



## Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry

Publication details, including instructions for authors and subscription information:

<http://www.tandfonline.com/loi/lcyc20>

### One Pot Synthesis of Acetylated Homoallyl Alcohols

S. Chandrasekhar <sup>a</sup>, Pradyumna K. Mohanty <sup>a</sup> & Abbas Raza <sup>a</sup>

<sup>a</sup> Indian Institute of Chemical Technology ,  
Hyderabad, 500 007, India  
Published online: 17 Sep 2007.

To cite this article: S. Chandrasekhar , Pradyumna K. Mohanty & Abbas Raza (1999) One Pot Synthesis of Acetylated Homoallyl Alcohols, *Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry*, 29:2, 257-262, DOI: [10.1080/00397919908085765](https://doi.org/10.1080/00397919908085765)

To link to this article: <http://dx.doi.org/10.1080/00397919908085765>

PLEASE SCROLL DOWN FOR ARTICLE

Taylor & Francis makes every effort to ensure the accuracy of all the information (the "Content") contained in the publications on our platform. However, Taylor & Francis, our agents, and our licensors make no representations or warranties whatsoever as to the accuracy, completeness, or suitability for any purpose of the Content. Any opinions and views expressed in this publication are the opinions and views of the authors, and are not the views of or endorsed by Taylor & Francis. The accuracy of the Content should not be relied upon and should be independently verified with primary sources of information. Taylor and Francis shall not be liable for any losses, actions, claims, proceedings, demands, costs, expenses, damages, and other liabilities whatsoever or howsoever caused arising directly or

indirectly in connection with, in relation to or arising out of the use of the Content.

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden. Terms & Conditions of access and use can be found at <http://www.tandfonline.com/page/terms-and-conditions>

## ONE POT SYNTHESIS OF ACETYLATED HOMOALLYL ALCOHOLS

S. Chandrasekhar,\* Pradyumna K. Mohanty and Abbas Raza

Indian Institute of Chemical Technology, Hyderabad-500 007, India.

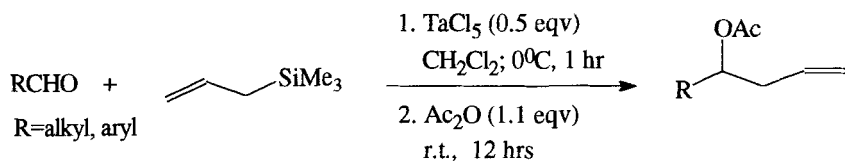
**Abstract :** An efficient one pot procedure for the preparation of acetylated homoallyl alcohols mediated by  $\text{TaCl}_5$  is described.

Sakurai reaction<sup>1</sup> has become a most versatile tool to prepare homoallyl alcohols starting from allylsilanes and aldehyde wherein a new C-C bond is also created. This reaction is generally mediated by Lewis acids. Various mediators have been reported in literature to-date which include  $\text{BF}_3 \cdot \text{OEt}_2$ ,  $\text{NbCl}_5$ ,  $\text{InI}_3$ ,  $\text{Sc}(\text{OTf})_3$ , besides others.<sup>2</sup> The recent additions include chiral ligand assisted reactions<sup>3</sup> wherein good ee's of homoallyl alcohols are obtained. In a long synthetic scheme it may be desirable to protect the newly generated free alcohol before proceeding further. It would be more advantageous if both allylation and acetylation reactions are achieved in one pot. Thus, as part of a programme initiated at development and utilisation of  $\text{TaCl}_5$  as Lewis acid,<sup>4</sup> we were interested in developing a new procedure for Sakurai reaction and insitu acetylation, the results of which are being presented herein (equation 1).

