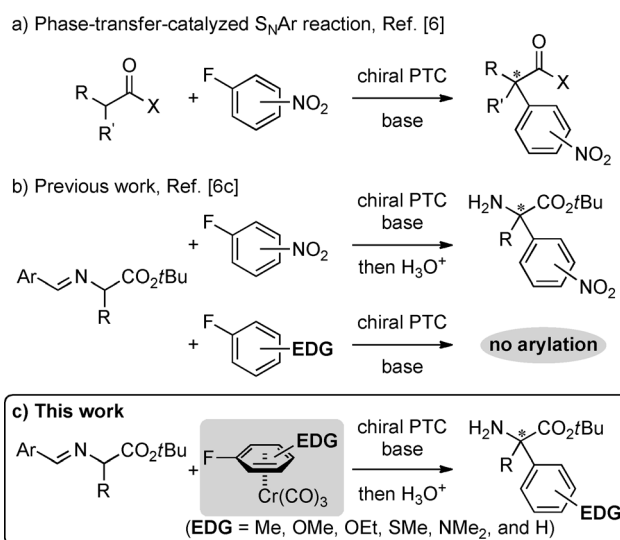


Phase-Transfer-Catalyzed Asymmetric S_NAr Reaction of α -Amino Acid Derivatives with Arene Chromium Complexes**

Seiji Shirakawa, Kenichiro Yamamoto, and Keiji Maruoka*

Abstract: Although phase-transfer-catalyzed asymmetric S_NAr reactions provide unique contribution to the catalytic asymmetric α -arylations of carbonyl compounds to produce biologically active α -aryl carbonyl compounds, the electrophiles were limited to arenes bearing strong electron-withdrawing groups, such as a nitro group. To overcome this limitation, we examined the asymmetric S_NAr reactions of α -amino acid derivatives with arene chromium complexes derived from fluoroarenes, including those containing electron-donating substituents. The arylation was efficiently promoted by binaphthyl-modified chiral phase-transfer catalysts to give the corresponding α,α -disubstituted α -amino acids containing various aromatic substituents with high enantioselectivities.

Catalytic asymmetric α -arylation of carbonyl compounds has been extensively studied over the last decade to prepare biologically interesting α -aryl carbonyl compounds.^[1,2] Several catalytic asymmetric methods for α -arylation have been developed using chiral metal complexes.^[3] As another method for asymmetric α -arylation, the phase-transfer-catalyzed nucleophilic aromatic substitution (S_NAr) reaction provides an efficient means to realize enantioselective α -arylations (Scheme 1).^[4,5] Jørgensen's and our groups have reported highly enantioselective phase-transfer-catalyzed S_NAr reactions of carbonyl compounds with nitro group-bearing fluoroarenes as electrophiles (Scheme 1a).^[6] The drawback of this method is the limited scope of fluoroarenes that can be used, as only electron-deficient arenes are effective substrates. The S_NAr reactions using fluoroarenes with electron-donating groups (EDG), did not work under the phase-transfer conditions (Scheme 1b). As a solution to this problem, we are interested in S_NAr reaction of the chromium complexes of fluoroarenes, which activate the arenes through η^6 -coordination to $Cr(CO)_3$.^[7,8] Here we report a valuable example of a phase-transfer-catalyzed asymmetric S_NAr reaction of α -amino acid derivatives with arene chromium



Scheme 1. Phase-transfer-catalyzed asymmetric S_NAr reactions. PTC = phase-transfer catalyst.

complexes, which produces enantioenriched α,α -disubstituted α -amino acids^[9] including those containing an electron-rich aromatic substituents (Scheme 1c).

