

Tetrahedron Letters 39 (1998) 3017-3020

TETRAHEDRON LETTERS

A Novel Protocol for the Stereoselective Synthesis of Variously Substituted (Z)-5-Ylidene-5H-furan-2-ones

Renzo Rossi,*a Fabio Bellina^a and Luisa Mannina^b

Dipartimento di Chimica e Chimica Industriale, Università di Pisa, Via Risorgimento 35, I-56126 Pisa, Italy,^a and Corso di Laurea in Scienze Ambientali, Università del Molise, Via Mazzini 8, I-86170 Isernia, Italy^b

Received 8 January 1998; revised 30 January 1998; accepted 6 February 1998

Abstract: The Pd(II)- or Ag(I)-catalyzed lactonization of easily available (E)-4-(1-alkynyl)-2bromopropenoic acids provides (Z)-3-bromo-5-ylidene-5H-furan-2-ones, 5. These compounds, which represent an unpreviously reported class of (Z)-alkylidenebutenolides, are able to undergo Pd-catalyzed cross-coupling reactions with arylzinc halides, tetraalkylstannanes or alkenylstannanes to provide the corresponding 3-substituted (Z)-5-ylidene-5H-furan-2-ones, 1. The new procedure for the preparation of compounds 1 has been employed in a new synthesis of the butter flavour component bovolide. © 1998 Elsevier Science Ltd. All rights reserved.

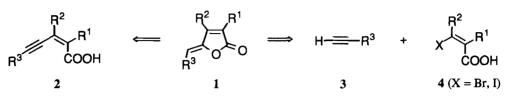
3-Substituted and 3,4-disubstituted (Z)-5-ylidene-5*H*-furan-2-ones, 1, are synthetic targets for which there is a continuous interest because they include a number of natural products which display a wide range of biological activities.¹ Very simple examples of these naturally-occurring substances include tetrenolin, which displays antibiotic activity,² the furanosesquiterpenoid freelingyne,³ the butter flavour component bovolide,⁴ and nostoclides I and II, which display cytotoxic activity.⁵



Since most of the methods developed earlier for the synthesis of compounds 1 produce Z and E mixtures,⁶ in recent years several highly selective organometallic methods have been developed.^{1a} Among these those which have proven to be very synthetically useful are based either on the Ag- ot the Pd-catalyzed lactonization of suitable substituted (Z)-2-en-4-ynoic acids, 2,⁷ or a tandem process involving a Pd-catalyzed cross-coupling between 1-alkynes, 3, and suitable substituted (Z)-3-halopropenoic acids, 4, under the Sonogashira conditions,⁸ and a subsequent Pd-catalyzed lactonization of the resulting (Z)-2-en-4-ynoic acids (Scheme 1).⁹ Obviously, these synthetic strategies afford compounds 1 characterized by a number of carbon atoms identical to that of compounds 2 which are used as starting materials or are formed *in situ* from 3 and 4.

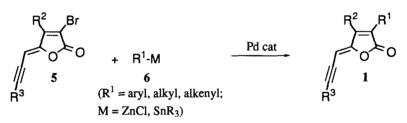
Unfortunately, in the case of compounds 1 characterized by a functionalized carbon chain in their 3-position (e.g. the (E)-3-hydroxy-1-propenyl unit), these strategies require the use of not easily available starting materials.

Scheme 1



We surmised that a solution to this problem might be represented by a convenient synthesis of (Z)-3bromo-5-ylidene-5*H*-furan-2-ones, 5, and the subsequent use of these new derivatives as electrophiles in Pdcatalyzed reactions with alkenyl-, aryl or alkylmetal derivatives which could contain a functional group (Scheme 2). We now wish to report some preliminary results obtained in the study on the synthesis of compounds 5 and the reactions of these versatile new reagents.

