond geometries, were isolated respectively. When each isolated 7ab was independently heated in refluxing toluene for 1 h, both isomers exclusively afforded the *trans*-isomer of 2,3-dihydropyrrole 5ab (eq 2).

1a + (E)-4b
$$\xrightarrow{\text{Rh}_2[(S)\text{-NTTL}]_4}$$
 (1.0 mol %) $\xrightarrow{\text{CHCl}_3}$ MS 4Å $\xrightarrow{\text{rt}, 1 \text{ h}}$ Ph $\xrightarrow{\text{Ph}}$ 120 °C, 1 h $\xrightarrow{\text{74}\%}$ (2) (trans)

1a + (Z)-4b $\xrightarrow{\text{CHCl}_3}$ MS 4Å $\xrightarrow{\text{rt}, 1 \text{ h}}$ Ph $\xrightarrow{\text{Ph}}$ 120 °C, 1 h $\xrightarrow{\text{81}\%}$ 81%

The rearrangement reaction of the 4-oxazolines 7ab was monitored at 50 °C by 1H NMR. When starting from (E)-7ab, only the remaining (E)-7ab and the *trans*-product 5ab were detected throughout the reaction course. On the other hand, when starting from (Z)-7ab, the (E)-7ab was detected in addition to the (Z)-7ab and the *trans*-product 5ab. These results suggested that the (Z)-7ab isomerized to (E)-7ab prior to the rearrangement process.

On the basis of these results and the previous study on the formation of 4-oxazolines from aldehydes, ¹⁰ we propose a mechanism for the formation of products **5** and **6** from triazole **1a** and the α , β -unsaturated aldehyde **4** as depicted in Scheme 1. Initially, a reversible ring—chain tautomerization of the triazole **1a** generates α -diazo imine **1a**′, ¹⁴ which reacts with rhodium(II) in an irreversible manner to afford an α -imino rhodium carbene **A** with extrusion of molecular nitrogen. Nucleophilic addition of

Scheme 1. Proposed Mechanism for the Formation of Products 5 and 6 from Triazole 1a and α,β -Unsaturated Aldehyde 4

the α,β -unsaturated aldehyde 4 to the electrophilic carbene center of A occurs to furnish the zwitterionic intermediate B. The anionic rhodium releases an electron pair, which induces the imino nitrogen to attack on either the α - or γ -carbon of the oxonium ion. Attack on the γ -carbon forms 4,5-dihydro-1,4oxazepine 6 (path a), whereas attack on the α -carbon forms the 4-oxazoline intermediate 7 (path b). The C–O bond of the N,Oaminal moiety of 7 is selectively cleaved, probably due to the higher electronegativity of oxygen and the higher electrondonating ability of nitrogen, 15 giving the zwitterionic intermediate C. Double bond isomerization occurs readily with the delocalized conjugate cation moiety of C, which converges into a form of the more stable (*E*)-isomer. Finally, the enolate moiety of (E)-C intramolecularly couples with the delocalized conjugate cation moiety through conformation E, which experiences less gauche interactions along the axis of the developing carboncarbon bond, to give the trans-configured 2,3-dihydropyrrole 5. Although a [3,3] sigmatropic rearrangement presents an alternative option for the mechanistic pathway, the ionic mechanism mentioned above is favored on the basis of the non-stereospecificity between the isolated enantiomerically enriched 4-oxazoline intermediate and the product trans-5aa, 13 and the production of the racemic dihydropyrroles in the other cases (vide infra).

The scope of α , β -unsaturated aldehydes 4 was examined using Rh₂[(S)-NTTL]₄ as the catalyst (Table 2). (E)- β -Monosubstituted substrates 4c-g, possessing a wide variety of alkyl and aryl groups, reacted cleanly with triazole 1a to afford products 5ac-ag in yields ranging from 71% to 90% (entries 1–5). In addition, the mono(dimethyl acetal) of fumaraldehyde (4h) and fumaraldehydic acid methyl ester (4i) successfully participated in the annulation reaction (entries 6 and 7). As in the case of (E)-and (Z)-cinnamaldehyde (4b) (eq 1), both (E)- and (Z)-isomers of hex-2-enal (4c) selectively gave the *trans* isomer of the product 5ac (entries 1 and 8). The reaction of (Z)-3-bromoacrylaldehyde (4j) was followed by the subsequent E1cB process, resulting in the formation of 2-benzoylpyrrole 8 (entry 9). Acyclic α , β -disubstituted substrates 4k-n were also effectively converted

