IR) and sulphate (ppt. with BaCl₂). ¹³C NMR spectra were recorded at 50.3 MHz using DMSO- d_6 as solvent and TMS as int. standard.

Compound **1.** Yellow crystals; R_f values: BAW (0.28), H_2O (0.89); 15% HOAc (0.77); colours at 366 nm: dark brown (without any treatment), dark yellow (NH₃), fluorescent yellow (AlCl₃), yellow (NA); UV λ_{max}^{MeOH} nm: 268, 283 sh, 350; + NaOMe: 275, 325, 400; + AlCl₃: 267, 295 sh, 350, 400; + AlCl₃ + HCl: 276, sh 295, 350, 400; + NaOAc: 275, 310, 390; + NaOAc + H₃BO₃: 268, 295, 352; IR v_{max}^{KBr} cm⁻¹: 3400, 3250, 1660, 1505, 1400, 1300, 1040 and 805.

spectra and Instituto Nacional de Investigação Científica for financial support (Fm P_3).

REFERENCES

- 1. Harborne, J. B. (1975) Phytochemistry 4. 1331.
- 2. Harborne, J. B. (1975) Phytochemistry 4, 1147.
- Mabry, T. J., Markham, K. R. and Thomas, M. B. (1970) The Systematic Identification of Flavonoids, pp. 44–49. Springer, Berlin.
- 4. Harborne, J. B. and King, L. (1976) Biochem. Syst. Ecol. 4, 111.
- Markham, K. R. (1982) in *Techniques of Flavonoid Identific*ation. Academic Press, London.
- 6. Harborne, J. B. (1965) Phytochemistry 4, 107.

Acknowledgements—We thank Professor Victor Gil, Faculty of Science and Technology, Coimbra University, for ¹³C NMR

Phytochemistry, Vol. 27, No. 9, pp. 3020-3022, 1988. Printed in Great Britain. 0031 9422/88 \$3.00 + 0.00 Pergamon Press plc.

SEMECARPETIN, A BIFLAVANONE FROM SEMECARPUS ANACARDIUM*

S. S. N. MURTHY

Organic Chemistry Laboratories, School of Chemistry, Andhra University, Waltair 530003, India

(Received 20 July 1987)

Key Word Index—Semecarpus anacardium; Anacardiaceae; biflavanones; ¹H NMR and mass spectra; semecarpetin.

Abstract—A new biflavanone, semecarpetin, has been recently isolated from the nut shells of *Semecarpus anacardium*. Its structure has been characterized on the basis of spectral and chemical data as 7"-hydroxy-7,3",4',4"'-tetramethoxybi(8",3') flavanone.

From part A of the acetone soluble fraction [1] of the alcoholic extract of the defatted nut shells of *Semecarpus anacardium* L., three new compounds [1–3] besides the three known biflavanones [4] were reported. Two new dimeric flavanones designated as nallaflavanone and semecarpetin have recently been isolated from part B of the acetone soluble fraction [1]. Structure 1 has already been assigned [5] to nallaflavanone. The present report deals with the structural determination of semecarpetin.

Semecarpetin (2) appeared as bright yellow crystals from chloroform and methanol mixture, $C_{34}H_{30}O_9$, mp 164–65°. It afforded a pink colour with magnesium– hydrochloric acid and a red colour with sodium borohydride-hydrochloric acid characteristic of a flavanone. The UV spectrum in alcohol exhibited maxima at 298 nm which on addition of sodium acetate to the test solution underwent a bathochromic shift (298 \rightarrow 331 nm) while with aluminium chloride no such shift was observed revealing the absence of chelated hydroxyl groups in the biflavonoid. Further, in its ¹H NMR spectrum no low field proton was noticed. The compound showed IR bands at 3430 (hydroxyl group), 2830 (methoxyl groups), 1680 (flavanone carbonyl), 1610 and 1560 (aromatic) cm⁻¹. The above observations clearly revealed that there was at least one 7-hydroxyflavanone system [6, 7] in the molecule.

