This article was downloaded by: [University of California Santa Cruz] On: 26 November 2014, At: 14:10 Publisher: Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Phosphorus, Sulfur, and Silicon and the Related Elements

Publication details, including instructions for authors and subscription information:

http://www.tandfonline.com/loi/gpss20

SYNTHESIS OF BIHETEROCYCLIC α-AMINOPHOSPHONIC ACID DERIVATIVES

S. Achamlale a , A. Elachqar a , A. El Hallaoui a , A. Alami a , S. Elhajji a , M. L. Roumestant $^{a\ c}$ & Ph. Viallefont b

^a Laboratoire de Chimie Organique, Faculté des Sciences Dhar El Mehraz, Université Sidi Mohamed Ben Abdellah, Fes, Maroc

^b UMR 5810, Université MONTPELLIER II, Montpellier, France

^c Lab. de Chimie Org. , Université Montpellier II , 5 Av. Ch. Flahaut, F-34060, Montpellier Cedex, France Published online: 24 Sep 2006.

To cite this article: S. Achamlale , A. Elachqar , A. El Hallaoui , A. Alami , S. Elhajji , M. L. Roumestant & Ph. Viallefont (1998) SYNTHESIS OF BIHETEROCYCLIC α -AMINOPHOSPHONIC ACID DERIVATIVES, Phosphorus, Sulfur, and Silicon and the Related Elements, 140:1, 103-111, DOI: <u>10.1080/10426509808035736</u>

To link to this article: <u>http://dx.doi.org/10.1080/10426509808035736</u>

PLEASE SCROLL DOWN FOR ARTICLE

Taylor & Francis makes every effort to ensure the accuracy of all the information (the "Content") contained in the publications on our platform. However, Taylor & Francis, our agents, and our licensors make no representations or warranties whatsoever as to the accuracy, completeness, or suitability for any purpose of the Content. Any opinions and views expressed in this publication are the opinions and views of the authors, and are not the views of or endorsed by Taylor & Francis. The accuracy of the Content should not be relied upon and should be independently verified with primary sources of information. Taylor and Francis shall not be liable for any losses, actions, claims, proceedings, demands, costs, expenses, damages, and other liabilities whatsoever or howsoever caused arising directly or indirectly in connection with, in relation to or arising out of the use of the Content.

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden. Terms & Conditions of access and use can be found at http://www.tandfonline.com/page/terms-and-conditions

Phosphorus, Sulfur and Silicon, 1998, Vol. 140, pp. 103-111 Reprints available directly from the publisher Photocopying permitted by license only © 1998 OPA (Overseas Publishers Association) Amsterdam N.V. Published under license by the Gordon & Breach Science Publishers imprint. Printed in Malaysia

SYNTHESIS OF BIHETEROCYCLIC α-AMINOPHOSPHONIC ACID DERIVATIVES

S. ACHAMLALE^a, A. ELACHQAR^a, A. EL HALLAOUI^a, A. ALAMI^a, S. ELHAJJI^a, M.L. ROUMESTANT^{b*} and PH. VIALLEFONT^b

^aLaboratoire de Chimie Organique, Faculté des Sciences Dhar El Mehraz Université Sidi Mohamed Ben Abdellah, Fes, Maroc and ^bUMR 5810, Université MONTPELLIER II, Montpellier, France

(Received March 19, 1998)

We report here the synthesis of biheterocyclic α -aminophosphonic acid derivatives by 1,3 dipolar cycloaddition of acetylenic compounds, prepared in the laboratory, on α -azido α -amino phosphonic esters $\mathbf{1}^{[1a]}$.

Keywords: a-aminophosphonic acid; biheterocycle; 1,2,3-triazole; tetrazole

INTRODUCTION

 α -aminophosphonic acids, analogues of α -amino acids display diverse and useful biological properties^[2–4]: these compounds are found to be substrates and inhibitors of enzymes, as well as plant growth regulators and herbicides. They also display antibacterial properties and neuronal activities^[3–4].