We first examined the asymmetric S_NAr reaction of alanine derivative **1a** with chromium complex **2a** derived from fluorobenzene, promoted by biaryl-modified chiral quaternary ammonium salts **4–6** as promising chiral phase-transfer catalysts (Table 1).^[5] Attempted reaction of **1a** and fluorobenzene derivative **2a** with solid KOH in toluene under the influence of chiral phase-transfer catalyst (*R,R*)-**4**^[10] at 0°C for 24 h, followed by treatment with aqueous HCl for hydrolysis of the imine moiety and removal of chromium, afforded the corresponding α,α -disubstituted α -amino ester **3a** with moderate yield and enantioselectivity (77% *ee*, entry 1). The use of simplified catalyst (*R*)-**5**,^[11] which is one of the most effective catalysts for asymmetric alkylation reactions of **1a**,^[11,12] improved the enantioselectivity to give product **3a** in moderate yield with high enantioselectivity (97% *ee*, entry 2). The use of biphenyl-modified catalyst (*R*)-**6**^[13] caused the decrease of enantioselectivity for the S_NAr reaction (39% *ee*, entry 3). Changing the base to CsOH improved the yield (entry 4), and the highest yield and enantioselectivity was attained by the use of saturated aqueous CsOH^[14] as base with catalyst (*R*)-**5** (entry 5).^[15] The absolute configuration of product **3a** was determined by comparison of the optical rotation of the *N*-acetylated derivative of **3a** with the literature value.^[6c,16]

[*] Prof. Dr. S. Shirakawa, K. Yamamoto, Prof. Dr. K. Maruoka
Department of Chemistry, Graduate School of Science
Kyoto University
Sakyo, Kyoto 606-8502 (Japan)
E-mail: maruoka@kuchem.kyoto-u.ac.jp

Prof. Dr. S. Shirakawa
Graduate School of Fisheries Science and Environmental Studies
Nagasaki University
1-14, Bunkyo-machi, Nagasaki 852-8521 (Japan)

[**] This work was supported by a Grant-in-Aid for Scientific Research from JSPS and MEXT (Japan).

Supporting information for this article is available on the WWW under <http://dx.doi.org/10.1002/anie.201409065>.

Table 1: Optimization of the reaction conditions.^[a]

Reaction scheme for Table 1: **1a** + **2a** (Ar = H, *p*-Me, *m*-Me, *o*-Me, *p*-OMe, *m*-OMe, *p*-OEt, *p*-SMe, *p*-NMe₂) → **3a** (Ar = H, *p*-Me, *m*-Me, *o*-Me, *p*-OMe, *m*-OMe, *p*-OEt, *p*-SMe, *p*-NMe₂)

Entry	Catalyst	Base	Yield [%] ^[b]	ee [%] ^[c]
1	(<i>R,R</i>)- 4	KOH	42	77
2	(<i>R</i>)- 5	KOH	39	97
3	(<i>R</i>)- 6	KOH	41	39
4	(<i>R</i>)- 5	CsOH·H ₂ O	70	91
5	(<i>R</i>)- 5	sat. aq. CsOH	70	97

[a] Reaction conditions: **1a** (0.30 mmol), **2a** (0.10 mmol), and base (1.5 mmol) in the presence of phase-transfer catalyst (10 mol%, 0.010 mmol) in toluene (2.0 mL) at 0 °C for 24 h. [b] Yields of isolated product **3a**, which were determined on the basis of the amount of **2a**. [c] Determined by HPLC analysis on a chiral stationary phase.

With the optimal reaction conditions identified, we next examined the substrate generality of the asymmetric S_NAr reaction of alanine derivative **1a** with arene chromium complexes **2** under the influence of chiral phase-transfer catalyst (*R*)-**5** (Table 2). The S_NAr reactions with *p*-, *m*-, and *o*-fluorotoluene derivatives (**2b–2d**) proceeded efficiently, and gave the products (**3b–3d**) in good yields with high enantioselectivities (94–98 % ee, entries 2–4). Other electron-donating group (EDG)-substituted fluoroarene derivatives (**2e–2i**) could also be used in this reaction, thereby providing the products (**3e–3i**) with excellent enantioselectivities (98–99 % ee, entries 5–9).^[17]

