

## Phosphorus, Sulfur, and Silicon and the Related Elements

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## Synthesis and Use of 6-, 7-and 8-Membered P-Heterocycles

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## Synthesis and Use of 6-, 7- and 8-Membered P-Heterocycles

GYÖRGY KEGLEVICH<sup>a</sup>, LÁSZLÓ TÖKE<sup>a</sup>, KINGA STEINHAUSER<sup>a</sup>,  
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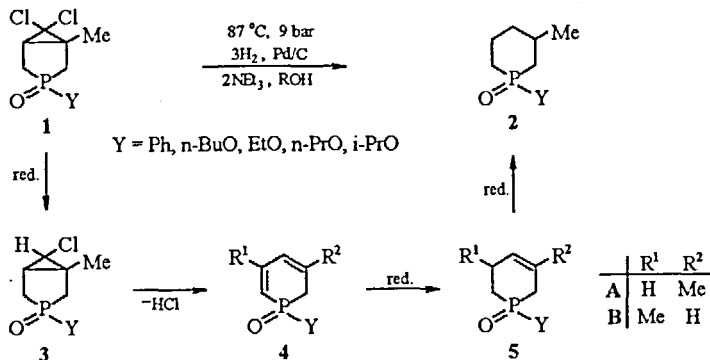
The synthesis of new families of P-heterocycles by reductive modifications, ring enlargement and Diels-Alder cycloaddition is described.

**Keywords:** P-heterocycles; ring enlargement; reductive methods; fragmentations

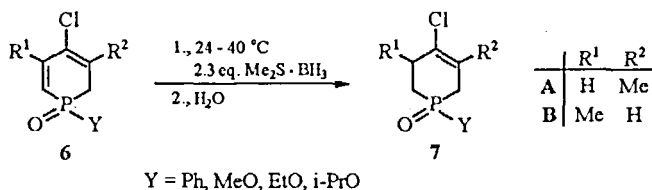
In this paper, the latest results of ours in P-heterocyclic chemistry are summarized.

### 1. Synthesis of six-membered P-heterocycles by reductive modifications

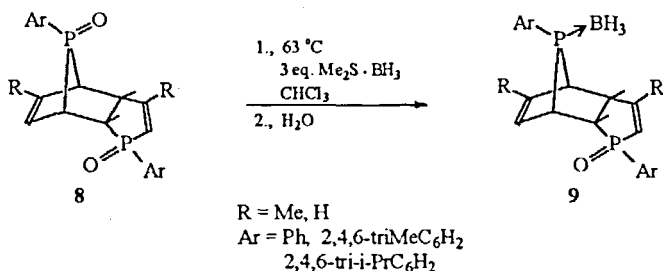
It was observed that the catalytic hydrogenation of phosphabicyclohexanes **1** led to the diastereomers of hexahydrophosphinine oxides **2**<sup>[1]</sup>. The formation of product **2** is the result of a consecutive series of reactions involving intermediates **3-5**. Decreasing the activity of the catalyst, we could achieve that monochloro species **3** should be the main component<sup>[2]</sup>.



We found that double-bonds with electron-withdrawing substituents at one end can be easily reduced by the borane-dimethyl sulfide (BMS) reagent. Thus, the reaction of the 2,3-dihydro-1H-phosphole oxides with BMS followed by hydrolysis afforded the corresponding tetrahydrophosphole oxides. Similar reaction of the 1,2-dihydrophosphinine oxides (6) gave the 1,2,3,6-tetrahydrophosphinines (7) in a regioselective manner<sup>[9]</sup>. The reductions take place certainly through hydroboration. Both the steric and the electronic reasons stand for the enhanced reactivity of the  $\alpha,\beta$ -double bond.



It was interesting to observe that the reaction of the phosphole-oxide dimers (8) with BMS resulted in a change in the functionality of the bridging P-center rather than selective reduction of the double-bond in the 2,3-dihydrophosphole moiety.

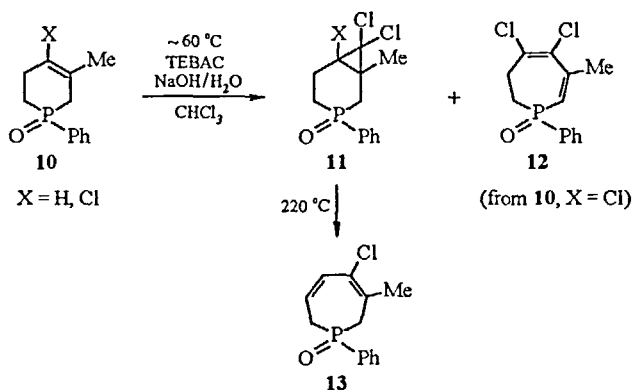


In the first step, the bridging phosphine oxide moiety in 8 is deoxygenated by the borane, then the phosphine so obtained is complexed by another unit of borane to give phosphine-borane 9.

