Tetrahedron Letters 55 (2014) 5443-5446

Contents lists available at ScienceDirect

Tetrahedron Letters

journal homepage: www.elsevier.com/locate/tetlet



Synthesis of arylsulfonyl-quinones and arylsulfonyl-1,4-diols as FabH inhibitors: Pd-catalyzed direct C-sulfone formation by C—S coupling of quinones with arylsulfonyl chloride



Bingyang Ge, Dawei Wang*, Weifu Dong, Piming Ma, Yongliang Li, Yuqiang Ding*

The Key Laboratory of Food Colloids and Biotechnology, Ministry of Education, School of Chemical and Material Engineering, Jiangnan University, Wuxi 214122, Jiangsu Province, China

ARTICLE INFO

Article history: Received 8 May 2014 Revised 22 July 2014 Accepted 8 August 2014 Available online 14 August 2014

Keywords: C-sulfone Coupling Quinones Sulfonyl-1,4-diols FabH inhibitors

ABSTRACT

The Pd-catalyzed direct C-sulfone formation by C—S coupling of quinones with arylsulfonyl chloride has been developed. This methodology provides an effective, convenient method for the synthesis of aryl-sulfonyl-quinones and arylsulfonyl-1,4-diols, which are potent inhibitors of FabH.

© 2014 Elsevier Ltd. All rights reserved.

2-Tosylnaphthalene-1,4-diol (**A**) is considered to be a potent inhibitor of β -*Ketoacyl-ACP-synthase III* (FabH),¹ which is a key condensing enzyme in bacterial fatty acid biosynthesis and a part of the dissociated *fatty acid synthase* (FAS).² In 2008, Reynolds and co-workers conducted research study in regard to the biological evaluation of several analogs of 2-tosylnaphthalene-1,4-diol and 2-tosylbenzene-1,4-diol (**B**, **C**, **D**, etc.).³ They found that the sulfonyl group and naphthalene-1,4 diol/quinones were required for activity against all enzymes (Scheme 1). One important discovery is that 2-tosylnaphthaquinone (**B**) was observed to have quite good activity against *Mycobacterium tuberculosis FabH* (mtFabH).

During the past twenty years, transition metal catalyzed carbon—heteroatom bond formation has been useful as a means to construct organic molecules by cross coupling from C—X or direct C—H activation reactions.⁴ For the C—H activation reactions, quinones and naphthoquinones are important oxidants.⁵ However, it is easy for scientists to ignore some quinones and naphthoquinone derivatives, which are practical intermediates for many drugs, medicines, and insecticides and can be synthesized by transition metal catalyzed couplings of quinones with boron reagents,⁶ indoles,⁷ anilines,⁸ and others.⁹ For example, tosylquinone, as a potent inhibitor of FabH mentioned in the previous paragraph,

E-mail addresses: wangdw@jiangnan.edu.cn (D. Wang), yding@jiangnan.edu.cn (Y. Ding).



Scheme 1. Several FabH inhibitors.

was synthesized by two-step reactions from two main methods (Scheme 2).¹⁰ Although there have been attempts with one step, reactions usually need harsh terms or strongly acidic conditions.¹¹ Based on our research on the synthesis of metal complexes and their photophysical properties and catalytic reactivity,¹² herein, we report Pd-catalyzed direct C-sulfone formation by C—S coupling of quinones with sulfonyl chloride, which could be easily converted to sulfonyl-1,4-diols.

Initially, simple quinone and TsCl were selected as model substrates in order to check the proposed concept. The reaction was

^{*} Corresponding authors. Tel./fax: +86 510 85917763.



Scheme 2. C-sulfone forming reaction from quinones and naphthoquinones.

carried out in dichloromethane at room temperature. It was found that the desired product was separated at only 10%, but it implied that the direct C-sulfone formation of guinones with sulfonyl chloride would be feasible. Next, the screening of reaction conditions was conducted in order to obtain a better yield. Results are summarized in Table 1. Generally, the reaction has great dependence on solvent and base. When potassium carbonate was used as the base in DCE (1,2-dichloroethane), 89% yield of the target product was separated (Table 1, entry 12). We also screened other palladium catalysts in this reaction, the results showed that Pd(OAc)₂ is the best catalyst. Blank testing showed that the reaction could not happen without Pd catalyst (Table 1, entry 16).

