

## Note

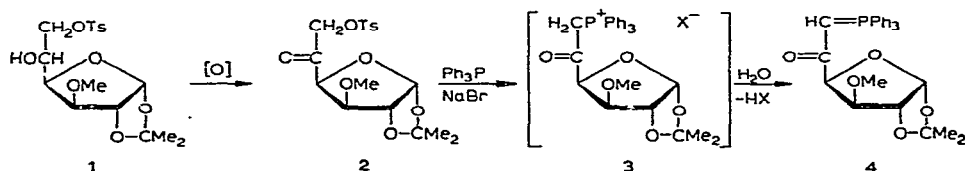
## A sugar phosphorane

YU. A. ZHDANOV AND V. A. POLENOV

Department of Chemistry, Rostov-on-Don State University, Rostov-on-Don (U. S. S. R.)

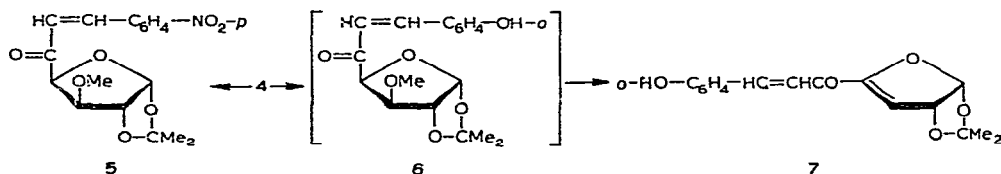
(Received February 6th, 1970; accepted for publication in revised form, June 24th, 1970)

The Wittig reaction is an effective method for the synthesis of carbohydrate derivatives containing new carbon-carbon bonds. In previous reports<sup>1,2</sup>, the carbohydrates were used only as the carbonyl components, but we report now the preparation of a carbohydrate phosphorane, using the sequence 1→4.



The ketone 2 did not react directly with triphenylphosphine in a variety of solvents, but reaction occurred readily in acetone in the presence of sodium bromide. Treatment of the resulting solution of the phosphonium salt 3 with water gave the phosphorane 4 which was recovered from the aqueous solution by extraction with chloroform. This behaviour reveals the relatively high acidity of the methylene group of compound 3 and the low reactivity of phosphorane 4. The constitution of phosphorane 4 follows from its mode of synthesis and from the similarity of its u.v. and i.r. spectra with those of acetylmethylenetriphenylphosphorane.

Compound 4 reacts with such active aldehydes as *p*-nitro- and *o*-hydroxybenzaldehyde, but does not react practically with *p*-dimethylamino-, *p*-hydroxy-, and 2,4-dihydroxy-benzaldehyde.



The product of the reaction of phosphorane 4 with *p*-nitrobenzaldehyde is the expected mixture of *cis*, *trans* isomers of the unsaturated ketone 5. The reaction of phosphorane 4 with salicylaldehyde leads to compound 7, which evidently originates from compound 6 during washing of the reaction mixture with aqueous sodium

hydroxide. The absence of the 3-proton singlet for  $\text{CH}_3\text{O}$  at  $\tau$  6.60 and the bathochromic shift (*cf.* the data for compound 8) of the three u.v. bands for compound 7 confirm its constitution.

#### EXPERIMENTAL

*1,2-O-Isopropylidene-3-O-methyl-6-O-tosyl- $\alpha$ -D-glucofuranose* (1). — *1,2-O-Isopropylidene-3-O-methyl- $\alpha$ -D-glucofuranose* (47 g) in a mixture of acetone (50 ml) and pyridine (40 ml) was treated with tosyl chloride (46 g) with ice-cooling. The mixture was kept for one day, water was then added with ice-cooling, and, after 1 h, the mixture was poured into water (500 ml) and extracted with chloroform ( $3 \times 100$  ml). The chloroform solution was washed with M sodium carbonate and dilute hydrochloric acid, dried, and evaporated to give 62 g (80%) of syrupy 1,  $[\alpha]_{\text{D}} -24^\circ$  (*c* 5.1),  $\nu_{\text{max}}^{\text{film}}$  1354s  $\text{cm}^{-1}$ .

*Anal.* Calc. for  $\text{C}_{17}\text{H}_{24}\text{O}_8\text{S}$ : C, 52.58; H, 6.19. Found: C, 52.30; H, 5.80.

*1,2-O-Isopropylidene-3-O-methyl-6-O-tosyl- $\alpha$ -D-xylo-hexofuranos-5-ulose* (2). — A solution of compound 1 (60 g) in *N,N*-dimethylformamide (100 ml) was added to a solution of phosphorus pentoxide (20 g) in a mixture of methyl sulphoxide (30 ml) and *N,N*-dimethylformamide (300 ml). The mixture was kept for 5 days, and was then poured into saturated, aqueous sodium hydrogen carbonate (1 litre) and extracted with chloroform ( $3 \times 200$  ml). The chloroform solution was dried and evaporated to give 47 g (78%) of syrupy 2,  $[\alpha]_{\text{D}} -62^\circ$  (*c* 5.6),  $\nu_{\text{max}}^{\text{film}}$  1748s  $\text{cm}^{-1}$ ;  $\lambda_{\text{max}}$  225 nm ( $\epsilon$  10,000).

