

Tetrahedron: Asymmetry 9 (1998) 1131-1135

TETRAHEDRON: ASYMMETRY

Diastereoselective palladium(0)-catalyzed azidation of 1-alkenylcyclopropyl esters: asymmetric synthesis of (-)-(1R,2S)-norcoronamic acid

Valérie Atlan,^a Sandrine Racouchot,^a Michael Rubin,^a Claudia Bremer,^{a,b} Jean Ollivier,^a Armin de Meijere^{*, b} and Jacques Salaün ^{a,*}

^aLaboratoire des Carbocycles (Associé au CNRS), Institut de Chimie Moléculaire d'Orsay, Bât. 420, Université de Paris-Sud, 91405 Orsay, France

^bInstitut für Organische Chemie der Georg August Universität, D-37077 Göttingen, Germany

Received 6 March 1998; accepted 12 March 1998

Abstract

Palladium(0)-catalyzed azidation of (1R,2S)-1-(1-alkenyl)2-methylcyclopropyl esters **10a**,**b** proceeds with complete retention of configuration to provide, after reduction of the azide and oxidative cleavage of the allylic double bond, the (1R,2S)-norcoronamic acid **22** (>99% e.e.). © 1998 Elsevier Science Ltd. All rights reserved.

Due to the physiological importance of 1-aminocyclopropanecarboxylic acids (2,3-methanoamino acids) considerable effort has been, and currently still is, devoted towards their total synthesis,¹ and their incorporation into peptidic chains which provide conformationally constrained peptidomimetics with enhanced biological activities.² Moreover, incorporation of 2,3-methanoamino acids increases the bioavailability of peptides by reducing their hydrolysis rate (proteolytic degradation).³ Substituted 1-aminocyclopropanecarboxylic acids also provide enzyme inhibitors, biological probes for mechanistic studies and allow the design of new drugs.^{1,2} The different methodologies recently reported for their asymmetric synthesis were mainly based on the cyclopropanation of chiral alkenes or on the enantio-selective cyclopropanation performed in the presence of chiral auxiliaries by means of the hazardous diazomethane or Et_2Zn/CH_2I_2 reagents.⁴ However, we have previously reported that the base-induced diastereoselective cyclization of 2-(*N*-benzylideneamino)-4-chlorobutyronitriles (d.e. 60–78%)⁵ and the palladium(0)-catalyzed substitution on (4*S*)-1-chloropent-2-ene-4-ol (readily available from ethyl (2*S*) lactate) with the anion of *N*-(diphenylmethyleneamino)acetonitrile, followed by a diastereo-selective S_N' cyclization under Mitsunobu conditions (DEAD, PMe₃) (d.e. 88%),⁶ provided chiral non-racemic (1*S*,2*S*)-2,3-methanoamino acids (ACCs) with 84–88% enantiomeric excesses.

^{*} Corresponding authors. Fax: +33(1)69156278; e-mail: jasalaun@icmo.u-psud.fr. Fax: +49 551399475; e-mail: ameijer@uni-goettingen.de



We now report that the palladium(0)-catalyzed azidation of non-racemic 1-(1-alkenyl)cyclopropyl esters can diastereoselectively lead to precursors of (1R,2S)-ACCs. While the palladium(0)-catalyzed amination with, for example, dibenzylamine of 1-ethenylcyclopropyl tosylate **1a**, which is readily available from cyclopropanone hemiacetal,⁷ led exclusively to *N*,*N*-dibenzylcyclopropylideneethylamine **3a** in 85% yield, the palladium(0)-catalyzed azidation of **1a**,**b**, on the other hand, with sodium azide in the presence of 15-crown-5 ether (10%) was reported to provide in 80% yield, the 1-(1-alkenyl)cyclopropyl azides **4a**,**b**, exclusively, suitable precursors of 2,3-methanoalanine **5**.⁸ This regioselectivity was shown to result from the non-symmetric charge distribution in the π -1,1-dimethyleneallylpalladium complex **2a** which led to substitution either at the primary allylic end by *soft* nucleophiles (e. g. stabilized carbanions) or on the cyclopropyl ring by *hard* nucleophiles (hydride donors, organometallics, etc., and also by azide).⁷



