

A Facile Synthesis of 2,3-Disubstituted 4-Oxo-3,4-dihydroquinazolines

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Several substituted 4-oxo-3,4-dihydroquinazolines [4(3*H*)-quinazolinones] are known to possess biological activities. In addition, the 4-quinazolinone moiety is found in several quinazoline alkaloids.

Different methods are known for the synthesis of substituted 4(3*H*)-quinazolinones. However, the largest number of 2,3-disubstituted 4(3*H*)-quinazolinones (**4**) have been synthesized by reacting an *N*-acylanthranilic acids with a primary amine in a suitable solvent in the presence of a catalyst, by reacting 3,1,4-benzoxazones with amines, or by thermal cyclization of *o*-acylaminobenzamides¹. In our previous communication², we reported the synthesis of 2-methyl-3-(2-methylphenyl)-4-oxo-3,4-dihydroquinazoline (**4d**) by refluxing *N*-acetylanthranilic acid with *o*-toluidine in a high-boiling solvent as bromobenzene.

We now describe the facile synthesis of 2,3-disubstituted 4-oxo-3,4-dihydroquinazolines (**4**) by the fusion of equimolar amounts of *N*-acylanthranilic acids (**1**) with primary amino compounds (**2**) such as amines, hydrazine and derivatives, semicarbazide, and thiosemicarbazide at 150–190° for 30–60 min. The yields are good. The structure of products **4** was confirmed by mixed melting points with authentic specimen, comparison of T.L.C. and spectral data, and (for new compounds) microanalyses. A further proof of the structure was obtained by conversion of compounds **4a–d** to the 2-styryl derivatives (**6**) by refluxing with substituted

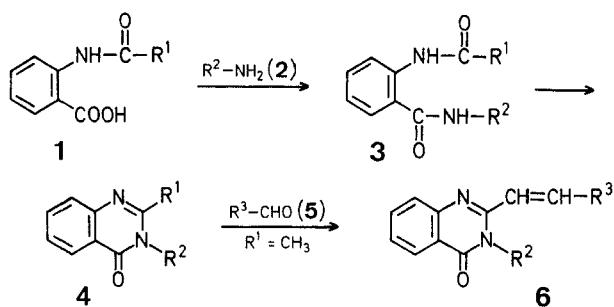












Table 1. 2,3-Disubstituted 4-Oxo-3,4-dihydroquinazolines (**4**)

4	R ¹	R ²	Yield [%]	m.p.	Molecular formula ^a or Lit. m.p.
a	CH ₃	-CH ₂ -COOH	57	259° (dec)	263° (dec) ³
b	CH ₃		54	248° (dec)	246–247° ⁴
c	CH ₃		53	276–277°	276° ⁴
d	CH ₃		60	112.5–113°	112.5–113° ²
e	CH ₃	-NH ₂	46	148–149°	148–149° ⁵
f	C ₂ H ₅	-NH ₂	52	152–153° ^b	C ₁₀ H ₁₁ N ₃ O ₆ (189.2)
g	CH ₃		55	231–232°	231–232° ⁶
h	CH ₃		75	183–185°	183–184° ⁶
i	C ₂ H ₅		55	280–281° ^b	C ₁₁ H ₁₂ N ₄ OS (248.3)
j	CH ₃		56	237–238° ^c	C ₁₆ H ₁₃ N ₃ O ₃ (295.3)
k	CH ₃		52	210–212°	210–212° ⁷
l	CH ₃		50	214–216°	214–216° ⁷
m	CH ₃	-NH- 	76	254–255° ^d	C ₁₃ H ₁₄ N ₄ O ₃ S (330.3)

^a The microanalyses of the new compounds were in satisfactory agreement with the calculated values: C, ± 0.40 ; H, ± 0.20 ; N, ± 0.40 ; S (**4m**), -0.10 .

^b Recrystallized from ethanol.








