

LETTERS  
 TO THE EDITOR

## Stereoselective Synthesis of 2,4-Dioxo-5-phenyl-1-phenylethylamino-4-phenoxy-1,3,4-diazaphospholidine

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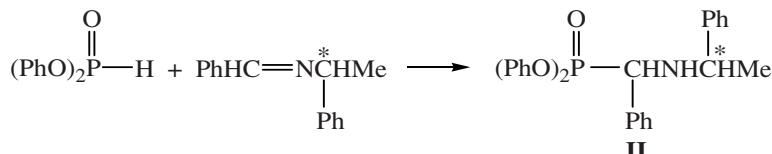
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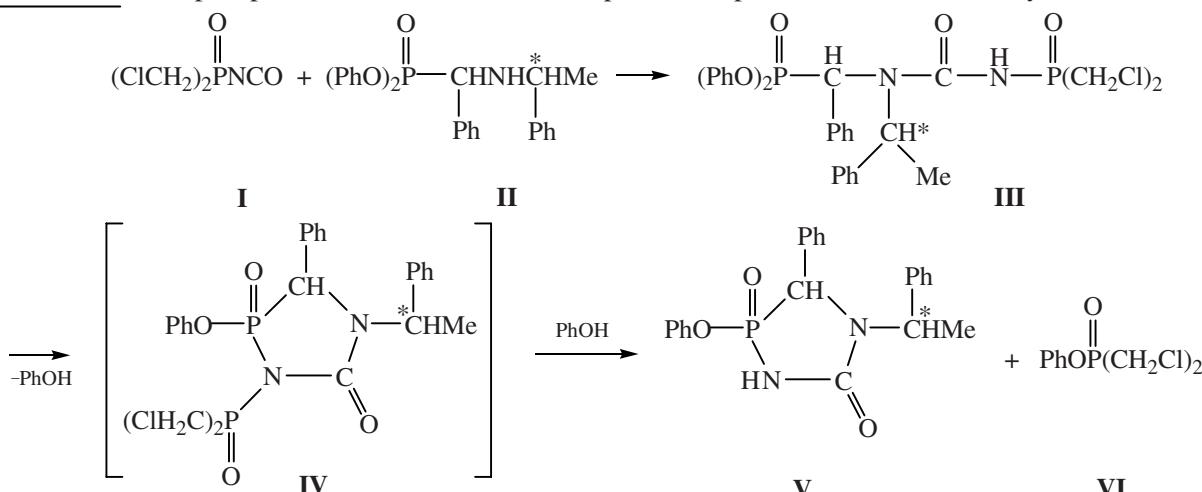
Recently we showed that diphenyl ( $\alpha$ -methylamino)benzylphosphonate easily added to phenyl isocyanate or phenyl isothiocyanate in the presence of a catalytic amount of triethylamine affording *N,N'*-substituted (thio)ureas; cyclization of the latter led to the formation of 1,3,4-diazaphospholidines [1]. Now we studied reaction of bis(chloromethyl)isocyanatophosphinate (**I**) with optically active diphenyl( $\alpha$ -methylamino)-

benzylphosphonate (**II**). Noteworthy that the formation of the latter compound was registered recently in the reaction of catalytic addition of diphenyl hydrogen phosphite to the optically active (*S*)(*–*)-benzal(1-phenyl)ethylamine (catalyst  $\text{BF}_3$ ) by  $^{31}\text{P}$  NMR spectroscopy as a mixture of stereoisomers which was not separated [2]. We obtained optically active aminophosphonate **II** in the same reaction using sodium phenolate as a catalyst:



The reaction mixture contained a mixture of amino-phosphonate **II** diastereomers in the ratio 74:26%, with chemical shifts of phosphorus nuclei 16.64 and 16.25 ppm respectively. By fractional crystallization one diastereomer of the phosphonate **II** was isolated in

enantiomerically pure form ( $[\alpha]_{D}^{20} -53.7$ ). By X-ray structural analysis we established that configurations of two chiral centers in this compounds were the same ( $S_C1S_C2$ ). Reaction of isocyanate **I** with aminophosphonate **II** proceeds stereoselectively.



