

Direct and Efficient Synthesis of Dimethylformamidrazones Using Bentriazole Vilsmeier Reagent

Alan R. Katritzky,* Tian-Bao Huang, and Michael V. Voronkov

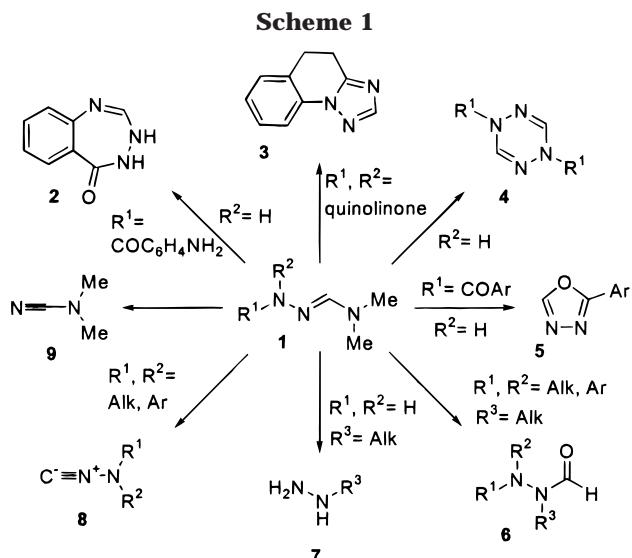
Department of Chemistry, Center for Heterocyclic Compounds, University of Florida, Gainesville, Florida 32611-7200

Received November 22, 1999

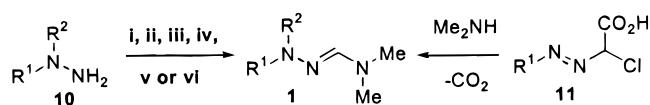
Dimethylformamidrazones **1** are versatile synthetic intermediates and are widely used as HN^-NCH^+ synthons (Scheme 1). Thus, **1** has been used (i) for the construction of various heterocycles including benzotriazepinones **2**,^{1a,b} triazoloquinolines **3** and their salts,² tetrazines **4**,^{3a,b} triazolopyrimidines,⁴ triazoles,⁵ and oxadiazoles **5**⁶ and (ii) for the preparation of formamidrazones **6**,⁷ hydrazines **7**,^{3b,8} isocyanides **8**,⁹ and cyanamides **9**.⁹

Further uses of **1** include synthesis of nucleosides,¹⁰ precursors for ureas, and other utilities.^{11a,b}

Several different preparative methods for dimethylformamidrazones **1** have been used previously (Scheme 2). Among these, the treatment of the corresponding hydrazine **10** with POCl_3 and DMF in benzene affords **1** in yields ranging from 63% to 71% and allows both alkyl and aryl substituents R^1 and R^2 .^{12a,b} Similarly, a mixture of DMF and $(\text{Me})_2\text{NSO}_2\text{Cl}$ was employed for preparation of **1** with electron-deficient substituents ($\text{R}^1 = \text{H}$, $\text{R}^2 = \text{SO}_2\text{Ar}$, $\text{NO}_2\text{C}_6\text{H}_4$, $(\text{NO})_2\text{C}_6\text{H}_3$, etc.) in yields of 71% to 88%. However, no synthesis of **1** with electron-rich substituents using such methods was reported.^{4,13a,b} Other approaches are less straightforward. For example, decarboxylation of arylazochloroacetic acids **11**¹⁴ affords **1** in 41–79% yield but requires a multistep preparation of the precursor that may have its own limitations on the generality of the overall approach. Similar problems arise in other examples such as a reaction of hydrazines



Scheme 2^a



^a Key: (i) DMF, $(\text{Me})_2\text{NSO}_2\text{Cl}$; (ii) DMF, POCl_3 ; (iii) $\text{Me}_2\text{N}^+\text{CHClCl}^-$. Bt=benzotriazole; (iv) $(\text{Me})_2\text{N}^+\text{CHSCH}_3\text{Cl}^-$; (v) $(\text{Me})_2\text{NCH}(\text{CN})\text{OCH}_3$; (vi) $(\text{Me})_2\text{N}^+\text{CHBtCl}^-$.

with *tert*-butylaminol esters¹⁵ or dimethylformamide diethyl acetal⁶ with amidrazines, which afford the selected **1** in 64–82% yield. Finally, reaction of hydrazines with dimethylaminoalkoxyacetonitrile¹⁶ produces toxic byproducts.

