## SYNTHESIS OF A NEW BRIDGED DIAMINE 3,6-DIABABICYCLO [3.2.0] HEPTANE: APPLICATIONS TO THE SYNTHESIS OF QUINOLONE ANTIBACTERIALS.

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**Abstract:** Diprotected 3,6-diazabicyclo [3.2.0] heptane has been prepared by an highly efficient process. Selective deprotection, followed by condensation with 7-halogenoquinolones led to the naphthyridones <u>10</u> and <u>13</u> and the quinolone <u>14</u>.

Our general interest<sup>1</sup> in azetidines, pyrrolidines and piperazines led us to examine new bridged diamines including pyrrolidine and azetidine moieties. We report here a convenient preparation of the 3,6-diazabicyclo [3.2.0] heptane.

Pyrrolidine 1 (cis and trans mixture)<sup>2</sup> was reacted with methanesulfonyl chloride in the presence of TEA to afford trans dimesylate<sup>3</sup> 2 (78%) purified by HPLC (silica gel, 4:1  $CH_2Cl_2$ -EtOAc). Treatment of 2 with an equimolecular amount of sodium azide in DMF at room temperature gave the monoazide 3. Reduction of 3 over PtO<sub>2</sub> under a hydrogen atmosphere followed by treatment with trifluoroacetic anhydride gave 5 (60% from 3). Cyclisation of 5 in DMF in presence of NaH at room temperature furnished the diprotected cis-diamine 6<sup>4</sup> in 97% yield.

1565



Deprotection of the azetidinyl part of 6 with ethanolic 1N NaOH at 80°C gave 7 which could be condensed with ethyl 7-chloro-1,8-naphthyridone-3-carboxylate to give 8, which was hydrogenated over Pd on C in presence of hydrochloric acid in methanol. Alkaline hydrolysis of 9 with 1N NaOH at room temperature afforded naphthyridone 10.



Deprotection of the pyrrolidinyl part of 6 was performed by catalytic hydrogenolysis ( $H_2$ , Pd/C) in ethanol to give 11 which was condensed with 7-chloro-1,8-naphthyridone or quinolone-3 carboxylic acids to afford 12. Basic hydrolysis of 12 in ethanol gave the corresponding naphthyridone and quinolone-3-carboxylic acids 13 and 14.



Some of these novel compounds displayed a substantial antibacterial activity: structure-activity relationships and biological data will be reported elsewhere.

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## References and notes

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  - c) Remuzon, P.; Bouzard, D.; Di Cesare, P.; Essiz, M.; Jacquet, J.P.; Kiechel, J.R.; Ledoussal, B.; Kessler, R.E.; J. Fung-Tomc, J. Med. Chem., 1991, <u>34</u>, 29.
- 2) Jaeger, E.; Biel, J.H.; J. Org. Chem., 1965, 30, 740.
- Trans and cis dimesylate were determined by NOE experiment on a 200MHz Bruker NMR apparatus.
- 4)a) This product 6 (oil) has been identified by highly-resolution 1H-NMR: (200MHz, CDCl<sub>3</sub>); 2.50-2.90 (CH<sub>2</sub>N, H<sub>1</sub>, 4H, 2m); 3.04 (CHN, 1H, dd); 3.56(CH<sub>2</sub>N, 2H, q); 3.68(CH<sub>2</sub>-Ph, 2H, s); 4.75(H<sub>4</sub>, 1H, m); 7.29(5H, Ar, m) and MS (70eV) 284 M<sup>+</sup>.
  - b) After completion of our work , the related diamine 3-benzyl-6-tertbutoxycarbonyl 3,6-diazabicyclo [3.2.0] heptane, synthetised via an other process appeared in a Japanese patent/JP 8956, 673; Chem. Abstr. 1989, <u>111</u>, 153779W.
  - c) Attempts to exchange the mesyl group of 5 with tetrabutyl ammonium fluoride in THF to give the corresponding fluoro- pyrrolidine failed. It was obtained exclusively the diamine 6.
- 5) Diamine 15 as dimaleate, white needles, m.p. 180°C, was characterized by 1H NMR, (200MHz, DMSO d6); 2.38(H<sub>1</sub>, 1H, m);2.87-3.43(CH<sub>2</sub>N, 6H,m); and by elemental analysis.

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