

SYNTHESIS OF 1-ARYLAZO-3-ACETYL-1H-CINNOLIN-4-ONE DERIVATIVES

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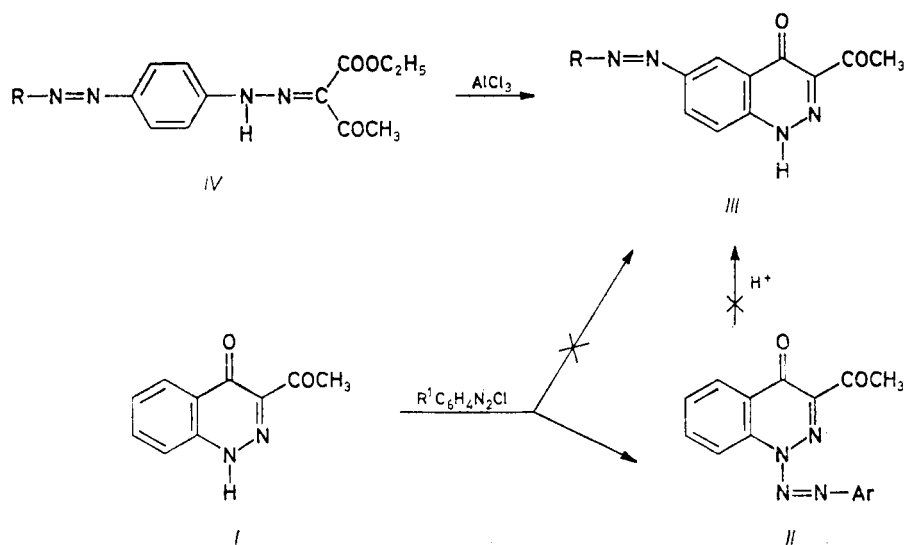
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The biological activity of azocinnoline derivatives especially in the chemotherapy of *Trypanosomiasis* was already reported¹⁻³. In 1971 for the first time pyridazine derivatives were found in nature⁴ and quite recently another naturally occurring derivative (nigellicine)⁵ has been described⁶.

Considering the foregoing and in continuation of our previous work⁷, 3-acetyl-1H-cinnolin-4-one (*I*) has been well exploited as a coupling component. Compound *I* has two potentially nucleophilic centres, i.e. position 1 and 6 and was expected to undergo azo-coupling reaction at either or both of these positions. In fact, reaction of *I* with aryldiazonium salts proceeds smoothly in aqueous ethanol in the presence

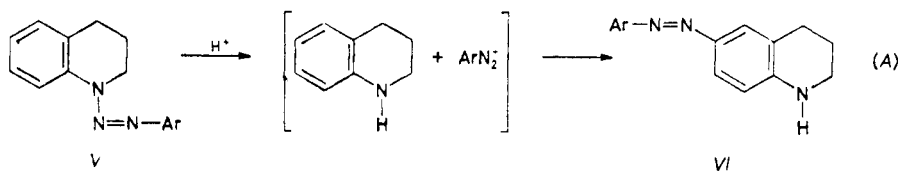


In formula II: a, $R^1 = H$ b, $R^1 = p-CH_3$ c, $R^1 = p-OCH_3$ d, $R^1 = p-NO_2$ e, $R^1 = p-Br$
 f, $R^1 = o-COOH$ g, $R^1 = p-SO_3H$

SCHEME 1

of sodium carbonate to give red coloured dyes identified as 1-arylaazo-3-acetyl-1*H*-cinnolin-4-ones (*II*) (Scheme 1).

On the other hand, it was reported that the diazoimino derivatives of tetrahydroquinoline (*V*) are rearranged in dilute mineral acid to the isomeric 6-areneazotetrahydroquinoline⁸ (*VI*) (Equation (*A*)). Attempts at rearrangement of *IIa* by boiling



dilute hydrochloric acid to give the isomeric 6-arylaazo-3-acetyl-1*H*-cinnolin-4-one (*III*) were unsuccessful. Compound *III* ($R = C_6H_5$) was prepared by another route, by intramolecular cyclization of ethyl acetoacetate (4-phenylazo)phenylhydrazone (*IV*), which was prepared by coupling of diazotized 4-aminoazobenzene with ethyl acetoacetate by anhydrous $AlCl_3$ in chlorobenzene (Scheme 1).