We now report a direct and general method for the preparation of dimethylformamidrazones **1** starting from corresponding hydrazines **10** ($\text{R}^1, \text{R}^2 = \text{Ar}, \text{COAr}, \text{Alk}, \text{H}$) in 66–99% yields. For the preparation of **1**, we utilized the novel versatile reagent **12**¹⁷ as a stable Vilsmeier reagent analogue. Previously, **12** was used in the synthesis of quinolines,^{18a} pyridines,^{18b} and pyrazoles^{18c} (Scheme 3).

Typically, to prepare compounds **1** ($\text{R}^1 = \text{H, Alk, Ar}$ and $\text{R}^2 = \text{H, Alk, Ar}$), 4 mmol of hydrazine **10** was reacted with 4 mmol of reagent **12** in 30 mL of THF under reflux to form the product **1** as a white precipitate. The desired product **1** is separated into its pure form by simple filtration. This procedure is applicable to hydrazines with a wide variety of substitutions (Table 1). For example, novel, or not readily available by other procedures,

- (1) (a) Scheiner, P.; Frank, L.; Giusti, I.; Arwin, S.; Pearson, S. A.; Excellent, F.; Harper, A. P. *J. Heterocycl. Chem.* **1984**, *21*, 1817. (b) Leiby, R. W. *J. Heterocycl. Chem.* **1984**, *21*, 1825.
- (2) Batori, S.; Sandor, P.; Messmer, A. *Heterocycles* **1990**, *31*, 289.
- (3) (a) Langdon, S. P.; Simmonds, R. J.; Stevens, M. F. G. *J. Chem. Soc., Perkin Trans. 1* **1984**, 993. (b) Zelenin, K. N.; Khrustalev, V. A.; Sergutina, V. P. *Zh. Org. Khim.* **1980**, *16*, 276.
- (4) Guillot, N.; Viehe, H. G.; Tinant, B.; Declercq, J. P. *Tetrahedron* **1990**, *46*, 3897.
- (5) Jones, S. D.; Kennewell, P. D.; Tulley, W. R.; Westwood, R.; Sammes, P. G. *J. Chem. Soc., Perkin Trans. 1* **1990**, 447.
- (6) Eilingsfeld, H. *Chem. Ber.* **1965**, *98*, 1308.
- (7) Khrustalev, V. A.; Zelenin, K. N.; Sergutina, V. P. *Zh. Org. Khim.* **1979**, *15*, 2288.
- (8) Hempel, U.; Lippmann, E.; Tenor, E. *Z. Chem.* **1990**, *30*, 320.
- (9) Zelenin, K. N.; Bazylevich, N. I.; Khrustalev, V. A. *Zh. Org. Khim.* **1977**, *13*, 2064.
- (10) Tutonda, M. G.; Fain, H. D.; Buckheit, R. W. Jr.; Broom, A. D. *Nucleosides Nucleotides* **1997**, *16*, 173.
- (11) (a) Becker, H. G. O.; Gwan, K. M. *J. Prakt. Chem.* **1992**, *334*, 231. (b) Kohn, H.; Olofson, R. A. *J. Org. Chem.* **1972**, *37*, 3504.
- (12) (a) Sergutina, V. P.; Zelenin, K. N.; Khrustalev, V. A. *Zh. Org. Khim.* **1978**, *14*, 622. (b) Bredereck, H.; Gompper, R.; Klemm, K.; Rempfer, H. *Chem. Ber.* **1959**, *92*, 837.
- (13) (a) Scott, F. L.; Barry, J. A. *Tetrahedron Lett.* **1968**, *20*, 2457. (b) Hunig, S.; Muller, F. *Liebigs Ann. Chem.* **1962**, *651*, 89.
- (14) Lozinskii, M. O.; Pel'kis, P. S. *Zh. Obshch. Khim.* **1962**, *32*, 526.

(15) Kantlehner, W.; Kapassakalidis, J. J.; Maier, T. *Liebigs Ann. Chem.* **1980**, 1448.

(16) Bredereck, H.; Simchen, G.; Kantlehner, W. *Chem. Ber.* **1971**, *104*, 932.

