February 1996 SYNTHESIS 209

A Short Asymmetric Synthesis of Both Enantiomers of Ramulosin and Its Analogues

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Received 6 September 1995

(+)-Ramulosin, a metabolite of *Pestalotia ramulosa*, and its enantiomer have been synthesized in high enantiomeric purity (ee $\geq 98\%$) via Michael addition of metalated acetone SAMP-hydrazone (S)-1a to the acetal protected cyclic enoate 2 and subsequent diastereoselective reductive lactonization with L-Selectride. 3-Substituted hexahydroisocoumarines (3S,4aR)-5, analogues of ramulosin, were also obtained with excellent stereoselectivities (de $\geq 98\%$, ee $\geq 96\%$) using this method.

(+)-Ramulosin was first isolated from the fungus *Pestalotia ramulosa* by Stodola et al. in the early sixties and inhibits the germination of seeds and spores of microorganisms.^{1,2} It is the simplest member of a class of biogenetically related δ-lactone antibiotics. The (3R,4aS)-configuration of (+)-ramulosin was proposed by Tanenbaum et al.³ and by Findlay et al.⁴ Three syntheses of the racemate,⁵ two 'ex-chiral pool' syntheses starting from ethyl (R)-3-hydroxybutanoate⁶ and (R)-O-benzylglycidol,⁷ respectively, and one enantioselective synthesis of ramulosin have been reported so far. We now wish to describe an alternative and very short asymmetric synthesis of (+)-ramulosin and its enantiomer, as well as of some analogues employing the SAMP/RAMP-hydrazone method.⁹

(+)-(3R, 4aS)-ramulosin

As depicted in the Scheme, the methyl ketone SAMPhydrazones (S)-1, readily available from the corresponding ketones and (S)-1-amino-2-methoxymethyl pyrrolidine (SAMP), were metalated with lithium diisopropylamide in tetrahydrofuran at 0°C and N,N,N',N'-tetramethylethylenediamine (TMEDA) was added at -78 °C to the azaenolate generated. Subsequent reaction with the protected cyclic β -keto enoate 2 in the presence of BF₃ · OEt₂ resulted in a clean 1,4-addition and the hydrazone Michael adducts (1R,6R,2'S)-3 could be obtained in good to excellent yield after purification by distillation (Tables 1 and 4). The acetal protection of the β -carbonyl group of the Michael acceptor 2 turned out to be essential. presumably because of a more favourable γ-deprotonation of the parent β -keto esters by the lithiated hydrazones instead of 1,4-addition and thus resulting in the isolation of only starting material after aqueous workup. Furthermore, to achieve conjugate addition the activation of the Michael acceptor with BF₃ · OEt₂ was also essential.

Scheme

For removal of the chiral auxiliary and the acetal protecting group the crude hydrazones 3 were dissolved in diethyl ether and treated with 6N HCl to effect acidic hydrolysis. The 6-substituted 2-hydroxycyclohex-1-enecarboxylates (R)-4 were obtained in practically quantitative yields as a mixture of keto/enol tautomers (NMR) and with excellent enantiomeric excess (ee $\geq 96\%$) (Tables 2 and 5). In the final step of our total synthesis of ramulosin and related 3-substituted 8-hydroxyhexahydroisocoumarines (3S,4aR)-5, the keto esters (R)-4 were diastereoselectively reduced with L-Selectride in tetrahydrofuran at -78 °C which, after aqueous quench, caused directly lactonization. (-)-Ramulosin [(3S,4aR)-5a] and its analogues 5b-e were obtained in moderate yield, but with excellent diastereomeric (de \geq 98 %) and enantiomeric excess (ee $\geq 96 - \geq 98\%$) (Tables 3 and 6). For the total synthesis of the naturally occurring (+)-ramulosin RAMP had to be used instead of SAMP as chiral auxiliary. The analytical and spectroscopic data were identical with those reported in the literature (Table 3, footnotes).

The enantiomeric excesses of (R)-4 and (3S,4aR)-5 were determined by ¹H NMR shift experiments with the chiral cosolvent (-)-(R)-9-anthryl-2,2,2-trifluoroethanol or more accurately in the case of 5a and 5b by GC analysis on a chiral stationary phase [heptakis(2,3,6-tri-O-meth-

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Table 1. SAMP-Hydrazones 3 Prepared by 1,4-Addition

Product	R	Yield (%)	$[\alpha]_{D}^{r.t.}$ (c, CHCl_3)
(1R, 6R, 2'S)-3a	Me	90	+ 86.39 (1.47)
(1R, 6R, 2'S)-3b	Et	92	+67.33(1.01)
(1R, 6R, 2'S)-3c	<i>i</i> -Pr	74	+43.31(1.57)
(1R, 6R, 2'S)-3d	Bu	81	+79.5(1.61)
(1R, 6R, 2'S)-3e	<i>i</i> -Bu	80	+ 46.81 (0.94)

Table 2. 6-Substituted 2-Hydroxycyclohex-1-enecarboxylates (R)-4

Product	R	Yield (%)	[α] _D ^{r.t.} (c, CHCl ₃)	ee ^a (%)
(R)-4a (S)-4a ^b (R)-4b (R)-4c (R)-4d (R)-4e	Me Me Et i-Pr Bu i-Bu	100 100 100 96 100 100	- 8.1 (0.86) + 8.8 (0.91) - 8.6 (1.16) - 4.5 (1.10) - 10.3 (0.39) - 11.9 (1.18)	≥ 96 ≥ 96 ≥ 96 ≥ 96 ≥ 96 ≥ 96 ≥ 96

^a Determined by ¹H NMR shift experiments with (-)-(R)-(9-anthryl)-2,2,2-trifluoroethanol.

