PHOTODIMERIZATION OF 3,3-BIPHENYLENE-1-BROMO-1-PHENYLALLENE

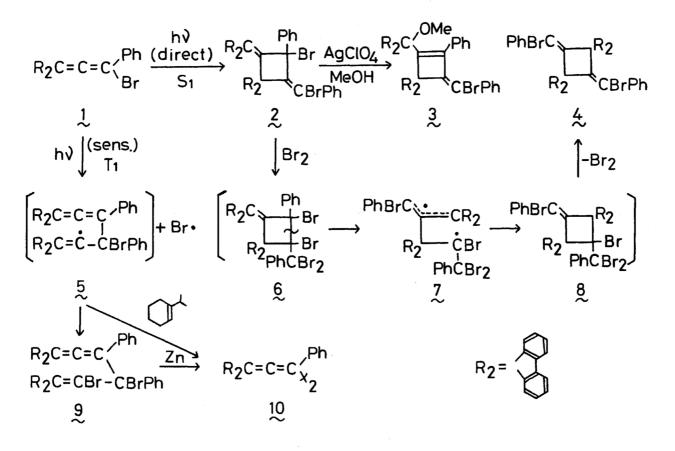
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Photodimerization of the title allene afforded 1,1-biphenylene-3-bromo-2bromobenzylidene-4-fluoren-9-ylidene-3-phenylcyclobutane $(\underline{2})$, whereas the photodimerization in the presence of fluorenone afforded 1,1,6,6-bis(biphenylene)-4,5-dibromo-3,4-diphenylhexa-1,2,5-triene (9).

Although thermal dimerization of allenes have been extensively studied,¹⁾ photodimerization of allenes has yet been reported.²⁾ We found that the photodimerization of the title allene (<u>1</u>) affords 1,3-dimethylenecyclobutane and hexa-1,2,5-triene, and that the photodimerization contrasts with thermal dimerization of allene which affords 1,2-dimethylenecyclobutane derivative.

A solution of <u>1</u> in benzene was irradiated³⁾ for 1.5 hr to afford 1,1-biphenylene-3-bromo-2-bromobenzylidene-4-fluoren-9-ylidene-3-phenylcyclobutane (2, 30%, mp 233-234°C, λ_{max} 372 nm (ϵ , 5,100), m/e 690 (M⁺, rel intensity 2), 611 (M⁺-Br, 12), 530 (M⁺-Br₂, 100), 453 (530-Ph, 25)) and 1,1,6,6-bis(biphenylene)-4,5-dibromo-3,4-diphenylhexa-1,2,5-triene (2, 10%, mp 181-181.5°C, λ_{max} 313 nm (ϵ , 10,200), ϑ_{max} 1930 cm⁻¹). The structures of these photodimers were elucidated on the basis of the spectral data shown in parentheses and of the following reactions. The reaction of <u>2</u> with AgClO₄ in THF-MeOH afforded 3,3-biphenylene-4-bromobenzylidene-2-(9-methoxyfluoren-9-yl)-1-phenylcyclobutene (<u>3</u>, mp 242-242.5°C, λ_{max} 352 nm (ϵ , 13,600), ϑ_{max} 1100 cm⁻¹, nmr (CDCl₃): 2.0-3.2 (m, 26H), 7.80 τ (s, 3H)). The debromination of <u>9</u> with Zn-powder afforded <u>10</u> (70%, mp 173.5-174°C).⁴) When <u>2</u> was treated with Br₂ in CHCl₃ at 40°C for 10 min, <u>2</u> was isomerized into 1,1,3,3-bis(biphenylene)-2,4-bis(bromobenzylidene)cyclobutane (<u>4</u>, mp 257-258°C, λ_{max} 357 nm (ϵ , 14,800), m/e 690 (M⁺)) in a quantitative yield. A plausible pathway of the rearrangement is as follows: Ring cleavage of <u>6</u> which formed initially by the addition of Br₂ to <u>2</u>, affords biradical intermediate (<u>7</u>). Debromination of <u>8</u> which produced by ring closure of <u>7</u> finally affords <u>4</u>.

When the photodimerization of $\underline{1}$ was carried out in the presence of fluorenone for 1 hr, $\underline{9}$ (46%) and $\underline{10}$ (5%) were obtained, but not any cyclic dimer. This reaction can be well interpreted by



assuming radical (5) as an intermediate. Recombination of 5 with Br• affords 9, and elimination of Br• from 9 affords 10. Because the photodimerization of 1 in the presence of 1-isopropylcyclohexene as a radical scavenger afforded 10 as the major product (40%) in addition to a small amount of 9 (10%), the above interpretation is not unreasonable.

References

- 1) See, e.g., D. R. Taylor, Chem. Rev., <u>67</u>, 342 (1967).
- 2) The only reported example in the photodimerization of cumulene is limited to the case of tetraphenylbutatriene in the solid state which affords 1,3-dialkenylidenecyclobutane. Z. Berkovitch-Yellin, M. Lahav, and L. Leiserowitz, J. Amer. Chem. Soc., <u>96</u>, 918 (1974).
- 3) Irradiation was carried out with a 400W high-pressure mercury lamp through a Pyrex filter.
- 4) F. Toda and Y. Takehira, Chem. Commun., in press.
- 5) The permanganate oxidation of 4 afforded benzoic acid and 2,2,4,4-bis(biphenylene)cyclobutane-1,3-dione.

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