Commercially available methyl (2S) 3-hydroxy-2-methylproponiate 6a (>99% e.e.)⁹ was reacted with mesyl chloride in the presence of NEt₃ to produce the mesylate (2S)-**6b** (91%) which was then treated with lithium bromide in NMP to give the bromide (2S)-6c (82%); alternatively, reaction of (2S)-6a with carbon tetrabromide/PPh₃ in CH₂Cl₂ led directly to the bromide (2S)-6c, $[\alpha]_D^{2D} = -18$ (c 1, CHCl₃), in 73% yield. Cyclization of (2S)-6c by treatment with sodium in diethyl ether in the presence of chlorotrimethylsilane (Et₂O at reflux) gave a 1:1 diastereomeric mixture of 1-ethoxy-2-methyl-1trimethylsiloxycyclopropanes (2S)-7, which on acid-catalyzed methanolysis (MeOH, ClSiMe) led to the hemiacetals (2S)-8 in 85% overall yield from (2S)-6c.¹⁰ Upon treatment with two equivalents of vinylmagnesium chloride of the 1:1 mixture of (2S)-8, followed by the addition of two equivalents of tosyl chloride in THF, the (1R,2S) 2-methyl-1-tosyloxycyclopropane 10a was obtained directly in 67% overall yield, as a single diastereomer, as revealed by its ¹H and ¹³C NMR spectra. Likewise, successive addition of one equivalent of ethylmagnesium bromide and of one equivalent of lithium phenylacetylide to (2S)-8 gave in 76% yield a 91:9 diastereomeric mixture of 2-methyl-1-(phenylethynyl)cyclopropanols (11), which on lithium aluminum hydride reduction led in 96% yield to the 2-methyl-1-styrylcyclopropanols 9b. While tosylation of 9b failed, mesylation was achieved under classical conditions (MeSO₂Cl, NEt₃, Et_2O) to produce a 90:10 diastereometric mixture of the expected mesulates **10b** in 96% yield.



Treatment of the cyclopropanone hemiacetal with one equivalent of Grignard reagent (i.e. methyl-, ethyl- or vinylmagnesium halides) has been shown to produce a magnesium salt which, contrary to the hemiacetal itself, is able to react with organolithium reagents as well as with a second Grignard reagent, to produce 1-alkyl(alkenyl)cyclopropanols in high yields.¹¹ The diastereoselectivity observed here for the nucleophilic substitutions (2*S*)-8 \rightarrow (1*R*,2*S*)-9a,b strongly suggests that the intermediate magnesium 1-ethoxycyclopropanolate 12 behaves more as the cyclopropanone 12', which is probably stabilized by ligation with MgXOEt. In any event, this reaction provides 2-methyl-1-alkenylcyclopropanols like 9a,b, diastereoselectively; non-racemic 1-alkenylcyclopropanols such as 9a,b have also been prepared from the dimethyl (2*S*)-2-methylsuccinate, via sodium-induced acyloin cyclization followed by a diastereoselective C₄ \rightarrow C₃ ring contraction, with total preservation of the chirality of the stereocenter.¹²



Palladium(0)-catalyzed reaction (Pd(dba)₂, 2PPh₃) of the tosylate (1*R*,2*S*)-**10a** and of the 90:10 diastereomeric mixture of mesylates **10b** with sodium azide in the presence of 10% 15-crown-5 ether, gave either the single azide (1*R*,2*S*)-**14a** (R=H) or a 96:4 diastereomeric mixture of azides (1*R*,2*S*)-**14b** and (1*S*,2*S*)-**15b** (R=C₆H₅), respectively, in 81–85% yields. COSY and NOESY experiments on the two-dimensional NMR spectra of the azide **14a** have confirmed the assigned configuration.

Alternatively, reaction of the hemiacetal **8** with ethoxycarbonylmethylenetriphenylphosphorane in benzene in the presence of a catalytic amount of benzoic acid, gave in 58% yield a 2:1 diastereomeric mixture of ethyl 2-(2-methylcyclopropylidene)ethyl acetate 16^{13} which was reduced to the allylic alcohol **17** with DIBAH (CH₂Cl₂, -78° C) and then esterified with acetic anhydride in the presence of DMAP and NEt₃ to give the allylic acetate **18** in 67% overall yield.^{8,14} Subsequent palladium(0)-catalyzed azidation of **18**, through the intermediate complex **13a**, provided in 85% yield the same azide **14a** as a single diastereomer.¹⁴



Reduction of the azides **14a,b** with 1,3-propanedithiol in MeOH containing NEt₃,¹⁵ gave the corresponding primary amines **20a,b** in 56 and 97% yield, respectively. The lower isolated yield of **20a** most probably was due to its volatility, therefore the total synthesis of the non-racemic (1*R*,2*S*)-norcoronamic acid **22** was performed with the 2-methyl-1-styrylcyclopropylamine **20b**, initially as a 96:4 diastereomeric mixture, as revealed by the ¹H NMR spectrum of the crude reduction product, but in fact obtained as a single diastereomer after flash chromatography, $[\alpha]_D^{20}$ =+94 (c 1.15, CHCl₃).

Subsequent treatment of **20b** with di-*t*-butyldicarbonate in water, containing KOH and *t*-BuOH, provided **21b** in quantitative yield. Finally, oxidative degradation of the styryl group in **21b** by ruthenium tetroxide (RuCl₃/NaIO₄),¹⁶ *N*-deprotection by treatment with HCl and ion-exchange chromatography (Dowex 50WX 18) led to the (1*R*,2*S*)-norcoronamic acid **22**,¹⁷ the enantiomeric purity (>99%) of which was determined by deuterium NMR in a cholesteric lyotropic liquid crystal of the deuterated methyl ester of the *N*-(diphenylmethylene)amino acid **22**, following a recently reported accurate method.¹⁸

It is known that the palladium(0)-catalyzed azidation of allyl esters occurs with *overall retention* of configuration,¹⁹ however, the substitution of the palladium moiety in the chiral complexes **13a**,**b** with the required *inversion* of configuration should lead as previously reported to (cyclopropylideneethyl)amine derivatives (*soft* nucleophile behaviour),⁷ while substitution of complexes **13a**,**b** with *retention* of configuration should lead to azides of type (1*S*,2*S*)-**15b** as major products (*hard* nucleophile behaviour).^{7,12} Therefore the exclusive formation of azides (1*R*,2*S*)-**14a**,**b** must result via primarily formed (cyclopropylideneethyl) azides and subsequent palladium(0)-induced isomerization¹⁹ which is stereocontrolled by the palladium moiety coordinated to the double bond.