Table 2. Rh(II)-Catalyzed Denitrogenative Annulation of Triazole 1a with Various α , β -Unsaturated Aldehydes 4c-p⁴

				•	-
	entry	α,β-unsaturated aldeh	yde 4	product 5	yield (%)
		R ² O		O Ms Ph	
	1	$R^2 = {}^nPr$	4c	5ac	75
	2	$R^2 = Cy$	4d	5ad	78
	3	$R^2 = {}^tBu$	4e	5ae	71
	4	$R^2 = 4\text{-MeO-C}_6H_4$	4f	5af	90
	5	$R^2 = 4-NO_2-C_6H_4$	4g	5ag	85
	6	$R^2 = CH(OMe)_2$	4h	5ah	73
	7	$R^2 = CO_2Me$	4i	5ai	85
		R^2		O Ms N R ²	
	8	$R^2 = {}^nPr$	4c ^c	5ac	74
	9	$R^2 = Br$	$\mathbf{4j}^d$	Ph N	65 ^e
		R ³ O		8 0 Ms N R ² R ³	
	10	$R^2 = Me$, $R^3 = Me$	4k	5ak	70^e
	11	$R^2 = Ph, R^3 = Me$	41	5al	88
	12	$R^2 = OEt$, $R^3 = Me$	4m	5am	91
	13	$R^2 = Ph, R^3 = Br$	4n	5an	56
		40		O Ms N N	
	14			5ao	35
		0 4n		O Ms N Ph	
	15	4 p		5ap	73^e
7.		. (2.22	/	1) 1 1	

"Conditions: 1a (0.20 mmol), 4 (0.22 mmol), and 4 Å MS (40 mg) were heated in toluene (1 mL) at 120 °C for 1 h in the presence of Rh₂[(S)-NTTL]₄ (2.0 μ mol). ^bIsolated yield (average of two runs). ^cE:Z = 9:91. ^dE:Z = 5:>95. ^eUsing 4 (0.40 mmol).

into the products **5ak—an** (entries 10–13). 1-Cyclohexene-1-carbaldehyde (**4o**) gave the bicyclic compound **5ao** in only 35% yield due to low product selectivity **5/6** (entry 14). Acrolein (**4p**) was also a suitable substrate, furnishing the product **5ap** in 73% yield (entry 15). The products **5ad**, **5ae**, **5af**, **5ag**, and **5ap** were analyzed by chiral HPLC and determined to be racemic.

Variation of triazoles 1 was also examined in the reaction with (*E*)-cinnamaldehyde (4b) (Table 3). Triazoles 1b–d, possessing aryl groups at the 4-position, afforded the corresponding products 5bb–db in yields ranging from 73% to 91% (entries 1–3). The reaction of the alkyl-substituted triazole 1e gave the product 5eb in moderate yield, due to 1,2-hydride migration occurring with the rhodium carbene intermediate to form *N*-mesyl-pent-2-en-1-imine (entry 4).¹⁷ The annulation reaction was amenable with respect to the R⁴ substituent on the sulfonyl group to give the products 5fb–ib in high yields (entries 5–8).

Table 3. Rh(II)-Catalyzed Denitrogenative Annulation of Various Triazoles 1b—i with (E)-Cinnamaldehyde (4b)^a

	1	riazole 1			
entry	R ¹	R ⁴		product 5	yield $(\%)^b$
1	4-MeO-C ₆ H ₄	Me	1b	5bb	91
2	$4-CF_3-C_6H_4$	Me	1c	5cb	87
3	3-thienyl	Me	1d	5db	73
4	"Pr	Me	1e	5eb	48 ^c
5	Ph	$(CH_2)_2TMS$	1f	5fb	89
6	Ph	4-Tol	1g	5gb	83
7	Ph	4-MeO-C ₆ H ₄	1h	5hb	88
8	Ph	4 -Br- C_6H_4	1i	5ib	77

"Conditions: 1a (0.20 mmol), 4b (0.22 mmol), and 4 Å MS (40 mg) were heated in toluene (1 mL) at 120 °C for 1 h in the presence of $Rh_2[(S)\text{-NTTL}]_4$ (2.0 μ mol). "Isolated yield (average of two runs). "Using 4b (0.60 mmol), $Rh_2[(S)\text{-NTTL}]_4$ (5.0 μ mol), and 4 Å MS (10 mg) in toluene (0.2 mL) for 2 h.

