## **Preliminary communication**

# Synthesis of methyl 6-deoxy-4-C-(hydroxymethyl)-5-O-methyl-2,3-O-methylene-L-idonate

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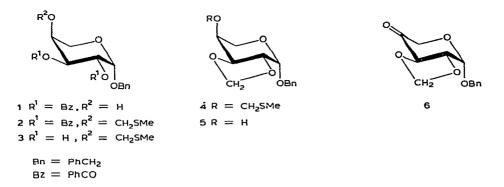
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As a part of studies on branched-chain sugars, syntheses of such new, branchedchain sugars as D-evermicose<sup>1</sup>, D-evalose<sup>2</sup>, and L-evernitrose<sup>3</sup> (found in certain oligosaccharide antibiotics, the everninomycins<sup>4</sup>) have been reported from our laboratory. The structure of the remaining, branched-chain sugar lactone, which is linked by an acetal interlinkage<sup>5</sup> at a terminal position of everninomycins B and D, was recently determined by Ganguly *et al.*<sup>6</sup> to be 6-deoxy-4-*C*-(hydroxymethyl)-5-*O*-methyl-2,3-*O*-methylene-L-idono-1,4<sup>1</sup>-lactone having the (*R*) configuration at C-1. Because the branched-chain sugar lactone was chemically characterized<sup>7</sup> as the methyl ester (20) of the free acid, the synthesis of 20 from L-arabinose is described in this communication.

Treatment of benzyl 2,3-di-O-benzoyl- $\beta$ -L-arabinopyranoside<sup>8</sup> (1) in acetic acid with dimethyl sulfoxide and acetic anhydride gave the corresponding 3-O-(methylthio)-methyl ether (2) as a syrup,  $[\alpha]_D$  +176.8° (c 1.24, CHCl<sub>3</sub>), in 86% yield; this was O-debenzoylated with sodium methoxide, to give benzyl 3-O-(methylthio)methyl- $\beta$ -L-arabino-pyranoside (3), m.p. 72–73°,  $[\alpha]_D$  +193.9° (c 0.72, CHCl<sub>3</sub>), in 57% yield. Treatment of 3 with an excess of dichloromethane and sodium hydride? gave the corresponding 2,3-O-methylene derivative (4) as a syrup,  $[\alpha]_D$  +110° (c 1.0, CHCl<sub>3</sub>), in 46% yield; this was O-de-(methylthio)methylated with mercuric chloride in the presence of calcium carbonate, to give benzyl 2,3-O-methylene- $\beta$ -L-arabinopyranoside (5) as a syrup,  $[\alpha]_D$  +212° (c 1.6, CHCl<sub>3</sub>), in 62% yield. The conformation of 5 was proved by its n.m.r. spectrum to be  ${}^4C_1$  (L) [ $\delta$  7.28–7.40 (m, 5 H, Ph), 5.36 (d, 1 H,  $J_{1,2}$  1.0 Hz, H-1), 5.11 and 5.04 (each d, 2 H,  $J_{gem}$  1.0 Hz, 2,3-O-methylene), 4.75 and 4.64 (ABq, 2 H, J 11.6 Hz, CH<sub>2</sub>Ph), 4.30 (m, 1 H, H-4), 3.89 (q, 1 H,  $J_{3,4}$  1.0,  $J_{2,3}$  10.0 Hz, H-3), 3.79 (dd, 1 H, H-2), 3.73 (dd, 1 H,  $J_{4,5e}$  1.8, H-5e), 3.58 (dd, 1 H,  $J_{4,5a}$  2.0,  $J_{5a,5e}$  13.2 Hz, H-5a), and 3.22 (broad s, OH)].

