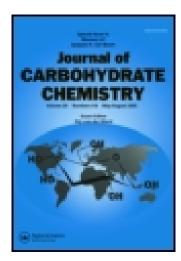
This article was downloaded by: [University of Newcastle, Australia]

On: 02 January 2015, At: 00:59 Publisher: Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office:

Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Journal of Carbohydrate Chemistry

Publication details, including instructions for authors and subscription information:

http://www.tandfonline.com/loi/lcar20

A Direct Route to a New Class of Acrylamide Thioglycosides

Galal H. Elgemeie $^{\rm a}$, Wafaa A. Zaghary $^{\rm b}$, Kamelia M. Amin $^{\rm c}$ & Tamer M. Nasr $^{\rm b}$

^a Faculty of Science, Department of Chemistry , Helwan University , Cairo, Egypt

^b Faculty of Pharmacy, Pharmaceutical Chemistry Department, Helwan University, Cairo, Egypt

^c Faculty of Pharmacy, Pharmaceutical Chemistry Department, Cairo University, Cairo, Egypt Published online: 07 Aug 2008.

To cite this article: Galal H. Elgemeie, Wafaa A. Zaghary, Kamelia M. Amin & Tamer M. Nasr (2008) A Direct Route to a New Class of Acrylamide Thioglycosides, Journal of Carbohydrate Chemistry, 27:6, 373-378, DOI: 10.1080/07328300802262786

To link to this article: http://dx.doi.org/10.1080/07328300802262786

PLEASE SCROLL DOWN FOR ARTICLE

Taylor & Francis makes every effort to ensure the accuracy of all the information (the "Content") contained in the publications on our platform. However, Taylor & Francis, our agents, and our licensors make no representations or warranties whatsoever as to the accuracy, completeness, or suitability for any purpose of the Content. Any opinions and views expressed in this publication are the opinions and views of the authors, and are not the views of or endorsed by Taylor & Francis. The accuracy of the Content should not be relied upon and should be independently verified with primary sources of information. Taylor and Francis shall not be liable for any losses, actions, claims, proceedings, demands, costs, expenses, damages, and other liabilities whatsoever or howsoever caused arising directly or indirectly in connection with, in relation to or arising out of the use of the Content.

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden. Terms & Conditions of access and use can be found at http://www.tandfonline.com/page/terms-and-conditions

Journal of Carbohydrate Chemistry, 27:373–378, 2008 Copyright © Taylor & Francis Group, LLC ISSN: 0732-8303 print 1532-2327 online DOI: 10.1080/07328300802262786



A Direct Route to a New Class of Acrylamide Thioglycosides

Galal H. Elgemeie, ¹ Wafaa A. Zaghary, ² Kamelia M. Amin, ³ and Tamer M. Nasr²

¹Faculty of Science, Department of Chemistry, Helwan University, Cairo, Egypt ²Faculty of Pharmacy, Pharmaceutical Chemistry Department, Helwan University, Cairo, Egypt

³Faculty of Pharmacy, Pharmaceutical Chemistry Department, Cairo University, Cairo, Egypt

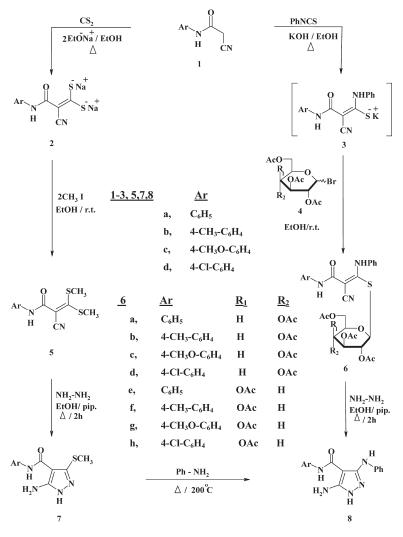
The preparation of a new class of acrylamide thioglycosides via one-pot reaction of the potassium 2-cyanoethylene-1-thiolate salts with 2,3,4,6-tetra-O-acetyl- α -D-gluco- and galactopyranosyl bromides has been studied. The E-configuration of these thioglycosides was proven by their transformations to the corresponding 5-aminopyrazoles.

