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Amidophenacylating Reagents in Synthesis of New Derivatives of 1,3-Oxazole- and 1,3-Thiazole-4-sulfonyl Chlorides and Corresponding Sulfonamides

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Abstract—Derivatives of 4-benzylsulfanyl-1,3-oxazole and 4-benzylsulfanyl-1,3-thiazole were synthesized using available amidophenacylating reagents. By the oxidative chlorination compounds obtained were converted to 1,3-oxazole-4-sulfonyl and 1,3-thiazol-4-sulfonyl chlorides. The latter were used to prepare the corresponding sulfonamides.

Keywords: 1,3-thiazole, 1,3-oxazole, sulfonyl chlorides, sulfonamides

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In the past decade, a significant number of natural and synthetic biologically active compounds containing 1,3-oxazole or 1,3-thiazole fragments were prepared [1–5]. It stimulates the studies of reactivity and search for the ways of heterocycles functionnalization. In the present work we developed new methods of the synthesis of 1,3-oxazole- and 1,3thiazole-4-sulfonyl chlorides as well as the corresponding sulfonamides.

Available amidophenacylating reagents I, which are convenient syntons for the preparation of a number of functionally substituted oxazoles and thiazoles [6, 7], were used as starting substrates. 2-Aryl-5-phenyl-4benzylsulfanyl-1,3-oxazoles III were obtained through simple sequence of the reactions $I \rightarrow II \rightarrow III$ (Scheme 1). The oxidative chlorination of these compounds (Cl₂, MeCOOH/H₂O, 0°C) afforded 2arvl-5-phenvl-1.3-oxazole-4-sulfonvl chlorides **IV**. The latter were purified by recrystallization from aprotic solvents and stored for a long time without any changes. Treating with excess aqueous ammonia, primary or secondary amines in the presence of triethylamine in dioxane resulted in high yields of the corresponding sulfonamides V-VII. 1,3-Oxazole-4sulfonyl chlorides IV and 1,3-oxazole-4-sulfonamides V–VII were synthesized for the first time.

Heating substrates II with the Lawesson's reagent in dioxane afforded the thiazole derivatives VIII (Scheme 2). The latter were converted into 2-aryl-(methyl)-5-phenyl-1,3-thiazole-4-sulfonyl chlorides IX by the action of chlorine in acetic acid. Chlorides IX were used to obtain a series of appropriate sulfonamides X–XII. It should be noted that similar 1,3-thiazole-4-sulfonyl chlorides and 1,3-thiazole-4sulfonamides have been known [8–17], however 2,5disubstituted 1,3-thiazole-4-sulfonyl chlorides IX and sulfonamides X–XII we prepared for the first time.

The composition and structure of new compounds were confirmed by elemental analysis (Table 1), IR, ¹H and ¹³C NMR spectra and by LC-MS spectrometry (Table 2). The formation of azole rings as a result of transformations II \rightarrow III and II \rightarrow VIII was proved by the disappearance of the signals of >CHNH group in ¹H NMR spectra and intensive absorption bands in the ranges of 1635–1649 (C=O), 1677–1686 (C=O), and 3256–3396 cm⁻¹ (N–H) in the IR spectra and by appearance of the absorption bands at 1143–1384 cm⁻¹ belonging to SO₂ group. ¹H and ¹³C NMR spectra of the synthesized compounds contain all the necessary signals. In GC-MS spectra there are peaks corresponding to protonated molecular ions [M + 1]⁺.



 $R = Ph (Ia-VIIa, VIc, VIIc, VIIe), 4-MeC_6H_4 (Ib-VIIb, VId, VIId, VIIf), Me (Ic, IIc); R^1NH = PhCH_2NH (VIa, VIb), PhNH (VIc, VId); R^2R^3N = Me_2N (VIIa, VIIb), (CH_2)_5N (VIIc, VIId), O(CH_2)_4N (VIIe, VIIf).$

Scheme 2.



