

Heteroarylboronates in Rhodium-Catalyzed 1,4-Addition to Enones

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Supporting Information

ABSTRACT: Rhodium(I)-catalyzed 1,4-addition of aryl and alkenylboronic acids to α,β -unsaturated carbonyl compounds is well established, but the transfer of heteroaryl residues in this reaction remains underdeveloped. We have studied heteroaryl MIDA and pinacol boronates as alternatives to the labile boronic acid counterparts. Under racemic conditions, 12 adducts with heteroaryl residues, among them unsubstituted 3- and 4-pyridinyl, 2-furanyl, thienyl, and pyrrolyl groups, were obtained in moderate to excellent yields. The enantioselective version of the reaction proved highly sensitive to the electronic character of the heteroaryl substituents, with boronates carrying electron-rich residues giving modest to high yields but consistently high enantiomeric excesses.

 $X_{n}B = HetAr \xrightarrow{Rh(l)} HetAr$ $X_{n}B = HetAr = Het$

R hodium(I)-catalyzed 1,4-addition of boronic acids (2) to α,β -unsaturated carbonyl compounds (1) has emerged as a powerful tool for C–C bond formation under the setup of a new stereocenter.^{1,2} While the reaction is well-documented for aryl and alkenylboronic acids, there are few examples for the transfer of heteroaryl residues³ (Scheme 1). This is due to the



lability of heteroarylboronic acids,⁴ resulting in rapid protodeboration⁵ under the aqueous basic conditions of rhodium-mediated 1,4-additions. Therefore, alternative reagents have to be sought.

Recently, Frost⁶ and Martin⁷ reported asymmetric 1,4addition of heteroarylzinc and titanium reagents under anhydrous conditions, but these organometallics are highly sensitive and not commercially available, rendering this approach inconvenient. A more attractive alternative is to replace heteroarylboronic acids with stabilized boronate reagents, e.g. pinacol boronates (4), trifluoroborates⁸ (5), cyclic triol borates⁹ (6), or *N*-methylimino diacetic acid (MIDA) boronates¹⁰ (7) (Figure 1). These compounds continuously generate boronic acids under basic aqueous conditions and are highly successful in Suzuki–Miyaura reactions,¹¹ but there are surprisingly few examples for heteroarylboronates in rhodium-catalyzed 1,4-additions.

Aryl pinacol boronates are well-known in 1,4-additions,¹² but only a few heteroaryl derivatives have been studied.¹³



Figure 1. Heteroarylboronates as alternatives to boronic acids.

Trifluoroborates were reported in racemic and stereoselective 1,4-additions by Batey¹⁴ and Genêt,¹⁵ but successful heteroaryl examples are restricted to the 3-thienyl group. Studies on heteroaryl triolborates were performed by Miyaura.¹⁶ With these, asymmetric 1,4-addition of 2-furanyl groups^{16b} and methoxy-substituted 2-thienyl as well as 3- and 2-pyridinyl residues was successful, while the reaction failed without additional electron-donating substituents on the heteroaryl cores.^{16a}

MIDA boronates were introduced to Suzuki–Miyaura couplings by Burke,¹⁷ and currently, they are the reagents of choice for the transfer of heteroaryl residues in this cross-coupling reaction, which also requires aqueous basic conditions.¹⁰ Surprisingly, there are no reports on heteroaryl MIDA boronates in 1,4-additions; to the best of our knowledge, only aryl¹⁸ and alkenyl¹⁹ MIDA boronates have been employed for this transformation.²⁰ Our group has introduced new carbohydrate-derived olefin ligands for stereoselective rhodium-catalyzed 1,4-addition,^{21–24} and in the course of this work, we became interested in studying the reaction of heteroaryl-boronates in more detail.