In this paper we describe new heterocyclic analogues of α -amino acids able to present an interesting biological activity due to the presence of heterocyclic cores as in the case of tetrazolic glycine^[5] and proline^[6] derivatives This approach consists of preparing, firstly, tetrazolic dipolarophiles by nucleophilic substitution of propargyl bromide by 5-aryl substituted tetrazoles.

^{*} Address to Author: Lab. de Chimie Org., Université Montpellier II, 5 Av. Ch. Flahaut, F-34060 Montpellier Cedex, France.

S. ACHAMLALE et al.

RESULTS

Tetrazole derivatives were obtained as described in the literature^[7], the yields (35 to 75%) being improved in our team^[1c,7]. There is literature^[8] report which illustrates the existence of the two tetrazole tautomeric forms (scheme 1).



Treatment of 5-aryl substituted tetrazoles with propargyl bromide in the presence of triethylamine in acetone at room temperature during 4 h afforded the two regioisomers (2,5-disubstituted / 1,5-disubstituted) in good yields (70–85%) in a ratio of approximately 80/20, separable by column chromatography, whose structures were determined unambiguously by X-Ray diffraction analysis (scheme 2)^[9].



SCHEME 2

The 2,5 and 1,5-disubstituted tetrazoles thus prepared were submitted to cycloaddition reaction with azide 1 (scheme 3). The azide 1 was obtained by reaction^[1a] of sodium azide on the α -bromo- α -amino phosphonic ester. Results are summarised in table I.



SCHEME 3

TABLE I Synthesis of biheterocyclic α-amino phosphonic acid derivatives (2-8) a,b

Product	Ar	m.p. (°C)	Time (h)	Yield (%)	Ratio of isomers 1,5/1,4
2b 2a	C ₆ H ₅	127–129 139–141	48	80	20 80
3b 3a	p-Me-C ₆ H ₄	145–147 116–118	48	75	47 53
4b 4a	p-MeO-C ₆ H ₄	110–112 98–100	48	73	40 60
5b 5a	α -thienyl	101–103 122–124	48	69	25 75
6a	o-HO-C ₆ H ₄	186-188	48	75	>98
7b 7a	α-furyl	151–153 139–140	48	72	10 90
8a	α-pyrryl	170-172	48	79	>98

S. ACHAMLALE et al.

If we consider now the triazole ring after cycloaddition reaction, two regioisomers (1,4-isomer and 1,5-isomer) were formed in good yields and separated. The regioselectivity was excellent, only the 1,4-isomer was obtained in two cases : from **6a** (Ar = o-HO-C₆H₄) and **8a** (Ar = α -pyrryl).

The structures of the two regioisomers (2-8) **a,b** were assigned on the basis of literature data^[1d,8,10–12] concerning the chemical shifts of triazolic protons^[1d,10,11] and the chemical shifts of the carbons of the triazole ring in positions 4 and 5. The studies carried out^[1d,10,11] have shown that the triazolic proton signal for the 1,5-isomer lies downfield from the corresponding signal for the 1,4-isomer but in ¹³C NMR^[8,12] the carbon signal (HC₄) for the 1,5-isomer lies upfield from the corresponding carbon signal (HC₅) for the 1,4-isomer. (Table II)

TABLE II The chemical shifts of triazolic protons and carbons in ¹H NMR and ¹³C NMR for 1,4 and 1,5-isomers

	Isomer 1,5	Isomer 1,4	¹ H NMR		¹³ C NMR	
Ar			^δ H ₄ ppm	^δ H ₅ ppm	^δ C₄H ppm	⁸ C ₅ H ppm
C ₆ H ₅	2b -	- 2a	7.76 -	8.28	135.04	- 58 124.58
p-Me-C ₆ H	3b -	- 3a	7.75	8.35	135.06	124.63
p-MeO-C ₆ H ₄	4b -	- 4 a	7.75 -	8.3 1	135.00	124.57
α -thienyl	5b	- 5a	7.75	8.30	135.03	124.65
o-HO-C ₆ H ₄		6	-	8.52	-	125.30
α-furyl	7b -	- 7 a	7.57 -	8.30	134.56	124.5 1
α-pyrryl	-	8	-	8.39	-	124.21

By this strategy we have obtained different biheterocyclic α -aminophosphonic acid derivatives in good yields.