R = Ph (IIa, VIIIa–XIIa, XId, XIIc, XIIf), 4-MeC₆H₄ (IIb, VIIIb–XIIb, XIe, XIId, XIIg), Me (IIc, VIIIc–XIc, XIf, XIIe, XIIh); R¹NH = PhCH₂NH (XIa–XIc), PhNH (XId–XIf); R²R³N = Me₂N (XIIa, XIIb), (CH₂)₅N (XIIc–XIIe),

C ~

O(CH₂)₄N (**XIIf-XIIh**); RL = MeO
$$P$$
 S P $-$ OMe.

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Comp. no.	Yield, %	mp, °C (solvent for recrystallization)	Found, %			Calculated, %	
			Ν	S	Formula	Ν	S
IIa	85	122–123 (MeCN)	3.95	8.81	$C_{22}H_{19}NO_2S$	3.87	8.87
IIb	83	102–104 (EtOH)	3.68	8.51	$C_{23}H_{21}NO_2S$	3.73	8.54
IIc	80	134–135 (EtOH)	4.55	10.73	$C_{17}H_{17}NO_2S$	4.68	10.71
IIIa	70	89–91 (EtOH)	4.00	9.36	C ₂₂ H ₁₇ NOS	4.08	9.34
IIIb	70	95–96 (EtOH)	3.89	8.95	C ₂₃ H ₁₉ NOS	3.92	8.97
IVa	75	147-148 (toluene)	4.21	9.94	$C_{15}H_{10}ClNO_3S^a$	4.38	10.03
IVb	72	102-103 (cyclohexane)	4.12	9.52	C ₁₆ H ₁₂ ClNO ₃ S ^b	4.20	9.61
Va	80	248-249 (MeCN)	9.39	10.65	$C_{15}H_{12}N_2O_3S$	9.33	10.68
Vb	81	254–256 (MeCN)	8.93	10.20	$C_{16}H_{14}N_2O_3S$	8.91	10.20
VIa	86	153–154 (MeCN)	7.21	8.21	$C_{22}H_{18}N_2O_3S$	7.17	8.21
VIb	85	175-176 (MeCN)	6.99	7.91	$C_{23}H_{20}N_2O_3S$	6.93	7.93
VIc	85	204–205 (MeCN)	7.41	8.56	$C_{21}H_{16}N_2O_3S$	7.44	8.52
VId	84	213-214 (MeCN)	7.19	8.24	$C_{22}H_{18}N_2O_3S$	7.17	8.21
VIIa	79	129–130 (EtOH)	8.42	9.79	$C_{17}H_{16}N_2O_3S$	8.53	9.76
VIIb	80	126–127 (EtOH)	8.11	9.33	$C_{18}H_{18}N_2O_3S$	8.18	9.36
VIIc	84	100–101 (EtOH)	7.56	8.79	$C_{20}H_{20}N_2O_3S$	7.60	8.70
VIId	84	142–143 (EtOH)	7.27	8.33	$C_{21}H_{22}N_2O_3S$	7.32	8.38
VIIe	87	119–121 (EtOH)	7.58	8.64	$C_{19}H_{18}N_2O_4S$	7.56	8.66
VIIf	87	122-123 (EtOH)	7.31	8.31	$C_{20}H_{20}N_2O_4S$	7.29	8.34
