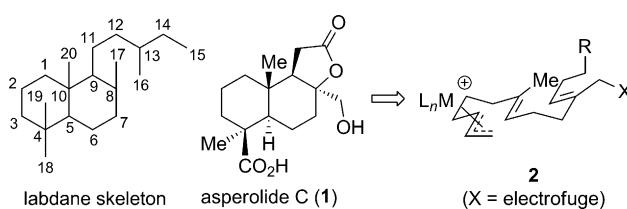


Total Synthesis of (+)-Asperolide C by Iridium-Catalyzed Enantioselective Polyene Cyclization**

Oliver F. Jeker, Alberto G. Kravina, and Erick M. Carreira*

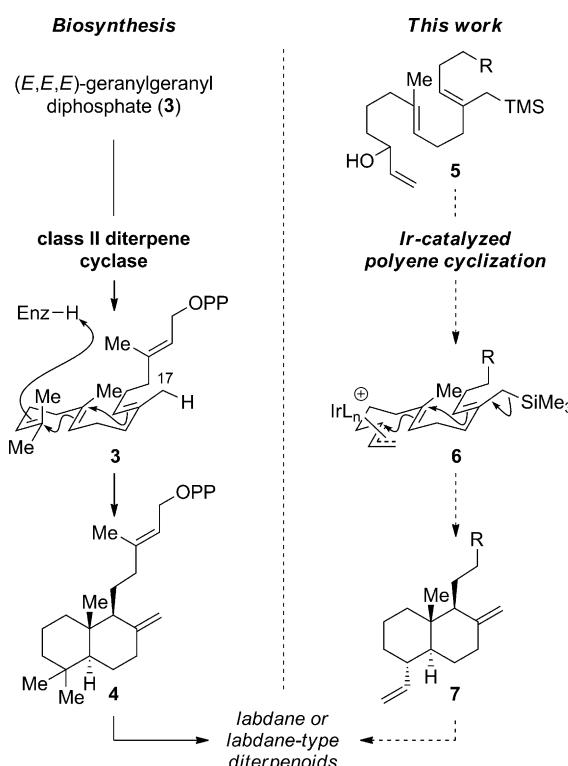
The labdane diterpenes encompass a structurally diverse class of natural products that are widely distributed in terrestrial and marine organisms.^[1] Many of these compounds exhibit notable biological properties, such as antibacterial,^[2] antimutagenic,^[3] cytotoxic,^[4] anti-inflammatory, and analgesic activities.^[5] Various procedures have been developed for their preparation,^[6] including isolation from natural sources and chemical or biological manipulation.^[7] Herein we report the first total synthesis of asperolide C (**1**, Scheme 1), a tetranor-



Scheme 1. Labdane skeleton, structure of asperolide C (**1**) and asymmetric catalytic polycyclization.

labdane diterpenoid isolated from *Aspergillus wentii* EN-48,^[8] through a unique asymmetric catalytic polycyclization cascade which is reminiscent of its biogenesis. A number of enantioselective polyolefin cyclizations have been reported in the literature.^[9] It is interesting to note, however, that applications of these processes in natural product synthesis are scarce, with only two reports emanating from the Yamamoto research group.^[10]

The carbon skeleton common to labdane-type diterpenoids is assembled biosynthetically from (*E,E,E*)-geranylgeranyl diphosphate (**3**) through a polyolefin cyclization pathway, which is consistent with the hypotheses of Stork and Eschenmoser.^[11,12] The currently accepted mechanism involves protonation-initiated cyclization of **3**, enabled by



Scheme 2. Proposed biosynthesis of labdane-type diterpenoids (labda-13-en-8-yl diphosphate cation omitted for conciseness) and iridium-catalyzed polyene cyclization.

class II diterpene cyclases, to generate a labda-13-en-8-yl diphosphate cation (Scheme 2).^[13] This intermediate typically suffers deprotonation of the C(17) methyl group to produce copalyl pyrophosphate **4**. The allylic diphosphate in **4** then engages in other rearrangement and/or cyclization reactions followed by a series of downstream biosynthetic modifications that lead to the introduction of additional functionality. It is estimated that a significant number of polycyclic diterpenoids (ca. 7000) arise from this initial cyclization event.