106

EXPERIMENTAL

Melting points were obtained on a electrothermal melting point apparatus and are uncorrected. ¹H NMR spectra were obtained on VARIAN EM – 360 (60 MHz) and BRUCKER (250 MHz) instruments, TMS as internal standard. ¹³C NMR spectra were obtained on BRUCKER (200 MHz) instrument. Microanalyses were performed by the ENCT (Toulouse). Mass spectra were measured on a JEOL-JMS-DX 300 FAB instrument and Desorption in chiemical ionisation with NH₃ D/CI instrument. 5-substituted tetrazoles have been prepared using Alami's method^[1c].

Cycloaddition reaction: General procedure

The azide 1 and the dipolarophile were magnetically stirred in a minimum volume of benzene at reflux (see table II for the reaction conditions). After evaporation of the solvent, the residue was chromatographied over silica gel or by preparative liquid chromatography (HPLC). The solid compounds are recrystallized from ether/dichloromethane and/or from acetone.

Minor isomer **2b**: Rf = 0.24 (ether).

¹H NMR (CDCl₃) δ : 1.08 (t, 3H, J = 7Hz), 1.31 (t, 3H, J = 7Hz), 3.75– 4.35 (m, 2x2H), 5.93 (s, 2H), 7.09 (dd, 1H, J₁ = 10Hz, J₂ = 15,5Hz), 7.35– 7.4 (m, 6H), 7.52 (dd, NH, J₁= 10Hz, J₃ = 4Hz), 7.76 (s,IH), 7.79–7.88 (m, 2H), 8.07–8.14 (m, 2H).

¹³C NMR (CDCl₃) δ : 16.32 (2xCH₃CH₂O), 48.03 (-CH₂-), 64.09 (d, -CH-, ¹J_{C-P} = 180.2Hz), 64.45 (2xCH₂O), 126.9, 127.1, 127.6, 128.8, 128.93, 130.52, 132.35,132.72 (2xC₆H₅), 135.04 (CH₄=C), 141.88 (CH₄=C), 165.63 (C₆H₅-CN₄), 166.8 (C=O). C₂₂H₂₅N₈O₄P (496): M.S [D.C.I/NH₃ (M+H)⁺ = 497 and (M+NH₄)⁺ = 514].

Major isomer **2a:** m.p. = $139-141^{\circ}$ C Rf = 0.16 (ether).

¹H NMR (CDCl₃) δ : 1.06 (t, 3H, J = 7Hz), 1.29 (t, 3H, J = 7Hz), 3.75– 4.35 (m, 2x2H), 5.98 (s, 2H), 7.08 (dd, 1H, J₁ = 10Hz, J₂ = 16Hz), 7.26– 7.5 (m, 6H), 7.64–7.69 (m, 2H), 8.04–8.09 (m, 2H), 7.28 (s, 1H), 8.6 (dd, NH, J₁ = 10Hz, J₃ = 4Hz).

¹³C NMR (CDCl₃) δ: 16.2 (2xCH₃CH₂O), 48.3 (-<u>C</u>H₂-), 60.4 (d, -<u>C</u>H-, ¹J_{C-P} = 182Hz), 64.76 (2x<u>C</u>H₂O), 124.58 (<u>C</u>H₅=C), 126.9, 127.2, 127.89, 128.6, 128.87, 130.41, 132.23,132.7 ($2x\underline{C}_{6}H_{5}$), 140.55 ($CH_{5}=\underline{C}$), 165.46 ($C_{6}H_{5}-\underline{C}N_{4}$), 167.5 (C=O). Anal. calcd. for $C_{22}H_{25}N_{8}O_{4}P$: C, 53.22, H,5.04, N,22.58.

Found.C, 52.93, H,4.96, N,22.50.

Minor isomer **3b**: Rf = 0.3 (ether).

¹H NMR (CDCl₃) δ : 1.06 (t, 3H, J = 7Hz), 1.31 (t, 3H, J = 7Hz), 2.38 (s, 3H), 3.7–4.31 (m, 2x2H), 5.91 (s, 2H), 7.1 (dd, 1H, J₁= 10Hz, J₂ = 15,5Hz), 7.2–7.6 (m, 5H), 7.75 (s, 1H), 7.8–8.03 (m, 5H).