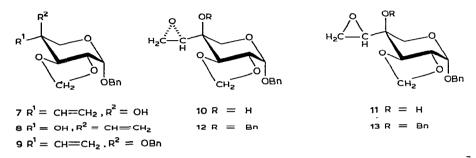
Oxidation of 5 with dimethyl sulfoxide-trifluoroacetic anhydride<sup>10</sup> gave benzyl 2,3-O-methylene- $\beta$ -L-threo-pentopyranosid-4-ulose (6) in good yield. Compound 6 could not be completely purified, but the n.m.r. spectrum could be analyzed:  $\delta$  7.52-7.21 (m, 5 H, Ph), 5.50 (d, 1 H,  $J_{1,2}$  3.0 Hz, H-1), 5 20 and 5.10 (each d,  $J_{gem}$  1.0 Hz, 2,3-O-methylene), 4.86 and 4.76 (ABq, J 11.5 Hz, CH<sub>2</sub>Ph), 4.72 (dd, 1 H,  $J_{3,5e} \sim$ 1.0 Hz, H-3), 4.02 (dd, 1 H, H-5e), 3.92 (d, 1 H,  $J_{5a,5e}$  15.0 Hz, H-5a), and 3.67 (dd, 1 H,  $J_{2,3}$  10.8 Hz, H-2).

PRELIMINARY COMMUNICATION



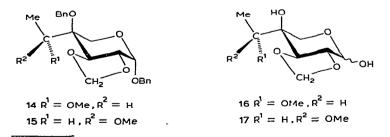
Reaction of 6 with vinylmagnesium bromide in oxolane gave a 4.1:1 mixture of benzyl 2,3-*O*-methylene-4-*C*-vinyl- $\beta$ -L-arabinopyranoside (7), as a syrup {[ $\alpha$ ]<sub>D</sub> +170° (*c* 1.2, CHCl<sub>3</sub>); n.m.r.: δ 7.42-7.24 (m, 5 H, Ph), 5.86 (dd, 1 H, J<sub>cis</sub> 10.0, J<sub>trans</sub> 16.7 Hz, CH in vinyl group), 5.53 (dd, 1 H, Jgem 2.0 Hz, trans-H in CH<sub>2</sub> of vinyl group), 5.38 (d, 1 H, J<sub>1,2</sub> 2.0 Hz, H-1), 5.30 (dd, 1 H, cis-H in CH<sub>2</sub> of vinyl group), 5.15 and 5.07 (each d, 2 H, Jgem 1.0 Hz, 2.3-O-methylene), 4.79 and 4.67 (ABq, 2 H, J 12.2 Hz, CH<sub>2</sub>Ph), 3.97 (d, 1 H, J<sub>2.3</sub> 9.4 Hz, H-3), 3.90 (dd, 1 H, H-2), 3.54 and 3.47 (ABq, 2 H, J 11.0 Hz, H-5e, 5a), and 2.50 (broad s, OH)}, and its 4-epimer (8) (of the  $\alpha$ -D-xylo configuration), as a syrup {[ $\alpha$ ]<sub>D</sub> +137° (c 3.0, CHCl<sub>3</sub>); n.m.r.: δ 7.41-7.20 (m, 5 H, Ph), 6.12 (dd, 1 H, J<sub>cis</sub> 11.0, J<sub>trans</sub> 17.6 Hz, CH in vinyl group), 5.52 (dd, 1 H, J<sub>gem</sub> 1.8 Hz, trans-H in CH<sub>2</sub> of vinyl group), 5.32 (dd, 1 H, cis-H in CH<sub>2</sub> of vinyl group), 5.26 (d, 1 H, J<sub>1,2</sub> 3.3 Hz, H-1), 5.09 and 5.03 (each d, Jgem 1.0 Hz, 2,3-O-methylene), 4.76 and 4.63 (ABq, J 12.0 Hz, CH<sub>2</sub>Ph), 4.01 (d, 1 H, J<sub>2,3</sub> 9.2 Hz, H-3), 3.62 and 3.56 (ABq, 2 H, J 10.8 Hz, H-5e, 5a), 3.37 (dd, 1 H, H-2), and 2.96 (broad s, OH) }, in 59% yield; 7 was separated from 8 on a column of silica gel (eluant, 4:1 hexane-ethyl acetate). The configuration at the quarternary carbon atom was determined by comparison of the chemical shifts of the methine carbon atom of the vinyl group [7,  $\delta$ 137.15 p.p.m. (from Me<sub>4</sub>Si); 8,  $\delta$  135.15 p.p.m.] in the <sup>13</sup>C-n.m.r. spectra<sup>11</sup>. Treatment of 7 with benzyl bromide and sodium hydride gave the corresponding 4-O-benzyl derivative (9) as a syrup,  $[\alpha]_{D}$  +101° (c 1.8, CHCl<sub>3</sub>), in 89% yield.