Other α-amino acid derivatives **1** could be employed in the S_NAr reaction with arene chromium complex **2a** (Table 3). Not only simple alkyl-substituted α-amino acid derivatives (**1a–1c**) but also phenylalanine (**1d**) and allylglycine (**1e**) derivatives were tolerated in the reaction to give the corresponding phenylation products **3j–3m** in high enantioselectivities (88–99 % ee, entries 1–5). Furthermore, heteroatom-containing α-amino acids, such as methionine (**1f**), serine (**1g**), and glutamic acid derivatives (**1h**), were also suitable for the reaction to give the corresponding products **3n–3p** in high enantioselectivities (90–96 % ee, entries 6–8).

Table 2: Asymmetric arylation of alanine derivative **1a**.^[a]

Reaction scheme for Table 2: **1a** + **2** (EDG = H, *p*-Me, *m*-Me, *o*-Me, *p*-OMe, *m*-OMe, *p*-OEt, *p*-SMe, *p*-NMe₂) → **3** (EDG = H, *p*-Me, *m*-Me, *o*-Me, *p*-OMe, *m*-OMe, *p*-OEt, *p*-SMe, *p*-NMe₂)

Entry	EDG (2)	Yield [%] ^[b]	ee [%] ^[c]
1	H (2a)	70 (3a)	97
2	<i>p</i> -Me (2b)	71 (3b)	98
3	<i>m</i> -Me (2c)	69 (3c)	98
4	<i>o</i> -Me (2d)	62 (3d)	94
5	<i>p</i> -OMe (2e)	70 (3e)	98
6	<i>m</i> -OMe (2f)	68 (3f)	99
7	<i>p</i> -OEt (2g)	61 (3g)	98
8	<i>p</i> -SMe (2h)	65 (3h)	99
9	<i>p</i> -NMe ₂ (2i)	63 (3i)	98

[a] Reaction conditions: **1a** (0.30 mmol), **2** (0.10 mmol), and saturated aqueous CsOH (1.5 mmol) in the presence of (*R*)-**5** (10 mol%, 0.010 mmol) in toluene (2.0 mL) at 0 °C for 24 h. [b] Yields of isolated products **3**, which were determined on the basis of the amount of **2**. [c] Determined by HPLC analysis on a chiral stationary phase.

Table 3: Asymmetric phenylation of α-amino acid derivatives **1**.^[a]

Reaction scheme for Table 3: **1** (R = Me, Et, *i*Bu, CH₂Ph, CH₂CH=CH₂, CH₂CH₂SMe, CH₂OTfBu, CH₂CH₂CO₂tBu) + **2a** → **3** (R = Me, Et, *i*Bu, CH₂Ph, CH₂CH=CH₂, CH₂CH₂SMe, CH₂OTfBu, CH₂CH₂CO₂tBu)

Entry	R (1)	Yield [%] ^[b]	ee [%] ^[c]
1	Me (1a)	70 (3a)	97
2	Et (1b)	78 (3j)	99
3	<i>i</i> Bu (1c)	60 (3k)	88
4	CH ₂ Ph (1d)	68 (3l)	95
5	CH ₂ CH=CH ₂ (1e)	74 (3m)	96
6	CH ₂ CH ₂ SMe (1f)	76 (3n)	90
7	CH ₂ OTfBu (1g)	73 (3o)	96
8	CH ₂ CH ₂ CO ₂ tBu (1h)	51 (3p)	92

[a] Reaction conditions: **1** (0.30 mmol), **2a** (0.10 mmol), and saturated aqueous CsOH (1.5 mmol) in the presence of (*R*)-**5** (10 mol%, 0.010 mmol) in toluene (2.0 mL) at 0 °C for 24 h. [b] Yields of isolated products **3**, which were determined on the basis of amount of **2a**. [c] Determined by HPLC analysis on a chiral stationary phase.