As part of our program on the study of iridium-catalyzed enantioselective transformations,^[14] we became interested in their application to the synthesis of complex molecules. A particularly useful set of reactions that we recently disclosed includes the asymmetric cyclization of polyunsaturated allylic alcohols.^[15,16] It was envisioned that this method, which employed arenes to conclude the cationic cascades, might be expanded in new ways to provide a general entry into the alicyclic labdane core. Significantly, this would require the use of an allyl silane as a terminating group. Accordingly, linear

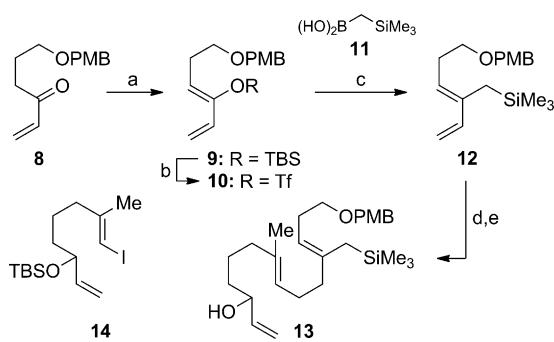
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allylic alcohol **5** (Scheme 2) would be subjected to conditions that generate π -allyl iridium complex **6**. This reactive intermediate would then undergo a series of stereoselective cyclizations before suffering loss of the Me_3Si group to form the exomethylene in **7**. The *trans*-decalin system thus obtained would be a valuable intermediate in the synthesis of various labdane or labdane-type diterpenoids.

The synthesis of asperolide C (**1**) commenced with the preparation of vinyl ketone **8** (Scheme 3), which was obtained from commercially available γ -butyrolactone in three steps.

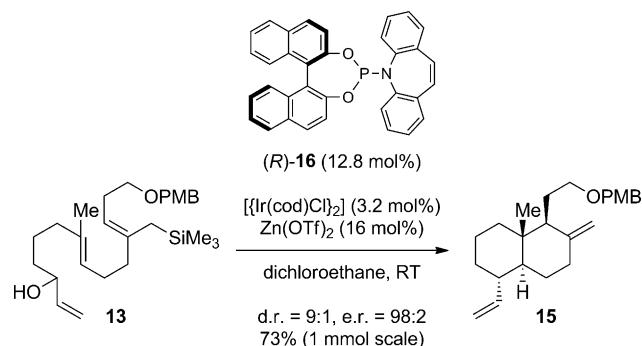


Scheme 3. Reagents and conditions: a) LiHMDS (1.25 equiv), *t*Bu-Me₂SiCl (1.25 equiv), THF/HMPA, -78°C, d.r. > 95:5, 95%; b) PhNTf₂ (1.5 equiv), CsF (2.5 equiv), (MeOCH₂)₂, RT, d.r. > 95:5, 96%; c) **11** (1.5 equiv), [Pd(dppf)Cl₂]·CH₂Cl₂ (10 mol %), Ph₃As (10 mol %), Cs₂CO₃ (2.5 equiv), THF/DMF/H₂O, 0°C to RT, d.r. = 10:1, 62%; d) 9-BBN (1.1 equiv), THF, 0°C to RT; then **14** (1.0 equiv), [Pd(dppf)Cl₂]·CH₂Cl₂ (2.7 mol %), NaOH (3.0 equiv), THF/H₂O, 0°C to RT, 61%; e) PPTS (10 mol %), MeOH, RT, 81% (2 iterations). PMB = *p*-methoxybenzyl, TBS = *tert*-butylsilyl, dppf = 1,1'-bis(diphenylphosphino)ferrocene, PPTS = pyridinium *p*-toluenesulfonate, HMPA = hexamethylphosphoramide, HMDS = hexamethyldisilazane, Tf = trifluoromethanesulfonyl, 9-BBN = 9-borabicyclo[3.3.1]nonane.