Epoxidation of the double bond in 7 with *m*-chloroperoxybenzoic acid in dichloromethane gave a 4.5:1 mixture of benzyl 2,3-*O*-methylene-4-*C*-[(*S*)-oxiran-2-yl]-β-L-arabinopyranoside (11) as a syrup,  $[\alpha]_D$  +182° (*c* 0.9, CHCl<sub>3</sub>), and the corresponding (*R*) isomer (10) as a syrup,  $[\alpha]_D$  +183.2° (*c* 1.1, CHCl<sub>3</sub>), in 52% yield. Similar oxidation of 9 gave the corresponding *R* epoxide (12) as a syrup { $[\alpha]_D$  +114.6° (*c* 1.1, CHCl<sub>3</sub>); n.m.r.:  $\delta$ 7.48–7.10 (m, 5 H, Ph), 5.39 (d, 1 H,  $J_{1,2}$  2.0 Hz, H-1), 5.14 and 5.07 (each d, 2 H,  $J_{gem}$ 1.0 Hz, 2,3-*O*-methylene), 4.82 and 4.78 (ABq, 2 H, *J* 15.2 Hz, 4-*O*-CH<sub>2</sub>Ph), 4.78 and 4.68 (ABq, 2 H, *J* 11.6 Hz, 1-*O*-CH<sub>2</sub>Ph), 4.10–3.98 (m, 2 H, H-2,3), 3.95 and 3.58 (ABq, *J* 12.2 Hz, H-5*e*,5*a*), and 3.16, 2.90, and 2.79 (each dd, 3 H, *J* 3.0, 4.0, and 5.0 Hz, epoxy H)}, and the (*S*) epoxide (13) as a syrup { $[\alpha]_D$  +114.7° (*c* 1.1, CHCl<sub>3</sub>), n.m.r.:  $\delta$  7.50–7.12 (m, 5 H, Ph), 5.34 (d, 1 H,  $J_{1,2}$  2.6 Hz, H-1), 5.08 and 5.04 (each d, 2 H,  $J_{gem}$  0.8 Hz, 2,3-*O*methylene), 4.94 and 4.85 (ABq, 2 H, *J* 11.6 Hz, 4-*O*-CH<sub>2</sub>Ph), 4.77 and 4.66 (ABq, 2 H, *J*  12.4 Hz, 1-O-CH<sub>2</sub>Ph), 4.00 (dd, 1 H,  $J_{2,3}$  10.0 Hz, H-2), 3.93 (d, 1 H, H-3), 3.78 and 3.66 (ABq, 2 H, J 12.5 Hz, H-5e,5a), and 3.11, 3.04, and 2.83 (each dd, 3 H, J 2.9, 4.2, and 5.0 Hz, epoxy-H) } in the ratio of 1:1, in 65% yield. The difference in the ratio of isomers, as between the products from 7 and 9, indicates the effect of the tertiary hydroxyl group on C-4, as had been observed for 2-cyclohexen-1-ol derivatives<sup>12</sup>. The configurations of the epoxides were deduced by comparison of the ratio with that of the products obtained by similar oxidation of the 2,3-di-O-benzyl analog of 7; unexpectedly, their absolute configurations could be determined by a chemical method.\*



Reduction of 12 and 13 in oxolane with lithium aluminum hydride gave the corresponding alcohols, which were characterized as the *O*-methyl derivatives: 14, a syrup,  $[\alpha]_D$  +123.9° (c 0.8, CHCl<sub>3</sub>), 74%; and 15, a syrup,  $[\alpha]_D$  +114.5° (c 0.9, CHCl<sub>3</sub>), 69%. Hydrogenation of 14 and 15 in the presence of palladium-on-charcoal gave 4-*C*-[(*R*)-1-methoxyethyl]- (16), a syrup,  $[\alpha]_D$  -28.7° (c 3.3, EtOH), and 4-*C*-[(*S*)-1-methoxyethyl]- 2,3-*O*-methylene-L-arabinose (17), m.p. 53–57°,  $[\alpha]_D$  -44.7° (c 1.9, EtOH), in 93 and 83% yield, respectively.