---

IICT Communication No. 4014

\* To whom correspondence should be addressed

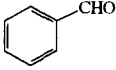
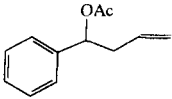
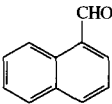
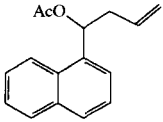
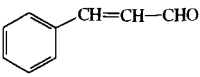
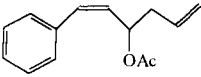
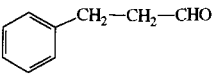
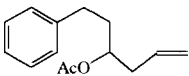
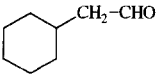
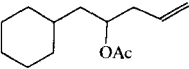
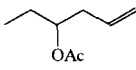
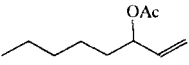
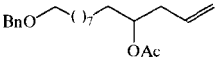
**equation 1:**

Initially benzaldehyde (entry 1, table 1) was treated with 0.5 eq. of tantalum chloride and 1.1 eq of allyltrimethylsilane in dry  $\text{CH}_2\text{Cl}_2$  at  $0^\circ\text{C}$  for 1 hour (monitored by TLC) followed by addition of 1.1 eq of acetic anhydride yielded the corresponding acetylated homoallyl alcohol in 70% isolated yield. Also, naphthaldehyde (entry 2, Table 1) responded well to the reaction protocol and good yield of acetylated alcohol was obtained. In the case of conjugated aldehyde (entry 3, Table 1) no traces of 1,4-product was observed. Few other aliphatic aldehydes (entries 5,6,7 & 8, Table 1) also were consistent to the reaction conditions which explains the mildness of the procedure.

**TYPICAL PROCEDURE :**

To a stirred solution of 0.5 mmol of  $\text{TaCl}_5$  (0.179 g) in 20 mL of dry  $\text{CH}_2\text{Cl}_2$ , which was cooled to  $0^\circ\text{C}$  was added 1.0 mmol of benzaldehyde (0.106 g) in dry  $\text{CH}_2\text{Cl}_2$  under  $\text{N}_2$  atmosphere. After stirring for 20 minutes 1.1 mmol of allyltrimethylsilane (0.125 g) in dry  $\text{CH}_2\text{Cl}_2$  was added dropwise and allowed to stir at  $0^\circ\text{C}$  for 1 hour. Freshly distilled 1.1 mmol of  $\text{Ac}_2\text{O}$  (0.112 g) was added dropwise at  $0^\circ\text{C}$  and allowed to stir at room temperature for 12 hours. The total reaction mixture was quenched with a saturated solution of  $\text{NaHCO}_3$  and stirred for 10 minutes. The reaction mixture was extracted with  $\text{CH}_2\text{Cl}_2$  (3 x 10 mL). The combined organic layers were washed with brine, dried over anhydrous sodium sulphate, filtered and concentrated to give crude product ( $R_f=0.5$ , 5% ethyl acetate in hexane as eluting mixture). The crude product was purified by column chromatography over  $\text{SiO}_2$  using hexane : ethylacetate (50 : 1) as eluting mixture, providing 0.114 g (70%) of acetylated homoallyl alcohol.

Table 1 :

Entry	Aldehyde	Product	Yield%*
1		1a 	1b 70
2		2a 	2b 66.7
3		3a 	3b 70.5
4		4a 	4b 50 4**
5		5a 	5b 60
6	$\text{CH}_3\text{CH}_2\text{-CHO}$	6a 	6b 55
7	$\text{CH}_3(\text{CH}_2)_5\text{-CHO}$	7a 	7b 66
8	$\text{BnO}(\text{CH}_2)_9\text{-CHO}$	8a 	8b 57

\* Yields based on isolation of chromatographically homogenous products

\*\* About 15% of alcohol was also isolated

**Analytical and spectroscopic data of compounds :**

**1b:** IR (neat) : 845 $\text{cm}^{-1}$ , 1737 $\text{cm}^{-1}$ .  $^1\text{H}$  NMR( $\text{CDCl}_3$ ) :  $\delta$  7.21-7.51 (m, 5H), 5.67 (t, 1H,  $J=7.01$  Hz), 5.62-5.71 (m, 1H), 5.15-5.21 (m, 2H), 2.45-2.51 (m, 2H), 2.05 (s, 3H); MS ( $\text{M}^+$ ) : 190; Anal. Calcd. for  $\text{C}_{12}\text{H}_{14}\text{O}_2$  : C, 75.76; H, 7.41; found : C, 75.36; H, 7.28.