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As many MIDA boronates (7) and pinacol boronates (4) with heteroaryl groups are commercially available, we performed our investigation on racemic 1,4-additions to cyclohexenone (1) with these substrates (Table 1). Using $[Rh(cod)OH]_2$ (cod: 1,5-cyclooctadiene) as the catalyst and

Table 1. Racemic 1,4-Addition of Heteroaryl MIDA and
Pinacol Boronates to Cyclohexenone

HetAr-B(MIDA) 7 (2 equiv)										
or │										
[Rh(cod)OH] ₂ (2.5 mol %)										
	1 dioxan	e/H ₂ O (2	;3), 16 h	3 Her	tAr					
	boronate 7	temp [°C]	pro	duct 3						
	HetAr		yield [%]ª							
1 ^b	ror contraction of the second s	7a	60	3a	91					
2	ros S	7 b	60	3b	71					
3	MeO N	7c	60	3c	90					
4	CIN	7 d	60	3d	88					
5	FN	7e	60	3e	46					
6	ج ^ح	7 f	60		42					
7		7f	rt	3f	-					
8	N ²	4f	60		73					
9	N	7g	rt	3g	-					
10	rrs .	7 h	60	.1	85					
11	N N	4h	60	3h	31					
12	کے	7i	60		34					
13	ra l	7i	rt	3i	74					
14 ^c	Ŭ Ó	4i	rt		95					
15	non and	7j	rt	2:	24					
16	s_/	4j	rt	3)	30					
17	nor -	7k	rt	34	36					
18	01	4k	rt	JK	32					
19	BocN	7 m	60	3m	75					
20	MeN	4n	rt	3n	99					
21°	NTIPS	40	rt	30	99					



phenyl MIDA boronate (7a) as a model substrate, we set out to develop suitable reaction conditions. After extensive optimization, K_3PO_4 (4 equiv) as the base, a 2:3 mixture of dioxane and water, and a reaction temperature of 60 °C were identified as optimal, giving adduct 3a in high yield (entry 1). With these optimized conditions, we started our study on heteroaryl MIDA boronates.

As the first example, we investigated 3-thienyl MIDA boronate 7b. Most known examples of heteroaryl transfer in 1,4-additions involve the 3-thienyl residue, ${}^{3a,b,13-16}$ and indeed, the reaction with 7b proceeded smoothly (entry 2). Pyridinylboronic acids are known for their lability,⁴ and so far pyridinyl triol borates with additional electron-donating substituents on the pyridine core were the only successful substrates.^{16a} Therefore, we were very pleased to find that, apart from activated 3-pyridinyl derivative 7c, 3-pyridinyl MIDA boronates 7d and 7e with chloro and fluoro substitution as well as derivative 7f carrying an unsubstituted pyridinyl group also gave the expected products (entries 3-6). Yields were strongly dependent on the electronic makeup of the 3-pyridinyl residue, with 90% for methoxy derivative 7c and 88% for chloro derivative 7d, but yields <50% were obtained for the addition products of fluorinated and unsubstituted 3-pyridinyl MIDA boronates 7e and 7f. An attempt to improve the yield for 7f by performing the reaction at rt resulted in no conversion (entry 7). Therefore, we repeated the experiment with 3-pyridinyl pinacol boronate 4f as an alternative reagent at 60 °C, which led to a substantially improved yield of 73% (entry 8). Boron reagents with unsubstituted 2-pyridinyl groups are notorious for their low stability,⁴ and unfortunately, MIDA derivative 7g failed to yield any product (entry 9). 4-Pyridinyl MIDA and pinacol boronates 7h and 4h produced 3h in 85% and 31% yield, respectively (entries 10, 11); in a related study, the corresponding boronic acid led to a very modest yield.³

Next, we turned our attention to unsubstituted fivemembered heteroarenes. At 60 °C, 3-furanyl MIDA derivative 7i gave product 3i in moderate yield, but reactions with 7i or pinacol boronate 4i at rt led to greatly improved results (entries 12-14).

Unsubstituted 2-furanyl-, thienyl-, and pyrrolylboronates are challenging substrates,⁴ and so far, there are only a few reports concerning 2-thienyl^{3c} and furanyl^{16b} residues. At rt, 2-thienyl MIDA boronate 7j and pinacol boronate 4j gave 3j in moderate yield (entries 15, 16). Similar results were obtained for 2-furanyl derivatives 7k and 4k (entries 17, 18). In contrast, *N*-Boc 2-pyrrolyl MIDA boronate 7m produced 3m in high yield, and more electron-rich *N*-methyl 2-pyrrolyl pinacol boronate 4n even gave 99% of 3n (entries 19, 20). The same excellent yield was obtained with 3-pyrrolyl boronate 4o²⁵ (entry 21).