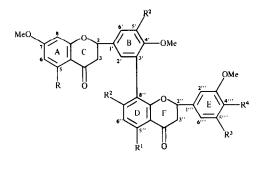
The ¹H NMR spectrum (80 MHz, CDCl₃, TMs as internal standard) of semecarpetin (**2**) displayed signals due to two benzylic methine protons (C-2, F-2") at δ 5.24 (*m*, 2H) and four methylene protons (C-3, F-3") at 2.86 (2H, *br*, *cis*-protons) and 3.12 (2H, *m*, *trans*-protons). The three protons corresponding to 5, 6 and 8 positions of ring A were observed respectively at δ 7.66 (1H, *d*, *J* = 8 Hz), 6.24 (1H, *dd*, *J* = 2,8 Hz), and 6.34 (1H, *d*, *J* = 2 Hz). There are two unresolvable multiplets between δ 6.68 and 6.92, and 7.12 and 7.34 integrating each for three protons; the former multiplet corresponded to 2", 5" and 6" protons of ring B. The two *ortho*-coupled protons corresponding to 5" and 6" positions of ring D were noticed respectively at δ 7.72 (1H, *d*, *J* = 8 Hz) and 6.56 (1H, *d*, *J* = 8 Hz). Further, the ⁻¹H NMR spectrum showed the

^{*}Part 10 in the series 'Naturally Occurring Biflavonoid Derivatives' For part 9, see ref. [5].

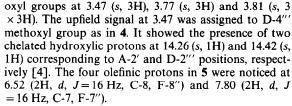
presence of one non-chelated D_2O exchangeable hydroxylic proton at 8.22 (s) and four methoxyl groups at 3.70 (s, 3H) and 3.76 (s, $3 \times 3H$) and these could be ascribed respectively to D-7", A-7, B-4', E-3"', 4"' positions.

All the nine oxygen atoms in 2 are accounted for by the four pyranone oxygen atoms, one non-chelated hydroxyl group and four methoxyl groups. Hence the diaryl linkage between the two flavanone units must be through a C-C linkage only. Oxidation of semecarpetin with neutral permanganate gave only one mole of veratric acid (identified by mmp and IR) suggesting that one of the side-phenyl rings is involved in the interflavonoid linkage.

On acetylation with acetic anhydride and pyridine on a steam bath for one hour, semecarpetin afforded an acetate (3), $C_{36}H_{32}O_{10}$, mp 126° whose ¹H NMR spectrum (CDCl₃) showed one acetoxyl group at 2.21 corresponding to D-7" position. Methylation of semecarpetin with diazomethane gave a monomethyl ether (4), $C_{35}H_{32}O_9$, mp 138° while with dimethyl sulphate and potassium carbonate in acetone under reflux, it furnished a bichalcone pentamethyl ether (5), $C_{35}H_{32}O_9$, mp 214–215°. The ¹H NMR spectrum of 4 (CDCl₃) indicated the presence of five methoxyl groups at 3.43 (s, 3H), 3.73 (s, 3H) and 3.79 (s, $3 \times 3H$). The upfield methoxy signal at 3.43 was assigned to the D-7" methoxyl group by comparison with glaucine [8]. Further, 5 showed in its ¹H NMR spectrum (CDCl₃), the presence of five meth-

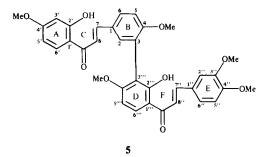


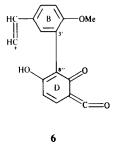
- **1** $R^1 = OH$, $R^2 = R^3 = OMe$, $R^4 = OH$ **2** $R^1 = R^3 = H$, $R^2 = OH$, $R^4 = OMe$
- **2** $R^1 = R^3 = H$, $R^2 = OH$, $R^4 = OMe$ **3** $R^1 = R^3 = H$, $R^2 = OAc$, $R^4 = OMe$
- **3** $R^1 = R^3 = H$, $R^2 = OAc$, $R^4 = G$ **4** $R^1 = R^3 = H$, $R^2 = R^4 = OMe$
- **4** $R^{2} = R^{2} H, R^{2} R^{4} Ome$

