

## Synthesis and Structure–Activity Relationships of Potent and Orally Active 5-HT<sub>4</sub> Receptor Antagonists: Indazole and Benzimidazolone Derivatives

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A series of indole-3-carboxamides, indazole-3-carboxamides, and benzimidazolone-3-carboxamides was synthesized and evaluated for antagonist affinity at the 5-HT<sub>4</sub> receptor in the rat esophagus. The *endo*-3-tropanamine derivatives in the indazole and benzimidazolone series possessed greater 5-HT<sub>4</sub> receptor affinity than the corresponding indole analogues. 5-HT<sub>4</sub> receptor antagonist affinity was further increased by alkylation at N-1 of the aromatic heterocycle. In a series of 1-isopropylindazole-3-carboxamides, replacement of the bicyclic tropane ring system with the monocyclic piperidine ring system or an acyclic aminoalkylene chain led to potent 5-HT<sub>4</sub> receptor antagonists. In particular, those systems in which the basic amine was substituted with groups capable of forming hydrogen bonds showed increased 5-HT<sub>4</sub> receptor antagonist activity. While some of these compounds displayed high affinity for other neurotransmitter receptors (in particular, 5-HT<sub>3</sub>,  $\alpha_1$ , and 5-HT<sub>2A</sub> receptors), as the conformational flexibility of the amine moiety increased, the selectivity for the 5-HT<sub>4</sub> receptor also increased. From this series of compounds, we identified LY353433 (1-(1-methylethyl)-*N*-[2-[4-[(tricyclo[3.3.1.1<sup>3,7</sup>]dec-1-ylcarbonyl)amino]-1-piperidinyl]ethyl]-1*H*-indazole-3-carboxamide) as a potent and selective 5-HT<sub>4</sub> receptor antagonist with clinically suitable pharmacodynamics.

### Introduction

The neurotransmitter serotonin (5-HT) modulates the activity of the central nervous system and peripheral tissues through its actions at a number of receptor subtypes.<sup>1</sup> Because of the wide range of physiologic and pathophysiologic systems in which 5-HT is known to play a role, considerable attention has centered around the identification of agents which act selectively at each of these receptor subtypes. One of these receptor subtypes, the 5-HT<sub>4</sub> receptor,<sup>2</sup> was first identified in 1988<sup>3</sup> and has recently been cloned.<sup>4</sup> Based on the localization of the 5-HT<sub>4</sub> receptor in gastrointestinal tissue, atrial tissue, urinary bladder tissue, and the central nervous system, a number of therapeutic indications have been proposed for both agonists and antagonists at this receptor. Proposed therapeutic applications for agents that block the 5-HT<sub>4</sub> receptor include the treatment of irritable bowel syndrome, atrial arrhythmias, urinary incontinence, and various diseases of the central nervous system. We were interested in identifying a potent and selective 5-HT<sub>4</sub> receptor antagonist as a possible therapeutic agent.

Although a number of compounds have been identified which block 5-HT<sub>4</sub> receptors, most do not have the requisite selectivity or pharmacodynamic properties which would make them ideal pharmacologic tools or therapeutic agents. Tropisetron (**1**) (Chart 1) was among the first compounds shown to possess 5-HT<sub>4</sub> receptor antagonist activity.<sup>5</sup> However, **1** has higher affinity at 5-HT<sub>3</sub> than 5-HT<sub>4</sub> receptors. The benzamide