VIIIa	70	88-89 (MeCN)	3.81	17.86	$C_{22}H_{17}NS_2$	3.90	17.84
VIIIb	70	105-106 (MeCN)	3.55	17.25	$C_{23}H_{19}NS_2$	3.75	17.17
VIIIc	65	Oil	4.75	21.52	$C_{17}H_{15}NS_2$	4.71	21.56
IXa	74	131–132 (cyclohexane)	4.02	19.00	$C_{15}H_{10}CINO_2S_2^{c}$	4.17	19.10
IXb	74	121–122 (cyclohexane)	3.35	18.25	$C_{16}H_{12}CINO_2S_2{}^d$	4.00	18.33
IXc	67	98–99 (hexane)	5.02	23.38	C ₁₀ H ₈ ClNO ₂ S ₂ ^e	5.12	23.42
Xa	80	236–237 (MeCN)	8.91	20.21	$C_{15}H_{12}N_2O_2S_2$	8.85	20.27
Xb	80	264–265 (MeCN)	8.52	19.33	$C_{16}H_{14}N_2O_2S_2$	8.48	19.41
Xc	72	148 (EtOH)	11.04	25.21	$C_{10}H_{10}N_2O_2S_2$	11.01	25.21
XIa	85	135–136 (MeCN)	6.92	15.81	$C_{22}H_{18}N_2O_2S_2 \\$	6.89	15.77
XIb	85	158–159 (MeCN)	6.70	15.20	$C_{23}H_{20}N_2O_2S_2\\$	6.66	15.25
XIc	80	174–175 (MeCN)	8.21	18.57	$C_{17}H_{16}N_2O_2S_2$	8.13	18.62
XId	83	162-163 (MeCN)	7.16	16.31	$C_{21}H_{16}N_2O_2S_2$	7.14	16.34
XIe	83	171–172 (MeCN)	6.84	15.76	$C_{22}H_{18}N_2O_2S_2 \\$	6.89	15.77
XIf	80	199–200 (MeCN)	8.37	19.44	$C_{16}H_{14}N_2O_2S_2\\$	8.48	19.41
XIIa	80	174–175 (MeCN)	8.10	18.66	$C_{17}H_{16}N_2O_2S_2\\$	8.13	18.62
XIIb	82	209–210 (MeCN)	7.75	17.87	$C_{18}H_{18}N_2O_2S_2\\$	7.81	17.89
XIIc	85	135–136 (EtOH)	7.11	16.72	$C_{20}H_{20}N_2O_2S_2$	7.29	16.68
XIId	85	146–147 (MeCN)	7.07	16.04	$C_{21}H_{22}N_2O_2S_2\\$	7.03	16.09
XIIe	80	131–132 (cyclohexane)	8.57	19.93	$C_{15}H_{18}N_2O_2S_2\\$	8.69	19.89
XIIf	86	173–174 (MeCN)	7.19	16.59	$C_{19}H_{18}N_2O_3S_2\\$	7.25	16.59
XIIg	87	132-133 (MeCN)	7.09	16.00	$C_{20}H_{20}N_2O_3S_2\\$	6.99	16.01
XIIh	80	171-173 (toluene)	8.60	19.71	$C_{14}H_{16}N_2O_3S_2$	8.63	19.77