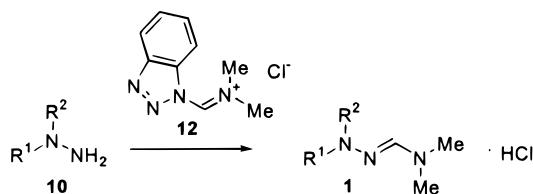
(17) Katritzky, A. R.; Cheng, D.; Leeming, P.; Chiviriga, I.; Hartshorn, C. M.; Steel, P. J. *J. Heterocycl. Chem.* **1996**, *33*, 1935.

(18) (a) Katritzky, A. R.; Arend, M. *J. Org. Chem.* **1998**, *63*, 9989. (b) Katritzky, A. R.; Denisenko, A.; Arend, M. *J. Org. Chem.* **1999**, *64*, 6076. (c) Katritzky, A. R.; Denisenko, A.; Arend, M.; Denisenko, S. N. Unpublished work.

(19) Winberg, H. E. US Patent 3,121,084, **1964**; *Chem. Abstr.* **1964**, *60*, 13197d.

(20) Bartlett, R. K.; Humphrey, I. R. *J. Chem. Soc. C* **1967**, 1664.

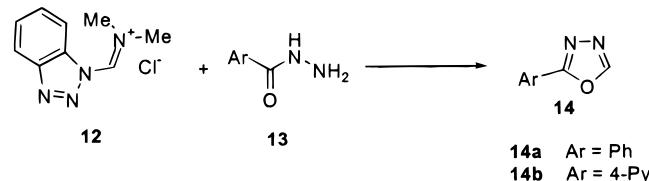
(21) Taylor, E. C.; Morrison, R. W., Jr. *J. Org. Chem.* **1967**, *32*, 2379.

Scheme 3**Table 1. Preparation of Dimethylformamidrazone Hydrochlorides 1**

entry	R ¹	R ²	yield (%)	published yield (%)
1a	C ₆ H ₅	H	99	
1b	4-BrC ₆ H ₄	H	84	
1c	4-NO ₂ C ₆ H ₄	H	98	81 ^{13a}
1d	2,4-(NO ₂) ₂ C ₆ H ₃	H	98	71 ^{13a}
1e	4-CH ₃ C ₆ H ₄	H	75	
1f	C ₆ F ₅	H	98	
1g	CH ₃	CH ₃	88 ^a	82 ¹⁶
1h	4-(N-oxide-pyridinyl)-CO	H	73 ^b	
1i	4-CH ₃ C ₆ H ₄ SO ₂	H	79	not given ¹⁹
1j	H	H	69 ^a	not given ²⁰
1k	H	H	98 ^c	80 ²⁰
1l	-CH=N-N=CH-		91	
1m	C ₆ H ₅ SO ₂	H	89	88 ^{e,13b}
1n	-COC ₆ H ₄ CO-		87	66 ^{e,21}
1o	C ₆ H ₅	CH ₃	90	71 ^{e,12b}
1p	-CH ₂ CH ₂ N(CH ₃)CH ₂ CH ₂ -		98 ^d	
1q	-CH ₂ CH ₂ OCH ₂ CH ₂ -		74	
1r	4-CH ₃ OC ₆ H ₄	H	91	

^a Extremely hydroscopic, no mp was obtained. ^b 9% of the corresponding oxadiazole was isolated along with the product.

^c Bisadduct was obtained when 2 equiv of **12** was used. ^d Was obtained as the bis hydrochloride salt. ^e The yield for the free amidrazones.

Scheme 4

formamidrazones with both electron-deficient aromatic substituents (entries **1c**, **1d**, and **1f**) and more electron-rich aromatic rings (entries **1a**, **1e**, **1o**, and **1r**) were prepared in 69–98% yields (Table 1). A number of novel heterocyclic formamidrazones (entries **1h**, **1l**, **1p**, and **1q**) are now available in 73–98% yields. In reactions of acyclic 1,1-disubstituted hydrazines, the yields of the corresponding compounds (entries **1g** and **1o**) were 88% and 90%. Similarly, compounds **1j** and **1k** were prepared from unsubstituted hydrazine giving 69% and 98% yields.

When we extended this procedure to hydrazides **13**, 1,3,4-oxadiazoles **14** were obtained as pure compounds in 95–98% yields (Scheme 4).

Conclusion. A simple and efficient procedure of preparation of dimethylformamidrazones was developed. Eighteen examples were prepared in 69–99% yields.