Table 3. 3-Substituted 8-Hydroxy-3,4,4a,5,6,7-hexahydroisocoumarines **5**

Product	R	Yield (%)		$[\alpha]_{D}^{r.t.}$ (c, EtOH)	de ^b (%)	ee (%)
(3S,4aR)-5a (3R,4aS)-5a ^e (3S,4aR)-5b (3R,4aR)-5c (3S,4aR)-5d (3S,4aR)-5e	Me Me Et i-Pr n-Bu i-Bu		117-118 117-118 ^f 87-88 98-99 98 101	$\begin{array}{l} -14.8 \ (1.01)^{c} \\ +18.2 \ (0.66)^{g} \\ -14.3 \ (0.63) \\ -13.6 \ (0.22) \\ -24.1 \ (0.29) \\ -9.5 \ (0.21) \end{array}$		$\geq 96^{d}$ $\geq 98^{d}$ $\geq 96^{d}$ $\geq 96^{h}$ $\geq 96^{h}$ $\geq 96^{h}$

^a Uncorrected, measured on a Büchi apparatus (Dr. Tottoli).

yl)-β-cyclodextrin in OV 1701]. The racemic keto esters 4 required for comparison were prepared via the corresponding dimethylhydrazone homocuprates according to the method previously described. The de values of the final products 5 were determined by H and T3C NMR spectroscopy.

The assignment of the relative and absolute configurations presented in the Scheme are not only based on the polarimetric data of **5a**, but also on the unambiguously assigned configuration (X-ray structure analysis) of related compounds prepared by MIRC reactions. The (R)-configuration of the stereogenic center generated in the first step of the asymmetric Michael addition is in agreement with previous results of 1,4-additions via metalated SAMP-hydrazones and the postulated mechanism of electrophilic substitutions with SAMP/RAMP-hydrazones. 9

Another possibility to synthesize the hydrazones (1R,6R,2'S)-3 is by using a Michael initiated ring closure (MIRC) reaction of metalated SAMP-hydrazones with ethyl 6-ethoxycarbonylhex-2-enoate, but this procedure gave lower yields of hydrazones $3.^{13}$

In summary, a very short, highly diastereo- and enantioselective synthesis of (+)-ramulosin and its enantiomer employing the SAMP/RAMP-hydrazone method has been developed. Key steps of the synthesis are the Michael addition of metalated methyl ketone SAMP-hydrazones to the protected cyclic enoate 2 and the diastereoselective reductive lactonization with L-Selectride. The new method offers an efficient entry to a variety of 3-substituted hexahydroisocoumarines and 6-substituted 2-hydroxycyclohex-1-enecarboxylates¹⁴ with high asymmetric induction, both valuable classes of compounds in natural product synthesis.

Solvents were dried and purified according to literature procedures prior to use. All reagents were of commercial quality from freshly opened containers or distilled before use. THF was freshly distilled from potassium under argon. Light petroleum refers to the fraction with bp 40-80°C. Analytical TLC plates (silica gel 60 F₂₅₄) and silica gel 60 (230-400 mesh) were purchased from Merck, Darmstadt. n-BuLi (1.6 N in n-hexane) and L-Selectride was purchased from Aldrich. All melting points (Büchi apparatus, system Dr. Tottoli) are uncorrected. Optical rotation values were measured using a Perkin-Elmer P 241 polarimeter. Microanalyses were obtained with a Heraeus CHN-O-RAPID element analyzer. MS were recorded on a Varian MAT 212 (70 eV, 1 mA) spectrometer with DEI ionisation. IR spectra were obtained using a Perkin-Elmer FT 1750 spectrometer. The ¹H and ¹³C NMR spectra were measured on a Varian VXR 300 (300 and 75 MHz) and Gemini 300 (300 and 75 MHz) using TMS as internal standard. (S)-1-amino-2-methoxymethylpyrrolidine (SAMP), (R)-1-amino-2-methoxymethylpyrrolidine (RAMP),15 the ketone SAMP/RAMP-hydrazones and the Michael acceptor16 were synthesized according to literature proce-

Ethyl (1R,6R)-2,2-(Ethylenedioxy)-6-[2-[[2'S')-2'-(methoxymethyl)-pyrrolidino|imino|alkyl|cyclohexanecarboxylate; SAMP-Hydrazone (1R,6R,2'S)-3; General Procedure:

A solution of 1.6 M BuLi in hexane (2.2 equiv) was added dropwise to a solution of i-Pr₂NH (2.2 equiv) in THF (typically 10 mL/mmol) under Ar at 0°C. The mixture was stirred to generate a clear and colorless solution of LDA (2.2 equiv). After dropwise addition of SAMP- or RAMP-hydrazone (S)- or (R)-1 (1.8 equiv) the mixture was stirred for 4 h to obtain a colored solution or suspension of the nucleophile. The mixture was subsequently cooled to $-78\,^{\circ}\mathrm{C}$ and TMEDA (5 equiv) was added. Stirring was continued for an additional 0.5 h. Then a solution of the Michael acceptor 2 (1 equiv) and BF₃·OEt₂ (1 equiv) in THF (5 mL) was added. The reaction mixture was stirred ca. 12 h at -78 °C. After addition of sat. NH₄Cl (10 mL), the reaction mixture was stirred 20 min at r.t. Then water (10 mL) and Et₂O (50 mL) were added and the phases were separated. The aqueous phase was extracted with Et₂O $(2 \times 50 \text{ mL})$. The collected organic phases were washed with brine (10 mL), dried (MgSO₄), concentrated in vacuo and the residue was purified by distillation.

b RAMP was used as auxiliary.