Keywords Acrylamide thioglycosides, Potassium 2-cyanoethylene-1-thiolate salts, 5-Aminopyrazoles

Nucleoside analogs have occupied a significant position in the search for effective antiviral agents, owing to the fact that a large number of unnatural nucleoside derivatives have been shown to inhibit infection caused by viruses. [1] Heterocyclic thioglycosides constitute a class of analogs with potential biological activity. [2] As part of our program directed toward the development of new, simple, and efficient procedures for the synthesis of antimetabolites [3,4] we described that pyridine thioglycosides exerted inhibitory effects on both DNA and RNA containing viruses. [5] Based on these findings, it was of interest to prepare modified cyclic and acyclic thioglycosides to search for more effective agents. Here we report novel syntheses of acrylamide thioglycoside derivatives. The potassium 2-cyanoethylene-1-thiolate salts 3a-d were chosen as the key

Received August 1, 2007; accepted June 9, 2008.

intermediate. The reaction sequence to prepare the compounds is summarized in Scheme 1. Substituted acetanilide derivatives $\mathbf{1a-d}$ reacted with phenyl isothiocyanate in KOH-EtOH with heating to give the corresponding stable potassium 2-cyano-ethylene-1-thiolate salts $\mathbf{3a-d}$. The latter react with 2,3,4,6-tetra-O-acetyl- α -D-gluco- and galacto-pyranosyl bromides $\mathbf{4}$ in ethanol at rt to give the corresponding S-glucosides $\mathbf{6a-d}$ or S-galactosides $\mathbf{6e-h}$, in high yield. Attempted removal of protecting groups in $\mathbf{6a-h}$ by methanolic ammonia did not result in formation of the corresponding free glycosides. We suggested that $\mathbf{6a-h}$ should be present in the E and not in the E form. This was shown by reacting $\mathbf{6a-h}$ with hydrazine in refluxing ethanol to give the corresponding



Scheme 1

5-aminopyrazole derivatives $\bf 8a-d$. Alternatively, $\bf 8a-d$ could be prepared by reaction of the 3-methylthiopyrazoles $\bf 7a-d$ with aniline. Compounds $\bf 7a-d$ were prepared by the reaction of substituted acetanilide derivatives $\bf 1$ with carbon disulfide in the presence of sodium ethoxide followed by the alkylation with methyl iodide to give the ketene S,S-acetals $\bf 5$; the latter reacts with hydrazine to give compounds $\bf 7a-d$. [6]

In summary, we have achieved the synthesis of acrylamide thioglycosides by the reaction of the potassium 2-cyanoethylene-1-thiolate salts with α -glycosyl halides. These acyclic glycosides can be utilized as starting materials for the synthesis of other carbohydrate derivatives.

EXPERIMENTAL

All melting points were uncorrected on a Gallenkamp melting point apparatus. The IR spectra were recorded (KBr disk) on a Perkin Elmer 1650 FT-IR instrument. The 1H NMR spectra were recorded on a Varian 400 MHz spectrometer in $(CD_3)_2SO$ using $Si(CH_3)_4$ as an internal standard. Mass spectra were recorded on a Varian MAT 112 spectrometer. Elemental analyses were obtained from the Microanalytical Data Center at Cairo University, Egypt.

Progress of the reactions was monitored by TLC using aluminum sheets coated with silica gel F254 (Merck). Viewing under a short-wavelength UV lamp effected detection. All evaporations were carried out under reduced pressure at 40°C .

Compounds 5 and 7 were prepared following reported procedures. [6]

(2e)-3-Anilino-N-Aryl-2-Cyano-3-(2',3',4',6'-Tetra-O-Acetyl- β -D-Gluco- and Galactopyranosylthio)acrylamides (6a – h)

General Procedure

A mixture of N-substituted cyanoacetamide derivatives $1\mathbf{a}-\mathbf{d}$ (0.01 mol) and phenyl isothiocyanate (0.01 mol) was heated for 10 to 20 min in ethanol (25 mL) containing potassium hydroxide (0.01 mol). After cooling, a solution of 2,3,4,6-tetra-O-acetyl-(-D-gluco- or galacto-pyranosyl bromide $4\mathbf{a}$,b (0.01 mol) in ethanol (20 mL) was added. The reaction mixture was stirred at rt until completion (TLC) and then evaporated under reduced pressure, and the residue was washed with distilled water to remove the formed potassium bromide. The resulting solid product was dried and crystallized from a mixture of EtOH/DMF.