Having established the optimal conditions: $Pd(OAc)_2$ (5%), K_2CO_3 as the base in DCE, we explored the reaction scope of the Pd-catalyzed direct C-sulfone coupling of quinones with sulfonyl chloride and the results are shown in Table 2. In general, all the substrates reacted smoothly to give the corresponding quinone derivatives. Moderate to excellent yields were obtained regardless of the steric hindrance of substituent groups in most cases.

Table 1

Screening of reaction conditions^a



Entry	Catalyst	Base	Solvent	Yield ^b (%)
1	$Pd(OAc)_2$	Na_2CO_3	CH ₂ Cl ₂	10
2	$Pd(OAc)_2$	Na_2CO_3	Toluene	<5
3	$Pd(OAc)_2$	Na_2CO_3	Benzene	<5
4	$Pd(OAc)_2$	Na_2CO_3	DMF	21
5	$Pd(OAc)_2$	Na_2CO_3	THF	18
6	$Pd(OAc)_2$	Na_2CO_3	Dioxane	14
7	$Pd(OAc)_2$	Na ₂ CO ₃	DCE	47
8	$Pd(OAc)_2$	NaHCO ₃	DCE	41
9	$Pd(OAc)_2$	KOH	DCE	27
10	$Pd(OAc)_2$	<i>t</i> BuOK	DCE	34
11	$Pd(OAc)_2$	Cs ₂ CO ₃	DCE	44
12	$Pd(OAc)_2$	K ₂ CO ₃	DCE	89
13	PdCl ₂	K ₂ CO ₃	DCE	41
14	$Pd(PPh_3)_4$	K ₂ CO ₃	DCE	23
15	Pd ₂ (dba) ₃	K ₂ CO ₃	DCE	37
16	-	K ₂ CO ₃	DCE	<5

^a Conditions: **1a** (0.5 mmol, 1.0 equiv), **2a** (1.5 equiv), Pd (5 mol %), base (1.5 equiv), 2 mL solvent, 24 h, reflux. ^b Isolated yields based on **1a**.

Table 2

Substrate expansion of quinones^a





^a Conditions: **1** (0.5 mmol, 1.0 equiv), **2** (1.5 equiv), Pd(OAc)₂ (5%), base (1.5 equiv), 2 mL DCE, 24 h, reflux.

Isolated vields based on 1.

^c 2,6-Dimethyl-quinone was used as substrate.

Next, the Pd-catalyzed C-sulfone formation of naphthoguinone substrates with sulfonyl chloride was explored under the optimized reaction conditions. A series of reactions could be successfully converted to afford corresponding naphthoquinone derivatives. As illustrated in Table 3, full conversions and high yields were achieved with different substituent groups. For the substrates bearing the alkyl chain, the yield was slightly affected. In the case of substrates with a strong electron withdrawing group, this methodology produced a disappointing result.

Table 3

Substrate expansion of naphthoquinones^a







^b Isolated yields based on **4**.

Additionally, naphthoquinone and quinone derivatives were very easily converted to corresponding sulfonyl-1,4-diols. The reaction was carried out under conditions of sodium borohydride in methanol, and thereby high yields of sulfonyl-1,4-diols were separated (Scheme 3).

Finally, a possible reaction mechanism for the reaction was proposed (Scheme 4). Initially, the oxidation addition of Pd with sulfonyl chloride was conducted, forming species **B** and followed with carbopalladation, whereby intermediate **C** was formed. After β -H



Scheme 3. Conversion of quinone derivatives to sulfonyl-1,4-diols.



Scheme 4. The proposed possible mechanism.

elimination, intermediate **C** released the coupling product to complete the catalytic cycle.

In conclusion, the direct C—S coupling of quinones and naphthoquinones with arylsulfonyl chloride by Pd-catalyzed C—H bond activation was developed with high yields for the first time. This methodology provides an effective, convenient method for the synthesis of arylsulfonyl-quinones and arylsulfonyl-1,4-diols. Further investigation to expand different substrate experiments and clearly understand this transformation is under way.

Acknowledgments

We gratefully acknowledge the financial support of this work by the National Natural Science Foundation of China (No. 21371080), the Natural Science Foundation of Jiangsu Province of China (BK20130125), 333 Talent Project of Jiangsu Province of China (BRA2012165), and MOE & SAFEA for the 111 Project (B13025).