¹³C NMR (CDCl₃) δ : 16.4 (2x<u>C</u>H₃CH₂O), 21.52 (<u>C</u>H₃), 47.96 (-<u>C</u>H₂-), 64.05 (d, -<u>C</u>H-, ¹J_{C-P} = 180.4Hz), 64.47 (2x<u>C</u>H₂O), 124.29, 126.81, 127.35, 127.64, 128.8, 129.63, 132.72, 140.73 (<u>C</u>₆H₅ and p-Me-<u>C</u>₆H₄), 135.06 (<u>C</u>H₄=C), 141.95 (CH₄=<u>C</u>), 165.7 (Me-C₆H₄-<u>C</u>N₄), 166.8 (C=O).

Anal. calcd. for C₂₃H₂₇N₈O₄P: C, 54.12, H,5.3, N,21.96. Found.C, 53.87, H,5.36, N,21.93.

Major isomer **3a:** Rf = 0.22 (ether).

¹H NMR (CDCl₃) δ : 1.02 (t, 3H, J = 7Hz), 1.26 (t, 3H, J = 7Hz), 2.35 (s, 3H), 3.64–4.28 (m, 2x2H), 5.97 (s, 2H), 7.15 (dd, 1H, J₁= 10Hz, J₂ = 16,7Hz), 7.16–7.56 (m, 5H), 7.84–7.96 (m, 4H), 8.35 (s, 1H), 8.95 (dd, NH, J₁ = 10Hz, J₃ = 4Hz).

¹³C NMR (CDCl₃) δ : 16.26 (2x<u>C</u>H₃CH₂O), 21.5 (<u>C</u>H₃), 48.22 (-<u>C</u>H₂-), 60.4 (d, -<u>C</u>H-, ¹J_{C-P} = 182.78Hz), 64.8 (2x<u>C</u>H₂O), 124.63 (<u>C</u>H₅=C), 124.38, 126.78, 128, 128.5, 129.56, 132.3, 132.6 140.56 (<u>C</u>₆H₅ and p-Me-<u>C</u>₆H₄), 140.6 (CH₅=<u>C</u>), 165.5 (Me-C₆H₅-<u>C</u>N₄), 167.6 (C=O).

 $C_{23}H_{27}N_8O_4P$ (510): M.S [D.C.I/NH₃ (M+H)⁺ = 511 and (M+NH₄)⁺= 528].

Minor isomer 4b: Rf = 0.33 (AcOEt/DCM 3/7).

¹H NMR (CDCl₃) δ : 1.07 (t, 3H, J = 7Hz), 1.31 (t, 3H, J = 7Hz), 3.84 (s, 3H), 3.7-4.31 (m, 2x2H), 5.9 (s, 2H), 7.09 (dd, 1H, J₁= 10Hz, J₂ = 15,5Hz), 7.3-7.87 (m, 5H+NH), 7.56 (AA'BB'System, 4H, J_{AB} = J_{AB} = 9Hz), 7.75 (s,1H).

¹³C NMR (CDCl₃) δ : 16.31 (2x<u>C</u>H₃CH₂O), 48.33 (-<u>C</u>H₂-), 55.4 (<u>C</u>H₃O), 64.05 (d, -<u>C</u>H, ¹J_{C-P} = 180.2 Hz), 64.45 (2x<u>C</u>H₂O), 114.33, 119.68, 127.61, 128.41, 128.68, 130.43, 132.35, 132.72, 161.43 (<u>C₆H₅ and p-MeO-C₆H₄), 135 (<u>C</u>H₄=C), 141.99 (CH₄=<u>C</u>), 165.5 (MeO-C₆H₄-<u>C</u>N₄), 166.8 (C=O).</u>

Anal. calcd. for C₂₃H₂₇N₈O₅P: C, 52.47, H,5.13, N,21.29. Found.C, 52.53, H,5.10, N,21.57.

Major isomer 4a: Rf = 0.2 (AcOEt/DCM 3/7).

