

Copper-Catalyzed Cross-Coupling of 1-Iodoalkynes with Organostannanes

Suk-Ku Kang,* Won-Yeob Kim, Xianghua Jiao

Department of Chemistry, Sung Kyun Kwan University, Natural Science Campus, Suwon 440-746, Korea
 Fax +82(331)290-7079; E-mail: skkang@chem.skku.ac.kr

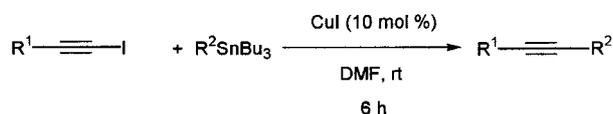
Received 5 January 1998; revised 23 February 1998

Abstract: Copper-catalyzed cross-coupling of 1-iodoalkynes with organostannanes was readily achieved in the presence of CuI (10 mol%) in DMF at room temperature for 6 hours by adding 1-iodoalkynes slowly to organostannanes.

Key words: copper-catalyzed cross-coupling, 1-iodoalkynes, organostannanes, 1-substituted acetylenes

The conjugated enyne moiety is an important structural unit in a number of natural products. The enynes have been generally prepared by the coupling reactions of terminal alkynes with aryl or vinyl halides, which is known as the Sonogashira reaction.¹ The synthesis of enynes can be also accomplished using haloalkynes with aryl and alkynyl copper compounds.²

The palladium-catalyzed cross-coupling (Stille reaction)³ of organic electrophiles such as vinyl halides or triflates with organostannanes has become an extremely powerful tool in organic synthesis. The cross-coupling of vinyl iodides or vinyl triflates with alkynylstannanes to obtain enynes in the presence of palladium catalyst has been reported.⁴ However, the cross-coupling of haloalkynes with organostannanes to synthesize enynes has rarely been reported.⁵ In 1990, Liebeskind et al.⁶ reported the first example of the palladium-catalyzed cross-coupling of 1-iodohexyne with cyclobutenylorganostannane. Recently, palladium-catalyzed cross-coupling of 1-iodoalkynes with (*E*)- α -selanylvinylstannanes was reported by Huang et al.⁷ In connection with our programs to utilize copper as a catalyst for cross-coupling,⁸ we have investigated the copper iodide catalyzed cross-coupling of 1-iodoalkynes with organostannanes. Here we report copper(I) iodide catalyzed cross-coupling of 1-iodoalkynes with aryl-, alkynyl-, and allylstannanes.

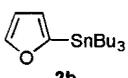
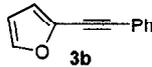
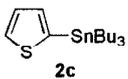
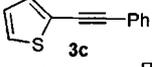
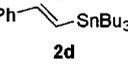
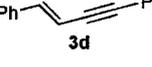
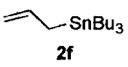
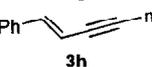
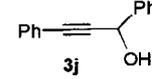
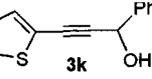
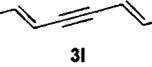


Initially, we examined the copper-catalyzed cross-coupling of 1-iodo-2-phenylacetylene (**1a**) and 2-furylstannane **2b** to determine the optimum reaction conditions. A series of experiments were performed. Of the catalysts tested CuI (10 mol%) was the most suitable and better than PdCl₂ in NMP at room temperature. As solvents among NMP, DMF, THF, THF/NMP (1:1), and 1,4-dioxane, NMP and DMF were preferable. The addition of additives to reduce homocoupling, such as LiCl and NaCl, did not improve the yields of the cross-coupled products, even if the addition of KCl and CsCl increased the yields of the cross-coupled products with limited success. Without additives when the above reaction was conducted in