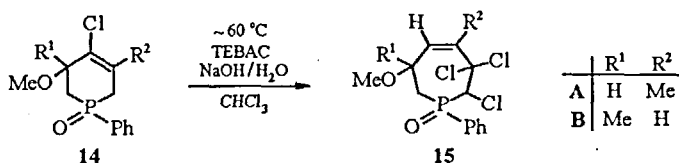
## 2. Synthesis of seven-membered P-heterocycles by ring enlargement of tetrahydrophosphinine oxides

We wished to utilize the tetrahydrophosphinine oxides in the synthesis of seven-membered P-heterocycles. The reaction of tetrahydrophosphinine oxides 10 with dichlorocarbene afforded phosphabicycloheptanes 11. In the reaction of the 4-chloro derivative (10, X = Cl) ca. 30% of 2,3-dihydrophosphepine oxide 12 was also formed. The thermolysis

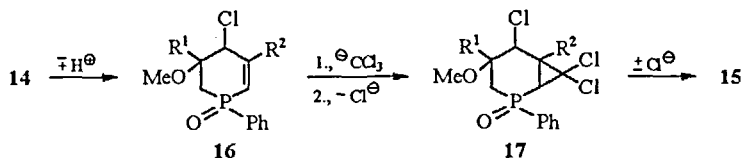
of the dichlorocarbene adducts (11) led, in both cases, to the same 2,7-dihydrophosphepine oxide (13)<sup>[4]</sup>.



Applying the dichlorocarbene ring enlargement to 3-methoxy-tetrahydrophosphinines 14, the reaction followed an unexpected route: 2,3,6,7-tetrahydrophosphepine oxides 15 could be isolated from the reaction mixture.

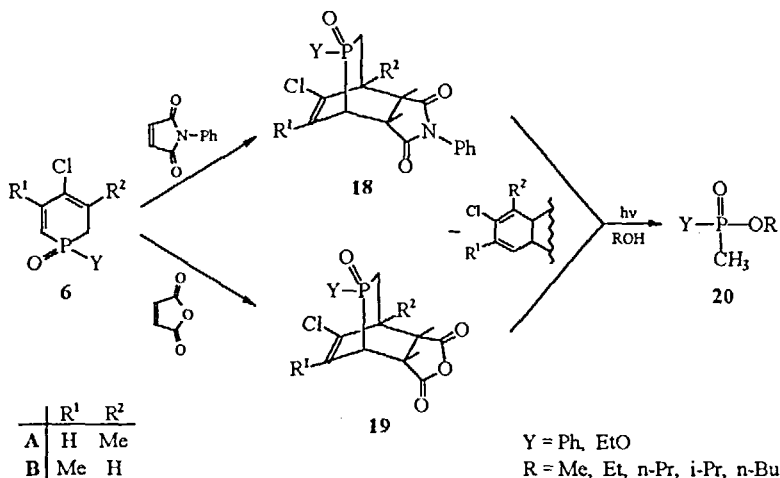


The key-step for the formation of products 15 is the isomerisation of 1,2,3,6-tetrahydrophosphinines 14 to 1,2,3,4-tetrahydro derivatives 16. The next step may involve Michael-type addition of a trichloromethyl anion at the end of the double-bond of 16 followed by cyclopropanation. Products 15 are formed by the opening of the cyclopropane ring in 17 by the effect of a chloride anion.



### 3. Synthesis of 2-phosphabicyclo[2.2.2]octene derivatives and their utilization in phosphorylations

We have synthesized Diels-Alder cycloadducts **18** and **19** to utilize them in the UV light-mediated phosphorylation of alcohols. The alcohols were added to the reaction mixture prior to irradiation.



Phosphinic and phosphonic esters **20** were obtained in good yields. The mechanistic investigations suggest that beside the E-A pathway involving methylene phosphine oxide (Y-P(O)(CH<sub>2</sub>)) as the intermediate, the A-E route is also realized. According to this, the alcohol is added to the phosphoryl group of the phosphabicyclooctene (**18** or **19**) to form an intermediate with a pentacoordinate phosphorus atom<sup>[5]</sup>.

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