

# An enantiocontrolled synthesis of a key intermediate to (+)-lactacystin

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An asymmetric synthesis of a key intermediate **16** to (+)-lactacystin **1** has been established starting from epoxide **2** via intramolecular mercurioamidation of allylic trichloroacetimidate **4** and concomitant addition-reduction of ester **13** by  $\text{Pr}^i\text{MgBr}$ , in which reduction of the intermediate ketone proceeded with complete stereoselectivity.

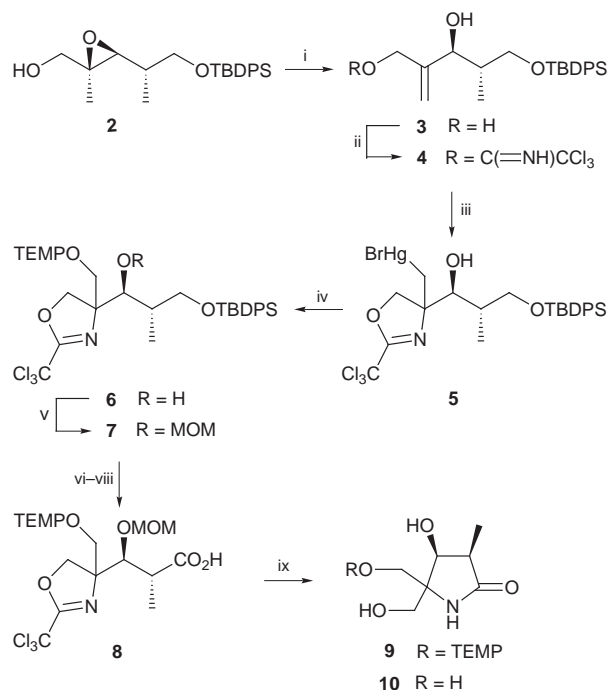
Since neurotrophic factors are responsible for the survival and function of neurons,<sup>1</sup> they might be useful in the treatment of various nerve diseases.<sup>2</sup> Omura *et al.* screened a number of microbial culture samples to isolate the first non-protein neurotrophic agent (+)-lactacystin **1** from *Streptomyces* sp. OM-6519.<sup>3</sup> Its structure, elucidated by NMR spectroscopy and X-ray crystallographic analysis, is composed of (*R*)-*N*-acetylcysteine and a unique pyroglutamic acid via a thioester linkage.<sup>4</sup> (+)-Lactacystin inhibits cell proliferation, induces neurogenesis and increases the intracellular cAMP level transiently in the Neuro 2A neuroblastoma cell line.<sup>3,5</sup> Its intriguing structural features as well as potential therapeutic utility have engendered considerable interest in the fields of synthetic and medicinal chemistry. Here we describe a stereoselective synthetic route to (+)-lactacystin.<sup>6–9</sup> The key steps of our synthesis comprise tertiary amination of the olefinic double bond in allylic trichloroacetimidate **4** via mercurioamidation,<sup>10</sup> facile differentiation of the hydroxymethyl groups in **10** by ring formation and diastereoselective derivatization of ester **13** into alcohol **14**.

The known epoxide **2**,<sup>11</sup>  $[\alpha]_{\text{D}}^{20} -24.7$  (*c* 1.15,  $\text{CHCl}_3$ ), was treated with LDA to give allylic alcohol **3**,  $[\alpha]_{\text{D}}^{21} +10.4$  (*c* 1.44,  $\text{CHCl}_3$ ), in 91% yield (Scheme 1). Only the primary hydroxy group of **3** was functionalized to a trichloroacetimidate. The crude monoimidate **4** was subjected to intramolecular mercurioamidation using mercuric trifluoroacetate with  $\text{K}_2\text{CO}_3$  to furnish a 1:1 diastereomeric mixture of oxazolines **5** in 92% overall yield after aqueous KBr work-up. Since oxidative demercuration<sup>12</sup> of **5** using  $\text{O}_2$  failed under a variety of reaction conditions, it was attempted by exposing **5** to TEMPO in the presence of  $\text{LiBH}_4$  to provide the oxidized products **6** in 78% yield. The secondary hydroxy groups of **6** were protected with  $\text{MeOCH}_2\text{Cl}$  (MOMCl) and then the silyl groups were removed to afford the corresponding primary alcohols in 84% overall yield. While PDC oxidation of the alcohols in DMF was sluggish, they were efficiently oxidized to carboxylic acids **8** in 78% yield by Swern oxidation<sup>13</sup> followed by  $\text{KMnO}_4$  oxidation.<sup>14</sup> Complete hydrolysis and the ensuing cyclization were effected by heating **8** at reflux with ethanolic HCl in AcOH. The 2,2,6,6-tetramethylpiperidyl (TEMP) groups of the generated pyrrolidinones **9** were reductively cleaved *in situ* by adding zinc to the hot reaction mixture to produce trihydroxy pyrrolidinone **10**,  $[\alpha]_{\text{D}}^{19} +9.5$  (*c* 0.95, MeOH), in 72% overall yield from **8**.