TABLE I
UV spectra of compounds *I–III* (Scheme 1)

Compound	λ_{\max}^a	$\epsilon_{\max} 10^{-3},^a$	λ_{\max}	$\epsilon_{\max} 10^{-3}$	λ_{\max}	$\epsilon_{\max} 10^{-3}$
<i>I</i>	350	23·333	296 sh 286 sh	4·666 3·999	242	11·833
<i>IIa</i>	440	23·666	300 260	17·333 10·333	250	11·666
<i>IIb</i>	442	24·833	306	17·166	254	11·666
<i>IIc</i>	450	26·833	322	18·333	252	13·999
<i>IId</i>	434	16·999	304	8·833	248	6·499
<i>IIf</i>	442	23·666	306 260 sh	15·999 9·666	254	10·666
<i>IIg</i>	434	25·333	300	19·666	244	10·166
<i>IIh</i>	358	32·333	308 sh 298 sh	6·333 5·499	253	18·333
<i>III</i>	402	40·500	300 br	4·250	246	10·500

^a λ_{\max} in nm and ϵ_{\max} in $\text{mol}^{-1} \text{cm}^2$.

EXPERIMENTAL

All melting points are uncorrected and were recorded on Fisher-Johns melting point apparatus. ^1H NMR were obtained on 90 MHz Varian spectrometer in δ ppm relative to TMS. Electronic spectra were recorded on a Shimadzu UV 200S spectrophotometer using 1 cm matched silica cells.

3-Acetyl-1*H*-cinnolin-4-one (*I*)

This compound was prepared by cyclization of ethyl 2-(phenylhydrazono)-3-oxobutanoate with AlCl_3 in chlorobenzene as cited in ref.⁷.

1-Arylazo-3-acetyl-1*H*-cinnolin-4-ones (*IIa–IIg*)

General procedure. A solution of diazotized aryl amine (0.01 mol) was added gradually with stirring to a solution 3-acetyl-1*H*-cinnolin-4-one (0.01 mol) in aqueous sodium carbonate (200 ml,

TABLE II
Physical and analytical data of compounds *II–IV*

Compound	M.p. °C	Yield %	Formula M.w.	Calculated/Found		
				% C	% H	% N
<i>IIa</i>	125–127	95	$\text{C}_{16}\text{H}_{12}\text{N}_4\text{O}_2$ 292.30	65.75	4.14	19.17
				66.05	4.30	18.95
<i>IIb</i>	132–135	93	$\text{C}_{17}\text{H}_{14}\text{N}_4\text{O}_2$ 306.32	66.66	4.61	18.29
				66.42	4.25	17.98
<i>IIc</i>	130	90	$\text{C}_{17}\text{H}_{14}\text{N}_4\text{O}_3$ 322.32	63.35	4.38	17.38
				63.50	4.30	17.27
<i>IId</i>	182–185	75	$\text{C}_{16}\text{H}_{11}\text{N}_5\text{O}_4$ 337.29	56.98	3.29	20.76
				56.78	2.95	20.70
<i>IIf</i>	158–160	80	$\text{C}_{16}\text{H}_{11}\text{N}_4\text{O}_2\text{Br}$ 371.19	51.77	2.99	15.09 ^a
				51.67	3.15	14.94
<i>IIg</i>	195	78	$\text{C}_{17}\text{H}_{12}\text{N}_4\text{O}_4$ 336.30	60.71	3.60	16.66
				60.87	3.65	16.64
<i>III</i>	160–162	69	$\text{C}_{16}\text{H}_{12}\text{N}_4\text{O}_5\text{S}$ 372.35	51.61	3.25	15.05 ^b
				51.82	3.37	15.25
<i>IV</i>	220	75	$\text{C}_{16}\text{H}_{12}\text{N}_4\text{O}_2$ 292.30	65.75	4.14	19.17
				65.70	4.10	19.25
<i>IV</i>	145–147	94	$\text{C}_{18}\text{H}_{18}\text{N}_4\text{O}_3$ 338.36	63.90	5.36	16.56
				63.78	5.23	16.51