6a: White, m.p. 198°C (from EtOH/DMF), yield (87%). IR (KBr) $\nu_{\rm max}/{\rm cm}^{-1}$ 3340 (NH), 2206 (CN), 1751 (CO). ¹H NMR (DMSO) δ 1.94–2.02 (4s, 12H, 4 × CH₃CO), 3.66 (m, 2H, H₂-6'), 4.08 (m, 1H, H-5'), 4.84 (m, 2H, H-4', H-3'),

5.20 (d, $J_{1'\text{-}2'}$ 9.62 Hz, 1H, H-1'), 5.25 (t, 1H, H-2'), 7.22–7.50 (m, 9H, C_6H_5 , C_6H_4), 9.88 (s, 1H, NH). $^{13}\mathrm{C}$ NMR (DMSO) δ 20.13–20.37 (4 × CH $_3$), 61.34 (CH $_2$, C-6'), 67.38 (C-4'), 69.65 (C-2'), 72.62 (C-3'), 74.54 (C-5'), 83.84 (C-1'), 117.85 (CN), 121.17–138.44 (2C $_6H_5$), 154.00 (C-2), 160.18 (C-3), 163.70 (C-1), 169.00–169.80 (4 × CO). $C_{30}H_{31}N_3O_{10}S$, Calcd: C, 57.58; H, 5.00; N, 6.71, Found: C, 57.24; H, 5.25; N, 6.37%.

6b: White, m.p. 205°C (from EtOH/DMF), yield (86%). IR (KBr) $\nu_{\rm max}/{\rm cm}^{-1}$ 3402 (NH), 2198 (CN), 1743 (CO). ¹H NMR (DMSO) (1.90–2.00 (4s, 12H, 4 × CH₃CO), 2.25 (s, 3H, CH₃), 3.65 (m, 2H, H₂-6'), 4.00 (m, 1H, H-5'), 4.85 (m, 2H, H-4', H-3'), 5.15 (m, 1H, H-2') 5.25 (d, $J_{1'-2'}$ 9.9, 1H, H-1'), 7.10–7.50 (m, 9H, C₆H₅, C₆H₄), 9.55 (s, 1H, NH). ¹³C NMR (DMSO) (20.13–20.37 (5 × CH₃), 61.34 (CH₂, C-6'), 67.38 (C-4'), 69.60 (C-2'), 72.62 (C-3'), 74.52 (C5'), 83.47 (C-2'), 83.87 (C-1'), 117.85 (CN), 121.24–138.42 (C₆H₄ and C₆H₅), 160.04 (C-3), 163.63(C-1), 169.00–169.79 (4 × CO). C₃₁H₃₃N₃O₁₀S, Calcd: C, 58.19; H, 5.20; N, 6.56, Found: C, 58.10; H, 5.35; N, 6.55%.

6c: White, m.p. 215°C (from EtOH/DMF), yield (77%). IR (KBr) $\nu_{\rm max}/{\rm cm}^{-1}$ 3402 (NH), 2206 (CN), 1743 (CO). ¹H NMR (DMSO) δ 1.90–2.00 (4s, 12H, 4 × CH₃CO), 3.70 (m, 2H, 6'-H₂), 3.75 (s, 3H, OCH₃), 4.00 (m, 1H, H-5'), 4.85 (m, 2H, H-4', H-3'), 5.15 (d, $J_{1'-2'}$ 9.75, 1H, H-1'), 5.25 (t, 1H, H-2'), 6.80–7.50 (m, 9H, C₆H₅, C₆H₄), 9.50 (s, 1H, NH). ¹³C NMR (DMSO) (20.14–20.37 (4 × CH₃), 55.07 (OCH₃), 61.35 (CH₂, C-6'), 67.40 (C-4'), 69.67 (C-2'), 72.63 (C-3'), 74.51 (C-5'), 83.90 (C-1'), 113.51 (CN), 117.85–138.41 (C₆H₄ and C₆H₅), 155.88(C-2), 159.94(C-3), 163.64(C-1), 169.01–169.80 (4 × CO). C₃₁H₃₃N₃O₁₁S, Calcd: C, 56.77; H, 5.08; N, 6.40, Found: C, 57.04; H, 5.20; N, 6.35%.

6d: White, m.p. 215°C (from EtOH/DMF), yield (77%). IR (KBr) $\nu_{\rm max}/{\rm cm}^{-1}$ 3402 (NH), 2198 (CN), 1743 (CO). ¹H NMR (DMSO) δ 1.90–2.05 (4s, 12H, 4 × CH₃CO), 3.70 (m, 2H, H₂-6′), 4.00 (m, 1H, H-5′), 4.90 (m, 2H, H-4′, H-3′), 5.20 (d, J_{1′-2′} 9.80, 1H, H-1′), 5.25 (t, 1H, H-2′), 7.20–7.55 (m, 9H, C₆H₅, C₆H₄), 9.80 (s, 1H, NH). ¹³C NMR (DMSO) δ 20.13–20.37 (4 × CH₃), 61.34 (CH₂, C-6′), 67.37 (C-4′), 69.63 (C-2′), 72.59 (C-3′), 74.56 (C-5′), 83.75 (C-1′), 117.78 (CN), 122.60–138.47 (C₆H₄ and C₆H₅), 160.31 (C-2), 163.58 (C-3), 168.99 (C-1), 169.35–169.79 (4 × CO). C₃₀H₃₀ClN₃O₁₀S, Calcd: C, 54.54; H, 4.54; N, 6.36, Found: C, 54.44; H, 4.80; N, 6.34%.