¹H NMR (CDCl₃) δ : 1.04 (t, 3H, J = 7Hz), 1.27 (t, 3H, J = 7Hz), 3.82 (s, 3H), 3.80–4.30 (m, 2x2H), 5.96 (s, 2H), 7.1 (dd, 1H, J₁= 10Hz, J₂ = 15,5Hz), 7.24–7.95 (m, 5H), 7.45 (AA'BB'System, 4H, J_{AB}=J_{AB} = 9Hz), 8.31 (s, 1H), 8.8 (dd, NH, J₁= 10Hz, J₃ = 4Hz)

¹³C NMR (CDCl₃) δ : 16.35 (2xCH₃CH₂O), 48.10 (-CH₂-), 55.38 (CH₃O), 60.4 (d, -CH, ¹J_{C-P} = 182.5Hz), 64.81 (2xCH₂O), 124.57 (CH₅=C), 114.27, 119.78, 127.74, 127.94, 128.38, 128.57, 128.72, 132.23, 132.66, 161.32 (C₆H₅and p-MeO-C₆H₄), 140.63 (CH₅=C), 165.3 (Me-C₆H₅-CN₄), 167.5 (C=O).

 $C_{23}H_{27}N_8O_5P$ (526): M.S [D.C.I/NH₃ (M+H)⁺= 527 and (M+NH₄)⁺= 544].

Minor isomer **5b**: Rf = 0.36 (AcOEt/Et₂O 1/2).

¹H NMR (CDCl₃) δ : 1.08 (t, 3H, J = 7Hz), 1.32 (t, 3H, J = 7Hz), 3.72– 4.35 (m, 2x2H), 5.9 (s, 2H), 7.1 (dd, 1H, J₁ = 10Hz, J₂ = 15,5Hz), 7.11– 7.13 (m, 1H), 7.39–7.9 (m, 8H), 7.75 (s,1H).

¹³C NMR (CDCl₃) δ: 16.31 (2xCH₃CH₂O), 48.01 (-CH₂-), 64.12 (d, -CH-, ¹J_{C-P}= 180.16 Hz), 64.47 (2xCH₂O), 127.63, 127.98, 128.1, 128.18, 128.8, 132.34, 132.7 (C₆H₅ and C₄H₃S), 135.03 (CH₄=C), 141.7 (CH₄=C), 161.9 (α-Thienyl-CN₄), 166.8 (C=O). Anal. calcd. for $C_{20}H_{23}N_8O_4PS$: C, 47.81, H,4.58, N,22.31.

Found.C, 47.41, H,4.55, N,22.16.

Major isomer **5a:** Rf = 0.28 (AcOEt/Et₂O 1/2).

¹H NMR (CDCl₃) δ : 1.06 (t, 3H, J = 7Hz), 1.28 (t, 3H, J = 7Hz), 3.74– 4.32 (m, 2x2H), 5.96 (s, 2H), 7.07–7.1 (m, 1H), 7.14 (dd, 1H, J₁= 10Hz, J₂ = 15,5Hz), 7.26–7.94 (m, 6H), 7.7–7.73 (m, 1H), 7.3 (s,1H), 8.77 (dd, NH, J₁ = 10Hz, J₃ = 4Hz).

¹³C NMR (CDCl₃) δ: 16.2 (2x<u>C</u>H₃CH₂O), 48.27 (-<u>C</u>H₂-), 60.45 (d, -<u>C</u>H-, ¹J_{C-P} = 182.43 Hz), 64.78 (2x<u>C</u>H₂O), 124.65 (<u>C</u>H₅=C), 127.42, 127.94, 128, 128.55, 128.82, 132, 132.26, 132.64 (<u>C</u>₆H₅and <u>C</u>₄H₃S), 140.36 (CH₅=<u>C</u>), 161.5 (α-Thienyl-<u>C</u>N₄), 167.5 (C=O).

Anal. calcd. for C₂₀H₂₃N₈O₄PS: C, 47.81, H,4.58, N,22.31. Found.C, 47.72, H,4.59, N,21.85. Major isomer **6a:** Rf = 0.2 (AcOEt/DCM 2/3).