Table 1. Yields, melting points, and elemental analyses of compounds II-XII

^a Found Cl, %: 11.14. Calculated Cl, %: 11.09. ^b Found Cl, %: 10.81. Calculated Cl, %: 10.62. ^c Found Cl, %: 10.72. Calculated Cl, %: 10.56. ^d Found Cl, %: 10.27. Calculated Cl, %: 10.13. ^e Found Cl, %: 13.16. Calculated Cl, %: 12.95.

 Table 2. Spectral data for compounds II–XII

Comp. no.	IR spectrum (KBr), v, cm ⁻¹	¹ H NMR spectrum (DMSO- d_6) ^a , δ , ppm	Mass spectrum, m/z
IIa	1347, 1524; 1635 (C=O), 1686 (C=O); 3297 (NH)	3.86 d, 4.02 d (2H, CH ₂ S, ${}^{2}J_{HH}$ 13.0 Hz), 6.65 d (1H, CH, ${}^{3}J_{HH}$ 7.8 Hz), 7.32–7.90 m (15H, 3C ₆ H ₅), 9.25 d (1H, NH, ${}^{3}J_{HH}$ 7.8 Hz)	$362 [M+1]^+$
IIb	1344,1485; 1649 (C=O), 1677 (C=O); 3396 (NH)	2.38 s (3H, CH ₃), 3.86 d, 4.02 d (2H, CH ₂ S, ${}^{2}J_{HH}$ 13.0 Hz), 6.63 d (1H, CH, ${}^{3}J_{HH}$ 7.8 Hz), 7.27–7.87 m (14H, 2C ₆ H ₅ , C ₆ H ₄), 9.14 d (1H, NH, ${}^{3}J_{HH}$ 7.8 Hz)	$376 [M+1]^+$
IIc	1331, 1522; 1647 (C=O), 1684 (C=O); 3256 (NH)	1.97 s (3H, CH ₃), 3.79 d, 3.94 d (2H, CH ₂ S, ${}^{2}J_{HH}$ 13.0 Hz), 6.44 d (1H, CH, ${}^{3}J_{HH}$ 8.0 Hz), 7.28–7.82 m (10H, 2C ₆ H ₅), 8.94 d (1H, NH, ${}^{3}J_{HH}$ 8.0 Hz)	$300 [M+1]^+$
III a ^b	1482, 1552, 1600	4.38 s (2H, CH ₂), 7.26–8.13 m (15H, 3C ₆ H ₅)	$344 [M+1]^+$
IIIb	1495, 1556, 1601	2.40 s (3H, CH ₃), 4.36 s (2H, CH ₂), 7.19–8.01 m (14H, 2C ₆ H ₅ , C ₆ H ₄)	$358 [M+1]^+$
IVa	1170, 1384 (SO ₂); 1487, 1559,	$7.28-8.19 \text{ m} (10\text{H}, 2\text{C}_6\text{H}_5)$	_
	1606		
IVb	1170, 1393 (SO ₂); 1498, 1564, 1611	_	_
Va	1157, 1353 (SO ₂); 1487, 1567; 3206, 3336 (NH ₂)	7.55–8.11 m (12H, NH ₂ , 2C ₆ H ₅)	$301 [M+1]^+$
Vb	1158, 1352 (SO ₂); 1496, 1568, 1612; 3198, 3325 (NH ₂)	2.41 s (3H, CH ₃), 7.42–8.00 m (11H, NH ₂ , C ₆ H ₅ , C ₆ H ₄)	$315 [M+1]^+$
VIa	1155, 1336 (SO ₂); 1448, 1485, 1547; 3243 (NH)	4.27 s (2H, CH ₂), 7.23–8.08 m (15H, 3C ₆ H ₅), 8.77 s (1H, NH)	391 $[M+1]^+$