Supplementary data

Supplementary data (detailed experimental procedures, IR, ¹H NMR, and ¹³C NMR for **3a–3f**, **5a–5f**, and **6a–6c**) associated with this article can be found, in the online version, at http://dx.doi.org/10.1016/j.tetlet.2014.08.023. These data include MOL files and InChiKeys of the most important compounds described in this article.

References and notes

- Lu, J. Z.; Lee, P. J.; Waters, N. C.; Prigge, S. T. Comb. Chem. High Throughput Screening 2005, 8, 15–26.
- (a) Tsay, J. T.; Oh, W.; Larson, T. J.; Jackowski, S.; Rock, C. O. J. Biol. Chem. 1992, 267, 6807–6814; (b) Qiu, X.; Janson, C. A.; Konstantinidis, A. K.; Nwagwu, S.; Silverman, C.; Smith, W. W.; Khandekar, S.; Lonsdale, J.; Abdel-Meguid, S. S. J. Biol. Chem. 1999, 274, 36465–36471; (c) Revill, W. P.; Bibb, M. J.; Scheu, A. K.; Kieser, H. J.; Hopwood, D. A. J. Bacteriol. 2001, 183, 3526–3530.
- Alhamadsheh, M.; Waters, N.; Sachdeva, S.; Lee, P.; Reynolds, K. Bioorg. Med. Chem. Lett. 2008, 18, 6402–6405.
- For selected recent reviews, see: (a) Balcells, D.; Clot, E.; Eisenstein, O. Chem. Rev. 2010, 110, 749–823; (b) Colby, D. A.; Bergman, R. G.; Ellman, J. A. Chem. Rev. 2010, 110, 624–655; (c) Coperet, C. Chem. Rev. 2010, 110, 656–680; (d) Mkhalid, I. A.; Barnard, J. H.; Marder, T. B.; Murphy, J. M.; Hartwig, J. F. Chem. Rev. 2010, 110, 890–931; (e) Arockiam, P. B.; Bruneau, C.; Dixneuf, P. H. Chem. Rev. 2012, 112, 5879–5918.
- For selected reviews and papers, see: (a) Chen, X.; Engle, K. M.; Wang, D.-H.; Yu, J-Q. Angew. Chem. Int. Ed. 2009, 48, 5094–5115; (b) Lyons, T. W.; Sanford, M. S. Chem. Rev. 2010, 110, 1147–1169; (c) Sun, C.-L.; Li, B.-J.; Shi, Z.-J. Chem. Commun. 2010, 677–685; (d) Mo, H.; Bao, W. J. Org. Chem. 2010, 75, 4856– 4859; (e) Wrigglesworth, J. W.; Cox, B.; Lloyd-Jones, G. C.; Booker-Milburn, K. I. Org. Lett. 2011, 13, 5326–5329; (f) Chu, J.-H.; Wu, C.-C.; Chang, D.-H.; Lee, Y.-M.; Wu, M.-J. Organometallics 2013, 32, 272–2827; (g) Zhang, C.; Ji, J.; Sun, P. J. Org. Chem. 2014, 79, 3200–3205; (h) Chu, J.-H.; Huang, H.-P.; Hsu, W.-T.; Chen, S.-T.; Wu, M.-J. Organometallics 2014, 33, 1190–1204.