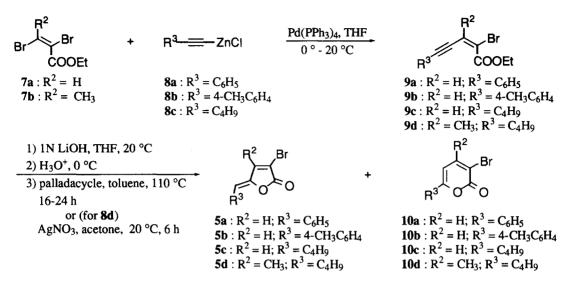
Scheme 2



Thus, according to a general procedure which we had previously developed for the regioselective and stereospecific monoarylation, monoalkynylation and monoalkylation of stereodefined 2,3-dibromo-2-alkenoates, ¹⁰ we prepared stereoisomerically pure compounds **9a**, **9b**, **9c** and **9d** in 64, 68, 80 and 49 % yield, respectively, by reaction of the corresponding (E)-2,3-dibromo-2-alkenoates, **7a**^{10b} and **7b**^{10c}, respectively, with 1.3 equiv of **6a**, **6b** and **6c**, respectively, in THF at 0-20 °C for 24-36 h in the presence of 5 mol % Pd(PPh₃)₄. (Scheme 3).

Compounds **9a-d** were then converted into the desired (Z)-3-bromo-5-ylidene-5*H*-furan-2-ones, **5a-d**, by saponification with 1N LiOH in THF at 20 °C followed by acidification and lactonization of the crude carboxylic acids so obtained by heating their deareated toluene solutions under argon at 110 °C for 16-24 h in the presence of 5 mol % *trans*-di(μ -acetato)-bis[(di-o-tolylphosphino)benzyl]dipalladium(II).¹¹ Compounds **5a**, **5b**, **5c** and **5d** were so obtained in 45, 36, 50 and 25 % yield, respectively (Scheme 3).¹² It is interesting to note that, analogously to what observed by Kotora and Negishi^{7b} for the synthesis of γ -alkylidenebutenolides by lactonization of 3-aryl substituted (Z)-2-en-4-ynoic acids in the presence of Pd(PPh₃)4, the new protocol that we used for the cyclization of the crude (*E*)-2-bromo-2-en-4-ynoic acids which derived from **9a-c**, afforded compounds **5a-c** contaminated by *ca*. 3-6 % of the corresponding 3-bromo-2*H*-pyran-2-ones, **10a-c**. However, quite surprisingly, the cyclization of the crude carboxylic acid derived from **9d** provided a mixture of **5d** and **10d** in a 38.5 : 61.5 ratio, respectively. Purification of this mixture by MPLC on silica gel allowed to isolate **5d** and **10d** in 25 and 45 % yield, respectively. Nevertheless, when the crude carboxylic acid derived from **9d** was reacted in acetone at 20 °C for 6 h in the presence of 20 mol % AgNO₃, a mixture of **5d** and **10d** in a *ca*. 79 : 21 ratio, respectively, was obtained and pure **5d** could be isolated in 52 % yield (Scheme 3).

Scheme 3



With compounds 5 now readily available on a multigram scale, some their uses as synthetic equivalents for incorporation of a (Z)-5-ylidene-5H-furan-2-one unit were investigated. It was so found that reaction of 5a with 1.5 equiv of 4-fluorophenylzinc chloride, 6a, in THF at 65 °C for 6h provided 1a in 67 % yield (entry 1, Table).

Table. Palladium-catalyzed cross-coupling reaction between compounds 5 and organometallic reagents.

		R^2 Br O R^3 5	+ R ¹ -M 6	P	d cat	/ R	$ \begin{array}{c} $			
Entry	Compound	Organometallic	Catalyst	Solvent	Reaction	Product				Yield
	5	reagent	system		conditions			_		(%)
		6			(h / °C)	1	R ¹	R ²	R ³	
1	5a	4-F-C6H4ZnCl	Pd(PPh3)4	THF	6/65	1a	4-F-C6H4	Н	C ₆ H ₅	67
2	5b	6a Bu₃Sn OH 6b	PdCl ₂ (PhCN) ₂ AsPh3, CuI	NMP	40 / 70	1b	√∕∕ОН	н	C4H9	36
3	5c		PdCl ₂ (PhCN) ₂ AsPh ₃ , CuI	NMP	23 / 80	1c	CH ₃	н	CH3	92
4	5d		PdCl ₂ (PhCN) ₂	NMP	72/80	1d	CH3	CH3	C4H9	54

Moreover, treatment of 5b with 1.5 equiv of (E)-3-hydroxy-1-propenyltributyltin, 6b, in NMP at 70 °C for 40 h, in the presence of 5 mol % PdCl₂(PhCN)₂, 10 mol % AsPh₃ and 10 mol % CuI provided 1b in 36 % yield (entry 2, Table). Interestingly, a much better yield into the desired cross-coupled product was obtained in a similar Pd-catalyzed reaction between 5c and 3 equiv of tetramethyltin, 6c. In fact, this reaction provided 1c in 92 % yield (entry 3, Table). Finally, treatment of 5d with 6c under experimental conditions similar to those employed for the synthesis of 1c, allowed to obtain bovolide, 1d,¹⁴ in 54 % yield (entry 4, Table).