6e: White, m.p. 179° C (from EtOH/DMF), yield (85%). IR (KBr) $\nu_{\rm max}/{\rm cm}^{-1}$ 3397 (NH), 2201 (CN), 1745 (CO). ¹H NMR (DMSO) δ 1.85–2.00 (4s, 12H, 4 × CH₃CO), 3.85 (m, 2H, H₂-6'), 4.10 (m, 1H, H-5'), 5.00 (t, 1H, H-4'), 5.10 (m, 2H, H-3', 2'-H), 5.25 (d, $J_{1'-2'}$ 9.8, 1H, H-1'), 7.05–7.70 (m, 10H, 2C₆H₅), 9.60 (s, 1H, NH). ¹³C NMR (DMSO) δ 20.01–20.40 (4 × CH₃), 60.83 (CH₂, C-6'), 66.77 (C-4'), 67.06 (C-2'), 70.64 (C-3'), 73. 81 (C-5'), 84.03 (C-1'), 118.03 (CN), 120.98–138.75 (2C₆H₅), 155.70 (C-2), 159.61 (C-3), 163.10 (C-1),

 $169.24-169.70~(4 \times CO).~C_{30}H_{31}N_3O_{10}S,~Calcd:~C,~57.58;~H,~5.00;~N,~6.71,~Found:~C,~57.53;~H,~5.00;~N,~7.06\%.$

6f: White, m.p. 210°C (from EtOH/DMF), yield (86%). IR (KBr) $\nu_{\rm max}/{\rm cm}^{-1}$ 3402 (NH), 2198 (CN), 1743 (CO). ¹H NMR (DMSO) δ 1.85–2.05 (4s, 12H, 4 × CH₃CO), 2.25 (s, 3H, CH₃), 3.85 (m, 2H, H₂-6'), 4.10 (m, 1H, H-5'), 5.00 (t, 1H, H-4'), 5.10 (m, 3H, H-3', H-2') 5.30 (d, $J_{1'-2'}$ 9.6, 1H, H-1'), 7.10–7.50 (m, 9H, C₆H₅, C₆H₄), 9.55 (s, 1H, NH). ¹³C NMR (DMSO) δ 20.04–20.47 (5 × CH₃), 60.82 (CH₂, C-6'), 66.80 (C-4'), 67.06 (C-2'), 70.65 (C-3'), 73.78 (C-5'), 84.11 (C-1'), 118.03 (CN), 119.66–138.71 (C₆H₄ and C₆H₅), 155.60 (C-2), 159.51 (C-3), 163.09 (C-1), 169.24–169.72 (4 × CO). C₃₁H₃₃N₃O₁₀S, Calcd: C, 58.19; H, 5.20; N, 6.56, Found: C, 57.90; H, 4.90; N, 6.64%.

6g: White, m.p. 191° C (from EtOH/DMF), yield (76%). IR (KBr) $\nu_{\rm max}/{\rm cm}^{-1}$ 3409 (NH), 2198 (CN), 1743 (CO). ¹H NMR (DMSO) δ 1.85–2.05 (4s, 12H, 4 × CH₃CO), 2.25 (s, 3H, CH₃), 3.85 (m, 2H, H₂-6'), 4.10 (m, 1H, H-5'), 5.00 (t, 1H, H-4'), 5.10 (m, 3H, H-3', H-2'), 5.30 (d, J_{1'-2'} 9.88, 1H, H-1'), 7.10–7.50 (m, 9H, C₆H₅, C₆H₄), 9.55 (s, 1H, NH). ¹³C NMR (DMSO) δ 20.05–20.40 (4 × CH₃), 55.06 (OCH₃), 60.80 (CH₂, C-6'), 66.81 (C-4'), 67.05 (C-2'), 70.66 (C-3'), 73.76 (C-5'), 84.14 (C-1'), 113.73 (CN), 118.02–138.69 (C₆H₄ and C₆H₅), 155.81 (C-2), 159.40 (C-3), 163.10 (C-1), 169.25–169.73 (4 × CO). C₃₁H₃₃N₃O₁₁S, Calcd: C, 56.77; H, 5.08; N, 6.40, Found: C, 57.07; H, 5.20; N, 6.40%.