the presence of CuI (10 mol%) in DMF at room temperature, the homocoupled products of bisphenylacetylene, bisfuran, and the cross-coupled product **3b** were formed in the ratio of 2:1:1. We assumed that 1-iodo-2-phenylacetylene (**1a**) is the reactive species which undergoes homocoupling. Finally, by adding **1a** slowly to organostannanes **2b** via syringe pump we could get the cross-coupled product **3b** as the major product. Thus, it is crucial to add 1-iodoalkynes slowly via a syringe pump to reduce homocoupling. The results of copper-catalyzed cross-coupling of 1-iodoalkynes with organostannanes are summarized in the Table. 1-Iodo-2-phenylacetylene (**1a**) was slowly added via a syringe pump to organostannane **2a** in DMF at room temperature for 6 hours to afford the cross-coupled product **3a**⁹ in 71% yield (entry 1, Table). The reaction of **1a** with 2-furylstannane **2b** proceeded smoothly to give 2-furyl-substituted alkyne **3b** in 75% yield by adding **1a** slowly (entry 2). In considering the mechanistic explanation, it is presumed that the slow addition of 1-iodoalkynes prevents the homocoupling of 1-iodoalkynes. As indirect evidence when the same reaction was conducted by the slow addition of organostannane **2b** to **1a**, the homocoupled product of **1a** was produced in ~25% yield along with 50% yield of the cross-coupled product **3b** and bisfuran (~25%). Accordingly, the iodoalkyne **1a** was subjected to coupling with 2-thienylstannane **2c** to give the product **3c**¹⁰ in 77% yield (entry 3). This coupling method was applied to alkenyl-, alkynyl-, and allylstannanes (entries 4–6). Thus, the iodoalkyne **1a** underwent facile cross coupling in the presence of CuI (10 mol%) with (*E*)- β -styrylstannane **2d**, 1-phenylethynylstannane **2e**, and allylstannane **2f** to afford the coupled products **3d**,¹¹ **3e**, and **3f** in 76, 80, and 90% yields, respectively (entries 4–6). For the alkyl-substituted 1-iodoalkyne **1b**, the reaction under the typical conditions with **2a** gave the phenyl-substituted alkyne **3g** in 88% yield (entry 7). Treatment of **1b** with alkenylstannane **2d** and alkynylstannane **2e** under the same conditions provided the coupled product **3h** and **3i** in 71 and 76% yields, respectively (entries 8 and 9). When 1-iodoalkyn-3-ol **1c** was reacted with **2a** and **2b**, the coupled products **3j** and **3k** were readily obtained (entries 10 and 11). Finally, 1-iodoalkyn-3-ene **1d** was smoothly coupled with the alkenylstannane **2d** to afford the yndiene compound **3l** in 82% yield (entry 12).

Reagent grade commercially available reagents and solvents were used. DMF was freshly distilled from CaH₂ prior to use. 1-Iodoalkynes **2a**, **2b**, and **2c** were prepared¹² from the corresponding 1-alkynes and BuLi followed by iodination with molecular iodine. The 1-iodoalkyne **2b** was prepared¹³ from the corresponding 1-alkyne by treatment with NIS and AgNO₃.

Table 1. Copper-Catalyzed Cross-Coupling of 1-Iodoalkynes with Organoastannanes^a

Entry	1-Iodoalkynes	Organostannanes	Products	Yield (%) ^b
1	PhC≡CI 1a	PhSnBu ₃ 2a	Ph—C≡C—Ph 3a	71
2	1a	 2b	 3b	75
3	1a	 2c	 3c	77
4	1a	 2d	 3d	76
5	1a	Ph—C≡C—SnBu ₃ 2e	Ph—C≡C—C≡C—Ph 3e	80
6	1a	 2f	 3f	90
7	n-BuC≡CI 1b	2a	Ph—C≡C—n-Bu 3g	88
8	1b	2d	 3h	71
9	1b	2e	Ph—C≡C—C≡C—n-Bu 3i	76
10	Ph(HCOH)C≡CI 1c	2a	 3j	72
11	1c	2b	 3k	76
12	 1d	2d	 3l	82

^a Reaction conditions: 1-iodoalkynes (1 equiv), organostannanes (1 equiv), CuI (10 mol %), DMF, r.t., 6 h.

^b The isolated yields.

Copper-Catalyzed Cross-Coupling of Organostannanes with 1-Iodoalkynes; 2-(2-Phenylethynyl)furan (**3b**); Typical Procedure:

To a stirred solution of 2-(tributylstannyl)furan (100 mg, 0.280 mmol), CuI (5.3 mg, 0.028 mmol) in DMF (3 mL) at r.t. was slowly added the 1-iodo-2-phenylacetylene (63.9 mg, 0.280 mmol) in DMF (3 mL) via syringe pump over 1.5 h. The mixture was stirred at r.t. for 6 h and extracted with Et₂O and washed with sat. KF (10 mL) and then brine (10 mL). The organic layer was dried (anhyd MgSO₄) and evaporated in vacuo. The crude product was separated by column chromatography (silica gel, hexanes, *R_f* 0.43) to afford **3b**; yield: 35.3 mg (75%).

¹H NMR (400 MHz, CDCl₃): δ = 6.44 (d, 1H), 6.67 (dd, 1H), 7.37 (m, 4H), 7.54 (m, 2H).

IR (neat): ν = 3105, 3056, 2953, 1500, 1415, 1070, 918, 690, 536 cm⁻¹.
MS (EI): *m/z* (%) = 169 (14), 168 (100), 140 (17), 139 (89), 114 (11), 70 (9), 63 (9).

Diphenylacetylene (**3a**):⁹

¹H NMR (400 MHz, CDCl₃): δ = 7.35 (m, 6H), 7.54 (m, 4H).

IR (KBr): ν = 3036, 1560, 1500, 1070, 918, 690, 536 cm⁻¹.