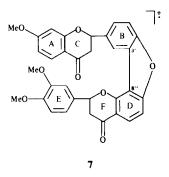


When the bichalcone pentamethyl ether (5) was oxidized with neutral permanganate, both veratric acid and 2-hydroxy-4-methoxy benzoic acid (identified by mmp and IR) were obtained. Hence, the two remaining methoxyl groups must be in the diaryl system which are assigned by analogy and ¹H NMR spectral data at the B-4 and D-4"' positions (chalcone numbering). Consequently the biflavonoid linkage in 5 must be either at the B-3-D-3"' position (B-3'-D-8") or at the B-3-D-5"' position (B-3'-D-6'') flavonoid numbering in the parenthesis). Since semecarpetin contains no chelated hydroxyl groups, the positions at A-5 and D-5" are free. Further the ¹H NMR spectra of the parent compound 2 and its three derivatives (3-5) clearly indicate the presence of two ortho-coupled protons which must correspond to ring D. On the basis of this observation, the C-C linkage at the B-3'-D-6" position can be eliminated and hence semecarpetin must have the biflavonoid linkage at the B-3'-D-8" position. This is also in good agreement with the biflavanones already reported from this plant [1-5].

Semecarpetin (2) in its mass spectrum displayed the molecular ion (M⁺) at m/z 582 (12%). The peak at m/z163 $[3,4-(OMe)_2-C_6H_3-CH=C^+H; 71\%]$ not only revealed that the rings E and F do not carry the biflavonoid linkage but also indicated that the 3" and 4" positions of ring E were occupied by the two methoxyl groups. The central fragment at m/z 267 (19%) corresponding to the ion 6 was formed after two RDA fragmentations. Compound 4 in its mass spectrum showed a peak at m/z 550 (6%) which is formed by the loss of 46 mass units. This fragment can be formulated as 7 in which the orthomethoxyl groups to the biphenyl linkage cyclize to a furan ring [9]. Perhaps the most significant feature is the formation of the two fragments 6 and 7 which could be diagnostic of the biphenyl linkage. It may be mentioned here that similar fragments have been reported in morelloflavone heptamethyl ether [10]. Overall the mass spectral fragmentation pattern of semecarpetin exhibits a close similarity with that of GB-2 [11]. Thus semecarpetin has structure 2.







Acknowledgements I am grateful to Professor L. R. Row for encouragement. My thanks are also due to Professor L. S. A. Dikshitulu, Drs Ch. Prabhakar and D. R. K. Sastry for the spectral data.

REFERENCES

- 1. Murthy, S. S. N. (1985) Phytochemistry 24, 1065.
- 2. Murthy, S. S. N. (1983) Phytochemistry 22, 1518.
- 3. Murthy, S. S. N. (1985) Indian J. Chem. 24B, 398.
- 4. Prakasa Rao, N. S., Ramachandra Row, L. and Brown, R. T. (1973) *Phytochemistry* 12, 671.

- Murthy, S. S. N. (1987) Proc. Indian Natl. Sci. Acad. 53A, 632.
- 6. Horowitz, R. M. and Jurd, L. (1961) J. Org. Chem. 26, 2446.
- Mabry, T. J., Markham, K. R. and Thomas, M. B. (1970) The Systematic Identification of Flavonoids, pp. 165–171. Springer, New York.
- Bick, I. R. C., Harley-Mason, J, Sheppord, N. and Vernengo, M. J. (1961) J. Chem. Soc., 1896.
- Natarajan, S., Murti, V. V. S. and Seshadri, T. R. (1969) Indian J. Chem. 7, 751.
- Karanjgaokar, C. G., Radhakrishnan, P. V. and Venkataraman, K. (1967) *Tetrahedron Letters* 3195.
- 11. Jackson, B., Locksley, H. D., Scheinmann, F. and Wolstenholme, W. A. (1971) J. Chem. Soc. (C) 3791.