*Anal.* Calc. for  $\text{C}_{17}\text{H}_{22}\text{O}_8\text{S}$ : C, 52.85; H, 5.72. Found: C, 52.30; H, 5.77.

*(1,2-O-Isopropylidene-3-O-methyl- $\alpha$ -D-xylofuranuronylmethylene)triphenylphosphorane* (4). — A mixture of compound 2 (31 g), triphenylphosphine (24 g), and sodium bromide (10 g) in dry acetone (200 ml) was refluxed for 3 h. Water (150 ml) was then added, and the mixture was extracted with chloroform ( $3 \times 150$  ml). The chloroform solution was washed with M sodium carbonate, dried, and evaporated to give 32 g (83%) of syrupy 4,  $[\alpha]_{\text{D}} -46^\circ$  (*c* 5.3);  $\nu_{\text{max}}^{\text{film}}$  1636 m, 1592 s, 1542 s, 1472 s, 1454 s  $\text{cm}^{-1}$ ;  $\lambda_{\text{max}}$  223 nm ( $\epsilon$  21,400); *cf.* acetylmethylenetriphenylphosphorane,  $\nu_{\text{max}}^{\text{Nujol}}$  1614 w, 1586 m, 1552 s, 1486 s, 1456 s  $\text{cm}^{-1}$ ,  $\lambda_{\text{max}}$  220 nm ( $\epsilon$  31,400).

*Anal.* Calc. for  $\text{C}_{28}\text{H}_{29}\text{O}_5\text{P}$ : C, 70.59; H, 6.08. Found: C, 70.04; H, 6.20.

*1,2-O-Isopropylidene-3-O-methyl-4-[3-(*p*-nitrophenyl)acryloyl]- $\alpha$ -D-xylo-tetrofuranose* (5). — A mixture of compound 4 (1.2 g) and *p*-nitrobenzaldehyde (0.4 g) in xylene (5 ml) was heated for 2 h at  $100^\circ$ . The solution was evaporated under diminished pressure, and the residue was fractionated on aluminium oxide ( $R_F$  0.80 in chloroform) to give 5 as a syrup (0.50 g, 51%),  $[\alpha]_{\text{D}} -30^\circ$  (*c* 4.8);  $\nu_{\text{max}}^{\text{film}}$  1706 s, 1610 s, 1522 s, 1346 s  $\text{cm}^{-1}$ ;  $\lambda_{\text{max}}$  218 ( $\epsilon$  4,300), 310 ( $\epsilon$  8,000) nm.

*Anal.* Calc. for  $\text{C}_{17}\text{H}_{19}\text{NO}_7$ : C, 58.45; H, 5.45. Found: C, 57.87; H, 5.68.

*3-Deoxy-1,2-O-isopropylidene-4-[3-(*o*-hydroxyphenyl)acryloyl]- $\alpha$ -D-glycero-tetra-3-enofuranose* (7). — A mixture of compound 4 (2.4 g) and salicylaldehyde (0.62 g) in xylene (10 ml) was heated for 3 h at  $100^\circ$ . Benzene (20 ml) was then added, and the

solution was extracted with 0.1M sodium hydroxide (4 × 20 ml). The aqueous layer was acidified with dilute hydrochloric acid and extracted with ether. The dried, ethereal solution was then evaporated to give greenish-yellow crystals of **7** (0.51 g, 30%), m.p. 148° (from carbon tetrachloride),  $[\alpha]_D +18^\circ$  (*c* 5.6);  $\nu_{\max}^{\text{Nujol}}$  1656 s, 1604 s, 1578 m  $\text{cm}^{-1}$ ;  $\lambda_{\max}$  256 ( $\epsilon$  6,300), 314 ( $\epsilon$  15,000), 364 ( $\epsilon$  17,000) nm.

*Anal.* Calc. for  $\text{C}_{16}\text{H}_{16}\text{O}_5$ : C, 66.66; H, 5.56. Found: C, 66.20; H, 5.60.

*1-(o-Hydroxyphenyl)-but-1-en-3-one (8).* — The reaction of salicylaldehyde with acetylmethylenetriphenylphosphorane, performed as above, gave **8**, m.p. 138°;  $\nu_{\max}^{\text{Nujol}}$  1675 m, 1640 s, 1604 s  $\text{cm}^{-1}$ ;  $\lambda_{\max}$  236 ( $\epsilon$  8,000), 285 ( $\epsilon$  16,000), 338 ( $\epsilon$  1,000) nm.

*Anal.* Calc. for  $\text{C}_{10}\text{H}_{10}\text{O}_2$ : C, 74.07; H, 6.18. Found: C, 73.90; H, 6.20.

U.v. spectra were measured for methanol solutions; specific rotations were measured for ethanol solutions

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