MS (EI): *m/z* (%) = 178 (100), 176 (20), 152 (9), 89 (13), 76 (12).

1-Phenyl-2-(2-thienyl)acetylene (**3c**):¹⁰

¹H NMR (400 MHz, CDCl₃): δ = 7.01 (m, 1H), 7.29 (m, 2H), 7.35 (m, 3H), 7.53 (m, 2H).

IR (neat): ν = 3058, 3022, 2954, 1600, 1381, 1215, 909, 735 cm⁻¹.
MS (EI): *m/z* (%) = 185 (11), 184 (100), 152 (13), 139 (30), 126 (9), 74 (11), 63 (11).

(*E*)-1,4-Diphenylbut-1-en-3-yne (**3d**):¹¹

¹H NMR (400 MHz, CDCl₃): δ = 6.40 (d, 1H, *J* = 16 Hz), 7.06 (d, 1H, *J* = 16 Hz), 7.33 (m, 6H), 7.38 (m, 2H), 7.62 (m, 2H).

IR (KBr): ν = 3041, 3010, 1605, 1471, 843 cm⁻¹.

MS (EI): *m/z* (%) = 205 (78), 203 (100), 102 (28).

1,4-Diphenylbuta-1,3-diyne (**3e**):

¹H NMR (400 MHz, CDCl₃): δ = 7.35 (m, 6H), 7.54 (m, 4H).

IR (KBr): ν = 3104, 2158, 1594, 1477, 920 cm⁻¹.

MS (EI): *m/z* (%) = 203 (16), 202 (100), 200 (23), 174 (3), 102 (22), 88 (13), 75 (6).

5-Phenylpent-1-en-4-yne (**3f**):

¹H NMR (400 MHz, CDCl₃): δ = 3.22 (m, 2H), 5.19 (dt, 1H), 5.43 (dd, 1H), 5.92 (m, 1H), 7.31 (m, 3H), 7.45 (m, 2H).

IR (neat): ν = 3082, 3059, 3018, 2982, 2921, 2886, 1641, 1598, 1489, 916 cm⁻¹.

MS (EI): *m/z* (%) = 143 (10), 142 (90), 141 (100), 115 (55), 89 (7), 63 (8).

1-Phenylhex-1-yne (**3g**):

¹H NMR (400 MHz, CDCl₃): δ = 0.96 (t, 3H, *J* = 7.3 Hz), 1.55 (m, 4H), 2.42 (t, 2H, *J* = 7.0 Hz), 7.34 (m, 5H).

IR (neat): $\nu = 3057, 1599, 1491, 1064, 909 \text{ cm}^{-1}$.
 MS (EI): m/z (%) = 159 (20), 158 (30), 143 (52), 129 (30), 116 (22), 115 (100), 113 (11), 102 (9).

(E)-1-Phenyloct-1-en-3-yne (3h):

$^1\text{H NMR}$ (400 MHz, CDCl_3): $\delta = 0.96$ (t, 3H, $J = 7.3$ Hz), 1.55 (m, 4H), 2.42 (t, 2H, $J = 7.0$ Hz), 6.17 (dt, 1H, $J = 16$ Hz), 6.88 (d, 1H, $J = 16$ Hz) 7.33 (m, 5H).

IR (neat): $\nu = 3062, 3010, 2932, 2872, 1486, 1463, 1404, 1104, 755, 689 \text{ cm}^{-1}$.

MS (EI): m/z (%) = 184 (34), 169 (9), 155 (20), 141 (100), 115 (59), 91 (20), 77 (10), 63 (10), 51 (7).

1-Phenylocta-1,3-diyne (3i):

$^1\text{H NMR}$ (400 MHz, CDCl_3): $\delta = 0.96$ (t, 3H), 1.55 (m, 4H), 2.42 (t, 2H), 7.34 (m, 5H).

IR (neat): $\nu = 3054, 2986, 2931, 1487, 1441, 1265, 738 \text{ cm}^{-1}$.

MS (EI): m/z (%) = 182 (26), 165 (76), 152 (53), 139 (100), 126 (40), 115 (36), 91 (21), 63 (10).

1,3-Diphenylprop-2-yn-1-ol (3j):

$^1\text{H NMR}$ (400 MHz, CDCl_3): $\delta = 2.36$ (s, 1H), 5.70 (s, 1H), 7.33 (m, 4H), 7.42 (m, 2H), 7.49 (m, 2H), 7.63 (m, 2H).

IR (neat): $\nu = 3377, 3118, 3063, 2199, 1707, 1490, 1286, 1032, 758, 694 \text{ cm}^{-1}$.

MS (EI): m/z (%) = 209 (11), 208 (74), 207 (100), 191 (21), 179 (67), 178 (64), 165 (29), 131 (24), 130 (39), 129 (43), 102 (41), 77 (46), 51 (15).