In summary, we have successfully overcome a major limitation of phase-transfer-catalyzed asymmetric S_NAr reactions by the use of arene chromium complexes. The S_NAr reaction of α-amino acid derivatives with chromium complexes derived from electron-rich fluoroarenes, was efficiently promoted by binaphthyl-modified chiral phase-transfer catalysts to give the corresponding α,α-disubstituted α-amino acids containing various aromatic substituents, in a highly enantioselective manner. This report offers a new option for

catalytic asymmetric α -arylation to produce biologically active α -aryl carbonyl compounds.

Received: September 13, 2014

Published online: ■ ■ ■ ■ ■ ■ ■ ■ ■ ■

Keywords: amino acids · arylation · asymmetric synthesis · organocatalysis · phase-transfer catalysis

- [1] For reviews on α -arylation, see: a) A. C. B. Burtoloso, *Synlett* **2009**, 320; b) C. C. C. Johansson, T. J. Colacot, *Angew. Chem. Int. Ed.* **2010**, 49, 676; *Angew. Chem.* **2010**, 122, 686; c) F. Bellina, R. Rossi, *Chem. Rev.* **2010**, 110, 1082; d) C. Mazet, *Synlett* **2012**, 1999.
- [2] For examples of biologically active α -aryl carbonyl compounds, see: a) H. U. Stilz, W. Guba, B. Jablonka, M. Just, O. Klingler, W. König, V. Wehner, G. Zoller, *J. Med. Chem.* **2001**, 44, 1158; b) H. Venkatesan, M. C. Davis, Y. Altas, J. P. Snyder, D. C. Liotta, *J. Org. Chem.* **2001**, 66, 3653; c) T. Hatanaka, Y. Nabuchi, H. Ushio, *J. Pharm. Pharmacol.* **2002**, 54, 549; d) C. Zha, G. B. Brown, W. J. Brouillette, *J. Med. Chem.* **2004**, 47, 6519; e) A. Natarajan, Y. Guo, F. Harbinski, Y.-H. Fan, H. Chen, L. Luus, J. Diercks, H. Aktas, M. Chorev, J. A. Halperin, *J. Med. Chem.* **2004**, 47, 4979; f) D. A. Neel, M. L. Brown, P. A. Lander, T. A. Grese, J. M. Defauw, R. A. Doti, T. Fields, S. A. Kelley, S. Smith, K. M. Zimmerman, M. I. Steinberg, P. K. Jadhav, *Bioorg. Med. Chem. Lett.* **2005**, 15, 2553; g) K. H. Mortell, D. J. Anderson, J. J. Lynch III, S. L. Nelson, K. Sarris, H. McDonald, R. Sabet, S. Baker, P. Honore, C.-H. Lee, M. F. Jarvis, M. Gopalakrishnan, *Bioorg. Med. Chem. Lett.* **2006**, 16, 1138; h) Y.-J. Kwon, M.-J. Sohn, C.-J. Zheng, W.-G. Kim, *Org. Lett.* **2007**, 9, 2449; i) M. K. Uddin, S. G. Reignier, T. Coulter, C. Montalbetti, C. Grånäs, S. Butcher, C. Krog-Jensen, J. Felding, *Bioorg. Med. Chem. Lett.* **2007**, 17, 2854; j) M. K. Christensen, K. D. Erichsen, C. Trojel-Hansen, J. Tjørnelund, S. J. Nielsen, K. Frydenvang, T. N. Johansen, B. Nielsen, M. Sehested, P. B. Jensen, M. Ikaunieks, A. Zaichenko, E. Loza, I. Kalvinsh, F. Björklund, *J. Med. Chem.* **2010**, 53, 7140; k) F. Nique, S. Hebbe, C. Peixoto, D. Annot, J.-M. Lefrançois, E. Duval, L. Michoux, N. Triballeau, J.-M. Lemoullec, P. Mollat, M. Thauvin, T. Prangé, D. Minet, P. Clément-Lacroix, C. Robin-Jagerschmidt, D. Fleury, D. Guédin, P. Deprez, *J. Med. Chem.* **2012**, 55, 8225.