VIb	1151, 1328 (SO ₂); 1448, 1493, 1555, 1612; 3148 (NH)	2.42 s (3H, CH ₃), 4.27 s (2H, CH ₂), 7.25–7.97 m (14H, 2C ₆ H ₅ , C ₆ H ₄), 8.72 s (1H, NH)	$405 [M+1]^+$
VIc	1152, 1343 (SO ₂); 1408, 1485, 1554; 3269 (NH)	7.05–8.04 m (15H, 3C ₆ H ₅), 10.80 s (1H, NH)	377 $[M+1]^+$
VId	1158, 1339 (SO ₂); 1413, 1497, 1556, 1612; 3251 (NH)	2.39 s (3H, CH ₃), 7.05–7.93 m (14H, 2C ₆ H ₅ , C ₆ H ₄), 10.78 s (1H, NH)	391 $[M+1]^+$
VIIa ^c	1150, 1338 (SO ₂); 1449, 1488, 1556	2.94 s (6H, 2CH ₃), 7.57–8.12 m (10H, 2C ₆ H ₅)	$329 [M+1]^+$
VIIb	1155, 1347 (SO ₂); 1449, 1499, 1557, 1612	2.41 s (3H, CH ₃), 2.93 s (6H, 2CH ₃), 7.40–7.99 m (9H, C ₆ H ₅ , C ₆ H ₄)	$343 [M+1]^+$
VIIc	1165, 1336 (SO ₂); 1447, 1488, 1554	1.48–1.59 m (6H, 3CH ₂ , piperidine), 3.28–3.33 m (4H, 2CH ₂ , piperidine), 7.57–8.12 m (10H, $2C_{6}H_{5}$)	$369 [M+1]^+$
VIId	1162, 1351 (SO ₂); 1442, 1498, 1561 1614	1.47–1.55 m (6H, 3CH ₂ , piperidine), 2.38 s (3H, CH ₃), 3.27– 3.37 m (4H, 2CH ₂ , piperidine), 7.41–8.00 m (9H, C ₆ H ₅ , C ₆ H ₄)	383 $[M+1]^+$
VIIe	1157, 1344 (SO ₂); 1448, 1487, 1555	3.30-3.33 m (4H, 2CH ₂ , morpholine), $3.67-3.70 m$ (4H, 2CH ₂ , morpholine), $7.57-8.13 m$ (10H, 2C(H ₂))	371 $[M+1]^+$
VIIf	1160, 1345 (SO ₂); 1455, 1497, 1556 1615	2.42 s (3H, CH ₃), $3.31-3.34$ m (4H, 2CH ₂ , morpholine), $3.66-3.69$ m (4H, 2CH ₂ morpholine), $7.41-8.00$ m (9H, C, H ₂ , C, H ₃)	$385 [M+1]^+$
VIIIa	1436 1451 1465 1497 1595	4 49 s (2H CH ₂) 7 21–8 02 m (15H 3C/H ₂)	$360 [M+1]^+$
VIIIh ^d	1413 1450 1468 1493 1596	$2.39 \text{ s} (3H \text{ CH}_2) + 4.49 \text{ s} (2H \text{ CH}_2) + 7.29-7.90 \text{ m} (14H \text{ 2C}_2\text{H}_2)$	$374 [M + 1]^+$
, 1110	1110, 1100, 1100, 1170, 1070	$C_{6}H_{4}$)	5/1[m ' 1]
IXa	1161, 1383 (SO ₂); 1441, 1464	$7.51-8.03 \text{ m} (10 \text{H}, 2 \text{C}_6 \text{H}_5)$	_