¹H NMR (CDCl₃+DMSO d₆) δ : 1.06 (t, 3H, J = 7Hz), 1.24 (t, 3H, J = 7Hz), 3.73–4.27 (m, 2x2H), 6.04 (s, 2H), 6.8–8 (m, 9H), 7.2 (dd, 1H, J₁ = 10Hz, J₂ = 15.5Hz), 8.52 (s, 1H), 9.68 (dd, NH, J₁ = 10Hz, J₃ = 4Hz), 11.5 (s, 1H).

¹³C NMR (CDCl₃+DMSO d₆) δ : 16.8 (2xCH₃CH₂O), 49.13 (-CH₂-), 62.96 (d, -CH-, ¹J_{C-P} = 180.1Hz), 64.7 (2xCH₂O), 125.3 (CH₅=C), 112.15, 118, 120.5, 126.4, 128.6, 128.8, 132.7, 132.9, 133.1, 156.64 (C₆H₅ and o-HO-C₆H₄), 140.67 (CH₅=C), 165.1 (o-HO-C₆H₄-CN₄), 168 (C=O). Anal. calcd. for C₂₂H₂₅N₈O₅P: C, 51.56, H,4.88, N,21.87.

Found.C, 51.40, H,4.71, N,21.55.

Minor isomer **7b:** Rf = 0.67 (ether).

¹H NMR (CDCl₃) δ : 1.01 (t, 3H, J = 7Hz), 1.29 (t, 3H, J = 7Hz), 3.63– 4.28 (m, 2x2H), 5.92 (s, 2H), 6.54–6.56 (m, 1H), 7.1 (dd, 1H, J₁= 10Hz, J₂ = 16Hz), 7.25–7.27 (m, 1H), 7.3–7.84 (m, 5H), 7.62–7.64 (m, 1H), 7.57 (s, 1H), 8.1 (dd, NH, J₁ = 10Hz, J₃= 4Hz).

¹³C NMR (CDCl₃) δ : 16.2 (2xCH₃CH₂O), 43.71 (-CH₂-), 64.04 (d, -CH-, ¹J_{C-P} = 180.52 Hz), 64.4 (2xCH₂O), 112.56, 115.44, 139.1, 145.5 (C₄H₃O), 127.50, 128.56, 132.23, 132.7 (C₆H₅), 134.56 (CH₄=C), 142.51 (CH₄=C), 145.91 (α-Furyl-CN₄), 167 (C=O). C₂₀H₂₃N₈O₅P (486): M.S [D.C.I/NH₃ (M+H)⁺ = 487 and (M+NH₄)⁺ = 504].

Major isomer **7a:** Rf = 0.48 (ether).

¹H NMR (CDCl₃) δ : 1.02 (t, 3H, J = 7Hz), 1.25 (t, 3H, J = 7Hz), 3.58– 4.32 (m, 2x2H), 5.97 (s, 2H),6.54–6.57 (m, 1H), 7.12 (dd, 1H, J₁= 10Hz, J₂ = 16Hz), 7.18–7.9 (m, 7H), 8.3 (s, 1H), 9 (dd, NH, J₁ = 10Hz, J₃ = 4Hz).

¹³C NMR (CDCl₃) δ: 16.3 (2x<u>C</u>H₃CH₂O), 43.8 (-<u>C</u>H₂-), 60.42 (d, -<u>C</u>H-, ¹J_{C-P} = 182.43 Hz), 64.7 (2x<u>C</u>H₂O), 124.51 (<u>C</u>H₅=C), 112.4, 115.36, 139.18, 145.8 (<u>C</u>₄H₃O), 128, 128.47, 132.3, 132.6 (<u>C</u>₆H₅ and <u>C</u>₄H₃S), 141.2 (CH₅=<u>C</u>), 146.4 (α-Furyl-<u>C</u>N₄), 167 (C=O).

Anal. calcd. for C₂₀H₂₃N₈O₅P: C, 49.38, H,4.73, N,23.04. Found.C, 48.98, H,4.66, N,22.71.

Major isomer 8a: Rf = 0.17 (AcOEt/DCM 1/1).