Initial attempts to directly transform enone **8** to enol triflate **10** (LiHMDS, 2-NTf₂-5-chloropyridine, THF, -78°C) proved unsuccessful. Therefore, a two-step protocol was examined, which involved conversion of **8** into enol silane **9** followed by exchange of the silyl group with the required triflate. Formation of **9** under standard conditions was hampered by polymerization of **8**. However, the desired intermediate could be obtained in good yield (95%) and excellent *Z* selectivity (d.r. > 95:5) when **8** was added to a premixed solution of LiHMDS and *tert*-butyldimethylsilyl chloride in THF at -78°C, using HMPA as a cosolvent. Enol triflate **10** was obtained with complete retention of the olefin geometry by treatment with triflic fluoride, generated *in situ* following the procedure of Corey and co-workers (96%).^[17] Cross-coupling of enol triflate **10** and boronic acid **11** was achieved under the conditions of Johnson and Braun, by using [Pd(dppf)Cl₂] (10 mol %) as a catalyst in combination with Ph₃As (10 mol %) as a coligand and Cs₂CO₃ as a base to produce the desired product **12** in 62% yield (*Z/E* = 10:1).^[18–20] The terminal olefin in diene **12** was hydroborated with 9-BBN and the resulting trialkylborane was directly subjected to *B*-alkyl Suzuki coupling with known vinyl iodide **14** to afford the desired polyene (61%).^[15,21] Removal of the TBS protective group required carefully chosen conditions. Whereas

attempts with TBAF or *para*-toluenesulfonic acid led to decomposition of the material, PPTS (10 mol %) was found to catalyze the desired transformation. However, as a result of the poor stability of the product in acidic media, the reaction was usually halted at 60% conversion.^[22] Re-subjection of the recovered starting material led to allylic alcohol **13** in 81% combined yield.

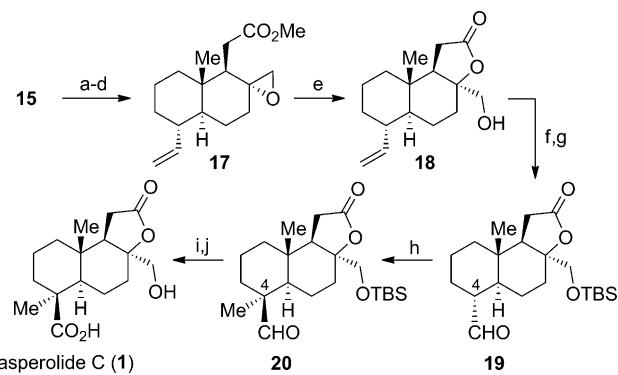
With polyene precursor **13** in hand, the pivotal iridium-catalyzed cyclization cascade was examined (Scheme 4). Gratifyingly, reaction of **13** under standard conditions with $[\{\text{Ir}(\text{cod})\text{Cl}\}_2]$ (3.2 mol %) and (*R*)-**16** (12.8 mol %) as catalyst precursors, along with Zn(OTf)₂ (16 mol %) as a Lewis acid delivered decalin **15** with excellent stereoselectivity (d.r. = 9:1, e.r. = 98:2) and in 73% yield.^[15,23]



Scheme 4. Iridium-catalyzed enantioselective polyene cyclization cascade.

The synthesis of asperolide C (**1**) continued with deprotection of **15** followed by stepwise oxidation of the liberated primary hydroxy group. The carboxylic acid thus obtained was treated with trimethylsilyldiazomethane to afford the corresponding methyl ester (62% over 4 steps, Scheme 5). Epoxidation of the exomethylene group with freshly prepared DMDO at -20°C proceeded from the sterically more accessible α face to deliver oxirane **17** in moderate yield (45%, 66% brsm). Exposure of **17** to trifluoroacetic acid in anhydrous CH₂Cl₂ at 0°C led to selective epoxide opening and efficient cyclization to furnish lactone **18** (70%).^[24]