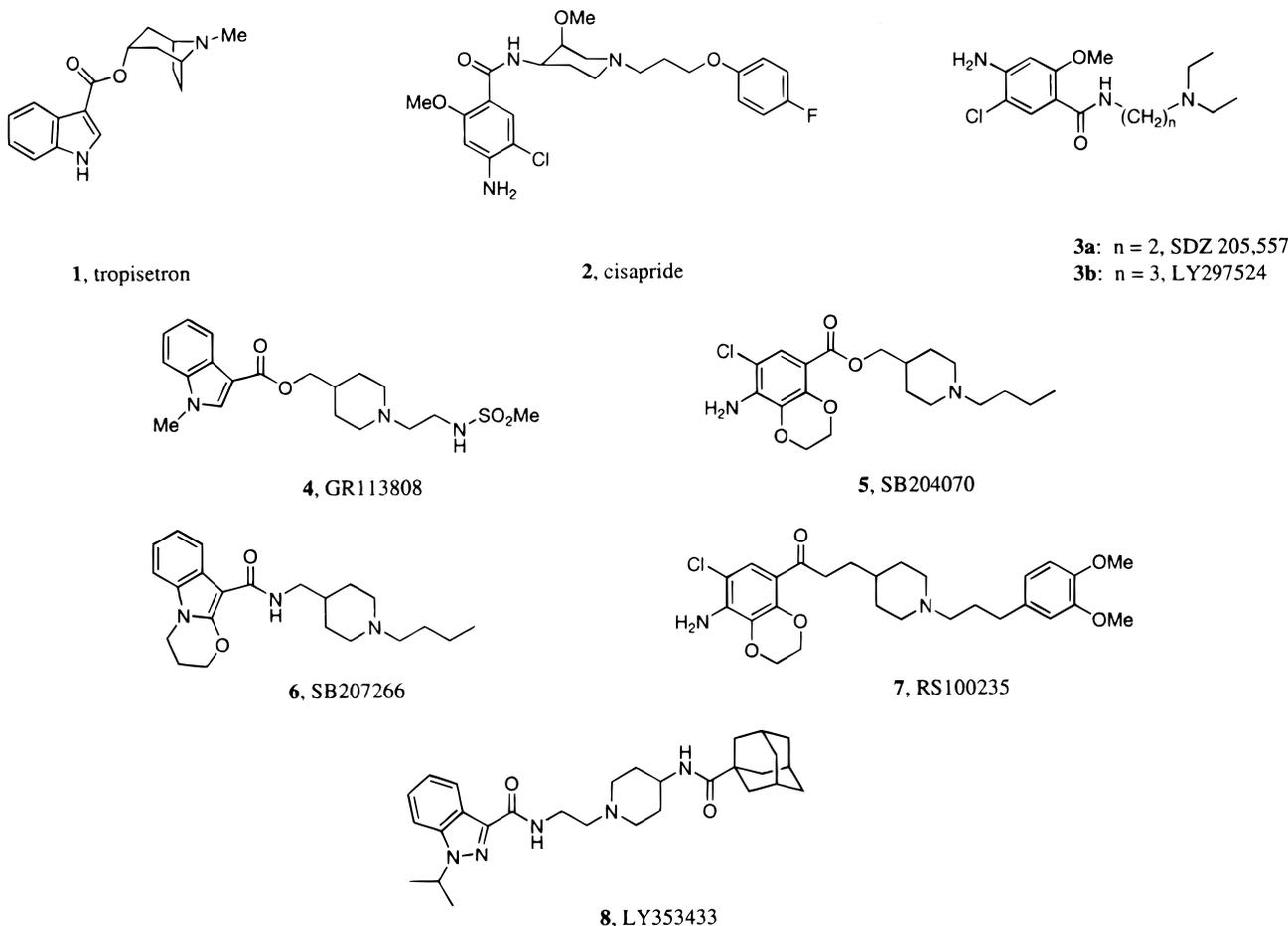
cisapride (**2**) has been shown to be a partial agonist at 5-HT<sub>4</sub> receptors but also has high affinity for 5-HT<sub>2A</sub> and  $\alpha_1$  receptors.<sup>6</sup> SDZ 205,557 (**3a**)<sup>7</sup> and LY297524 (**3b**),<sup>8</sup> both esters related to metoclopramide, are reported to be 5-HT<sub>4</sub> receptor antagonists but also have considerable 5-HT<sub>3</sub> receptor antagonist activity.<sup>8</sup> GR113808 (**4**)<sup>9</sup> and SB204070 (**5**)<sup>10</sup> are the first reported 5-HT<sub>4</sub> receptor antagonists with high affinity and selectivity for the 5-HT<sub>4</sub> receptor. However, neither of these compounds has a clinically suitable duration of action following oral administration, presumably because the ester linkage present in both molecules is readily cleaved in vivo.<sup>2c,11</sup> SB207266 (**6**)<sup>12</sup> and RS100235 (**7**)<sup>13</sup> are more recently reported 5-HT<sub>4</sub> receptor antagonists. Both compounds demonstrated improved pharmacodynamics over **4** and **5**, presumably because the amide and ketone systems of **6** and **7**, respectively, are more stable than the ester linkage found in **4** and **5**.