In the  $^{31}\text{P}$  NMR spectrum of the reaction mixture appeared two signals of phosphorus nuclei, at 21.51 and 22.83 ppm, indicating the formation of diaza-phospholidine **V** as a diastereomeric mixture in the ratio 83:17 %, and a signal of phenyl bis(chloromethyl)phosphinate (**VI**) ( $\delta_{\text{P}}$  37.5 ppm). By the fractional crystallization we isolated enantiomerically pure 1-phenylethyl-2,4-dioxo-4-phenoxy-5-phenyl-1,3,4-dizaphospholidine (**V**) with  $[\alpha]_D^{20} +31.0$ . We suggest that the first step in the addition of aminophosphonate **II** to isocyanate **I** leads to the formation of diastereomerically uniform phosphorylated urea **III**. The subsequent intramolecular nucleophilic substitution at the tetracoordinated phosphorus atom forms a five-membered cyclic frame with appearance of the third chiral center on the phosphorus atom. Since the cyclisation process proceeds stereoselectively, we obtain final product as a diastereomeric mixture: **Va**,  $S_{\text{C}1}S_{\text{C}2}S_{\text{P}}$ , and **Vb**,  $S_{\text{C}1}S_{\text{C}2}R_{\text{P}}$ ; the prevailing form **Va** was isolated.

**Diphenyl ( $\alpha$ -phenylethylamino)benzylphosphonate (**II**).** To a solution of 4.9 g of diphenylphosphite in 20 ml of anhydrous acetonitrile was added at stirring 4.38 g of (*S*)-(-)-benzalphenylethylimoine and 1 drop of sodium phenolate. The reaction mixture was heated for 4.5 h at 80°C, then the solvent was removed and the residue was recrystallized from ether. 0.95 g (10%) of phosphonate **II** was obtained, mp 122–123°C. IR spectrum (KBr),  $\nu$ ,  $\text{cm}^{-1}$ : 1195, 1220 (P=O–Ph), 1275 (P=O), 1590 (Ph), 3325 (NH).  $^1\text{H}$  NMR spectrum  $[(\text{CD}_3)_2\text{CO}]$ ,  $\delta$ , ppm ( $J$ , Hz): 1.4 d (3H, MeC,  $^3J_{\text{HH}}$  6.3), 3.73 q (1H, NCH,  $^3J_{\text{HH}}$  6.7), 4.16 d (1H, PCH,  $^2J_{\text{PH}}$  23.6), 7.3 m (20 H, Ph).  $^{31}\text{P}$  NMR spectrum,  $\delta_{\text{P}}$ , ppm: 16.55.  $[\alpha]_D^{20} -53.7$ . Found, %: C 72.85; H 6.07; N 3.33; P 6.88.  $\text{C}_{27}\text{H}_{26}\text{NO}_3\text{P}$ . Calculated, %: C 73.12; H 5.91; N 3.16; P 6.99.

**2,4-Dioxo-5-phenyl-1-phenylethylamino-4-phenoxy-1,3,4-diazaphospholidine (**Va**).** To a solution of 0.5 g of aminophosphonate **II** in 20 ml of anhydrous acetonitrile was added at stirring 0.21 g of bis(chloromethyl)isocyanatophosphinate (**I**). After 45 days a crystalline precipitate separated and was filtered off. After triple recrystallization from acetonitrile 0.2 g (45%) of diastereomer **Va** was obtained, mp 182–183°C. IR spectrum (KBr),  $\nu$ ,  $\text{cm}^{-1}$ : 1190, 1215 (P=O–Ph), 1255 (P=O), 1595 (Ph), 1695 (C=O), 3070 (NH).  $^1\text{H}$  NMR spectrum  $[(\text{CD}_3)_2\text{CO}]$ ,  $\delta$ , ppm ( $J$ , Hz): 1.62 d (3H, MeC,  $^3J_{\text{HH}}$  4.6), 4.86 q (1H, NCH,  $^3J_{\text{HH}}$  5.0), 5.07 d (1H, PCH,  $^2J_{\text{PH}}$  13.1), 7.2 m (15H, Ph), 8.54 s (1H, NH).  $^{31}\text{P}$  NMR spectrum,  $\delta_{\text{P}}$ , ppm: 21.60.  $[\alpha]_D^{20} +31.0$ . Mass spectrum,  $m/z$ : 392 [ $M^+$ ]. Found, %: C 67.05; H 5.16; N 6.80; P 7.72.  $\text{C}_{22}\text{H}_{21}\text{N}_2\text{O}_3\text{P}$ . Calculated, %: 67.33; H 5.39; N 7.13; P 7.89.

IR spectra were registered on a UR-20 spectrophotometer in the range 400–3600  $\text{cm}^{-1}$  in mineral oil.  $^1\text{H}$  NMR spectra were registered on a Bruker WM-250 instrument (250.132 MHz), internal reference TMS. The  $^{31}\text{P}$  NMR spectra were registered on a Fourier NMR spectrometer Bruker MSL-400 (162.0 MHz).

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#### REFERENCES

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