Table 4. One-Pot Synthesis of 2,3-Dihydropyrroles 5 Starting from Phenylacetylene $(2a)^a$

	azide	3	α, $β$ -unsaturated aldehyde 4				
entry	R ⁴		R ²	\mathbb{R}^3		product 5	yield $(\%)^b$
1	Me	3a	Me	Н	4a	5aa	68
2	Me	3a	Ph	Н	4b	5ab	70
3	4-Tol	3g	Ph	Н	4b	5gb	74
4	Me	3a	Ph	Me	41	5al	69
5	Me	3a	Me	Н	4a	5aa	68 ^c

^aConditions: **2a** (0.20 mmol), 3 (0.20 mmol), 4 (0.22 mmol), CuTC (20 μ mol), Rh₂[(S)-NTTL]₄ (2.0 μ mol), and 4 Å MS (40 mg) in toluene (1 mL) were stirred at rt for 6 h, then heated at 120 °C for 1 h. ^bIsolated yield (average of two runs). ^cOn a 10 mmol scale using **4a** (15 mmol).

With the detailed study on the transformation of the triazoles 1 into the 2,3-dihydropyrroles 5 finalized, we next carried out a one-pot synthesis of compounds 5 from terminal alkyne 2 in order to demonstrate the practical convenience of the present transformation (Table 4). Phenylacetylene (2a, 0.20 mmol), N-sulfonyl azides 3 (0.20 mmol), α , β -unsaturated aldehydes 4 (0.22 mmol), CuTC (10 mol %), Rh₂[(S)-NTTL]₄ (1.0 mol %), 4 Å MS, and toluene (1 mL) were placed in a reaction vessel, and the reaction mixture was simply stirred at room temperature. Both 2a and 3 were consumed after 6 h. The reaction mixture was subsequently stirred at 120 °C for an additional 1 h. After chromatographic separation, the compounds 5 were isolated in overall yields ranging from 68% to 74% (entries 1–4). An experiment using 1.0 g of 2a (10 mmol) also gave a comparable result (entry 5).

The synthetic utility of the dihydropyrrole products was exemplified by further transformations. Deprotective aromatiza-

Scheme 2. Synthetic Derivatization of 2,3-Dihydropyrroles

tion took place on treatment with 1,8-diazabicycloundec-7-ene (DBU) through an E1cB/prototropy sequence (Scheme 2). Tetrahydropyrrole (2,3-disubstituted pyrrolidine) 10 was obtained when 5aa was hydrogenated using Crabtree's catalyst. Furthermore, 2,3,5-trisubstituted pyrrolidine 12 was diastereoselectively synthesized through a sequence of hydration under acidic conditions, Horner—Wadsworth—Emmons olefination, and an aza-Michael reaction. 18

In summary, we have disclosed an interesting and useful reactivity of α -imino rhodium carbenes toward α,β -unsaturated aldehydes, providing an efficient method for the diastereoselective synthesis of *trans*-2,3-disubstituted 2,3-dihydropyrroles from terminal alkynes.

■ ASSOCIATED CONTENT

Supporting Information

Experimental procedures, spectral data for the new compounds, and details of the X-ray analysis. This material is available free of charge via the Internet at http://pubs.acs.org.

AUTHOR INFORMATION

Corresponding Authors

tmiura@sbchem.kyoto-u.ac.jp murakami@sbchem.kyoto-u.ac.jp

Notes

The authors declare no competing financial interest.