We initiated studies with the goal of identifying structurally distinct, potent, selective, and orally active 5-HT<sub>4</sub> receptor antagonists with a clinically suitable in vivo duration of action. The present report details the structure–activity relationships (SAR), synthesis, and pharmacological characterization of LY353433 (**8**), 1-(1-methylethyl)-*N*-[2-[4-[(tricyclo[3.3.1.1<sup>3,7</sup>]dec-1-ylcarbonyl)amino]-1-piperidinyl]ethyl]-1*H*-indazole-3-carboxamide) a potent, selective, and long-acting compound from this series.<sup>6</sup>

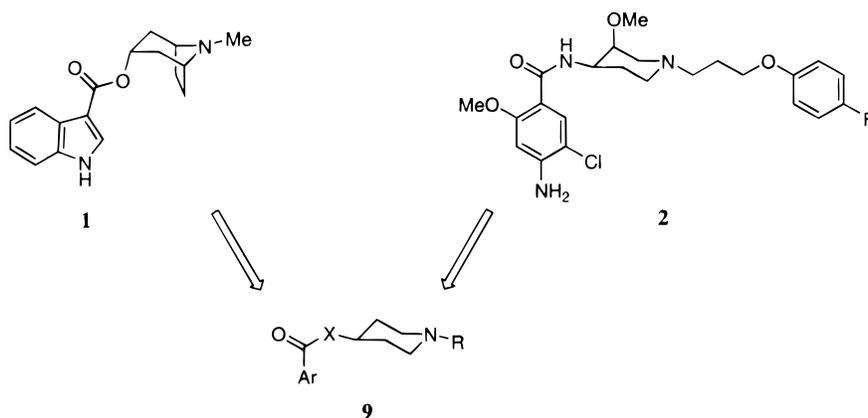
Tropisetron (**1**) and cisapride (**2**) served as starting points for our structure–activity studies. While both compounds have affinity for the 5-HT<sub>4</sub> receptor, neither is particularly potent and both have high affinity for

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## Chart 1



## Chart 2



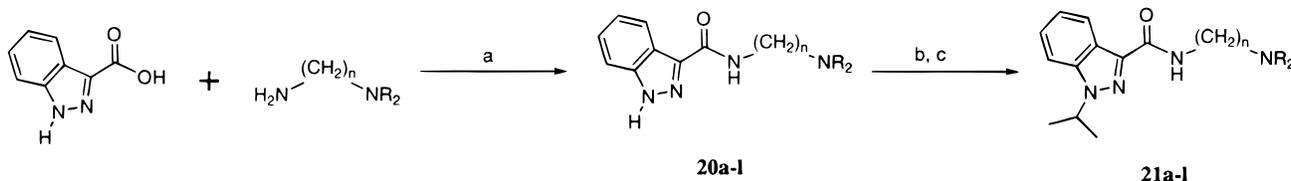
other receptors. A common pharmacophore (**9**) can be drawn for the two compounds which is centered around a piperidine ring system (Chart 2). For tropisetron, the piperidine is further rigidified into a tropane ring system. In each case, an aromatic acyl group is connected to the 4-position of the piperidine through either an ester or amide linkage. The basic nitrogen of the piperidine is also a site for substitution. We explored the modification of each of these elements of the proposed pharmacophore: the aromatic acyl group, the piperidine ring system, and the substituent on the basic nitrogen. Because of the expected lability of the ester linkage in compounds such as tropisetron, we focused our synthetic efforts on compounds in which the acyl

group was connected to the piperidine ring through an amide linkage.