^a Calculated: 21.53% Br, found: 21.66% Br. ^b Calculated: 8.61% S, found: 8.68% S.

2.5%) and ethanol (20 ml) at 0–5°C during 15 min. The resulting red precipitate was filtered off and recrystallized from ethanol. Compound *II* precipitated after acidification with acetic acid. The ^1H NMR spectrum of *IIa* in CDCl_3 exhibited signals at δ 2.55 s, 3 H (CH_3); δ 7.25 to 7.65 m, 9 H (CH-Ar), and the disappearance of the signal at δ 14 due to NH. Results are summarized in Tables I and II.

Ethyl 2-[(4-phenylazo)phenylhydrazono]-3-oxobutanoate (*IV*)

A solution of 4-aminoazobenzene (0.025 mol) in acetic acid (10 ml) and hydrochloric acid (20 ml, 50% (v/v)) was diazotized at 0–5°C by addition of sodium nitrite (0.025 mol) dissolved in water (10 ml). The resulting suspension was added with stirring to a cold solution of ethyl acetoacetate (0.025 mol) in aqueous ethanol (40 ml, 50%) containing sodium acetate (10 g) during 15 min. The resulting solid was recrystallized from ethanol, giving orange red needles of m.p. 145–147°C. The ^1H NMR of *IV* in CDCl_3 showed signals at δ 1.25–1.40 t, 3 H (CH_3 -ester); δ 2.50 s ($\text{CH}_3\text{—COCH}_3$), δ 4.15–4.40 q, 2 H (C_2H -ester); δ 7.30–7.90 m, 9 H (CH-Ar), and at δ 14.65 s, 1 H (NH).

6-Phenylazo-3-acetyl-1*H*-cinnolin-4-one (*III*)

A mixture of ethyl 2-[(4-phenylazo)phenylhydrazono]-3-oxobutanoate (*IV*) (0.01 mol), anhydrous aluminium chloride (0.02 mol) and chlorobenzene (30 ml) was refluxed on a water bath for 1 h. The resulting complex was cooled and decomposed by addition of concentrated hydrochloric acid (20 ml). The product was recrystallized from benzene, affording orange crystals of m.p. 220°C. The ^1H NMR spectrum of *III* in CDCl_3 exhibited signals at δ 2.65 s, 3 H (COCH_3); 7.45–8.05 m, 8 H (CH-Ar), and at δ 14.00 s, 1 H (NH).

REFERENCES

1. Macey P. E., Simpson J. C. E.: *J. Chem. Soc.* 1952, 2602.
2. McIntyre J., Simpson J. C. E.: *J. Chem. Soc.* 1952, 2606.
3. McIntyre J., Simpson J. C. E.: *J. Chem. Soc.* 1952, 2615.
4. Bevan K., Davies J., Hassall C., Morton R., Phillips D.: *J. Chem. Soc., C* 1971, 514.
5. Rahman Atta-Ur, Malik S., Cun-heng He, Clardy J.: *Tetrahedron Lett.* 26, 2759 (1985).
6. Suortti T., Von Wright A., Koskinen A.: *Phytochemistry* 22, 2873 (1983).
7. Youssef M. S. K., Kamal El-Dean A. M., Abbady M. S., Hassan K. M.: *Collect. Czech. Chem. Commun.* 56, 1768 (1991).
8. Saunders K. H., Allen R. L. M.: *Aromatic Diazo Compounds*, 3rd ed., p. 405. E. Arnold, Brisbane 1985.