Experimental Section

General Methods. Melting points were determined with a MEL-TEMP capillary melting point apparatus equipped with a Fluka 51 digital thermometer. NMR spectra were taken in DMSO-d₆, D₂O, and CDCl₃ with tetramethylsilane as the internal standard for ¹H (300 MHz) or solvent as the internal standard for ¹³C (75 MHz). THF was distilled from sodium/

benzophenone under nitrogen immediately prior to use. All reactions with air-sensitive compounds were carried out under an argon atmosphere. Column chromatography was conducted on silica gel 230–400 mesh.

General Procedure for the Preparation of **1a–**l**.** A mixture of 4 mmol (0.85 g) of *N*-(1*H*-1,2,3-benzotriazol-1-ylmethylene)-*N*-methylmethanaminium and 4 mmol of the corresponding hydrazine in 30 mL of dry THF was refluxed for 4 h under dry nitrogen. The solution was cooled to room temperature, and the resulting white precipitate was quickly filtrated off and was washed once with 10 mL of dry THF to give **1** as a hydrochloride salt.

***N,N*-Dimethyl-*N*-phenylhydrazoneformamide hydrochloride (**1a**):** 98% yield; white solid; mp 185–186 °C; ¹H NMR δ 3.08 (s, 3H), 3.18 (s, 3H), 6.88 (d, 2H, J = 8.0 Hz), 6.97 (t, 1H, J = 7.5 Hz), 7.28 (t, 2H, J = 8.13 Hz), 8.03 (s, 1H); ¹³C NMR δ 38.3, 44.7, 115.3, 123.7, 131.0, 147.9, 160.4. Anal. Calcd for C₉H₁₄ClN₃: C, 54.13; H, 7.07; N, 21.04. Found: C, 53.67; H, 7.22; N, 21.07.

***N*-(4-Bromophenyl)-*N,N*-dimethylhydrazoneformamide hydrochloride (**1b**):** 84% yield; pale yellow solid; mp 215–216 °C; ¹H NMR δ 3.19 (br. s, 6H), 6.85 (d, 2H, J = 8.8 Hz), 7.38 (d, 2H, J = 8.8 Hz), 8.47 (s, 1H); ¹³C NMR δ 38.8, 39.9, 114.4, 115.2, 131.2, 146.9, 159.2. Anal. Calcd for C₉H₁₃BrClN₂: C, 38.80; H, 4.70; N, 15.08. Found: C, 38.66; H, 4.64; N, 14.83.

***N,N*-Dimethyl-*N*-4-nitrophenylhydrazoneformamide hydrochloride (**1c**):** 98% yield; yellow solid; mp 241–242 °C; ¹H NMR δ 3.19 (br.s, 6H), 6.90 (d, 2H, J = 8.0 Hz), 8.04–8.18 (m, 2H), 8.90 (s, 1H); ¹³C NMR δ 42.0, 43.0, 113.7, 127.6, 142.2, 154.2, 160.8. Anal. Calcd for C₉H₁₃ClN₄O₂: C, 44.18; H, 5.36; N, 22.90. Found: C, 43.94; H, 5.61; N, 22.76.

***N,N*-Dimethyl-*N*-2,4-dinitrophenylhydrazoneformamide hydrochloride (**1d**):** 98% yield; yellow solid; mp 238–239 °C; ¹H NMR δ 3.22 (br.s, 6H), 7.64 (d, 1H, J = 9.4 Hz), 8.38 (d, 1H, J = 9.3 Hz), 8.53 (s, 1H), 8.9 (s, 1H), 10.96 (s, 1H); ¹³C NMR δ 23.1, 38.6, 40.3, 116.0, 125.3, 129.8, 136.7, 145.7, 157.7. Anal. Calcd for C₉H₁₂ClN₅O₄: C, 37.32; H, 4.18; N, 24.18. Found: C, 37.09; H, 4.26; N, 23.88.

***N*-(4-Methylphenyl)-*N,N*-dimethylhydrazoneformamide hydrochloride (**1e**):** 75% yield; white solid; mp 212–213 °C; ¹H NMR δ 2.21 (s, 3H), 3.21 (br. S, 6H), 6.78 (d, 2H, J = 7.2 Hz), 7.04 (d, 2H, J = 7.2 Hz), 8.44 (s, 1H); ¹³C NMR δ 19.9, 41.9, 113.4, 129.0, 129.1, 145.1, 159.1; HRMS calcd for C₁₀H₁₆N₃(M + 1) 178.1344, found 178.1347.