**2b:** IR (neat) : 848 $\text{cm}^{-1}$ , 1734 $\text{cm}^{-1}$ .  $^1\text{H}$  NMR( $\text{CDCl}_3$ ) :  $\delta$  7.41-8.21 (m, 7H), 6.65 (t, 1H,  $J=5$  Hz), 5.72-5.92 (m, 1H), 5.15-5.23 (m, 2H), 2.80 (t, 2H,  $J=7.5$  Hz), 2.15 (s, 3H); MS ( $\text{M}^+$ ) : 240; Anal. Calcd. for  $\text{C}_{16}\text{H}_{16}\text{O}_2$  : C, 79.98; H, 6.70; found : C, 79.38; H, 6.25.

**3b:** IR (neat) : 848 $\text{cm}^{-1}$ , 1736 $\text{cm}^{-1}$ .  $^1\text{H}$  NMR( $\text{CDCl}_3$ ) :  $\delta$  7.21-7.45 (m, 5H), 6.60 (d, 1H,  $J=15.5$  Hz), 6.15 (dd, 1H,  $J=15.5$  Hz), 5.60-5.80 (m, 1H), 5.15-5.62 (m, 2H), 2.50 (t, 1H,  $J=6$  Hz), 2.05 (s, 3H); MS ( $\text{M}^+$ ) : 216; Anal. Calcd. for  $\text{C}_{14}\text{H}_{16}\text{O}_2$  : C, 77.75; H, 7.45; found : C, 77.25; H, 7.38.

**4b:** IR (neat) : 849 $\text{cm}^{-1}$ , 1735 $\text{cm}^{-1}$ .  $^1\text{H}$  NMR( $\text{CDCl}_3$ ) :  $\delta$  6.95-7.22 (m, 5H), 5.51-5.72 (m, 1H), 4.91-5.08 (m, 2H), 4.75-4.85 (m, 1H), 2.45-2.62 (m, 2H), 2.15 (t, 2H,  $J=7$  Hz), 2.01 (s, 3H); 1.82-1.95 (m, 2H) MS ( $\text{M}^+$ ) : 218; Anal. Calcd. for  $\text{C}_{14}\text{H}_{18}\text{O}_2$  : C, 77.03; H, 8.31; found : C, 76.97; H, 8.25.

**5b:** IR (neat) : 848 $\text{cm}^{-1}$ , 1735 $\text{cm}^{-1}$ .  $^1\text{H}$  NMR( $\text{CDCl}_3$ ) :  $\delta$  5.60-5.80 (m, 1H), 4.95-5.12 (m, 2H), 4.12-4.55 (m, 1H), 2.25-2.35 (m, 2H), 2.05 (s, 3H), 0.81-1.82 (m, 13H); MS ( $\text{M}^+$ ) : 210; Anal. Calcd. for  $\text{C}_{13}\text{H}_{22}\text{O}_2$  : C, 74.25; H, 10.55; found : C, 74.01; H, 10.23.

**6b:** IR (neat) : 848 $\text{cm}^{-1}$ , 1735 $\text{cm}^{-1}$ .  $^1\text{H}$  NMR( $\text{CDCl}_3$ ) :  $\delta$  5.65-5.75 (m, 1H), 5.05-5.15 (m, 2H), 4.82-4.91 (m, 1H), 2.35-2.45 (m, 2H), 0.85-1.00 (m, 5H), 2.08 (s, 3H), 0.81-1.00 (m, 5H); MS ( $\text{M}^+$ ) : 142; Anal. Calcd. for  $\text{C}_8\text{H}_{14}\text{O}_2$  : C, 67.57; H, 9.92; found : C, 67.25; H, 9.61.

**7b:** IR (neat) : 847 $\text{cm}^{-1}$ , 1735 $\text{cm}^{-1}$ .  $^1\text{H}$  NMR( $\text{CDCl}_3$ ) :  $\delta$  5.62-5.75 (m, 1H), 5.12-5.15

(m, 2H), 4.85-4.95 (m, 1H), 2.25-2.45 (m, 2H), 2.01 (s, 3H), 0.82-1.15 (m, 13H); MS ( $M^+$ ) : 180; Anal. Calcd. for  $C_{11}H_{16}O_2$  : C, 73.30; H, 12.21; found : C, 73.15; H, 12.01.