With the tricyclic scaffold of the target constructed, the final stages of the total synthesis were addressed. After masking the primary hydroxy group in **18** as a TBS ether (TBSCl, imidazole, DMAP, 89%), Lemieux-Johnson oxidation afforded aldehyde **19** in 81% yield. Introduction of the quaternary center at C(4) by enolate alkylation was complicated by the presence of a γ -lactone ($\text{pK}_a \approx 20$). We reasoned that the direct alkylation of an aldehyde ($\text{pK}_a \approx 17$) could offer the necessary chemoselectivity in the deprotonation event. In the experiment, treatment of a solution of **19** in THF at -20°C with *t*BuOK (1.25 equiv),^[25] followed by addition of iodomethane (1.25 equiv) and warming to 0°C delivered **20** as a single isolable product. Pinnick oxidation of aldehyde **20** to the corresponding carboxylic acid (76%) and cleavage of the TBS group (74%) completed the first total synthesis of asperolide C (**1**). The ¹H and ¹³C NMR spectra of the synthetic material were in agreement with those reported



Scheme 5. Reagents and conditions: a) DDQ (1.1 equiv), pH 7 buffer, CH_2Cl_2 , RT, 98%; b) DMP (1.5 equiv), CH_2Cl_2 , RT, 80%; c) NaClO_2 (4.0 equiv), NaH_2PO_4 (6.0 equiv), 2-methyl-2-butene (70 equiv), $t\text{BuOH}/\text{H}_2\text{O}$, RT; then Me_3SiCH_2 (1.1 equiv), MeOH , 0°C to RT, 79%; d) DMDO (1.1 equiv), acetone, -78°C to -20°C, 45% (66% brsm); e) $\text{CF}_3\text{CO}_2\text{H}$ (1.2 equiv), CH_2Cl_2 , -20°C to 0°C, 70%; f) $t\text{Bu}-\text{Me}_2\text{SiCl}$ (3.0 equiv), imidazole (6.0 equiv), DMAP (10 mol%), CH_2Cl_2 , RT, 89%; g) OsO_4 (20 mol%), NaO_4 (5.0 equiv), 2,6-lutidine (2.0 equiv), 1,4-dioxane/ H_2O , RT, 81%; h) $t\text{BuOK}$ (1.25 equiv), MeI (1.25 equiv), THF, -20°C to 0°C, 36%; i) NaClO_2 (6.0 equiv), NaH_2PO_4 (6.0 equiv), 2-methyl-2-butene (30 equiv), $t\text{BuOH}/\text{H}_2\text{O}$, RT, 76%; j) TBAF (1.5 equiv), THF, 0°C, 74%. DDQ = 2,3-dichloro-5,6-dicyanobenzoquinone, DMP = Dess–Martin periodinane, DMDO = dimethyldioxirane, DMAP = 4-dimethylaminopyridine, TBAF = tetra-*n*-butylammonium fluoride, brsm = based on recovered starting material.

for the natural product. It should be noted that asperolide C (**1**) was originally isolated as an inseparable 3:4 mixture with the known terpene botryosphaerin, which precluded thorough characterization. With a pure sample of **1** in hand, we could measure its optical rotation for the first time ($[\alpha]_D^{26} = +2.5$ ($c = 0.25$, MeOH)).

In conclusion, the first total synthesis of the tetranorlabdane diterpenoid asperolide C (**1**) has been achieved. This study represents a rare example of the use of an enantioselective polyene cyclization reaction in a natural product synthesis and the first that strategically relies on modern iridium catalysis to construct the carbocyclic core scaffold. Additionally, the described route features a series of cross-coupling reactions to efficiently assemble the linear polyene precursor. Specifically, the Pd-mediated coupling of a dienol triflate with $\text{Me}_3\text{SiCH}_2\text{B}(\text{OH})_2$ provides a novel access route to allylic silanes. Moreover, a chemo- and diastereoselective alkylation of an aldehyde enolate was employed to complete the target structure. The synthetic strategy provides a general entry into the labdane-type diterpenoids and has significant potential for enabling the total synthesis of other terpene natural products.