- [3] For examples of catalytic asymmetric α -arylations, see: a) J. Åhman, J. P. Wolfe, M. V. Troutman, M. Palucki, S. L. Buchwald, *J. Am. Chem. Soc.* **1998**, 120, 1918; b) D. J. Spielvogel, S. L. Buchwald, *J. Am. Chem. Soc.* **2002**, 124, 3500; c) G. Chen, F. Y. Kwong, H. O. Chan, W.-Y. Yu, A. S. C. Chan, *Chem. Commun.* **2006**, 1413; d) X. Liao, Z. Weng, J. F. Hartwig, *J. Am. Chem. Soc.* **2008**, 130, 195; e) A. M. Taylor, R. A. Altman, S. L. Buchwald, *J. Am. Chem. Soc.* **2009**, 131, 9900; f) A. Bigot, A. E. Williamson, M. J. Gaunt, *J. Am. Chem. Soc.* **2011**, 133, 13778; g) J. S. Harvey, S. P. Simonovich, C. R. Jamison, D. W. C. MacMillan, *J. Am. Chem. Soc.* **2011**, 133, 13782; h) A. E. Allen, D. W. C. MacMillan, *J. Am. Chem. Soc.* **2011**, 133, 4260; i) J. Guo, S. Dong, Y. Zhang, Y. Kuang, X. Liu, L. Lin, X. Feng, *Angew. Chem. Int. Ed.* **2013**, 52, 10245; *Angew. Chem.* **2013**, 125, 10435.
- [4] F. Terrier, *Modern Nucleophilic Aromatic Substitution*, Wiley-VCH, Weinheim, **2013**.
- [5] For recent reviews on asymmetric phase-transfer catalysis, see: a) M. J. O'Donnell, *Aldrichimica Acta* **2001**, 34, 3; b) K. Maruoka, T. Ooi, *Chem. Rev.* **2003**, 103, 3013; c) M. J. O'Donnell, *Acc. Chem. Res.* **2004**, 37, 506; d) B. Lygo, B. I. Andrews, *Acc. Chem. Res.* **2004**, 37, 518; e) J. Vachon, J. Lacour, *Chimia* **2006**, 60, 266; f) T. Ooi, K. Maruoka, *Angew. Chem. Int. Ed.* **2007**, 46, 4222; *Angew. Chem.* **2007**, 119, 4300; g) T. Ooi, K. Maruoka, *Aldrichimica Acta* **2007**, 40, 77; h) T. Hashimoto, K. Maruoka, *Chem. Rev.* **2007**, 107, 5656; i) K. Maruoka, *Org. Process Res. Dev.* **2008**, 12, 679; j) S.-s. Jew, H.-g. Park, *Chem. Commun.* **2009**, 7090; k) K. Maruoka, *Chem. Rev.* **2010**, 10, 254; l) S. Shirakawa, K. Maruoka, *Angew. Chem. Int. Ed.* **2013**, 52, 4312; *Angew. Chem.* **2013**, 125, 4408.
- [6] a) M. Bella, S. Kobbelaard, K. A. Jørgensen, *J. Am. Chem. Soc.* **2005**, 127, 3670; b) S. Kobbelaard, M. Bella, K. A. Jørgensen, *J. Org. Chem.* **2006**, 71, 4980; c) S. Shirakawa, K. Yamamoto, T. Tokuda, K. Maruoka, *Asian J. Org. Chem.* **2014**, 3, 433; d) S. Shirakawa, K. Koga, T. Tokuda, K. Yamamoto, K. Maruoka, *Angew. Chem. Int. Ed.* **2014**, 53, 6220; *Angew. Chem.* **2014**, 126, 6334.