^c Recrystallized from aqueous ethanol.

^d Recrystallized from aqueous acetic acid.

M.S.: $m/e = 330$ (M^+ , 100%).

¹H-N.M.R. (DMSO- d_6): $\delta = 2.40$ (s, 3H); 6.6–8.5 (m, H_{arom}); 9.40 ppm (s, NH).

Table 2. 4-Oxo-2-styryl-3,4-dihydroquinazolines (**6**)

6	R ²	R ³	Yield [%]	m.p. (solvent)	Molecular formula ^a
a	-CH ₂ -COOH		36	209–211° (H ₂ O/CH ₃ OH)	C ₁₉ H ₁₆ N ₂ O ₄ (336.3)
b			67	206–207° (CH ₃ OH)	C ₂₄ H ₁₈ N ₂ O ₅ (414.4)
c			60	287–288° (AcOH)	C ₂₃ H ₁₆ N ₂ O ₃ (368.4)
d			52	161–162° (CH ₃ OH)	C ₂₃ H ₁₈ N ₂ O (338.4)

^a The microanalyses were in agreement with the calculated values.

benzaldehydes (**5**) in glacial acetic acid in the presence of concentrated sulfuric acid.

The I.R. (nujol) spectra of compounds **4** showed the C=O absorption at 1670–1680 cm^{-1} , whereas the 2-styryl derivatives **6** showed this absorption at 1650 cm^{-1} . An additional carbonyl band appeared at 1700 cm^{-1} in spectra of the acids **6a** and **6c**.

The assumption that the cyclocondensation proceeds via the *o*-acylaminobenzamides **3** as intermediates was proven by the thermal cyclization of separately prepared **3** to give the quinazoline derivatives **4**.

All melting points were determined in open glass capillaries and are uncorrected.

2,3-Disubstituted 4-Oxo-3,4-dihydroquinazolines (**4**); General Procedure:

An intimate mixture of the *N*-acylanthranilic acid **1** (0.01 mol) and the primary amine, hydrazine, acyl hydrazide, semicarbazide hydrochloride, or thiosemicarbazide (**2**; 0.01 mol) is fused at 150–190° for 30–60 min; the crude product obtained after cooling is recrystallized from a suitable solvent.

3-Substituted 4-Oxo-2-styryl-3,4-dihydroquinazolines (**6a–d**); General Procedure:

A solution of the 3-substituted 2-methyl-4-oxo-3,4-dihydroquinazoline (**4a–d**; 5 mmol) and a substituted benzaldehyde (**5**; 5 mmol) in glacial acid (25 ml) containing concentrated sulfuric acid (5 drops) is refluxed for 3 h and then allowed to cool. The crystalline product is isolated by suction and recrystallized from a suitable solvent.

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¹ W. L. F. Armarego, in: A. Weissberger, *The Chemistry of Heterocyclic Compounds, Fused Pyrimidines Part I, Quinazolinones*, Interscience Publishers, New York, 1967, p. 74.

² F. S. G. Soliman, R. M. Shafik, E. A. Elnenaey, *J. Pharm. Sci.* **67**, 411 (1978).

³ B. R. Baker, M. V. Querry, R. E. Schaub, J. H. Williams, *J. Org. Chem.* **17**, 58 (1952).

G. Serventi, R. Marziesi, *Boll. Sci. Fac. Chim. Ind. Bologna* **15**, 117 (1957); *C. A.* **52**, 9147 (1958).

⁵ D. J. Anderson, T. L. Gilchrist, D. C. Horwell, C. W. Rees, *J. Chem. Soc. [C]* **1970**, 576.

⁶ W. S. Dighe, S. Sonasekhara, G. Bagavant, S. L. Mukherjee, *Current Sci. (India)* **33**, 78 (1964); *C. A.* **60**, 10681 (1964).

⁷ J. Klosa, *J. prakt. Chem.* [4] **31**, 140 (1966).