**8b:** IR (neat) :  $846\text{cm}^{-1}$ ,  $1737\text{cm}^{-1}$ .  $^1\text{H}$  NMR( $\text{CDCl}_3$ ) :  $\delta$  7.21-7.35 (m, 5H), 5.66-5.82 (m, 1H), 5.01-5.15 (m, 2H), 4.85-4.95 (m, 1H), 4.49 (s, 2H), 3.35 (t, 2H,  $J=5.88$  Hz), 2.30-2.35 (m, 2H), 2.01 (s, 3H), 1.12-1.75 (m, 16H); MS ( $M^+$ ) : 346; Anal. Calcd. for  $C_{22}H_{34}O_3$  : C, 76.26; H, 9.88; found : C, 76.14; H, 9.58.

### Acknowledgments :

P.K.M. thanks CSIR, New Delhi for a Senior Research Fellowship and A.R. thanks, INSA, New Delhi for financial assistance.

### References :

1. Sakurai, H. *Pure and Appl. Chem.*, **1985**, *57*, 1759-70.
2. a) Veenstra, S.J.; Schmid, P. *Tetrahedron Lett.*, **1987**, *38*, 997; b) Kobayashi, S.; Iwamoo, S.; Nagayama, S. *Synlett.*, **1997**, 1099; c) Hosomi, A.; Sakurai, H., *Tetrahedron Lett.*, **1976**, *16*, 1295; d) Kira, M.; Hino, T.; Sakurai, H. *Tetrahedron Lett.*, **1989**, *30*, 1099; e) Yanagisawa, a.; Morodome, M.; Nakashima, H.; Yamanisto, H. *Synlett.*, **1997**, 1309; f) Kobayashi, S.; Basujima, T.; Nagayama, S. *Chem. Commun.*, **1998**, *19*; g) Macta, H.; Nagaswa, T.; Handa, Y.; Takei, T.; Osamura, Y.; Suzuki, K. *Tetrahedron Lett.*, **1995**, *36*, 899.
3. a) Yu, C.; Choi, H.; Jung, W.; Kim, H.; Shin, J. *Chem. Commun.*, **1997**, 761; b) Zhang, L.; Sakurai, H.; Kira, M. *Chem. Lett.*, **1997**, 129; c) Miyai, T.; Inoue, K.; Yasuda, M.; baba, A. *Synlett.*, **1997**, 699; d) Ishinara, K.; Mouri, M.; Gao, Q.; Maruyama, T.; Furuta, K.; Yamamoto, H. *J. Am. Chem. Soc.*, **1993**, *115*, 11490; e) Yanagisawa, A.; nakashima, H.; Ishiba, A.; Yamamoto, H. *J. Am. Chem. Soc.*, **1996**, *118*, 4723; f) Faller, J.W.; Sams, D.W.I.; Liu-X. *J. Am. Chem. Soc.*, **1996**, *118*, 1277; g) Hafner, A.; Duthaler, R.O.; Marti, R.O.; Matri, R.; Rins, G.; Striet, P.R.; Schwarzenbach, F. *J. Am. Chem. Soc.*, **1992**, *114*, 2321; h) Keck, G.E.;

- Tarbet, K.H.; Geroci, L.S. *J. Am. Chem. Soc.*, **1993**, *115*, 8467; i) Costa, A.L., Piazza, M.G.; Taglirini, E.; Trombini, C.; Umani-Rohchi, A. *J. Am. Chem. Soc.*, **1993**, *115*, 7001; j) Wang, Z.; Wang, D.; Sui, X. *Chem. Commun*, **1996**, 2261; k) Keck, G.E.; Krishnamurthy, D.K.; Grier, M.C. *J. Org. Chem.*, **1993**, *58*, 6543.
4. a) Chandrasekhar, S.; Takhi, M.; Reddy, Y.R.; Mohapatra, S.; Rao, C.R.; Reddy, K.V. *Tetrahedron*, **1997**, *53*, 14997; b) Chandrasekhar, S.; Takhi, M.; Uma, G. *Tetrahedron Lett.*, **1997**, 8089; c) Chandrasekhar, S.; Ramchander, T.; Takhi, M. *Tetrahedron, Lett.*, (In Press).

(Received in the USA 28 June 1998)