1-Phenyl-3-(2-thienyl)prop-2-yn-1-ol (3k):

$^1\text{H NMR}$ (400 MHz, CDCl_3): $\delta = 2.36$ (s, 1H), 5.71 (s, 1H), 6.99 (dd, 1H), 7.27 (m, 2H), 7.42 (m, 3H), 7.60 (m, 2H).

IR (neat): $\nu = 3300, 3055, 2986, 2189, 1637, 1450, 1266, 1030, 990, 737 \text{ cm}^{-1}$.

MS (EI): m/z (%) = 213 (11), 212 (82), 183 (77), 152 (14), 139 (21), 135 (100), 107 (16), 92 (19), 77 (67), 63 (37).

(E,E)-1,6-Diphenylhexa-1,5-dien-3-yne (3l):

$^1\text{H NMR}$ (400 MHz, CDCl_3): $\delta = 6.30$ (d, 2H, $J = 16$ Hz), 7.00 (d, 2H, $J = 16$ Hz), 7.37 (m, 10H).

IR (KBr): $\nu = 3063, 1807, 1550, 1500, 1069, 915, 705, 687, 524 \text{ cm}^{-1}$.

MS (EI): m/z (%) = 230 (42), 229 (39), 228 (34), 215 (16), 115 (100), 101 (11).

We acknowledge with thanks the generous financial support from KOSEF (97-0501-02-01-3), KOSEF-OCRC, and Ministry of Education (BSRI-97-3420).

- (1) Fleming, I. In *Comprehensive Organic Synthesis*, Vol. 3; Trost, B. M., Fleming, I., Eds.; Pergamon: Oxford, 1991; p 551–561 and p 521–549.
- (2) Oliver, R.; Walton, D. R. M. *Tetrahedron Lett.* **1972**, 5209.
- (3) Normant, J. F.; Commercon, A.; Villieras, J. *Tetrahedron Lett.* **1975**, 1465.
- (4) Commercon, A.; Normant, J. F.; Villieras, J. *Tetrahedron* **1980**, *36*, 1215.
- (5) Brown, H. C.; Molander, G. A. *J. Org. Chem.* **1981**, *46*, 645.
- (6) Stille, J. K. *Angew. Chem., Int. Ed. Engl.* **1986**, *25*, 508.
- (7) Mitchell, T. N. *Synthesis* **1992**, 803.
- (8) Scott, W. J.; Crisp, G. T.; Stille, J. K. *J. Am. Chem. Soc.* **1984**, *106*, 4630.
- (9) Rudisill, D. E.; Castonguay, L. A.; Stille, J. K. *Tetrahedron Lett.* **1988**, *29*, 1509.
- (10) The palladium-catalyzed coupling reaction of tributyl(prop-1-enyl)tin with functionalized styryl bromides is described: Zapata, A. J.; Ruiz, J. *J. Organomet. Chem.* **1994**, *479*, C6.
- (11) Liebeskind, L. S.; Fengl, R. W. *J. Org. Chem.* **1990**, *55*, 5359.
- (12) Ma, Y.; Huang, X. *Synth. Commun.* **1997**, *27*, 3441.
- (13) Kang, S.-K.; Yamaguchi, T.; Kim, T.-H.; Ho, P.-S. *J. Org. Chem.* **1996**, *61*, 9082.
- (14) Kang, S.-K.; Kim, T.-H.; Pyun, S.-J. *J. Chem. Soc., Perkin Trans. I* **1997**, 797.
- (15) Kang, S.-K.; Kim, J.-S.; Choi, S.-C. *J. Org. Chem.* **1997**, *62*, 4208.
- (16) Kang, S.-K.; Lim, K.-H.; Ho, P.-S.; Kim, W.-Y. *Synthesis* **1997**, 874.
- (17) *The Merck Index*; Vol. 11, p 9426.
- (18) Brittain, J. M.; Jones, R. A.; Taheri, S. A. N. *Tetrahedron* **1992**, *48*, 7609.
- (19) Thomas, S.; Ariel, H.; Moris, S. E. *J. Am. Chem. Soc.* **1995**, *117*, 6364.
- (20) Kim, J.-I.; Lee, J. T.; Yeo, K.-D. *Bull. Korean Chem. Soc.* **1985**, *6*, 366.
- (21) Brandsma, L. In *Preparative Acetylene Chemistry*; Elsevier: Amsterdam, 1971; p 99.
- (22) Chauhan, Y. S.; Chandraratna, R. A. S.; Miller, D. A.; Kondratte, R. W.; Reischl, W.; Okamura, W. H. *J. Am. Chem. Soc.* **1985**, *107*, 1028.
- (23) Hofmeister, H.; Annen, K.; Laurent, H.; Wiechert, R. *Angew. Chem., Int. Ed. Engl.* **1984**, *23*, 727.