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Deacetylation of α -Bromo- α -substituted-acetoacetates

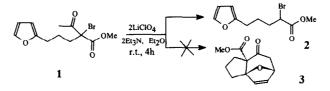
Youhong Hu and Donglu Bai*

Shanghai Institute of Materia Medica, Chinese Academy of Science, 294 Taiyuan Road, Shanghai 200031, China

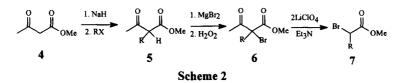
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Abstract: A new and efficient method for the deacetylation of α -bromo- α -substituted-acetoacetates to obtain α -bromoesters has been achieved by treatment with LiClO₄/Et₃N in ether at room temperature. © 1998 Elsevier Science Ltd. All rights reserved.

During the course of the study of [4+3] intramolecular cycloaddition, we used α -bromo- α -3-(2furyl)propylacetoacetate(1) as one of the precursors. Under the conditions for the formation of allylic cation (Scheme 1), only deacetylated product 2 was obtained instead of the desired [4+3] cycloadduct 3. It is wellknown that the hydrolysis of acetoacetates is readily accompanied with decarboxylation,¹ but the efficient removal of the acetyl group from acetoacetates is rare.² α -Bromoesters are versatile intermediates in organic synthesis for the preparation of phosphoranes and phosphonates in Wittig reaction or of α -amino acids and esters. This communication describes a facile method for the preparation of α -bromoesters by means of deacetylation of α -bromo- α -substituted-acetoacetates (Scheme 2).



Scheme 1



Methyl acetoacetate was converted to 5 in high yields by alkylation. The α -methine in 5 was brominated

by using MgBr₂ in the presence of H₂O₂, affording α -bromides 6³ in 72-82% yields. Deacetylation of 6 was achieved by treatment with LiClO₄/Et₃N in ether at room temperature for 4h to obtain α -bromoesters. The reaction sequence in Scheme 2 offers the possibility for preparation of α -bromoesters containing sensitive functionality in the substituent R to bromine and Ba(OH)₂.^{2a}

A typical procedure is as follows: Methyl α -bromo- α -3-(2-furyl)propylacetoacetate(1) (303mg, 1mmol) was dissolved in freshly distilled ether (20mL), LiClO₄ (203mg, 2mmol)and freshly distilled (from CaH₂) triethylamine (0.28mL, 2mmol) were added. The mixture was stirred at room temperature for 4h. The reaction was quenched with H₂O (5mL), and the mixture was extracted with Et₂O (15mL×3). The combined organic layer was washed with H₂O (10mL), brine (10mL), and dried over anhydrous Na₂SO₄. The solvent of the organic layer was removed in vacuum. The residual oil was purified by column chromatography on silica gel(petroleum ether : EtOAc = 20 : 1) to give compound **2** as a colorless oil (215mg) in 83% yield. In order to verify the generality of this reaction, the other five α -bromo- α -substituted-acetoacetates were tested and the yields of the deacetylated products **7** is in a range of 70 ~ 82% (**Table 1**). All the compounds obtained were characterized by ¹H NMR and IR spectra and new compounds were also characterized by MS, HRMS or elemental analysis.

Table 1. Deacetylation of α -bromo- α -substituted-acetoacetate 6

R		Reaction Time	Product	Yield%
$\overline{\mathbb{Q}}$	1	4h	2*	83
\bigcirc	6a	6h	7a	76
CH3(CH2)2CH2-	6b	5h	7b	72
ζ_{o}^{o}	6c	4h	7c*	82
	6d	2d	7d*	70

* New compound

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References

- 1. Schaefer, J. P.; Bloomfield, J. J., Org. Reactions, 1967, 15, 1.
- (a) Stotter, P. L.; Hill, K. A., *Tetrahedron Lett.*, **1972**, 4067. (b) Mignani, G.; Morel, D.; Grass, F., *Tetrahedron Lett.*, **1987**, 5505. (c) Fujimoto, K.; Maekawa, H.; Matsubara, Y.; Nishiguchi, I., *Chem. Lett.*, **1996**, 143.
- 3. Inukai, N.; Iwamoto, H.; Tamura, T.; Yanagisawa, I.; Ishii, Y.; Murakami, M., Chem. Pharm. Bull., 1976, 24, 820.