[†]S.G.S. is on leave from School of Chemistry and Biochemistry, The University of Western Australia

ACKNOWLEDGMENTS

We thank Dr. Yuuya Nagata (Kyoto University) for helping with X-ray analysis. This work was supported by MEXT (Grant-in-Aid for Scientific Research on Innovative Areas Nos. 22105005 and 24106718, Scientific Research (B) No. 23350041, Young Scientists (A) No. 23685019), JST (ACT-C), and Takeda Science Foundation. S.G.S. is grateful for the JSPS Invitation Fellowship for Research in Japan.

REFERENCES

(1) (a) Marti, C.; Carreira, E. M. J. Am. Chem. Soc. 2005, 127, 11505. (b) Magedov, I. V.; Luchetti, G.; Evdokimov, N. M.; Manpadi, M.; Steelant, W. F. A.; Van Slambrouck, S.; Tongwa, P.; Antipin, M. Y.; Kornienko, A. Bioorg. Med. Chem. Lett. 2008, 18, 1392. (c) Li, W.; Khullar, A.; Chou, S.; Sacramo, A.; Gerratana, B. Appl. Environ. Microbiol. 2009, 75, 2869.

(2) (a) Humphrey, J. M.; Liao, Y.; Ali, A.; Rein, T.; Wong, Y.-L.; Chen, H.-J.; Courtney, A. K.; Martin, S. F. J. Am. Chem. Soc. 2002, 124, 8584. (b) Herzon, S. B.; Myers, A. G. J. Am. Chem. Soc. 2005, 127, 5342. (c) Martin, R.; Jäger, A.; Böhl, M.; Richter, S.; Fedorov, R.; Manstein, D.

J.; Gutzeit, H. O.; Knölker, H.-J. *Angew. Chem., Int. Ed.* **2009**, *48*, 8042. (d) Wegner, J.; Ley, S. V.; Kirschning, A.; Hansen, A.-L.; Garcia, J. M.; Baxendale, I. R. *Org. Lett.* **2012**, *14*, 696.

(3) (a) Bressy, C.; Menant, C.; Piva, O. Synlett 2005, 577. (b) Evans, P.; McCabe, T.; Morgan, B. S.; Reau, S. Org. Lett. 2005, 7, 43. (c) Pathak, T. P.; Sigman, M. S. Org. Lett. 2011, 13, 2774. (d) Gigant, N.; Gillaizeau, I. Org. Lett. 2012, 14, 4622.

(4) For selected recent papers on 2,3-dihydropyrrole synthesis, see:
(a) Wender, P. A.; Strand, D. J. Am. Chem. Soc. 2009, 131, 7528.
(b) Brawn, R. A.; Panel, J. S. Org. Lett. 2009, 11, 473. (c) Monge, D.; Jensen, K. L.; Franke, P. T.; Lykke, L.; Jørgensen, K. A. Chem.—Eur. J. 2010, 16, 9478. (d) Zhang, G.; Zhang, Y.; Jiang, X.; Yan, W.; Wang, R. Org. Lett. 2011, 13, 3806. (e) Liu, C.-R.; Zhu, B.-H.; Zheng, J.-C.; Sun, X.-L.; Xie, Z.; Tang, Y. Chem. Commun. 2011, 47, 1342. (f) Cheng, J.; Jiang, X.; Zhu, C.; Ma, S. Adv. Synth. Catal. 2011, 353, 1676. (g) Tian, J.; Zhou, R.; Sun, H.; Song, H.; He, Z. J. Org. Chem. 2011, 76, 2374. (h) Polindara-García, L. A.; Miranda, L. D. Org. Lett. 2012, 14, 5408. (i) Ghorai, M. K.; Tiwari, D. P. J. Org. Chem. 2013, 78, 2617.

(5) (a) Yoo, E. J.; Ahlquist, M.; Kim, S. H.; Bae, I.; Fokin, V. V.; Sharpless, K. B.; Chang, S. Angew. Chem., Int. Ed. 2007, 46, 1730. (b) Raushel, J.; Fokin, V. V. Org. Lett. 2010, 12, 4952. (c) Liu, Y.; Wang, X.; Xu, J.; Zhang, Q.; Zhao, Y.; Hu, Y. Tetrahedron 2011, 67, 6294.