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- [1] a) I. Chinou, *Curr. Med. Chem.* **2005**, *12*, 1295–1317; b) A. F. Barrero, M. M. Herrador, P. Arteaga, J. F. Arteaga, A. F. Arteaga, *Molecules* **2012**, *17*, 1448–1467.
- [2] M. Singh, M. Pal, R. P. Sharma, *Planta Med.* **1999**, *65*, 2–8.
- [3] M. Miyazawa, H. Shimamura, S. Nakamura, H. Kameoka, *J. Agric. Food Chem.* **1995**, *43*, 3012–3015.
- [4] H. Itokawa, H. Morita, I. Katou, K. Takeya, A. J. Cavalheiro, R. C. B. de Oliveira, M. Ishige, M. Motidome, *Planta Med.* **1988**, *54*, 311–315.
- [5] H. Morita, H. Itokawa, *Planta Med.* **1988**, *54*, 117–120.
- [6] B. Z. S. Awen, M. Nozawa, H. Hagiwara, *Org. Prep. Proced. Int.* **2008**, *40*, 317–363.
- [7] L. M. T. Frija, R. F. M. Frade, C. A. M. Afonso, *Chem. Rev.* **2011**, *111*, 4418–4452.
- [8] H.-F. Sun, X.-M. Li, L. Meng, C.-M. Cui, S.-S. Gao, C.-S. Li, C.-G. Huang, B.-G. Wang, *J. Nat. Prod.* **2012**, *75*, 148–152.
- [9] a) K. Ishihara, S. Nakamura, H. Yamamoto, *J. Am. Chem. Soc.* **1999**, *121*, 4906–4907; b) Y.-J. Zhao, B. Li, L.-J. S. Tan, Z.-L. Shen, T.-P. Loh, *J. Am. Chem. Soc.* **2010**, *132*, 10242–10244; c) K. Surendra, E. J. Corey, *J. Am. Chem. Soc.* **2012**, *134*, 11992–11994; d) A. Sakakura, A. Ukai, K. Ishihara, *Nature* **2007**, *445*, 900–903; e) S. Rendler, D. W. C. MacMillan, *J. Am. Chem. Soc.* **2010**, *132*, 5027–5029; f) R. R. Knowles, S. Lin, E. N. Jacobsen, *J. Am. Chem. Soc.* **2010**, *132*, 5030–5032; g) C. A. Mullen, A. N. Campbell, M. R. Gagné, *Angew. Chem.* **2008**, *120*, 6100–6103; *Angew. Chem. Int. Ed.* **2008**, *47*, 6011–6014; h) S. G. Sethofer, T. Mayer, F. D. Toste, *J. Am. Chem. Soc.* **2010**, *132*, 8276–8277.
- [10] a) K. Ishihara, H. Ishibashi, H. Yamamoto, *J. Am. Chem. Soc.* **2001**, *123*, 1505–1506; b) H. Ishibashi, K. Ishihara, H. Yamamoto, *J. Am. Chem. Soc.* **2004**, *126*, 11122–11123.
- [11] a) G. Stork, A. W. Burgstahler, *J. Am. Chem. Soc.* **1955**, *77*, 5068–5077; b) A. Eschenmoser, L. Ruzicka, O. Jeger, D. Arigoni, *Helv. Chim. Acta* **1955**, *38*, 1890–1904; c) P. A. Stadler, A. Eschenmoser, H. Schinz, G. Stork, *Helv. Chim. Acta* **1957**, *40*, 2191–2198; d) A. Eschenmoser, D. Arigoni, *Helv. Chim. Acta* **2005**, *88*, 3011–3050.
- [12] For more recent discussions, see a) L. Kürti, R.-J. Chein, E. J. Corey, *J. Am. Chem. Soc.* **2008**, *130*, 9031–9036; b) R. A. Yoder, J. N. Johnston, *Chem. Rev.* **2005**, *105*, 4730–4756.
- [13] R. J. Peters, *Nat. Prod. Rep.* **2010**, *27*, 1521–1530.
- [14] For selected examples, see a) J. Y. Hamilton, D. Sarlah, E. M. Carreira, *Angew. Chem.* **2013**, *125*, 7680–7683; *Angew. Chem. Int. Ed.* **2013**, *52*, 7532–7535; b) S. Krautwald, D. Sarlah, M. A. Schafroth, E. M. Carreira, *Science* **2013**, *340*, 1065–1068.
- [15] M. A. Schafroth, D. Sarlah, S. Krautwald, E. M. Carreira, *J. Am. Chem. Soc.* **2012**, *134*, 20276–20278.
- [16] For pioneering studies from other research groups involving Ir-catalyzed cyclizations with C-nucleophiles, see a) Q.-L. Xu, L.-X. Dai, S.-L. You, *Org. Lett.* **2012**, *14*, 2579–2581; b) S. Streiff, C. Welter, M. Schelwies, G. Lipowsky, N. Miller, G. Helmchen, *Chem. Commun.* **2005**, 2957–2959.
- [17] Y. Mi, J. V. Schreiber, E. J. Corey, *J. Am. Chem. Soc.* **2002**, *124*, 11290–11291.
- [18] C. R. Johnson, M. P. Braun, *J. Am. Chem. Soc.* **1993**, *115*, 11014–11015.
- [19] Cross-coupling under the conditions reported by Negishi and co-workers (X. Zeng, M. Qian, Q. Hu, E. Negishi, *Angew. Chem.* **2004**, *116*, 2309–2313; *Angew. Chem. Int. Ed.* **2004**, *43*, 2259–2263) delivered the *E* isomer of **12** as the major product (X. Zeng, Q. Hu, M. Qian, E. Negishi, *J. Am. Chem. Soc.* **2003**, *125*, 13636–13637).
- [20] The olefin geometry of **12** and the corresponding *E* isomer were unambiguously assigned by 2D NMR experiments (NOESY).
- [21] For a review on the *B*-alkyl Suzuki–Miyaura cross-coupling reaction, see S. R. Chemler, D. Trauner, S. J. Danishefsky,