AsPh₃, CuI

6c

In conclusion, it has been shown that the Pd(II)- or Ag(I)-catalyzed lactonization of easily available (E)-4-(1-alkynyl)-2-bromopropenoic acids provides (Z)-3-bromo-5-ylidene-5H-furan-2-ones, 5. These compounds, which represent an unpreviously reported class of (Z)- γ -ylidenebutenolides, have been shown to be able to undergo Pd-catalyzed cross-coupling reactions with organozinc or organotin derivatives to provide 3-aryl, 3-(1alkenyl) and 3-methyl substituted (Z)-5-ylidene-5H-furan-2-ones, 1. Moreover, the new procedure for the synthesis of compounds 1 has been used in a new synthesis of bovolide.

Acknowledgements. We gratefully acknowledge the financial support from the Progetto Finalizzato Beni Culturali (CNR, Roma) and the Ministero dell'Università e della Ricerca Scientifica e Tecnologica (MURST). We wish also to thank Prof. Annalaura Segre for the use of NMR facilities of the Servizio NMR dell'Area della Ricerca di Roma (CNR, Roma).

References and Notes

- (a) Negishi, E.; Kotora, M. Tetrahedron 1997, 53, 6707-6738; (b) Knight, D. W. Contemp. Org. Synth. 1. 1994. J. 287-315.
- Gallo, G. G.; Coronelli, C.; Vigevani, A.; Lancini, G. C. Tetrahedron 1969, 25, 5677-5680. 2
- 3. (a) Massy-Westropp, R. A.; Reynolds, G. D.; Spotswood, T. M. Tetrahedron Lett. 1966, 1939-1946; (b) Knight, D. W.; Pattenden, G. J. Chem. Soc. Perkin Trans I 1975, 641-644.
- Lardelli, G.; Dijkstra, G.; Harkes, P. D.; Boldingh, J. Rec. Trav. Chim. Pays Bas 1966, 85, 43-55. 4.
- 5.
- Yang, X.; Shimizu, Y.; Steiner, J. R.; Clardy, J. *Tetrahedron Lett.* **1993**, *34*, 761-764. For reviews on this subject, see: (a) Rao, Y. S. *Chem. Rev.* **1976**, *76*, 625-694; (b) Rao, Y. S. *Chem. Rev.* **1964**, *64*, 353-388; (c) Ref 1b; (d) Avetisyan, A. A.; Danggyan, M. T. Usp. Khim. **1977**, *46*, 1250-1278. For leading references on this subject, see: (a) Ref 1a; (b) Kotora, M.; Negishi, E. Synthesis **1997**, 121-6.
- 7. 129; (c) Mori, H.; Kubo, H.; Hara, H.; Katsumura, S. Tetrahedron Lett. 1997, 38, 5311-5312.
- 8. Sonogashira, K.; Tohda, Y.; Hagihara, N. Tetrahedron Lett. 1975, 4467-4470.
- (a) Ref 7b; (b) Kotora, M.; Negishi, E. Tetrahedron Lett. 1996, 37, 9041-9042.
- (a) Bellina, F.; Carpita, A.; De Santis, M.; Rossi, R. Tetrahedron Lett. 1994, 35, 6913-6916; (b) Rossi, R.; 10. Bellina, F.; Carpita, A.; Gori, R. Gazz. Chim. Ital. 1995, 125, 381-382; (c) Rossi, R.; Bellina, F.; Carpita, A. Mazzarella, F. Tetrahedron 1996, 52, 4095-4110; (d) Rossi, R.; Bellina, F. Bechini, C.; Mannina, L.; Vergamini, P. Tetrahedron 1998, 54, 135-156.