After the encouraging results of the racemic study, we set out to develop an asymmetric protocol. As a catalyst system we chose $[RhCl(C_2H_4)_2]_2$ in the presence of *fuco*-enoPhos (8) or *galacto*-enoPhos (*ps*-ent-8) (Table 2). Both chiral ligands are accessible from inexpensive monosaccharides^{23,24} and act as *pseudo* enantiomers in 1,4-additions of arylboronic acids.²⁴ In the presence of the rhodium complex with *fuco*-ligand 8, and under otherwise unchanged conditions, phenyl MIDA boronate **7a** smoothly gave adduct **3a** as the expected (*R*)-enantiomer²⁴ in high yield and excellent ee (entry 1). Reaction of 3-thienyl MIDA derivative **7b** in the presence of 8 at 60 °C yielded 65% of (*R*)-**3b** in 91% ee, while *galacto*-ligand *ps*-ent-**8b** produced the (*S*)-enantiomer in 56% yield and 96% ee (entries 2, 3). At rt, the yield and stereoselectivity improved for the reaction with

	HetAr-B(MIDA) 7 (2 equiv)				fuco-enoPhos galacto-enoPhos		
	HetAr-Bpin 4 (2 equiv)				Me O or O		
		$[RhCl(C_2H_4)_2]_2$ (2.5 r	nol %)	\square	OPPh ₂	F F	ξ ¹
	1	K ₃ PO ₄ (4 equiv) dioxane/H ₂ O (2:3), 1	6 h	HetAr 3	8	ps-ent- 8a R ¹ ps-ent- 8b R ¹	= OEt = H
entry	boronate 7 or 4		enoPhos ligand temp [°C			product 3	
	HetAr					yield [%]ª	ee [%] ^b
1 ^c	P. P. S.	7a	8	60	3a	88	97 (R)
2			8	60		65	91 (R)
3	non a	\	ps-ent- 8b	60		56	96 (<u>S</u>)
4	Ls	> 7b	8	rt	3b	77	95 (<i>R</i>)
5			ps-ent- 8b	rt		56	98 (<i>S</i>)
6			8	60		33	90 (+)
7	Parts 1	-	ps-ent- 8a	60	2.	44	94 (-)
8 ^d	MeO	v /c	8	rt	3c	80	95 (+)
9			ps-ent- 8a	rt		37	97 (-)
10	ş	7i	8			24	91 (+)
11	rr II	> 7i	ps-ent- 8b	rt	3i	19	93 (-)
12	<u> </u>) 4i	8			37	89 (+)
13	کر	7j	8			15	90 (-)
14	P22	> 7j	ps-ent- 8b	rt	3j	10	95 (+)
15	S-//	4j	8			11	93 (-)
16	کر	7k	8			14	81 (-)
17	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~		ps-ent- 8b	rt	3k	16	90 (+)
18	0-1	4k	8			19	71 (-)
19	MeN		8	rt	3n	81	82 (+)
20 ^e	NT	40	ps-ent- 8b	rt	30	56	94 (+)

Table 2. Asymmetric 1,4-Addition of Heteroaryl MIDA Boronates and Pinacol Boronates to Cyclohexenone

^{*a*}Isolated yield after chromatography. ^{*b*}Determined by chiral GC. ^{*c*}Reaction with 1.5 mol % $[RhCl(C_2H_4)_2]_2$ and 3.3 mol % of ligand 8. ^{*d*}Reaction for 6 days. ^{*c*}The pinacol boronate was added via syringe pump.

ligand 8 (77% and 95% ee), while ligand *ps-ent-*8b gave 3b in higher stereoselectivity but in the same yield as previously (entries 4, 5).

As observed for arylboronic $acids^{24}$ and as in the case of boronate 7b, ligands 8 and *ps-ent-*8b consistently yielded 1,4adducts in opposite configuration (cf. Table 2). At 60 °C, methoxy-substituted 3-pyridinyl MIDA boronate 7c led to a reduced yield but high enantioselectivity (entries 6, 7). Reduction of the temperature and a longer reaction time greatly improved the yield and stereoselectivity for the run with ligand 8 (80%, 95% ee) (entry 8). The experiment with *ps-ent-*8b at rt but for 16 h led to 98% ee but only a moderate yield (entry 9). In contrast to the racemic reactions, 3- and 4pyridinyl MIDA boronates 7d–f and 7h as well as pinacol boronates 4f and 4h did not yield any product in the presence of 8 or *ps-ent-*8b. Unfortunately, neither a longer reaction time nor altered conditions led to any improvement.