***N*-(Pentafluorophenyl)-*N,N*-dimethylhydrazoneformamide hydrochloride (**1f**):** 98% yield; white solid; mp 202–204 °C; ¹H NMR δ 3.13 (s, 6H), 8.38 (s, 1H); ¹³C NMR δ 38.7, 40.4, 116.2, 128.1, 137.0, 137.4, 137.6, 139.6, 140.7, 143.0, 160.1. Anal. Calcd for C₉H₉ClF₅N₃: C, 37.32; H, 3.13; N, 14.51. Found: C, 36.87; H, 2.97; N, 14.01.

***N,N,N,N*-Tetramethylhydrazoneformamide hydrochloride (**1g**):** 88% yield; colorless solid; mp 144–145 °C; ¹H NMR δ 2.62 (s, 3H), 2.67 (s, 3H), 3.00 (br, 6H), 7.94 (br.s, 1H); ¹³C NMR δ 38.0, 44.0, 48.5, 157.5. Anal. Calcd for C₅H₁₄ClN₃: C, 39.60; H, 9.37; N, 27.71. Found: C, 39.54; H, 9.77; N, 27.44.

***N*-(4-N'-oxideisonicotinoyl)-*N,N*-dimethylhydrazoneformamide hydrochloride (**1h**):** 73% yield; white solid; mp 155–156 °C; ¹H NMR δ 3.20 (s, 3H), 3.22 (s, 3H), 7.89 (d, 2H, J = 3.9 Hz), 8.27 (s, 1H), 8.77 (s, 2H); ¹³C NMR δ 42.3, 123.9, 141.8, 150.0, 160.0, 167.1; HRMS calcd for C₉H₁₂N₄O₂ 208.0960, found 208.0963. Anal. Calcd for C₉H₁₃ClN₄O₂: N, 22.90. Found: N, 22.75.

***N,N*-Dimethyl-*N*-toluenolsulfonylformamidrazone hydrochloride (**1i**):** 79% yield; white solid; mp 215–216 °C; ¹H NMR δ 2.32 (s, 3H), 2.90 (s, 3H), 3.17 (s, 3H), 7.37 (d, 2H, J = 8.0 Hz), 7.66 (d, 2H, J = 8.0 Hz), 8.03 (s, 1H); ¹³C NMR δ 22.5, 38.8, 45.2, 129.6, 131.9, 133.2, 148.1, 160.6. Anal. Calcd for C₁₀H₁₆ClN₃O₂S: C, 43.24; H, 5.81; N, 15.13. Found: C, 43.23; H, 5.93; N, 15.15.

***N,N*-Dimethylhydrazoneformamide hydrochloride (**1j**):** 69% yield; colorless crystals; mp 215–216 °C; ¹H NMR δ 2.57 (s, 6H), 8.20 (s, 1H); ¹³C NMR δ 35.9, 145.9.

***N,N*-Dimethylhydrazoneformamide (**1k**):** 98% yield; colorless oil; ¹H NMR δ 2.64 (s, 3H), 2.66 (s, 3H), 2.71 (s, 3H), 7.83 (s, 1H), 7.86 (s, 1H); ¹³C NMR δ 36.9, 39.9, 153.7, 163.2.

***N,N*-Dimethyl-*N*-1,2,4-triazol-4-ylformamide hydrochloride (**1l**):** 91% yield; white solid; mp 176–177 °C; ¹H NMR

δ 2.81 (s, 3H), 2.94 (s, 3H), 7.99 (s, 1H), 9.06 (s, 2H); ^{13}C NMR δ 36.3, 42.6, 141.7, 163.1. Anal. Calcd for $\text{C}_5\text{H}_{10}\text{ClN}_5$: N, 39.88. Found: N, 40.07.

N,N-Dimethyl-N-(phenylsulfonyl)hydrazoneformide hydrochloride (1m): 89% yield; white solid; mp 198–200 °C; ^1H NMR δ 2.87 (s, 3H), 3.15 (s, 3H), 7.54 (t, 2H, J = 7.8 Hz), 7.66 (t, 1H, J = 7.8 Hz), 7.80 (d, 2H, J = 7.8 Hz), 8.03 (s, 1H); ^{13}C NMR δ 39.1, 45.6, 129.9, 131.8, 136.5, 137.0, 161.0. Anal. Calcd for $\text{C}_9\text{H}_{14}\text{ClN}_3\text{O}_2\text{S}$: C, 40.99; H, 5.36; N, 15.94. Found: C, 40.87; H, 5.58; N, 15.96.