- 11. Recently, we successfully used this complex described by Herrmann et al. [Herrmann, W. A.; Brossner, C.; Öfele, K.; Reisinger, C-P.; Priermeier, T.; Beller, M.; Fischer, H. Angew. Chem. Int. Ed. Engl. 1995, 34, 1844-1848] for the cyclization of (E)-2-(1-alkynyl)-3-aryl/alkylpropenoic acids to the corresponding (E)-3-vlidene-3H-furan-2-ones (see: Ref 10d).
- 12. All the new products in this study gave satisfactory spectral and microanalytical data. Some characteristic All the new products in this study gave satisfactory spectral and increating from data. Come characteristic data for compounds **5a**, **5b**, **5c** and **5d** are as follows. **5a**: orange liquid; MS, m/z (%): 252 (26), 250 (24), 115 (100), 105 (18), 77 (39); ¹H NMR (CDCl₃, 600 MHz): δ 7.77 (br d, J = 7.7 Hz, 2H), 7.57 (br s, J = 1.5 Hz, 1H), 7.39 ((m, 2H), 7.37 (m, 1H), 6.07 (br s, 1H). ¹³C NMR (CDCl₃, 150 MHz): δ 165.58, 146.94, 142.90, 132.43; 130.87; 129.75, 128.99, 114.91, 111.13.5b: m.p. 159-161 °C; MS, m/z (%): 266 (72), 264 (72), 157 (51), 129 (100), 104 (74). IR (KBr): 1756, 987, 930, 903 cm⁻¹; ¹H NMR (CDCl₃, 600 MHz): 87.67 (br s, 2H), 7.56 (br s, 1H), 7.21 (br s, 2H), 6.05 (br s, 1H), 2.37 (s, 3H); ¹³C NMR (CDCl₃, 150 MHz): 165.78, 146.37, 142.91, 140.38, 130.91, 129.91, 129.70, 115.17, 110.42, 21.56. 5c: yellow liquid; MS, m/z (%): 232 (29), 230 (31), 123 (71), 81 (58), 41 (100); ¹H NMR (CDCl₃, 600 MHz): 7.43 (s, 1H), 5.38 (t, J = 7.9 Hz, 1H), 2.35 (dt, J = 7.9 and 7.9 Hz, 2H), 1.44 (quint, J = 7.9 Hz, 2H), 1.33 (sext, (5, 11), 5.36 (t, J = 7.9 Hz, 1H), 2.35 (tt, J = 7.9 and 7.9 Hz, 2H), 1.44 (quint, J = 7.9 Hz, 2H), 1.35 (sext, J = 7.9 Hz, 2H), 0.89 (t, J = 7.9 Hz, 3H); ¹³C NMR (CDCl₃, 150 MHz): 165.61, 147.96, 141.51, 118.79, 111.63, 30.79, 25.93, 22.26, 13.65. **5d**: yellow liquid; MS, m/z (%): 246 (26), 244 (27), 190 (1007, 67 (96), 55 (73). IR (neat): 1780, 996, 754 cm⁻¹; ¹H NMR (CDCl₃, 200 MHz): δ 5.46 (t, J = 7.9 Hz, 1H), 2.37 (dt, J = 7.9 and 7.9 Hz, 2H), 2.14 (s, 3H), 1.60-1.50 (br m, 4H), 0.92 (t, J = 7.9 Hz, 3H); ¹³C NMR (CDCl₃, 50 MHz): δ 165.34, 151.18, 149.28, 114.78, 110.12, 30.98, 25.67, 22.47, 13.79, 17.68.
- 13. We also attempted the synthesis of 5a from (E)-2,3-dibromopropenoic acid and phenylacetylene using PdCl₂(PPh₃)₂, PPh₃, CuI and Et₃N according to the general procedure described by Lu et al. for the synthesis of (Z)-y-ylidenebutenolides [Lu, X.; Huang, X.; Ma, S. Tetrahedron Lett. 1993, 34, 5963-5966]. Unfortunately, this reaction did not provide 5a but afforded 1,4-diphenylbutadiyne and a compound to
- which, on the basis of its MS spectrum, we tentatively assigned the structure of 1,2,4-triphenylbenzene. For previous syntheses of bovolide, see: (a) Ref 4; (b) Coperét, C.; Sugihara, T.; Wu, G.; Shimoyama, I.; Negishi, E. J. Am. Chem. Soc. 1995, 117, 3422-3431; (c) Wulff, W. D.; Gilbertson, S. R.; Springer, J. P. 14. J. Am. Chem. Soc. 1986, 108, 520-522.