All asymmetric reactions involving five-membered heteroraryl residues were conducted at rt. With 3-furanyl MIDA boronate 7i, the reaction led to adduct 3i in yields around 20% with >90% ee; by replacing 7i with pinacol boronate 4i, the yield was almost doubled while the ee was slightly lower (entries 10–12). For the challenging 2-thienyl MIDA ester 7j, the yields were modest, while high levels of stereoinduction were retained (entries 13, 14); pinacol boronate 4j gave very similar results (entry 15). With 2-furanyl MIDA and pinacol esters 7k and 4k, the yields did not exceed 20%, and for the experiments with ligand 8, a drop in stereoselectivity was observed, while *ps-ent-*8b produced 3k with 90% ee (entries 16–18). N-Boc 2-pyrrolyl MIDA boronate 7m did not yield any product, but pinacol boronate 4n with an electron-donating *N*-methyl group gave 3n in 81% yield with 82% ee (entry 19). Finally, pinacol boronate 4o with the *N*-TIPS 3-pyrrolyl residue yielded 56% of 3o with 94% ee (entry 20).

The results from our studies can be interpreted as follows: Both heteroaryl MIDA and pinacol boronates are suitable for racemic 1,4-addition, but the efficiency of the reaction strongly depends on the electronic nature of the heteroaromatic system and on the position of the boronate group. In the pyridinyl series, the reaction is successful for 3- and 4-pyridinylboronates and electron-donating substituents lead to improved results. In the series of five-membered heteroaryl groups, pinacol boronates tend to give higher yields than MIDA boronates and 3-pyrrolyl, thienyl, and furanyl derivatives produce the best results. For 2-boronated pyrrolyl, thienyl, and furanyl compounds, yields vary greatly with the heteroaryl group. Good to excellent results are obtained for electron-rich 2pyrrolylboronates, while thienyl and furanyl reagents at best give moderate yields. The asymmetric reaction is even more sensitive toward the substrate. Of the pyridine series, only the methoxy-substituted 3-pyridinyl derivative gives the desired product. For five-membered heteroaryl residues, high vields are obtained with 3-thienyl and pyrrolyl derivatives, while those for 3-furanylboronates are moderate. 2-Thienyl and furanyl derivatives produce modest yields, and for 2-pyrrolylboronates, only the electron-rich substrate was successful. Further, it should be noted that all but three of the asymmetric reactions gave more than 90% ee.

The modest yields for many reactions from Table 2 may be caused by two distinctive factors: (1) Protodeboration of the heteroarylboronic acids liberated from the respective boronate may be faster than transmetalation to the rhodium catalyst or (2) after transmetalation, protoderhodation of the heteroarylrhodium complex may become a competing side reaction to the desired 1,4-addition. We are currently conducting NMR studies to elucidate which of these factors is crucial for 1,4-additions with heterorarylboronates. Adjusting the decomposition rate of the boronate reagent may to some extent avoid unproductive protodeboration of the liberated heteroaylboronic acid. The transmetalation and/or the carborhodation steps may become faster by employing modified catalysts, e.g. rhodium complexes of electron-deficient phosphines.²⁶

In conclusion, we have presented an extensive study on heteroarylboronates in racemic and stereoselective rhodiumcatalyzed 1,4-additions. Further studies, aiming to improve product yields and the scope of the asymmetric version, are currently in progress in our laboratory.

ASSOCIATED CONTENT

Supporting Information

Full experimental details, characterization data, copies of spectra and chromatograms. This material is available free of charge via the Internet at http://pubs.acs.org.

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Notes

The authors declare no competing financial interest.