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Table 2. (Contd.)

Comp. no.	IR spectrum (KBr), v, cm ⁻¹	¹ H NMR spectrum (DMSO- d_6) ^a , δ , ppm	Mass spectrum, m/z
IXb	1157, 1372 (SO ₂); 1463	_	_
IXc	1168, 1384 (SO ₂); 1466, 1498	2.81 s (3H, CH ₃), 7.48–7.53 m (5H, C ₆ H ₅)	_
Xa	1146, 1334 (SO ₂); 1439, 1457, 1470; 3222, 3333 (NH ₂)	7.49–8.03 m (12H, NH ₂ , 2C ₆ H ₅)	$317 [M+1]^+$
Xb	1151, 1347 (SO ₂); 1455, 1472, 1566; 3192, 3350 (NH ₂)	2.37 s (3H, CH ₃), 7.38–7.92 m (11H, NH ₂ , C ₆ H ₅ , C ₆ H ₄)	$331 [M+1]^+$
Xc	1149, 1346 (SO ₂); 1442, 1469, 1492, 1555; 3282, 3346 (NH ₂)	2.71 s (3H, CH ₃), 7.43–7.55 m (7H, NH ₂ , C ₆ H ₅)	$255 [M+1]^+$
XIa	1145, 1325 (SO ₂); 1427, 1455, 1469; 3294 (NH)	4.29 s (2H, CH ₂), 7.25–7.98 m (15H, 3C ₆ H ₅), 8.62 s (1H, NH)	$407 [M+1]^+$
XIb	1143, 1340 (SO ₂); 1408, 1454, 1470; 3308 (NH)	2.41 s (3H, CH ₃), 4.29 s (2H, CH ₂), 7.21–7.87 m (14H, 2C ₆ H ₅ , C ₆ H ₄), 8.58 s (1H, NH)	$421 [M+1]^+$
XIc	1145, 1338 (SO ₂); 1420, 1443, 1499; 3290 (NH)	2.67 s (3H, CH ₃), 4.16 s (2H, CH ₂), 7.22–7.52 m (10H, $2C_6H_5$), 8.47 s (1H, NH)	$345 [M+1]^+$
XId	1148, 1349 (SO ₂); 1408, 1474, 1502, 1598; 3241 (NH)	7.15–7.92 m (15H, 3C ₆ H ₅), 10.65 s (1H, NH)	393 $[M+1]^+$
XIe	1149, 1348 (SO ₂); 1412, 1457, 1473, 1494, 1597; 3241 (NH)	2.38 s (3H, CH ₃), 7.16–7.80 m (14H, 2C ₆ H ₅ , C ₆ H ₄), 10.62 s (1H, NH)	$407 [M+1]^+$
XIf	1155, 1356 (SO ₂); 1426, 1487, 1595; 3127 (NH)	2.67 s (3H, CH ₃), 7.03–7.46 m (10H, 2C ₆ H ₅), 10.49 s (1H, NH)	$331 [M+1]^+$
XIIa	1145, 1352 (SO ₂); 1440, 1459	2.95 s (6H, 2CH ₃), 7.51–8.07 m (10H, 2C ₆ H ₅)	$345 [M+1]^+$
XIIb ^e	1145, 1352 (SO ₂); 1460	2.41 s (3H, CH ₃), 2.92 s (6H, 2CH ₃), 7.38–7.89 m (9H, C ₆ H ₅ , C ₆ H ₄)	$359 [M+1]^+$
XIIc	1166, 1334 (SO ₂); 1440, 1455, 1470	1.50–1.58 m (6H, 3CH ₂ , piperidine), 3.30–3.34 m (4H, 2CH ₂ , piperidine), 7.51–8.00 m (10H, 2C ₆ H ₅)	$385 [M+1]^+$
XIId	1158, 1343 (SO ₂); 1473	1.52–1.57 m (6H, 3CH ₂ , piperidine), 2.40 s (3H, CH ₃), 3.29–3.33 m (4H, 2CH ₂ , piperidine), 7.40–7.89 m (9H, C ₆ H ₅ , C ₆ H ₄)	$399 [M+1]^+$
XIIe	1146, 1344 (SO ₂); 1442, 1470, 1500	1.47–1.53 m (6H, 3CH ₂ , piperidine), 2.72 s (3H, CH ₃), 3.17–3.33 m (4H, 2CH ₂ , piperidine), 7.46–7.50 m (5H, C ₆ H ₅)	323 $[M+1]^+$
XIIf	1157, 1356 (SO ₂); 1442, 1458	3.27–3.31 m (4H, 2CH ₂ , morpholine), 3.65–3.68 m (4H, 2CH ₂ , morpholine), 7.52–8.01 m (10H, 2C ₆ H ₅)	$387 [M+1]^+$
XIIg	1158, 1356 (SO ₂); 1452	2.40 s (3H, CH ₃), 3.30–3.34 m (4H, 2CH ₂ , morpholine), 3.67–3.70 m (4H, 2CH ₂ , morpholine), 7.38–7.89 m (9H, C ₆ H ₅ , C ₆ H ₄)	$401 [M+1]^+$
XIIh	1157, 1350 (SO ₂); 1451, 1499	2.74 s (3H, CH ₃), 3.18–3.22 m (4H, 2CH ₂ , morpholine), 3.60–3.64 m (4H, 2CH ₂ , morpholine), 7.46–7.51 m (5H, C_6H_5)	$325 [M+1]^+$