- [7] For reviews on arene chromium complexes in organic synthesis, see: a) K. Kamikawa, M. Uemura, *Synlett* **2000**, 938; b) A. Berger, J.-P. Djukic, C. Michon, *Coord. Chem. Rev.* **2002**, 225, 215; c) S. E. Gibson, H. Ibrahim, *Chem. Commun.* **2002**, 2465; d) A. Salzer, *Coord. Chem. Rev.* **2003**, 242, 59; e) M. Rosillo, G. Domínguez, J. Pérez-Castells, *Chem. Soc. Rev.* **2007**, 36, 1589.
- [8] a) M. Chaari, J.-P. Laverne, P. Viallefont, *Synth. Commun.* **1989**, 19, 1211; b) F. Rose-Munch, K. Aniss, E. Rose, *J. Organomet. Chem.* **1990**, 385, C1; c) M. Chaari, A. Jenhi, J.-P. Laverne, P. Viallefont, *Tetrahedron* **1991**, 47, 4619.
- [9] For reviews on asymmetric syntheses of α,α -disubstituted α -amino acids, see: a) C. Cativiela, M. D. Díaz-de-Villegas, *Tetrahedron: Asymmetry* **1998**, 9, 3517; b) C. Cativiela, M. D. Díaz-de-Villegas, *Tetrahedron: Asymmetry* **2000**, 11, 645; c) Y. Ohfuné, T. Shinada, *Eur. J. Org. Chem.* **2005**, 5127; d) H. Vogt, S. Bräse, *Org. Biomol. Chem.* **2007**, 5, 406.
- [10] a) T. Ooi, M. Takeuchi, M. Kameda, K. Maruoka, *J. Am. Chem. Soc.* **2000**, 122, 5228; b) T. Ooi, M. Kameda, K. Maruoka, *J. Am. Chem. Soc.* **2003**, 125, 5139.
- [11] a) M. Kitamura, S. Shirakawa, K. Maruoka, *Angew. Chem. Int. Ed.* **2005**, 44, 1549; *Angew. Chem.* **2005**, 117, 1573; b) M. Kitamura, S. Shirakawa, Y. Arimura, X. Wang, K. Maruoka, *Chem. Asian J.* **2008**, 3, 1702.
- [12] S. Shirakawa, K. Yamamoto, K. Liu, K. Maruoka, *Org. Synth.* **2013**, 90, 121.
- [13] a) Z. Han, Y. Yamaguchi, M. Kitamura, K. Maruoka, *Tetrahedron Lett.* **2005**, 46, 8555; b) Y.-G. Wang, M. Ueda, X. Wang, Z. Han, K. Maruoka, *Tetrahedron* **2007**, 63, 6042.
- [14] For the preparation of saturated aqueous CsOH (80 wt % aq. CsOH), see Ref. [12].
- [15] The reaction on larger scale (10 times) gave a similar result (74 % yield, 97 % ee).
- [16] For details of the determination of the absolute configuration of **3a**, see the Supporting Information.
- [17] Racemic arene chromium complexes **2c**, **2d**, and **2f** were used for the reaction.

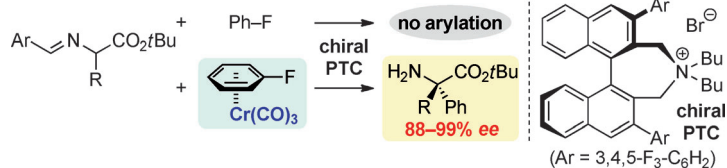
Communications



Organocatalysis

S. Shirakawa, K. Yamamoto,
K. Maruoka* ———— ■■■■-■■■■

Phase-Transfer-Catalyzed Asymmetric
 S_NAr Reaction of α -Amino Acid
Derivatives with Arene Chromium
Complexes



Increased substrate scope in phase-transfer-catalyzed asymmetric S_NAr reactions was achieved by the use of arene chromium complexes as electrophiles. An

efficient asymmetric synthesis of α,α -disubstituted α -amino acids containing various aromatic substituents is shown. PTC = phase-transfer catalyst.