Oxidation of 16 in 85% aqueous methanol with bromine, treatment with silver oxide, and separation of the products by preparative t.l.c., gave methyl 6-deoxy-4-C-(hydroxy-methyl)-5-O-methyl-2,3-O-methylene-D-gluconate (18) as a syrup (see Table I), together with the dimethyl acetal (19) of 16 as a syrup  $\{[\alpha]_D -23.7^\circ (c \ 2.2, \text{CHCl}_3); \text{ n.m.r.: } \delta \ 5.07$  and 4.93 (each s, 2 H, 2,3-O-methylene), 4.44 (d, 1 H, H-1), 4.35 (t, 1 H,  $J_{1,2} = J_{2,3} =$ 



<sup>\*</sup>For the 2,3-di-O-benzyl derivatives, the absolute configuration of the 4-C-(oxiran-2-yl) derivatives was determined by comparison of the corresponding, reduced products with that of those obtained by the epoxidation of the corresponding (Z)- and (E)-4-C-ethylene derivatives followed by alkaline cleavage of the epoxide ring.

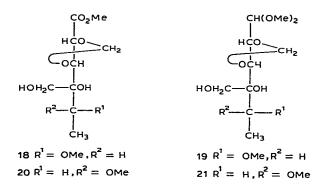
TABLE I

	[¤] <sub>D</sub> in CHCl, (degrees)	Chemical shifts (8) and coupling constants (Hz)								
		H-2	H-3	H-4'	H-5	H-6	OCH <sub>2</sub> O	ОМе	CO <sub>2</sub> Me	OH
18	-70.0	4.74d	4.41d	3.60 ABq 3.78	3.67q	1.28d	4.96s 5.21s	3.38	3.81	3.00
		J <sub>2,3</sub> 5.0		J 12.4	J <sub>5,6</sub> 6.5					
20	-26.1	4.85d	4.17d	3.59 ABq 3.78	3.69q	1.26d	4.98s 5.23s	3.40	3.80	2.66
		J <sub>2,3</sub> 5.0		J 12.2	J <sub>5,5</sub> 6.5					
20 (reported <sup>7</sup> )	-28	4.85d	4.19d	3.6-3	8.9m	1.25d	5.0s 5.21s	3.35	3.81	-
		$J_{2,3} 5.0 \qquad J_{5,6} 6.5$								

COMPARISON OF ROTATIONAL VALUES AND N.M.R. PARAMETERS

4.8 Hz, H-2), 4.27 (d, 1 H, H-3), 3.68 (q, 1 H,  $J_{5,6}$  6.4 Hz, H-5), ~3.6 (m, 2 H, 4'-CH<sub>2</sub>), 3.50, 3.47, and 3.36 (each s, 3 OMe), 2.81 (broad s, 2 OH), and 1.26 (d, 3 H, H-6)}, in 10 and 13% yield, respectively. Similar oxidation of 17 gave the C-5 epimer (20) of 18 as a syrup (see Table I), and the dimethyl acetal (21) of 17 as a syrup {[ $\alpha$ ]<sub>D</sub> +3.8° (c 2.3, CHCl<sub>3</sub>); n.m.r.:  $\delta$  5.10 and 4.98 (each d, 2 H,  $J_{gem}$  0.8 Hz, 2,3-O-methylene), 4.50-4.36 (m, 2 H, H-1,3), 4.05 (dd, 1 H, J 1.2 and 3.0 Hz, H-2), 3.71 (q, 1 H,  $J_{5,6}$  6.0 Hz, H-5), 3.68 and 3.54 (ABq, 2 H, J 12.4 Hz, 4'-CH<sub>2</sub>), 3.51, 3.49, and 3.42 (each s, 3 OMe), 3.02 (broad s, 2 OH), and 1.28 (d, 3 H, H-6)}, in 8 and 6% yield, respectively.

Comparison of the rotational value and n.m.r. parameters (see Table I) of 18 and 20 with those reported<sup>7</sup> indicated that the determination and deduction of the configuration of C-4 and of the  $\alpha$ -carbon atom in the branch chain were correct. Thus, synthesis of all of the new, branched-chain sugars found in the eveninomycins is now complete.



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