 ^a ¹H NMR spectra of **IVa, IXa, IXc** were recorded in CDCl₃. ^b ¹³C NMR spectrum (DMSO-*d*₆), δ_C, ppm: 37.59 (SCH₂), 125.61, 126.65, 126.68, 127.63, 127.69, 128.85, 128.92, 129.33, 129.70, 130.41, 131.54, 138.21, 147.89 (C⁴, oxazoline), 159.76 (C², oxazoline). ^c ¹³C NMR spectrum (DMSO-*d*₆), δ_C, ppm: 38.46 [N(CH₃)₂], 126.00, 126.15, 127.06, 128.97, 129.12, 129.78, 131.06, 132.19, 134.34 (C⁴, oxazoline), 151.95, 159.71 (C², oxazoline). ^d ¹³C NMR spectrum (DMSO-*d*₆), δ_C, ppm: 21.51 (CH₃), 37.86 (SCH₂), 126.39, 127.55, 128.72, 128.83, 128.99, 129.33, 130.36, 130.42, 130.95, 131.67, 138.65, 141.13, 144.14 (C⁴, oxazoline), 165.26 (C², oxazoline). ^e ¹³C NMR spectrum (DMSO-*d*₆), δ_C, ppm: 21.54 (CH₃), 38.65 [N(CH₃)₂], 126.66, 128.89, 129.57, 130.10, 130.53, 141.94, 142.06, 146.62, 155.41 (C⁴, oxazoline), 166.35 (C², oxazoline).

Some 1,3-thiazole-4-sulfonamides were screened for anticancer activity within the International Scientific Program of the National Cancer Institute (USA) [18-21]. The screening was performed in vitro at 60 cancer cell lines under the action of the substance in a concentration of 1×10^{-5} M, whereby the cancer cells growth index (GI) was determined in comparison with reference (100%). Among the investigated compounds, sulfonamide XIb showed moderate activity as inhibitor of the growth of leukemia cell lines (K-562, GI = 43.77%; MOLT-4, GI = 44.21%), prostate cancer (PS-3, GI = 40.44%), breast cancer (T-47D, GI = 42.51%); sulfonamide XId showed moderate activity as inhibitor of the growth of leukemia cell lines (RPMI, GI = 48.16%), lung cancer (A549/ATCC, GI = 48.18%), intestines cancer (HCT-116, GI = 31.46%), breast cancer (T-47D, GI = 46.42%) and a significant activity against renal cancer lines (UO-31, GI = -56.83%).

In summary, starting from available amidophenacylating reagents we have developed new methods for the synthesis of 1,3-oxazole- and 1,3thiazole-4-sulfonyl chlorides and sulfonamides, which are potential bioactive compounds.

EXPERIMENTAL

IR spectra were recorded on a Vertex 70 instrument from KBr pellets. ¹H and ¹³C NMR spectra were Bruker AVANCE obtained on а **DRX-500** spectrometer operating at 500 and 125 MHz, respectively, internal reference TMS. LC-MS spectra were registered on an Agilent 1100 Series HPLC chromatograph equipped with a diode array with a mass selective detector Agilent LC\MSD SL. GC-MS analysis parametersare as follows: column Zorbax SB-C18, 1.18 µm, 4.6 × 15 mm (PN 821975-932); acetonitrile-water (95 : 5), 0.1% aqueous trifluoroacetic acid, flow rate of the eluent 3 mL min⁻¹, injection volume 1 µL; UV detecting at 215, 254, 285 nm; chemical ionization at atmospheric pressure (APCI), scan range m/z 80–1000. Melting points were measured on a Fisher-Johns instrument.

Carboxylic acids *N*-(1-benzylsulfanyl-2-oxo-2phenylethyl)amides (IIa–IIc). To a solution of 0.05 mol of compound Ia–Ic [6, 22] in 200 mL of anhydrous acetonitrile were added benzylthiol (0.05 mol) and Et_3N (0.05 mol). The mixture was refluxed and then maintained at 20–25°C for 12 h. To the solution was added 400 mL of water. The precipitate was filtered off, dried, and purified by recrystallization.

2-Aryl-4-benzylsulfanyl-5-phenyl-1,3-oxazoles (IIIa, IIIb). A solution of 0.025 mol of compound IIa or IIb in 30 mL of POCl₃ was refluxed for 5 h, then maintained at 20–25°C for 12 h and poured into ice. The formed precipitate was filtered off, dried, and purified by recrystallization.