^b Determined by ¹H and ¹³C NMR spectroscopy.

c Relatively large deviation, reason not clear.

^d Determined by GC analysis on a chiral stationary phase.

RAMP was used as chiral auxiliary.

f Ref⁸ mp 118–119°C, ref⁶ 118–119°C; ref¹ 120–121°C and ref⁷ 121–122°C.

⁸ Ref⁸ $[\alpha]_D^{22}$ 17.1 (c=0.56, EtOH); ref.⁶ $[\alpha]_D^{22}$ 18.2 (c=1.15, EtOH); ref.¹ $[\alpha]_D^{25}$ 18 \pm 2 (c=2.90, EtOH); ref.⁷ $[\alpha]_D^{26}$ 18.1 (c=1.03, EtOH).

^h Determined by ¹H NMR shift experiments with (−)-(R)-(9-anthryl)-2,2,2-trifluoroethanol.

Table 4. Spectroscopic Data of the SAMP Hydrazones 3^a

Compound	IR (CHCl ₃) v (cm ⁻¹)	1 H NMR (CDCl $_{3}$ /TMS) δ , J (Hz)	13 C NMR (CDCl $_3$ /TMS) δ	MS (70 eV) m/z (%)
3a	2940, 2870, 1730, 1680, 1674, 1448, 1369, 1245, 1203, 1181, 1097, 1059, 1034	1.30 (t, J = 7.0, 3 H, CH ₂ CH ₃), 1.40–2.40 [m, 17 H, CH ₂ CH ₂ CH ₂ C(O) ₂ , CH ₂ CN, CHCH ₂ CN, CHCO ₂ , NCH ₂ CH ₂ , CH ₃], 3.34 (s, 3 H, OCH ₃), 2.9–3.5 (m, 6 H, OCH ₂ CH ₂ O,NCH ₂), 3.6–3.85 (m, 2 H, NCH, CHHO), 3.98 (m, 1 H, CHHO), 4.2 (q, J = 7.1, 2 H, CH ₂ CH ₃)	14.3 (CH ₂ CH ₃), 17.7 (CH ₃), 21.6 (NCH ₂ CH ₂), 26.4 (NCHCH ₂), 27.1 (CH ₂ CH), 31.7 (CHCH ₂ CN), 31.8, 34.5, 36.0 [CH ₂ CH ₂ C(O) ₂ , CH ₂ CN], 54.8 (NCH ₂), 59.0 (OCH ₃), 59.2 (CHCO ₂), 60.9 (CH ₂ CH ₃), 65.6, 66.1 (OCH ₂ CH ₂ O), 68.1 (NCH), 75.6 (CH ₂ O), 107.1 [C(O) ₂], 162.8 (CN), 167.7 (CO ₂)	383 (M ⁺⁺ +1, 0.3), 382 (M ⁺⁺ , 1), 337 (12), 197 (17), 123 (100), 111 (10), 87 (17), 83 (11), 73 (100), 70 (35), 61 (61), 60 (12)
3b	2940, 2864, 1697, 1630, 1460, 1448, 1371, 1244, 1216, 1181, 1107, 1057, 926, 666, 534, 475	0.9 (t, $J=7.4$, 3 H, CNCH ₂ CH ₃), 1.3 (t, $J=7.0$, 3 H, CH ₂ CH ₃), 1.0–2.5 [m, 16 H, CH ₂ CH ₂ CH ₂ C(O) ₂ , CH ₂ CN, CHCH ₂ CN, CHCO ₂ , NCH ₂ CH ₂ CH ₂ C, CNCH ₂ CH ₃], 3.33 (s, 3 H, OCH ₃), 2.9–3.45 (m, 6 H, OCH ₂ CH ₂ O, NCH ₂), 3.7–3.84 (m, 2 H, NCH, CHHO), 3.98 (m, 1 H, CHHO), 4.23 (q, $J=7.0$, 2 H, CH ₂ CH ₃)	12.2 (CNCH ₂ CH ₃), 14.3 (CH ₂ CH ₃), 22.1 (CNCH ₂ CH ₃), 25.3 (NCH ₂ CH ₂), 26.7 (NCHCH ₂), 27.2 (CH ₂ CH), 31.6 (CHCH ₂ CN), 32.2 [CH ₂ CH ₂ C(O) ₂], 33.5 [CH ₂ CH ₂ C(O) ₂], 35.1 (CH ₂ CN), 54.8 (NCH ₂), 58.9 (CHCO ₂), 59.2 (OCH ₃), 60.9 (CH ₂ CH ₃), 65.6, 65.7 (OCH ₂ CH ₂ O), 66.1 (NCH), 75.6 (CH ₂ O), 108.6 [C(O) ₂], 163.4 (CN), 167.6 (CO ₂)	397 (M** + 1, 6), 396 (M**, 22), 359 (10), 352 (22), 351 (100), 305 (10), 184 (17), 183 (33), 167 (19), 139 (14), 137 (14), 124 (24), 114 (23), 99 (11), 70 (31), 57 (11), 55 (16), 45 (34), 43 (91)