N-(1,3-Dioxo-1,3-dihydro-2H-isindol-2-yl)-N,N-dimethyliminoformamide hydrochloride (1n): 87% yield; white solid; mp 247–248 °C; ^1H NMR δ 3.18 (s, 3H), 3.27 (s, 3H), 7.77 (dd, 4H, J = 8.0 Hz, 3.1 Hz), 8.20 (s, 1H); ^{13}C NMR δ 39.0, 45.6, 125.0, 130.3, 137.3, 160.4, 167.1.

N-Methyl-N-phenyl-N,N-dimethylhydrazoneformide hydrochloride (1o): 90% yield; white solid; mp 223–224 °C; ^1H NMR δ 3.04 (s, 3H), 3.10 (s, 3H), 3.19 (s, 3H), 6.91–7.02 (m, 3H), 7.29 (t, 2H, J = 6.8 Hz), 8.10 (s, 1H); ^{13}C NMR δ 38.5, 44.4, 44.9, 116.5, 123.7, 131.0, 150.7, 159.8. Anal. Calcd for $\text{C}_{10}\text{H}_{15}\text{ClN}_3$: C, 56.20; H, 7.56; N, 19.67. Found: C, 55.77; H, 8.03; N, 19.70.

N,N-Dimethyl-N-(4-methylpiperazino)iminoformamide bishydrochloride (1p): 87% yield; white solid; mp 250–251 °C; ^1H NMR δ 2.79 (s, 3H), 2.93 (s, 3H), 3.13 (s, 3H), 3.22 (t, 4H, J = 7.2 Hz), 3.40 (t, 4H, J = 7.2 Hz), 8.08 (s, 1H); ^{13}C NMR

δ 38.4, 44.0, 44.8, 53.7, 54.5, 158.5. Anal. Calcd for $\text{C}_8\text{H}_{20}\text{Cl}_2\text{N}_4$: C, 39.51; H, 8.29; N, 23.04. Found: C, 39.12; H, 8.65; N, 22.71.

N,N-Dimethyl-N-morpholinoiminoformamide hydrochloride (1q): 74% yield; white solid; mp 145–146 °C; ^1H NMR δ 2.92 (t, 2H, J = 8.0 Hz), 2.96 (s, 3H), 3.17 (s, 3H), 3.72 (s, 3H), 8.07 (s, 1H); ^{13}C NMR δ 38.2, 44.6, 57.1, 67.4, 157.9. Anal. Calcd for $\text{C}_7\text{H}_{16}\text{ClN}_3\text{O}$: N, 21.70. Found: N, 21.28.

N-(4-Methoxyphenyl)-N,N-dimethylhydrazoneformamide hydrochloride (1r): 91% yield; white solid; mp 185–186 °C; ^1H NMR δ 3.03 (s, 3H), 3.19 (s, 3H), 3.70 (s, 3H), 6.84–6.85 (br, s, 4H), 8.04 (s, 1H); ^{13}C NMR δ 38.3, 44.8, 57.2, 116.5, 117.4, 141.8, 156.0, 160.2. Anal. Calcd for $\text{C}_{10}\text{H}_{16}\text{ClN}_3\text{O}$: N, 18.29. Found: N, 18.00.

2-Phenyl-1,3,4-oxadiazole (14a): 89% yield; white solid; mp 34–36 °C; ^1H NMR δ 7.75–7.90 (m, 3H), 8.20 (d, 2H, J = 8.4 Hz), 9.57 (s, 1H); ^{13}C NMR δ 123.2, 126.6, 129.3, 131.9, 154.4, 163.6.

2-(4-Pyridinyl)-1,3,4-oxadiazole (14b): 95% yield; white solid; mp 118 °C; ^1H NMR δ 7.98 (d, 2H, J = 6.1 Hz), 8.66 (s, 1H), 8.88 (d, 2H, J = 6.1 Hz); ^{13}C NMR δ 120.4, 130.6, 150.9, 153.3, 163.0. Anal. Calcd for $\text{C}_7\text{H}_5\text{N}_3\text{O}$: C, 57.14; H, 3.43; N, 28.56. Found: C, 57.25; H, 3.44; N, 28.47.

JO991807G