Phytochemistry, Vol. 27, No. 9, pp. 3022-3024, 1988. Printed in Great Britain. 0031 9422/88 \$3.00 + 0.00 © 1988 Pergamon Press plc.

EMEHETERONE, A PYRAZINONE DERIVATIVE FROM *EMERICELLA HE-TEROTHALLICA**

NOBUO KAWAHARA, KOOHEI NOZAWA, SOICHI NAKAJIMA and KEN-ICHI KAWAI

Faculty of Pharmaceutical Sciences, Hoshi University, Ebara 2-4-41, Shinagawa-ku, Tokyo 142, Japan

(Received 8 December 1987)

Key Word Index-Emericella hetrothallica; Eurotiaceae; 2(1H)-Pyrazinone; emeheterone; stellatin.

Abstract—Emeheterone, a novel pyrazinone derivative, has been isolated from the culture filtrate of the fungus *Emericella heterothallica*, along with stellatin. Its molecular structure has been investigated by spectroscopic means.

INTRODUCTION

Recently the antifungal epidithiodioxopiperazines, emestrin [2] and dithiosilvatin [3], were isolated from *Emericella striata* (Rai, Tewari & Mukerji) Malloch & Cain and *Aspergillus silvaticus* Fennell & Raper, respectively. In the course of screening for dioxopiperazine derivatives from *Emericella* spp., a novel pyrazinone derivative designated emeheterone (1) was isolated from the dichloromethane extract of the culture filtrate of *Emericella heterothallica* (Kwon, Fennell & Raper) Malloch & Cain (anamorph: *Aspergillus heterothallicus* Kwon, Fennell & Raper) (mating type a), strain ATCC 16824, along with a dihydroisocoumarin, stellatin (2).

RESULTS AND DISCUSSION

Emeheterone (1), mp 215–217°, gave molecular ions at m/z 322 in EIMS and at m/z 323 in CIMS, and elemental analysis confirmed the molecular formula as

 $C_{19}H_{18}N_2O_3$. The strong ion at m/z 91 $[C_6H_5CH_2]^+$ in the EIMS suggested the presence of benzyl groups in the molecule of 1. The ¹H NMR signals at δ 7.20–7.40 (10H), 3.93 (2H), and 4.20 (2H), the 10 ¹³C NMR signals at δ 126–137 and the two ¹³C NMR signals at δ 30.39 and 34.01 were assigned to two benzyl groups. The other ¹H NMR signals at δ 3.92 (3H) and 3.73 (1H) in 1 were assigned to a methoxy group, which appeared at δ 61.73 in the ¹³C NMR spectrum, and a hydroxyl group, respectively.

The IR absorption maximum at 1650 cm⁻¹ and the ¹³C NMR signal at δ 158.14 (St) in emeheterone (1) suggested the presence of a conjugated carbonyl, probably an amide. Compound 1 gave a positive coloration (greenish brown) with 0.5% aq. copper chloride [4], which suggested the presence of an oxime-like structure. The UV absorption maxima at 228, 276, 334 (sh), and 352 nm of 1 were closely similar to those at 225, 265, 279 (sh), 347 (sh), and 364 nm for deferriastechrome **3** [5], which also has a maximum at 291 nm due to the indole moiety. The above results are consistent with structure 1 for emeheterone, but do not suggest the orientation of the methoxy and hydroxyl groups located at N-1 and C-5.

^{*} Part 20 in the series 'Studies on Fungal Products'.For Part 19 see ref. [1].