3c ^b	2977, 2939, 2873, 1692, 1447, 1379, 1347, 1324, 1174, 1125, 1099, 950, 846	1.09 [d, J = 7.1, 3 H, CH(C H ₃) ₂], 1.12 [d, J = 6.7, 3 H, CH(C H ₃) ₂], 1.31 (t, J = 7.4, 3 H, CH ₂ C H ₃), 1.56–2.4 [m, 14 H, C H ₂ C H ₂ C H ₂ C(O) ₂ , C H ₂ CN, CHC ₂ CN, CHCO ₂ , NCH ₂ C H ₂ C H ₂], 2.56, 2.58 [q, J = 6.7, q, J = 7.1, 1 H, C H (CH ₃) ₂], 2.9–3.45 (m, 6H, OC H ₂ C H ₂ O, NC H ₂), 3.33 (s, 3 H, OC H ₃), 3.7–3.88 (m, 2 H, NC H , C H HO), 4.0 (m, 1 H, CH H O), 4.23 (q, J = 7.1, 2 H, C H ₂ C H ₃)	14.3 (CH ₂ CH ₃), 19.5 [CH(CH ₃) ₂], 21.9 (NCH ₂ CH ₂), 22.1 [CH(CH ₃) ₂], 26.5 (NCHCH ₂), 27.1 (CH ₂ CH), 31.1 [CH ₂ CH ₂ C(O) ₂], 32.0, 32.2, 33.6 [CH ₂ CH ₂ C(O) ₂], 32.0, 32.2, 33.6 [CH ₂ CH ₂ C(O) ₂ , CHCH ₂ CN, CH ₂ CN], 54.5 (NCH ₂), 58.9 (OCH ₃), 59.1 (CHCO ₂), 60.9 (CH ₂ CH ₃), 65.7 (NCH), 66.1, 68.1 (OCH ₂ CH ₂ O), 76.2 (CH ₂ O), 113.6 [C(O) ₂], 163.3 (CN), 167.7 (CO ₂)	411 (M ⁺⁺ + 1, 5), 410 (M ⁺⁺ , 18), 366 (22), 365 (100), 319 (27), 208 (17), 206 (12), 167 (16), 164 (16), 153 (15), 137 (11), 123 (23), 114 (19), 70 (33), 55 (15), 45 (23), 43 (17), 42 (20), 41 (14)
3d ^b	2970, 2930, 2860, 1700, 1445, 1375, 1320, 1245, 1120, 1095, 1045, 950, 880, 845, 800	$J = 7.1$, 2H, CH_2CH_3) 0.92 [t, $J = 7.0$, 3H, $(CH_2)_3CH_3$], 1.3 (m, 7H, $CH_2CH_2CH_2CH_3$, CH_2CH_3), $1.40-2.60$ [m, 15H, $CH_2CH_2CH_2CH_2$, CH_2CO_2 , CH_2CN , $NCH_2CH_2CH_2$, $2.9-3.5$ (m, 7H, OCH_2CH_2O , NCH_2 , $CHCH_2CN$), 3.7-3.85 (m, 2H, NCH , $CHCH_2CN$), 3.99 (m, 1H, $CHHO$), 4.23 (q, $J = 7.1$, 2H, CH_2CH_3)	13.9 [(CH ₂) ₃ CH ₃], 14.3 (CH ₂ CH ₃), 22.6 (NCH ₂ CH ₂), 22.9 [(CH ₂) ₂ CH ₂ CH ₃], 26.6 (NCHCH ₂), 27.1 (CH ₂ CH), 29.8 (CH ₂ CH ₂ C ₂ H ₅), 31.7 (CHCH ₂ CN), 33.7 [CH ₂ CH ₂ C(O) ₂], 34.6 [CH ₂ CH ₂ C(O) ₂], 34.9 (CH ₂ C ₃ H ₇), 40.1 (CH ₂ CN), 54.8 (NCH ₂), 59.0 (OCH ₃), 60.3 (CHCO ₂), 60.9 (CH ₂ CH ₃), 65.9 (NCH), 65.6, 68.1 (OCH ₂ CH ₂ O), 75.6 (CH ₂ O), 113.6 [C(O) ₂], 163.5 (CN), 167.6 (CO ₂)	425 (M ⁺⁺ + 1, 5), 424 (M ⁺⁺ , 17), 380 (26), 379 (100), 333 (12), 211 (13), 167 (16), 123 (13), 114 (17), 70 (18), 57 (13), 45 (13), 43 (20), 41 (13)
3e ^c	2978, 2940, 2873, 1700, 1447, 1379, 1347, 1424, 1297, 1262, 1172, 1125, 1098, 1049, 949, 847	0.92 (t, $J = 6.4$, 3H, CH_2CH_3), 0.98	14.2 (CH ₂ CH ₃), 21.9 (NCH ₂ CH ₂), 22.1, 22.2 [CH(CH ₃) ₂], 25.8 [CH(CH ₃) ₂], 26.3 (NCHCH ₂), 27.1 (CH ₂ CH), 31.7 (CHCH ₂ CN), 33.6 [CH ₂ CH ₂ C(O) ₂], 34.0 [CH ₂ CH ₂ C(O) ₂], 35.7 (CH ₂ CN), 45.2 [CH ₂ CH(CH ₃) ₂], 54.8 (NCH ₂), 59.0 (OCH ₃), 59.2 (CHCO ₂), 60.9 (CH ₂ CH ₃), 65.6, 66.1 (OCH ₂ CH ₂ O), 65.9 (NCH), 75.5 (CH ₂ OCH ₃), 104.3 [C(O) ₂], 163.5 (CN), 167.6 (CO ₂)	425 (M ⁺⁺ + 1, 4), 424 (M ⁺⁺ , 16), 380 (26), 370 (100), 333 (18), 220 (11), 212 (12), 211 (15), 181 (12), 167 (29), 137 (16), 123 (36), 114 (40), 99 (12), 95 (13), 70 (59), 57 (30), 55 (23), 45 (45), 43 (35), 41 (38)

