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Application of Copper Catalysis in a One-Pot Procedure for 1-Alkyl-3-methoxy-2-methylthiopyrroles Starting from Methoxyallene and Alkyl Isothiocyanates

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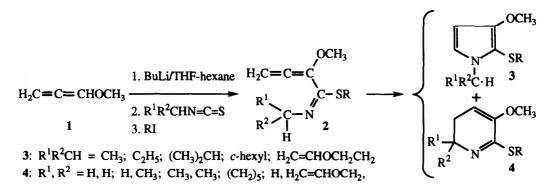
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Abstract: 1-Alkyl-3-methoxy-2-methylthiopyrroles are obtained with high purities and in high yields by Cu(I)halide-catalyzed cyclization of the products from addition of 1-lithiomethoxyallene to alkyl isothiocyanates and subsequent methylation. © 1998 Elsevier Science Ltd. All rights reserved.

Methoxyallene $CH_3OCH=C=CH_2$, readily available by base-catalyzed isomerization of methyl propargyl ether¹⁻³ has been used as starting compound in a wide variety of syntheses (for a recent review see⁴).



We recently reported the simultaneous formation of 2,3-dihydropyridines 4 and derivatives of pyrrole 3 by heating of the Schiff's bases 2 obtained by quenching of the adducts from 1-lithio-methoxyallene and alkyl isothiocyanates with alkyl iodide⁵. These cyclizations proceeded without any added catalyst. The ratio of 3 and 4 strongly depended upon the alkyl group of the isothiocyanate. Starting with CH₃N=C=S the major product (~80 rel.%) was the pyrrole derivative, but in the cases of $c-C_6H_{11}-N=C=S$ and $(CH_3)_2CH-N=C=S$ the 2,3-dihydropyridines predominated.

We here report the specific formation of pyrrole derivatives 3 by treatment of 2 with a catalytic amount of

copper(I)bromide or iodide at 20 to ~45 °C. The synthesis of 3 can be carried out in a one-pot procedure with good to excellent yields. The nickel(II) catalysts NiCl₂.dppf and NiCl₂.dppp and the palladium catalysts $PdCl_2(PPh_3)_2$ and $Pd(PPh_3)_4$ showed no significant activity in qualitative experiments.

EXPERIMENTAL PROCEDURE

Methoxyallene (0.12 mol) was added at -100 °C in one portion to a mixture of 70 ml of THF and 65 ml (0.10 mol) of hexane solution of *n*-BuLi. After allowing the temperature to rise to -70 °C, the solution was cooled to -90 °C and a mixture of 0.10 mol of ethyl isothiocyanate and 15 ml of THF was added portionwise over 5 min while maintaining the temperature between -70 and -80 °C. After an additional 10 min (at -70 °C) 0.14 mol of methyl iodide was added and the temperature was allowed to rise to +10 °C. The solution was divided in two equal parts: one of them was treated with water and after the usual work-up the product was distilled *in vacuo*, affording a 1 : 3 mixture of 3 and 4 (R = Me, R¹ = Me, R² = H) in -75% yield. The mixture was separated into the components by shaking the ethereal solution with cold, 2 M hydrochloric acid and treating the acidic layer with aqueous potassium hydroxide. Extraction with ether and distillation gave the pure (99%) dihydropyridine 4 (R = Me, R¹ = Me, R² = H), b.p. 90 °C/2 mm Hg, n²⁰_D 1.5376 in ~50% yield. NMR-spectrum (300 MHz, CD₃Cl: 1.90, 2.15 (m, H-3), 2.18 (s, SMe), 3.47 (m, H-2), 3.49 (s, OMe) and 4.92 (dd, H-4) ppm.

To the other part copper(I)bromide (1.5 g) or copper(I)iodide (2.5 g) was added with efficient stirring. The mixture was stirred for half an hour at 15-20 °C, after which the temperature was raised to 45 °C. After stirring for an additional 60 min at 40-50 °C a solution of 10 g of potassium cyanide and 15 g of ammonium chloride in 100 ml of water was added. The mixture was vigorously stirred for ~20 min under air after which the layers were separated and the aqueous phase was extracted twice with ether. The dried extracts were concentrated *in vacuo* and the remaining liquid distilled through a short Vigreux column. The compound **3**, R = Me, R¹R²CH = Et, b.p ~70 °C/0.5 mm Hg, n²⁰_D 1.5390, was obtained in ~80% yield. NMR-spectrum (90 MHz, CCl₄): 1.35 (t, Me), 2.20 (s, SMe), 3.70 (s, OMe), 3.92 (q, NCH₂), 5.70 (d, H-4) and 6.47 (d, H-5) ppm.

Other pyrrole derivatives 3 were obtained in yields of at least 70% overall. The physical and spectral data of the intermediate 2 and pyrrole 3 having R = Me and $R^1R^2CH = Me$, corresponded with the data published recently⁵.

The investigations are being continued with other terminally (e.g. t-Bu-CH=C=CHLi) and non-terminally (CH₂=C=C(Li)CH₃) lithiated allenes. In these cases pyrrole derivatives were obtained in satisfactory yields, whereas in the absence of copper catalysts *only* 2,3-dihydropyridines were formed at more elevated temperatures

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