6h: White, m.p. 197°C (from EtOH/DMF), yield (87%). IR (KBr) $\nu_{\rm max}/{\rm cm}^{-1}$ 3394 (NH), 2198 (CN), 1743 (CO). ¹H NMR (DMSO) δ 1.90–2.05 (4s, 12H, 4 × CH₃CO), 3.85 (m, 2H, H₂-6'), 4.10 (m, 1H, H-5'), 5.05 (t, 1H, H-4'), 5.10 (m, 3H, H-3', H-2'), 5.25 (d, J_{1'-2'} 9.77, 1H, H-1'), 7.10–7.70 (m, 9H, C₆H₅, C₆H₄), 9.75 (s, 1H, NH), 11.00 (s, 1H, CONH). ¹³C NMR (DMSO) δ 19.95–20.40 (4 × CH₃), 60.90 (CH₂, C-6'), 66.73 (C-4'), 67.09 (C-2'), 70.64 (C-3'), 73.88 (C-5'), 83.90 (C-1'), 117.98 (CN), 122.34–138.82 (C₆H₄ and C₆H₅), 155.70 (C-2), 159.74 (C-3), 162.92 (C-1), 169.25–169.69 (4 × CO). C₃₀H₃₀ClN₃O₁₀S, Calcd: C, 54.59; H, 4.54; N, 6.36, Found: C, 54.70; H, 4.40; N, 6.65%.

5-Amino-3-(Methylthio)-1 *H*-Pyrazole-4-Carboxanilides (7a-D)⁽⁶⁾

General Procedure

A mixture of compounds **5a-d** (0.01 mol) and hydrazine (0.01 mol) in ethanol (30 mL) was heated at reflux for 2 h. After cooling, the reaction mixture was diluted with cooled water and the resulting solid product was filtered off and recrystallized from ethanol.

5-Amino-3-Anilino-1 H-Pyrazole-4-Carboxanilides (8a-D)

General Procedures

Method (a)

A mixture of compounds **6a-d** (0.01 mol) and hydrazine (0.01 mol) in ethanol (30 mL) was heated at reflux for 2 h. The resulting solid product was collected by filtration and recrystallized from ethanol.

Method (b)^[7]

A mixture of compounds 7a-d (0.01 mol) and aniline (0.01 mol) was heated for 2 to 3 h at 200°C in an oil bath. The reaction mixture was dissolved in ethanol and the resulting solid product was filtered off and recrystallized from ethanol.

REFERENCES

- [1] Elgemeie, G.H.; Mansour, O.A.; Metwally, N.H. Synthesis and anti-HIV activity of different novel nonclassical nucleosides. Nucleosides Nucleotides 1999, 18, 113–123.
- [2] Elgemeie, G.H.; Attia, A.M.; Hussain, B.A. A synthetic strategy to a new class of cycloalkane ring-fused pyridine nucleosides as potential anti-HIV agents. Nucleosides Nucleotides 1998, 17, 855–868.
- [3] Elgemeie, G.H.; Hussain, B.A. A convenient synthesis of 5-deaza nonclassical antifolates: reaction of cyanothioacetamide with sodium salts of 2-(hydroxymethylene)-1-cycloalkanones. Tetrahedron **1994**, *50*, 199–204.
- [4] Elgemeie, G.H.; El-Ezbawy, S.R.; El-Aziz, H.A. The design and synthesis of structurally related mercaptopurine analogues: reaction of dimethyl *N*-cyano-dithioiminocarbonate with 5-aminopyrazoles. Synth. Commun. **2001**, *31*, 3453–3458.
- [5] Scala, S.; Akhmed, N.; Rao, U.S.; Paull, K.; Lan, L.; Dickstein, B.; Lee, J.; Elgemeie, G.E. H.; Stein, W.D.; Bates, S.E. P-Glycoprotein substrates and antagonists cluster into two distinct groups. Mol. Pharmacol. 1977, 51, 1024–1033.
- [6] Elgemeie, G.H.; Elghandour, A.H.; Elzanate, A.M.; Ahmed, S.A. Synthesis of some novel α -cyanoketene S,S-acetals and their use in heterocyclic synthesis. J. Chem. Soc. Perkin Trans. 1 **1997**, 3285–3289.
- [7] Elgemeie, G.H.; Elghandour, A.H.; Abd Elaziz, G.W. Potassium 2-cyanoethylene-1-thiolate: a new preparative route to 2-cyanoketene SN-acetals and pyrazole derivatives. Synth. Commun. 2004, 34, 3281–3291.