^a Satisfactory HRMS obtained for 3a-e. The NMR data are those of the major geometric isomer.

6-Substituted 2-Hydroxycyclohex-1-enecarboxylates (R)-4; General Procedure:

The crude or purified hydrazone 3 was dissolved in $\rm Et_2O$ (10 mL), then 6 N HCL (5 mL) was added. After stirring at r.t. for 30 min, $\rm Et_2O$ (50 mL) was added and the phases were separated. The aqueous phase was extracted with $\rm Et_2O$ (2 × 50 mL). The collected organic phases were washed with pH 7 buffer and water, dried (MgSO₄), concentrated in vacuo, and the residue was purified by flash column chromatography (silica gel, light petroleum/ $\rm Et_2O$ 1:1).

3-Substituted 8-Hydroxy-3,4,4a,5,6,7-hexahydroisocoumarines (3S,4aR)-5; General Procedure:

L-Selectride (2 equiv) was added dropwise to a solution of the purified cyclohexenecarboxylate 4 in THF under Ar at $-78\,^{\circ}$ C. Stirring was continued for 1 h. Then the reaction mixture was quenched with water (2 mL) at $-78\,^{\circ}$ C and stirring was continued until the reaction mixture reached r.t. After addition of Et₂O (20 mL), the phases were separated. The organic phase was washed with brine, dried (MgSO₄), concentrated in vacuo and was purified by flash column chromatography (silica gel, light petroleum/Et₂O).

^b IR spectrum determined in Et₂O.

^c IR spectrum determined as film.

Table 5. Spectroscopic Data of the 6-Substituted 2-Hydroxycyclohex-1-enecarboxylates (R)-4^a