Acknowledgements

This work was financially supported by the CNRS and the Université de Paris-Sud (XI), by the Deutsche Forschungs-gemeinschaft (SFB 416) as well as by the Fonds der Chemischen Industrie. A.d.M. is grateful to the French MENRT for the Alexander von Humboldt–Gay Lussac prize which allowed him to stay in Orsay; M.R. is indebted to the International Association for the Promotion of Cooperation with Scientists from the Independant States of the Former Soviet Union (INTAS) for a fellowship and C.B. to the Student Mobility Programme (ERASMUS) for a grant. Finally, we are very grateful to Cecile Canlet and Jacques Courtieu who recorded the ²H NMR spectra of **22** derivatives in a cholesteric lyotropic liquid crystal.

References

- For recent reviews see: (a) Stammer, C. A. *Tetrahedron* 1990, 46, 2231; (b) Alami, A.; Calmes, M.; Daunis, J.; Jacquier, R. *Bull. Soc. Chim. Fr.* 1993, 130, 5; (c) Burgess, K.; Ho, H.-H.; Moye-Sherman, D. *Synlett* 1994, 575; (d) Salaün, J.; Baird, M. S. *Curr. Med. Sci.* 1995, 2, 511.
- (a) Leete, E.; Louters, L. L., Prakash, Rao, H. S. *Phytochemistry* 1986, 25(12), 2753; (b) Liu, H.-W.; Walsh, C. T. The Chemistry of Functional Groups, *The Biochemistry of the Cyclopropyl Group*; Patai, S., Ed; Wiley, New York, 1987, Vol. 2, p. 959.
- 3. Mapelli, C.; Elrod, L. F.; Holt, E. M.; Switzer, F. L.; Stammer, C. H. Biopolymers 1989, 28, 123.
- 4. Charette, A. B.; Côté, B. J. Am. Chem. Soc. 1995, 87, 12721, and references cited therein.
- Gaucher, A.; Ollivier, J.; Marguerite, J.; Paugam, R.; Salaün, J. Can. J. Chem. 1994, 72, 1312; Salaün, J.; Marguerite, J.; Karkour, B. J. Org. Chem. 1990, 55, 4276.
- 6. Dorizon, Ph.; Ollivier, J.; Salaün, J. Synlett 1996, 1071.
- 7. Stolle, A.; Ollivier, J.; Piras, P. P.; Salaün, J.; de Meijere, A. J. Am. Chem. Soc. **1992**, 114, 4015; Ollivier, J.; Dorizon, Ph.; Piras, P. P.; de Meijere, A.; Salaün, J. Inorg. Chim. Acta **1994**, 222, 37.
- 8. Aufranc, P.; Ollivier, J.; Stolle, A.; Bremer, C.; Es-Sayed, M.; de Meijere, A.; Salaün, J. Tetrahedron Lett. 1993, 34, 4193.
- 9. Aldrichimica Acta 1984, 17, 42.
- 10. Salaün, J.; Marguerite, J. Org. Synth. 1984, 63, 147; Fadel, A.; Canet, J. L.; Salaün, J. Synlett 1990, 89.
- 11. Salaün, J. Chem. Rev. 1983, 83, 619, and references cited therein.
- 12. Salaün, J.; Karkour, B.; Ollivier, J. *Tetrahedron* 1989, 45, 3151; Salaün, J. *Chem. Rev.* 1989, 89, 1247; Chevtchouk, T.; Ollivier, J.; Salaün, J. *Tetrahedron: Asymmetry* 1997, 8, 1005.
- 13. Spitzner, D.; Swoboda, H. Tetrahedron Lett. 1986, 27, 1281.

- 14. Bremer, C. Dissertation, Universität Göttingen (Germany), 1995.
- 15. Baley, H.; Standring, D. N.; Knowler, J. R. Tetrahedron Lett. 1978, 3633.
- 16. Carlsen, P. H. J.; Hatsuki, T.; Martin, V. S.; Sharpless, K. B. J. Org. Chem. 1981, 46, 3936.
- 17. Baldwin, J. E.; Adlington, R. M.; Rawlings, B. J.; Jones, R. H. Tetrahedron Lett. 1985, 26, 481.
- 18. Canet, I.; Meddour, A.; Courtieu, J.; Canet, J. L.; Salaün, J. J. Am. Chem. Soc. 1994, 116, 2155.
- 19. Murahashi, S.-I.; Taniguchi, Y.; Imada, Y.; Tanigawa, Y. J. Org. Chem. 1989, 54, 3292.