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REFERENCES

(1) Takaya, Y.; Ogasawara, M.; Hayashi, T.; Sakai, M.; Miyaura, N. J. Am. Chem. Soc. **1998**, 120, 5579. (2) Reviews: (a) Hayashi, T.; Yamasaki, K. *Chem. Rev.* 2003, 103, 2829. (b) Tian, P.; Dong, H.-Q.; Lin, G.-Q. ACS Catal. 2012, 2, 95.
(3) Examples of heteroarylboronic acids in 1,4-addition: (a) Yoshida, K.; Hayashi, T. *Heterocycles* 2003, 59, 605. (b) Tokunaga, N.; Hayashi,

T. Adv. Synth. Catal. 2007, 349, 513. (c) Konno, T.; Tanaka, T.; Miyabe, T.; Morigaki, A.; Ishihara, T. Tetrahedron Lett. 2008, 49, 2106. (d) Collier, P. N. Tetrahedron Lett. 2009, 50, 3909.

(4) Tyrell, E.; Brookes, P. Synthesis 2003, 469.

(5) (a) Kuivila, H. G.; Reuwer, J. F., Jr.; Mangravite, J. A. Can. J. Chem. 1963, 41, 3081. (b) Billingsley, K.; Buchwald, S. L. J. Am. Chem. Soc. 2007, 129, 3358.

(6) Le Nôtre, J.; Allen, J. C.; Frost, C. G. Chem. Commun. 2008, 3795.

(7) Smith, A. J.; Abbott, L. K.; Martin, S. F. Org. Lett. 2009, 11, 4200.

(8) Darses, S.; Genêt, J.-P. Chem. Rev. 2008, 108, 288.

(9) Yamamoto, Y.; Takizawa, M.; Yu, X.-Q.; Miyaura, N. Angew. Chem., Int. Ed. 2008, 47, 928.

(10) Gillis, E. P.; Burke, M. D. Aldrichimica Acta 2009, 42, 17.

(11) Lennox, A. J. J.; Lloyd-Jones, G. C. Chem. Soc. Rev. 2014, 43, 412.

(12) Sakuma, S.; Sakai, M.; Itooka, R.; Miyaura, N. J. Org. Chem. 2000, 65, 5951.

(13) Tajuddin, H.; Shukla, L.; Maxwell, A. C.; Marder, T. B.; Steel, P. G. Org. Lett. **2010**, *12*, 5700.

(14) Batey, R. A.; Thadani, A. N.; Smil, D. V. Org. Lett. 1999, 1, 1683.
(15) Pucheault, M.; Darses, S.; Genêt, J.-P. Eur. J. Org. Chem. 2002,

3552. (16) (a) Yu, S.-Q; Yamamoto, Y.; Miyaura, N. Synlett **2009**, 994.

(b) Yu, X.-Q.; Shirai, T.; Yamamoto, Y.; Miyaura, N. Chem.—Asian J. 2011, 6, 932.

(17) Knapp, D. M.; Gillis, E. P.; Burke, M. D. J. Am. Chem. Soc. 2009, 131, 6961.

(18) Ohlmann, D. M.; Gooßen, L. J.; Dierker, M. Chem.-Eur. J. 2011, 17, 9508.

(19) Pattison, G.; Piraux, G.; Lam, H. W. J. Am. Chem. Soc. 2010, 132, 14373.

(20) A single report has described the stereoselective 1,2-addition of a heteroaryl MIDA to an imine: Brak, K.; Ellman, J. A. J. Org. Chem. **2010**, 75, 3147.

(21) Minuth, M.; Boysen, M. M. K. Org. Lett. 2009, 11, 4212.

(22) Grugel, H.; Minuth, T.; Boysen, M. M. K. Synthesis **2010**, 3248. (23) Grugel, H.; Albrecht, F.; Minuth, T.; Boysen, M. M. K. Org. Lett.

2012, *14*, 3780.

(24) Grugel, H.; Albrecht, F.; Boysen, M. M. K. Adv. Synth. Catal. 2014, DOI: 10.1002/adsc.201400459.

(25) Preparation of pinacol boronate **40** from pyrrole: (a) Muchowski, J. M.; Solas, D. R. *Tetrahedron Lett.* **1983**, *24*, 3455. (b) Morrison, M. D.; Hanthorn, J. J.; Pratt, D. A. Org. Lett. **2009**, *11*, 1051. (c) Ayats,

C.; Soley, R.; Albericio, F.; Álvarez, M. Org. Biomol. Chem. 2009, 7, 860.

(26) Genêt, J. P.; Ayad, T.; Ratovelomanana-Vidal, V. Chem. Rev. 2014, 114, 2824.