Com- pound	IR (KBr) ν (cm ⁻¹)	1 H NMR (CDCl ₃ /TMS) δ , J (Hz)	$^{13}\text{C NMR (CDCl}_3/\text{TMS)}$ δ	MS (70 eV) m/z (%)
4a	2980, 2940, 2871, 1740, 1714, 1646, 1613, 1423, 1403, 1368, 1288, 1257, 1224, 1179, 1164, 1148	1.28 (t, $J=7.0$, 3H, CH_2CH_3), 1.40–1.80 (m, 4H, CH_2CH_2CH , CH_2CH_2CO), 1.80–2.10 (m, 1H, $CHCHH$), 2.20–2.74 (m, 4H, CH_2CO , $CHCHH$, $CHCH_2$), 2.14 (s, 3H, $COCH_3$), 3.09–3.16 (m, $CHCO_2$, keto), 4.22 (q, $J=7.0$, 2H, CH_2CH_3), 12.43 (s, OH, enol)	14.2 (CH ₂ CH ₃), 26.4 (CH ₂ CH ₂ CH), 29.4 (CH ₂ CH), 30.4 (COCH ₃), 36.9 (CHCH ₂), 41.1 [CH ₂ C(OH)], 48.6 (CH ₂ CO), 61.2 (CH ₂ CH ₃), 62.4 (CHCO, keto), 100.9 (C = C(OH), enol), 173.5 (CO ₂), 205.4 (COCH ₃), 206.5 (COH), 208.3 (C=O)	226 (M ⁺ *, 4), 180 (21), 169 (25), 165 (26), 137 (11), 124 (15), 123 (50), 95 (15), 55 (16), 45 (13), 44 (18), 43 (100), 41 (15), 39 (10)
4 b	2978, 2940, 2877, 1742, 1713, 1646, 1613, 1460, 1421, 1402, 1365, 1289, 1257, 1224, 1210, 1180, 1074	(a), CH, CH, CH, CH, CH, CH, CH, CH, CH, CH	7.9 (COCH ₂ CH ₃), 14.2 (CH ₂ CH ₃), 26.4 (CH ₂ CH ₂ CH), 29.1 (CH ₂ CH), 36.1 (COCH ₂ CH ₃), 36.9 (CHCH ₂), 41.1 [CH ₂ C(OH)], 47.1 (CH ₂ CO), 60.4 (CH ₂ CH ₃), 62.5 (CHCO, keto), 101.1 [C=C(OH), enol], 173.4 (CO ₂), 205.5 (COC ₂ H ₅), 209.3 (COH), 210.9 (C=O)	240 (M ⁺⁺ , 13), 194 (13), 169 (44), 165 (49), 137 (16), 123 (89), 97 (11), 95 (21), 83 (13), 81 (15), 71 (13), 69 (15), 67 (16), 57 (27), 55 (26), 45 (13), 43 (100)
4c ^b	3406, 2970, 2938, 2873, 1741, 1713, 1647, 1613, 1466, 1450, 1402, 1383, 1368, 1286, 1256, 1225, 1211, 1179, 1152, 1078	1.09 [d, J = 6.7, 3 H, CH(C H ₃) ₂], 1.10 [d, J = 6.7, 3 H, CH(C H ₃) ₂], 1.28 (t, J = 7.0, 3 H, CH ₂ CH ₃), 1.45–1.78 (m, 4H, CH ₂ CH ₂ CH, CH ₂ CH ₂ CO), 2.00–2.75 (m, 5H, CHCH ₂ CO, CH ₂ CO, CHCH ₂), 2.60, 2.61 [q, J = 7.0, q, J = 7.0, 1 H, CH(CH ₃) ₂], 3.15 (m, CHCO ₂ , keto), 4.22 (q, J = 7.0, 2 H, CH ₂ CH ₃), 12.44 (s, OH, enol)	14.2 (CH ₂ CH ₃), 18.3, 18.2 [CH(CH ₃) ₂], 26.4 (CH ₂ CH ₂ CH), 29.1 (CH ₂ CH), 36.3 (CHCH ₂), 40.8 [CH(CH ₃) ₂], 41.1 [CH ₂ C(OH)], 44.9 (CH ₂ CO), 60.4 (CH ₂ CH ₃), 62.5 (CHCO, keto), 101.1 (C=C(OH), enol), 173.5 (CO ₂), 205.7 [COCH(CH ₃) ₂], 212.6 (COH), 214.0 (C=O)	255 (M ⁺⁺ + 1, 254 (M ⁺⁺ , 9), 211 (9), 209 (10), 208 (11), 169 (100), 165 (64), 137 (14), 123 (99), 95 (16), 71 (28), 67 (11), 55 (38), 43 (58), 41 (25), 39 (14)
4d	3450, 2970, 2930, 2870, 1705, 1445, 1375, 1340, 1320, 1175, 1120, 1095, 1045, 945, 840, 800	0.91 [t, $J = 7.4$, 3 H, (CH ₂) ₃ CH ₃], 1.29 (t, $J = 7.0$, 3 H, CH ₂ CH ₃), 1.29 [m, 2 H, (CH ₂) ₂ CH ₂ CH ₃], 1.48–1.75 (m, 5 H, CH ₂ CH ₂ CH, CH ₂ CH ₂ CO, CHCHH), 2.03 (m, 1 H, CHCH ₂), 2.2–2.74 (m, 7H, CH ₂ CO, CHCHH, CH ₂ CO	13.9 [(CH ₂) ₃ CH ₃], 14.2 (CH ₂ CH ₃), 22.4 [(CH ₂) ₂ CH ₂ CH ₃], 25.8 (CH ₂ CH ₂ C ₂ H ₅), 26.4 (CH ₂ CH ₂ CH), 27.6 [(CH ₂) ₃ CH ₃], 36.9 (CHCH ₂), 41.1 — CH ₂ C(OH)], 42.7 (CH ₂ C ₃ H ₇), 47.4 (CH ₂ CO), 60.4 (CH ₂ CH ₃), 62.5 (CHCO, keto), 101.0 [C = C(OH), enol], 173.4 (CO ₂), 205.5 (COC ₄ H ₉), 208.6 (COH), 210.6 (C = O)	268 (M+*, 10), 222 (13), 169 (84), 168 (14), 166 (16), 165 (76), 137 (18), 124 (14), 123 (100), 122 (11), 95 (21), 94 (17), 85 (39), 81 (10), 79 (10), 67 (11), 57 (53), 55 (48), 45 (12), 43 (55), 41 (45), 39 (18)
4e ^b	3405, 2957, 2872, 1742, 1713, 1647, 1613, 1466, 1422, 1403, 1367, 1334, 1285, 1257, 1223, 1180, 1146, 1095, 1080, 1064, 1047	0.91 [d, J = 6.6, 3 H, $CH(CH_3)_2$], 0.93 [d, J = 6.4, 3 H, $CH(CH_3)_2$], 1.29 (t, J = 7.4, 3 H, CH_2CH_3), 1.50–1.74 (m, 4H, CH_2CH_2CH , CH_2CH_2CO), 1.98–2.72 [m, 8 H, CH_2CO , $CHCH_2$, $CHCH_2CO$, $CH_2CH(CH_3)_2$], 3.08–3.18 (m, $CHCO_2$, keto), 4.22 (q, J = 7.0, 2 H, CH_2CH_3), 12.52 (s, OH , enol)	20.5 (CH ₂ -CH ₃), 22.5/22.6 [CH(CH ₃) ₂], 24.8 [CH ₂ CH(CH ₃) ₂], 26.3 (CH ₂ CH ₂ CH), 29.1 (CH ₂ CH), 36.8 (CHCH ₂), 41.1 [CH ₂ C(OH)], 47.9 (CH ₂ CO), 51.9 [CH ₂ CH(CH ₃) ₂], 60.4 (CH ₂ CH ₃), 62.5 (CHCO, keto), 100.9 [C=C(OH), enol], 173.5 (CO ₂), 201.9 (COC ₄ H ₉), 208.6 (COH), 210.2 (C=O)	(16) (M ⁺⁺ +1, 2) 268 (M ⁺⁺ , 17), 222 (19), 179 (15), 169 (87), 168 (12), 166 (18), 165 (83), 137 (14), 123 (100), 95 (16), 94 (12), 85 (40), 57 (49), 55 (34)