- Angew. Chem.* **2001**, *113*, 4676–4701; *Angew. Chem. Int. Ed.* **2001**, *40*, 4544–4568.
- [22] As determined by ^1H NMR analysis of the crude reaction mixture.
- [23] The relative and absolute configuration of the polyene cyclization product were determined by X-ray diffraction analysis of a derivative (see the Supporting Information).
- [24] Precedence for this transformation came from a report by Barrero et al. who described a closely related process as a side

reaction: A. F. Barrero, J. F. Sánchez, J. Elmerabet, D. Jiménez-González, F. A. Macías, A. M. Simonet, *Tetrahedron* **1999**, *55*, 7289–7304.

- [25] a) E. Alvarez-Manzaneda, R. Chahboun, E. Alvarez, J. M. Ramos, J. J. Guardia, I. Messouri, I. Chayboun, A. I. Mansour, A. Dahdouh, *Synthesis* **2010**, 3493–3503; b) T. N. Thompson, M. G. Sierra, J. D. McChesney, *J. Org. Chem.* **1985**, *50*, 4447–4450.

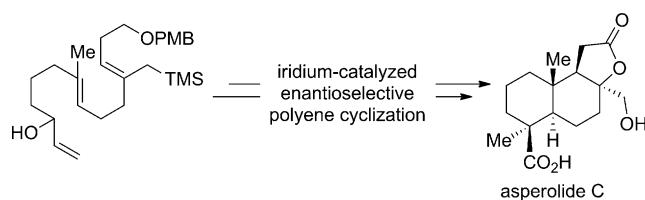
Communications



Natural Product Synthesis

O. F. Jeker, A. G. Kravina,
E. M. Carreira*

Total Synthesis of (+)-Asperolide C by
Iridium-Catalyzed Enantioselective
Polyene Cyclization



Domino rings: A general synthetic entry into labdane-type diterpenoids has been developed based on an iridium-catalyzed enantioselective polyene cyclization cascade. The potential of this process is

demonstrated in the first total synthesis of the tetranorlabdane diterpene asperolide C ($\text{PMB} = p\text{-methoxybenzyl}$, $\text{TMS} = \text{trimethylsilyl}$).