¹H NMR (CDCl₃+DMSO d₆) δ : 1.03 (t, 3H, J = 7Hz), 1.21 (t, 3H, J = 7Hz), 3.65–4.27 (m, 2x2H), 5.85 (s, 2H), 6.19–6.28 (m, 1H), 6.86–6.87 (m, 1H), 6.97–6.98 (m, 1H), 7.14 (dd, 1H, J₁ = 10Hz, J₂ = 16Hz),

7.3–7.8 (m, 5H), 8.39 (s, 1H), 9.75 (dd, NH, $J_1 = 10Hz$, $J_3 = 4Hz$), 11.88 (m, NHpyrryl).

¹³C NMR (CDCl₃+DMSO d₆) δ: 16.49 (2x<u>C</u>H₃CH₂O), 43.42 (-<u>C</u>H₂-), 60.67 (d, -<u>C</u>H-, ${}^{1}J_{C-P} = 182.43$ Hz), 64.22 (2x<u>C</u>H₂O), 110.38, 112.53, 114.32, 123.4 (C4H4N), 124.21 (CH5=C), 128.35, 128.52, 132.5, 132.84 $(\underline{C}_{6}H_{5})$, 141.56 (CH₅=<u>C</u>), 148.53 (α -Pyrryl-<u>C</u>N₄), 167.3 (C=O).

Anal. calcd. for C₂₀H₂₄N₉O₄P: C, 49.48, H,4.95, N,25.98. Found.C, 49.14, H,4.83, N,25.95.

References

- [1] a) A. Elachgar, A. El Hallaoui, M. L. Roumestant, Ph. Viallefont, Synthetic Comm., 24, 1279 (1994). b) S. Achamlale, A. Elachgar, A. El Hallaoui, M. L. Roumestant, Ph. Viallefont, Aminoacids, 12, 257 (1997). c) A. Alami, A. El Hallaoui, A. Elachgar, M. L. Roumestant, Ph.Viallefont, Bull. Soc. Chim. Belg., 105, 769 (1996). d) A. Atmani, S. Elhajji, A. El Hallaoui, M. L. Roumestant, Ph. Viallefont, Synthetic Comm., 21. 2383 (1991). e) F. Zaïd, Thesis, Fès, (1996). [2] P. Kafarski, B. Lejczak, Phosphorus, Sulfur and Silicon, 63, 193 (1991).
- [3] B. Lejczak, P. Kafarski, H. Sztajer, P. Mastalerz, J. Med. Chem., 29, 2212 (1986). V. P. Kukhar, N. M. Solodenko, V. A. Solodenko, Ukr. BioKhim. Zh., 60, 95 (1988). [4]
- [5] a) W. H. W. Lunn, D. D. Schoepp, D. O. Calligaro, R. T. Vasileff, L. J. Heinz, C. R. Salhoff, P. J. O'Malley, J. Med. Chem., 35, 4608 (1992). b) D. D. Schoepp, C. L. Smith, D. Lodge, J.D. Millar, J.D. Leander, A.I. Sacaan,
- W.H.W. Lunn, European J. Pharmacology, 203, 237 (1992). [6] J. A. Monn, M. J. Valli, R. A. True, D. D. Schoepp, J. D. Leander, D. Lodger, Bioorg. Med. Chem. Lett., 3, 95 (1993).
- [7] A. Antonowa, S. Hauptmann, Z. Chem., 16, 17 (1976).
- [8] J. Elguero, C. Marzin, J. D. Roberts, J. Org. Chem., 39, 357 (1974).
- S. Achamlale, A. Elachqar, A. El Hallaoui, S. Elhajji, A. Alami, M. L. Roumestant, Ph. [9] Viallefont, J. Heterocyclic Chem., in press.
- [10] L. Birkofer, A. Ritter, H. Uhlenbravick, Chem. Ber., 96, 3280 (1963).
- [11] G. I. Tsypin, T. N. Timofeeva, V. V. Mel'Nikov, B. V. Gidaspov, Zh. Org. Khim., 11, 1395 (1975). and 13, 2275 (1977).
- [12] M. Begtrup, J. Chem. Soc. Perkin Trans. II, 736 (1976).