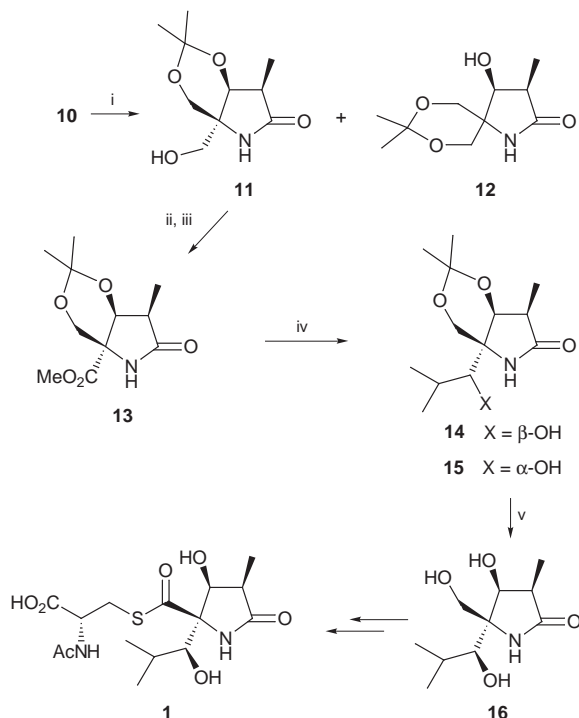
For the appropriate elaboration of the  $\alpha$ -hydroxymethyl groups in **10**, it was chemoselectively reacted with acetone under acidic conditions to give a 7:1 mixture of acetonides **11** and **12** in 95% combined yield (Scheme 2). After chromatographic separation, the primary alcohol **11**,  $[\alpha]_{\text{D}}^{20} +31.4$  (*c* 1.10,  $\text{CHCl}_3$ ), was oxidized under Swern conditions and the resulting aldehyde reacted with  $\text{Pr}^i\text{MgBr}$  under various reaction conditions to furnish a 1:1 mixture of alcohols **14** and **15** along with an appreciable amount of the reduced starting alcohol **11**.

Owing to the inefficient Grignard addition, **11** was converted into ester **13**,  $[\alpha]_{\text{D}}^{21} +57.1$  (*c* 1.70,  $\text{CHCl}_3$ ), in 90% yield. Subjection of **13** to 1 equiv. of  $\text{Pr}^i\text{MgBr}$  provided the corresponding isopropyl ketone in 80% yield, the stereoselective reduction of which was attempted employing several reducing agents such as oxazaborolidine,<sup>15</sup>  $\text{Ipc}_2\text{Cl}$ ,<sup>16</sup> sodium triacetoxyborohydride,<sup>17</sup>  $\text{NaBH}_4$  in the presence of diethylmethoxyborane,<sup>18</sup> and so forth. However, the best stereoselectivity turned out to be 5:1 in favor of **14** with  $\text{NaBH}_4$  in MeOH at 0 °C. Some experimentation revealed that an excess amount of  $\text{Pr}^i\text{MgBr}$  reduced the generated isopropyl ketone to the alcohol **14**. Accordingly, **13** was treated with >2 equiv. of  $\text{Pr}^i\text{MgBr}$  to give selectively only the desired diastereomeric alcohol **14**,  $[\alpha]_{\text{D}}^{20} +40.5$  (*c* 1.20,  $\text{CHCl}_3$ ), in 91% yield. Acidic hydrolysis of **14** yielded trihydroxy pyrrolidinone **16**, mp 198–199 °C (decomp.),  $[\alpha]_{\text{D}}^{20} +16.2$  (*c* 0.62, MeOH), quantitatively, the spectroscopic data of which are identical to those reported in the literature and which is a known intermediate to (+)-lactacystin **1**.<sup>8,19</sup>

We have developed an enantioselective synthetic route to (+)-lactacystin **1** via several crucial steps, including animo hydroxylation of the olefinic double bond in **3**, the hydrolytic cyclization of **8**, and the regio- and stereo-selective functionalization of one hydroxymethyl group in **10**; these should have versatility in the synthesis of its analogues.



**Scheme 1** Reagents and conditions: i, LDA, THF, 0–24 °C; ii, Cl<sub>3</sub>CCN, DBU, EtCN, –78 °C; iii, Hg(O<sub>2</sub>CCF<sub>3</sub>)<sub>2</sub>, K<sub>2</sub>CO<sub>3</sub>, THF, 0 °C, then aq. KBr; iv, TEMPO, LiBH<sub>4</sub>, THF, 24 °C; v, MOMCl, Pr<sub>2</sub>NEt, CH<sub>2</sub>Cl<sub>2</sub>, 0–24 °C; vi, Bu<sub>4</sub>NF, H<sub>2</sub>O, THF, 45 °C; vii, (COCl)<sub>2</sub>, DMSO, Et<sub>3</sub>N; viii, 1 M KMnO<sub>4</sub>, 1.25 M NaH<sub>2</sub>PO<sub>4</sub>, Bu<sup>t</sup>OH, 24 °C; ix, conc. HCl, EtOH, AcOH, reflux, then Zn, reflux



**Scheme 2** Reagents and conditions: i, TsOH, acetone, 24 °C; ii, Jones' reagent, acetone, 0 °C; iii, CH<sub>2</sub>N<sub>2</sub>, THF, 0 °C; iv, Pr<sup>i</sup>MgBr, THF, −20 to 0 °C; v, TsOH, MeOH, 60 °C

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## Notes and References

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- 19 All new compounds showed satisfactory spectral data. *Selected data for 16*: δ<sub>H</sub>(300 MHz, CD<sub>3</sub>OD) 0.94 (3H, d, *J* 6.7), 1.02 (3H, d, *J* 6.7), 1.11 (3H, d, *J* 7.5), 1.91–2.00 (1H, m), 2.79 (1H, p, *J* 7.5), 3.49 (1H, d, *J* 3.6), 3.78 (2H, s) and 4.40 (1H, d, *J* 7.5); δ<sub>C</sub>(75.5 MHz, CD<sub>3</sub>OD) 9.5, 17.6, 22.7, 30.6, 42.6, 63.4, 70.7, 74.4, 79.1 and 181.6.

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