2-Aryl-5-phenyl-1,3-oxazole-4-sulfonyl chlorides (IVa, IVb). To a solution of 0.01 mol of compound IIIa or IIIb in 30 mL of 95% acetic acid was bubbled at stirring Cl_2 for 0.5 h maintaining the reaction temperature in the range of 0–5°C. Then the mixture was maintained at this temperature for 12 h and poured into ice. The formed precipitate was filtered off, dried in a vacuum desiccator over phosphorus pentoxide and purified by recrystallization.

2-Aryl-5-phenyl-1,3-oxazole-4-sulfonamides (Va, Vb). To a solution (50 mL) of 25% aqueous ammonia while stirring and cooling with ice water was added dropwise a solution of 0.005 mol of compound **IVa** or **VIb** in 15 mL of anhydrous dioxane. The slurry was stirred at 20–25°C for 0.5 h. The formed precipitate was filtered off, dried, and purified by recrystallization.

2-Aryl-*N*-benzyl(phenyl)-5-phenyl-1,3-oxazole-4sulfonamides (VIa–VId). To a solution of 0.005 mol of sulfonyl chloride IVa or IVb in 15 mL of anhydrous dioxane was added 0.005 mol of the appropriate amine (benzylamine or aniline) and 0.005 mol of Et_3N . The mixture was refluxed for 2 h, and then maintained at 20–25°C for 12 h. The solvent was removed in a vacuum, and the residue was treated with water. The formed precipitate was filtered off, dried, and purified by recrystallization.

2-Aryl-*N*,*N***-dimethyl-5-phenyl-1,3-oxazole-4-sulfonamides (VIIa, VIIb).** To 50 mL of 40% aqueous dimethylamine with stirring and cooling with ice water was added a solution of 0.005 mol of compound **IVa** or **IVb** in 15 mL of anhydrous dioxane. The slurry was stirred at 20–25°C for 0.5 h, and the formed precipitate was filtered off, dried and purified by recrystallization.

2-Aryl-*N*,*N*-R¹R²-5-phenyl-1,3-oxazole-4-sulfonamides (VIIc-VIIf) were prepared similarly to compounds VIa-VId from IVa or IVb and morpholine or piperidine.

2-Aryl(methyl)-4-benzylsulfanyl-5-phenyl-1,3thiazoles (VIIIa–VIIIc). A mixture of 0.025 mol of compound IIa–IIc and 0.025 mol of the Lawesson's reagent in 120 mL of anhydrous dioxane was refluxed for 8 h, and then maintained at 20–25°C for 12 h. The solvent was removed in a vacuum. The residue was treated with 5% aqueous sodium hydroxide solution. The precipitate was filtered off and dried. Compounds **VIIIa, VIIIb** were purified by recrystallization. Compound **VIIIc** was extracted with chloroform, the extract was dried over anhydrous magnesium sulfate and concentrated in a vacuum to afford **VIIIc** which was used in the next steps without further purification.

2-Aryl(methyl)-5-phenyl-1,3-thiazole-4-sulfonyl chlorides (IXa–IXc) were prepared similarly to compounds IVa, IVb from corresponding thiazoles VIIIa–VIIIc.

2-Aryl(methyl)-5-phenyl-1,3-thiazole-4-sulfonamides (Xa–Xc) were prepared similarly to compounds Va, Vb from corresponding sulfonyl chlorides IXa–IXc and aqueous ammonia solution.

2-Aryl(methyl)-*N*-benzyl(phenyl)-5-phenyl-1,3thiazole-4-sulfonamides (XIa–XIf) were prepared similarly to compounds VIa–VId from corresponding sulfonyl chlorides IXa–IXc and benzylamine or aniline.

2-Aryl-*N*,*N*-dimethyl-5-phenyl-1,3-thiazole-4sulfonamides (XIIa, XIIb) were prepared similarly to compounds VIIa, VIIb from corresponding sulfonyl chlorides IXa, IXb and dimethylamine aqueous solution.

2-Aryl(methyl)-*N*,*N*-R¹R²-5-phenyl-1,3-thiazole-4-sulfonamides (XIIc–XIIh) were prepared similarly to compounds VIa–VId from corresponding sulfonyl chlorides IXa–IXc and piperidine or morpholine.

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