^a Satisfactory microanalyses obtained for 4a-e: C \pm 0.5, H \pm 0.23. NMR data of keto and enol form are given.

This work was supported by the Deutsche Forschungsgemeinschaft (Leibniz prize) and the Fonds der Chemischen Industrie. We thank Bayer AG, BASF AG, Degussa AG and Hoechst AG for the donation of chemicals.

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b IR spectrum determined in CHCl₃.

Table 6. Spectroscopic Data of the 3-Substituted 8-Hydroxy-3,4,4a,5,6,7-hexahydroisocoumarines (3S,4aR)-5a

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Com- pound	IR (KBr) ν (cm ⁻¹)	1 H NMR (CDCl ₃ /TMS) δ , J (Hz)	$^{13}\text{C NMR (CDCl}_3/\text{TMS)}$ δ	MS (70 eV) m/z (%)
5a ^b	3448, 2977, 2927, 2875, 2857, 1642, 1619, 1447, 1407, 1387, 1355, 1305, 1270, 1235, 1200, 1170, 1145, 1105, 1065, 1020, 960, 890	$\begin{array}{llll} 1.09-1.34 & [m, & 2H, & CH_{ax}HCHCH_3, \\ CH_{ax}HCH_2CH_2C(OH)=], & 1.38 & (d, & J=6.3, \\ 3H, & CH_3), & 1.56-1.74 & [m, & 1H, \\ CH_{ax}HCH_2C(OH)=], & 1.84-1.94 & [m, & 2H, \\ CH_{eq}CH_2CH_2C(OH)=, & CH_{eq}CH_2C(OH)=], \\ 1.95-1.98 & (dd, & J=2.4, 3.7, 1H, CH_{eq}CHCH_3), \\ 2.34-2.41 & [m, & 2H, & CH_2C(OH)=], & 2.45-2.58 \\ & (m, & 1H, & CHCH_2CH), & 4.40-4.52 & (ddq, & J=2.4, \\ 12.8, & 6.4, & 1H, & CHCH_3), & 13.26 & (s, & 1H, & OH) \\ \end{array}$	20.9 [CH ₂ CH ₂ C(OH)=], 21.7 (CH ₃), 29.5, 29.1 [CH ₂ CHCH ₂ , CH ₂ C(OH)=], 33.0 (CHCH ₂ CH), 37.5 (CH ₂ CHOCO), 76.6 (CHO-CO), 96.8 (CCO ₂), 171.8 (CO ₂), 174.8 [C(OH)=]	183 (M ^{+*} + 1, 10), 182 (M ^{+*} , 71), 154 (39), 141 (28), 139 (17), 136 (29), 126 (55), 123 (100), 121 (17), 112 (15), 96 (12), 95 (14), 84 (14), 79 (11), 68 (12), 67 (17), 55 (37), 43 (19), 41 (24), 39 (20)
5b°	3364, 2964, 2936, 2867, 1636, 1455, 1412, 1378, 1323, 1300, 1227, 1209, 1196, 1173, 1132, 1104, 1077, 990, 959, 871	1.00 (t, $J = 7.7$, 3 H, CH_3), 1.09 – 1.37 [m, 2 H, CH_{ax} HCHC ₂ H ₅ , CH_{ax} HCH ₂ CH ₂ C(OH)=], 1.58 – 1.82 [m, 3 H, CH_2 CH ₃ , CH_{ax} HCH ₂ C(OH)=], 1.85 – 1.96 [m, 3 H, CH_{eq} CHC ₂ H ₅ , CH_{eq} CH ₂ CH ₂ C(OH)=, CH_{eq} CH ₂ C(OH)=], 2.38 [m, 2 H, CH_2 C(OH)=], 2.50 (m, 1 H, CH CH ₂ CH), 4.25 (m, 1 H, CH C ₂ H ₅), 13.27 (s, 1 H, OH)	9.2 (CH ₃), 20.9 [CH ₂ CH ₂ C(OH)=], 28.7, 29.1, 29.7 [CH ₂ CHCH ₂ , CH ₂ C(OH)=, CH ₂ CH ₃], 32.9 (CHCH ₂ CH), 35.1 (CH ₂ CHOCO), 81.4 (CHOCO), 97.2 (CCO ₂), 171.7 (CO ₂), 174.6 [C(OH)=]	197 (M ⁺ +1, 6), 196 (M ⁺ , 45), 178 (17), 141 (22), 140 (14), 123 (100), 95 (12), 94 (31), 55 (10)