- The coupling of quinones with boron reagents, see: (a) Molina, M. T.; Navarro, C.; Moreno, A.; Csaky, A. C. Org. Lett. 2009, 11, 4938–4941; (b) Demchuk, O. M.; Pietrusiewicz, K. M. Synlett 2009, 1149–1153; (c) Fujiwara, Y.; Domingo, V.; Seiple, I. B.; Gianatassio, R.; Bel, M. D.; Baran, P. S. J. Am. Chem. Soc. 2011, 133, 3292–3295; (d) Wang, J.; Wang, S.; Wang, G.; Zhang, J.; Yu, X.-O. Chem. Commun. 2012, 11769–11771; (e) Singh, P. P.; Aithagani, S. K.; Yadav, M.; Singh, V. P.; Vishwakarma, R. A. J. Org. Chem. 2013, 78, 2639–2648; (f) Komeyama, K.; Kashihara, T.; Takaki, K. Tetrahedron Lett. 2013, 54, 1084–1086; (g) Deb, A.; Manna, S.; Maji, A.; Dutta, U.; Maiti, D. Eur. J. Org. Chem. 2013, 5251–5256.
- (a) Pirrung, M. C.; Park, K.; Li, Z. Org. Lett. 2001, 3, 365–367; (b) Pirrung, M. C.; Deng, L.; Li, Z.; Park, K. J. Org. Chem. 2002, 67, 8374–8388; (c) Knölker, H.-J.; Fröhner, W.; Reddy, K. R. Synthesis 2002, 557–564; (d) Yadav, J. S.; Reddy, B. V. S.; Swamy, T. Tetrahedron Lett. 2003, 44, 9121–9124.
- Honraedt, A.; Callonnec, F. L.; Grognec, E. L.; Fernandez, V.; Felpin, F.-X. J. Org. Chem. 2013, 78, 4604–4609.
- (a) Zhang, H.-B.; Liu, L.; Chen, Y.-J.; Wang, D.; Li, C.-J. Adv. Synth. Catal. 2006, 348, 229–235; (b) Miyamura, H.; Shiramizu, M.; Matsubara, R.; Kobayashi, S. Angew. Chem. Int. Ed. 2008, 47, 8093–8095; (c) Lockner, J. W.; Dixon, D. D.; Risgaard, R.; Baran, P. S. Org. Lett. 2011, 13, 5628–5631; (d) Ilangovan, A.; Saravanakumar, S.; Malayappasamy, S. Org. Lett. 2013, 15, 4968–4971; (e) Zhang, S.; Song, F.; Zhao, D.; You, J. Chem. Commun. 2013, 4558–4560.
- (a) Carreno, M. C.; Ruano, J. L. G.; Urbano, A.; Remor, C. Z.; Arroyo, Y. J. Org. Chem. 2000, 65, 453–458; (b) Kraus, G. A.; Kim, I. J. Org. Chem. 2003, 68, 4517– 4518; (c) Tandon, V. K.; Singh, R. V.; Yadav, D. B. Bioorg. Med. Chem. Lett. 2004, 14, 2901–2904; (d) Tandon, V. K.; Chhor, R. B.; Singh, R. V.; Rai, S.; Yadav, D. B. Bioorg. Med. Chem. Lett. 2004, 14, 1079–1083; (e) Ryu, C.-K.; Choi, I. H.; Park, R. E. Synth. Commun. 2006, 36, 3319–3328; (f) Ryu, C.-K.; Shim, J.-Y.; Chae, M. J.; Choi, I. H.; Han, J. Y.; Jung, O.-J.; Lee, J. Y.; Jeong, S. H. Eur, J. Med. Chem. 2005, 40, 438–444; (g) Mulchin, B. J.; Newton, C. G.; Baty, J. W.; Grasso, C. H.; Martin, W. J.; Walton, M. C.; Dangerfield, E. W.; Plunkett, C. H.; Berridge, M. V.; Harper, J. L.; Timmer, M. S. M.; Stocker, B. L. Bioorg. Med. Chem. 2010, 18, 3238–3251.
- (a) Spinner, I. H.; Raper, W. D.; Metanomski, W. Can. J. Chem. 1963, 41, 483–494;
 (b) Bruce, J. M.; Lloyd-Williams, P. J. Chem. Soc., Perkin Trans. 1 1992, 2877–2884;
 (c) Allgeier, D. E.; Herbert, S. A.; Nee, R.; Schlecht, K. D.; Finley, K. T. J. Org. Chem. 2003, 68, 4988–4990;
 (d) Yadav, J. S.; Reddy, B. V. S.; Ramireddy, T. S. N. Synthesis 2004, 1849–1853.
- For selected recent papers in our group, see: (a) Li, L; Wu, F.; Zhang, S.; Wang, D.; Ding, Y.; Zhu, Z. Dalton Trans. 2013, 42, 4539–4543; (b) Yang, W.; Wang, D.; Song, Q.; Zhang, S.; Wang, Q.; Ding, Y. Organometallics 2013, 32, 4130–4135; (c) Zhang, S.; Ding, Y. Organometallics 2011, 30, 633–641; (d) Zhang, S.; Shi, L.; Ding, Y. J. Am. Chem. Soc. 2011, 133, 20218–20229; (e) Yang, W.; Zhang, S.; Ding, Y.; Shi, L.; Song, Q. Chem. Commun. 2011, 5310–5312.