(6) For reviews, see: (a) Chattopadhyay, B.; Gevorgyan, V. Angew. Chem., Int. Ed. **2012**, 51, 862. (b) Gulevich, A. V.; Gevorgyan, V. Angew. Chem., Int. Ed. **2013**, 52, 1371.

(7) Miura, T.; Biyajima, T.; Fujii, T.; Murakami, M. J. Am. Chem. Soc. **2012**, 134, 194.

(8) Miura, T.; Tanaka, T.; Biyajima, T.; Yada, A.; Murakami, M. Angew. Chem., Int. Ed. 2013, 52, 3883.

(9) (a) Horneff, T.; Chuprakov, S.; Chernyak, N.; Gevorgyan, V.; Fokin, V. V. J. Am. Chem. Soc. 2008, 130, 14972. (b) Miura, T.; Yamauchi, M.; Murakami, M. Chem. Commun. 2009, 1470. (c) Grimster, N.; Zhang, L.; Fokin, V. V. J. Am. Chem. Soc. 2010, 132, 2510. (d) Chattopadhyay, B.; Gevorgyan, V. Org. Lett. 2011, 13, 3746. (e) Chuprakov, S.; Kwok, S. W.; Fokin, V. V. J. Am. Chem. Soc. 2013, 135, 4652. (f) Schultz, E. E.; Sarpong, R. J. Am. Chem. Soc. 2013, 135, 4696. (g) Parr, B. T.; Green, S. A.; Davies, H. M. J. Am. Chem. Soc. 2013, 135, 4716. (h) Spangler, J. E.; Davies, H. M. L. J. Am. Chem. Soc. 2013, 135, 6802.

(10) Zibinsky, M.; Fokin, V. V. Angew. Chem., Int. Ed. 2013, 52, 1507. (11) (a) Müller, P.; Allenbach, Y.; Robert, E. Tetrahedron: Asymmetry 2003, 14, 779. (b) Ghanem, A.; Gardiner, M. G.; Williamson, R. M.; Müller, P. Chem.—Eur. J. 2010, 16, 3291. (c) DeAngelis, A.; Boruta, D. T.; Lubin, J.-B.; Plampin, J. N., III; Yap, G. P. A.; Fox, J. M. Chem. Commun. 2010, 46, 4541.

(12) Davies, H. M. L.; DeMeese, J. Tetrahedron Lett. 2001, 42, 6803.

(13) The isolated (*E*)-7ab was racemic. On the other hand, the reaction of triazole 1a with (*E*)-crotonaldehyde (4a) under analogous conditions afforded enantiomerically enriched 4-oxazoline (73% ee), in accordance with Fokin's report. However, the isolated 4-oxazoline facilely underwent racemization even at room temperature. These results indicate that the ring opening of the *N*,*O*-aminal moiety and recyclization occur in a reversible way to racemize 4-oxazolines.

(14) McKinney, M. A.; Patel, P. P. J. Org. Chem. 1973, 38, 4059.

(15) For an aza-Cope/Mannich reaction triggered by the C-O bond cleavage of the *N,O*-aminal moiety of 5-vinyloxazolidines, see: (a) Overman, L. E.; Kakimoto, M.; Okawara, M. *Tetrahedron Lett.* **1979**, *20*, 4041. (b) Johnson, B. F.; Marrero, E. L.; Turley, W. A.; Lindsay, H. A. *Synlett* **2007**, 893. (c) Carballo, R. M.; Purino, M.; Ramírez, M. A.; Martín, V. S.; Padrón, J. I. *Org. Lett.* **2010**, *12*, 5334.

(16) The byproduct **6ao** was easily deprotected in situ to afford 1,4-oxazepine derivative in 20% yield.

(17) (a) Miura, T.; Funakoshi, Y.; Morimoto, M.; Biyajima, T.; Murakami, M. J. Am. Chem. Soc. **2012**, 134, 17440. (b) Selander, N.; Worrell, B. T.; Fokin, V. V. Angew. Chem., Int. Ed. **2012**, 51, 13054.

(18) (a) Collado, I.; Ezquerra, J.; Vaquero, J. J.; Pedregal, C. *Tetrahedron Lett.* **1994**, 35, 8037. (b) Cheng, T.; Meng, S.; Huang, Y. *Org. Lett.* **2013**, 15, 1958.