5e°	3365, 2973, 2944, 2892, 2872, 1642, 1453, 1417, 1382, 1355, 1323, 1311, 1264, 1254, 1237, 1182, 1166, 1132, 1085, 1001, 947, 881, 840	0.98 [d, J = 7.0, 3 H, CH(C H_3) ₂], 1.00 (d, J = 6.7, 3 H, CH(C H_3) ₂], 1.1–1.34 (m, 1 H, C H_a HCHC ₃ H ₇), 1.58–1.74 [m, 2 H, C H_a HCHC ₂ C(OH)=, C H_a HCHC ₂ C(OH)=], 1.84–1.98 [m, 4 H, C H_e HCHC ₃ H ₇ , C H_e HCH ₂ CH ₂ C(OH)=, C H_e HCHC ₂ C(OH)=], CH(CH ₃) ₂], 2.37 [m, 2 H, C H_2 C(OH)=], 2.48 (m, 1 H, C H CH), 4.11 (ddd, J = 2.3, 5.0, 5.4, 1 H, C H C ₃ H ₇), 13.27 (s, 1 H, OH)	17.6, 17.8 $[CH(CH_3)_2]$, 20.9 $[CH_2CH_2C(OH)=]$, 29.0, 29.7 $[CH_2CHCH_2$, $CH_2C(OH)=]$, 32.1 (CH_2CHOCO) , 32.6 $[CH(CH_3)_2]$, 32.9 $(CHCH_2CH)$ 84.8 $(CHOCO)$, 97.3 (CCO_2) , 172.0 (CO_2) , 174.5 $[C(OH)=]$	211 (M ⁺⁺ + 1, 3), 210 (M ⁺⁺ , 16), 141 (26), 125 (14), 123 (100), 108 (47), 95 (11), 93 (11), 55 (24), 43 (29), 41 (19), 39 (13)
5d	3433, 2955, 2937, 2862, 1637, 1468, 1457, 1415, 1364, 1330, 1305, 1252, 1241, 1172, 1107, 1045, 1006, 871	0.91 (t, $J = 7.4$, 3 H, CH_3), 1.08–1.74 [m, 9 H, $CH_{ax}HCHC_4H_9$, $CH_{ax}HCH_2CH_2C(OH) =$, $CH_{ax}HCH_2C(OH) =$, $CH_2C_3H_7$, $CH_2CH_2C_2H_5$, $(CH_2)_2CH_2CH_3$], 1.84–1.98 [m, 3 H, $CH_{eq}HCHC_4H_9$	13.9 $[(CH_2)_3CH_3]$, 20.9 $[CH_2CH_2CH_2C(OH)]$, 22.5 $[(CXH_2)_2CH_2CH_3]$, 26.9 $(CH_2CH_2C_2H_5)$, 29.0, 29.6 $[CH_2CHCH_2$, $CH_2C(OH)]$, 32.9 $(CHCH_2CH)$, 35.5 $(CH_2C_3H_7)$, 35.6 (CH_2CHOCO) , 80.2 $(CHOCO)$, 97.2 (CCO_2) , 171.9 (CO_2) , 174.6 $[C(OH)]$	225 (M ⁺⁺ + 1, 3), 224 (M ⁺⁺ , 20), 206 (9), 141 (37), 123 (100), 122 (30), 95 (11), 79 (9), 67 (8), 55 (19), 43 (24), 41 (14)
5e	3360, 2951, 2897, 2872, 1642, 1457, 1412, 1385, 1368, 1306, 1260, 1236, 1171, 1152, 1109, 1067, 1047, 989	0.93 [d, $J = 6.4$, 3 H, $CH(CH_3)_2$], 0.94 [d, $J = 6.7$, 3 H, $CH(CH_3)_2$], 1.08 – 1.39 [m, 2 H, $CH_{ax}HCHC_4H_9$, $CH_{ax}HCH_2CH_2C(OH) = $], 1.55 – 1.73 (m, 3 H, $CH_{ax}HCH_2C(OH) = $, $CH_2C_3H_7$], 1.84 – 1.97 [m, 4 H, $CH_{eq}HCHC_4H_9$, $CH_{eq}HCH_2CH_2C(OH) = $, $CH_{eq}CH_2C(OH) = $, $CH(CH_3)_2$], 2.37 [m, 2 H, $CH_2C(OH) = $], 2.51 (m, 1 H, $CHCH_2CH$), 4.37 (m, 1 H, CHC_4H_9), 13.28 (s, 1 H, OH)	20.9 [$CH_2CH_2C(OH) =]$, 22.1, 23.0, 23.9 [$CH_2CH(CH_3)_2]$, 29.1 ($CH_2CHCH_2)$, 29.6 [$CH_2C(OH) =]$, 33.0 ($CHCH_2CH)$, 36.2 ($CH_2CHO-CO$), 45.0 ($CH_2C_3H_7$), 78.4 ($CHO-CO$), 97.2 (CCO_2), 171.9 (CO_2), 174.6 $-C(OH) =]$	225 (M ⁺⁺ + 1, 2), 224 (M ⁺⁺ , 16), 206 (10), 141 (31), 123 (100), 122 (17), 95 (11), 67 (12), 57 (10), 55 (28), 43 (20), 41 (36), 39 (16)

^a Satisfactory microanalyses obtained for 5a-e: $C \pm 0.28$, $H \pm 0.17$.

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