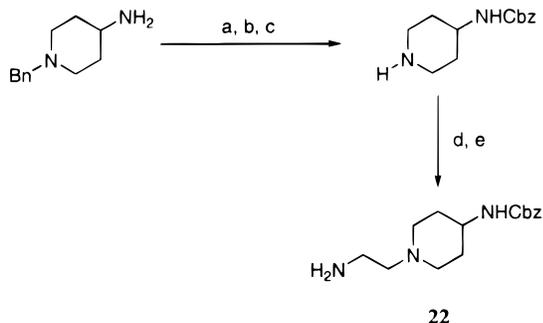
## Chemistry

The synthesis of the indolecarboxamides **11a–f** and the indazolecarboxamides **12a–f** is outlined in Scheme 1. Treatment of indole-3-carboxylic acid with oxalyl chloride provided the corresponding acid chloride which was reacted with the known tropanamine **5**<sup>14</sup> to give **11a**. Alkylation at N-1 was achieved by sequential treatment with sodium hydride and an alkyl halide to yield **11b–f**. The analogous indazoles were prepared by carbonyldiimidazole-mediated coupling between indazole-3-carboxylic acid<sup>15</sup> and **10** to give **12a**. Alkylation

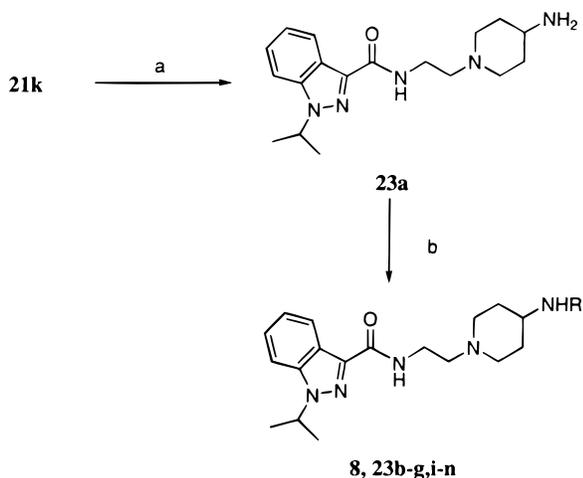


Scheme 3<sup>a</sup>

<sup>a</sup> (a) 1,1-Carbonyldiimidazole, DMF; (b) NaH, DMF; (c) *i*-PrI.

Scheme 4<sup>a</sup>

<sup>a</sup> (a) Cbz-Cl, NaOH, H<sub>2</sub>O, CH<sub>2</sub>Cl<sub>2</sub>; (b) MeCH(Cl)OCOCl, CH<sub>2</sub>Cl<sub>2</sub>; (c) MeOH, heat; (d) *N*-(2-bromoethyl)phthalimide, K<sub>2</sub>CO<sub>3</sub>, DMF; (e) NH<sub>2</sub>NH<sub>2</sub>·H<sub>2</sub>O, EtOH, heat.

Scheme 5<sup>a</sup>

<sup>a</sup> (a) H<sub>2</sub>, Pd/C, MeOH; (b) RCOCl, NEt<sub>3</sub>, THF; or MsCl, NEt<sub>3</sub>, THF.

chloride to yield **19i**, **19j**, and **19k**, respectively. Amides **21a-l** were prepared by carbonyldiimidazole-mediated acylation of the corresponding aminoalkyl derivative with indazole-3-carboxylic acid followed by regioselective alkylation with sodium hydride and isopropyl iodide (Scheme 3).

For the synthesis of **21k**, the required aminoethyl derivative, **22**, was prepared according to Scheme 4. Schotten-Bauman acylation of 4-amino-1-benzylpiperidine with benzyl chloroformate followed by removal of the benzyl group using von Braun conditions produced 4-((carbobenzyloxy)amino)piperidine. Alkylation with *N*-(2-bromoethyl)phthalimide and treatment with hydrazine to remove the phthalimide protecting group yielded **22**. Hydrogenolytic cleavage of the carbobenzyloxy protecting group of **21k** provided **23a** which served as a key synthetic intermediate (Scheme 5). Functionalization of **23a** under standard conditions provided compounds **8** and **23b-g,i-n**.

## Results and Discussion

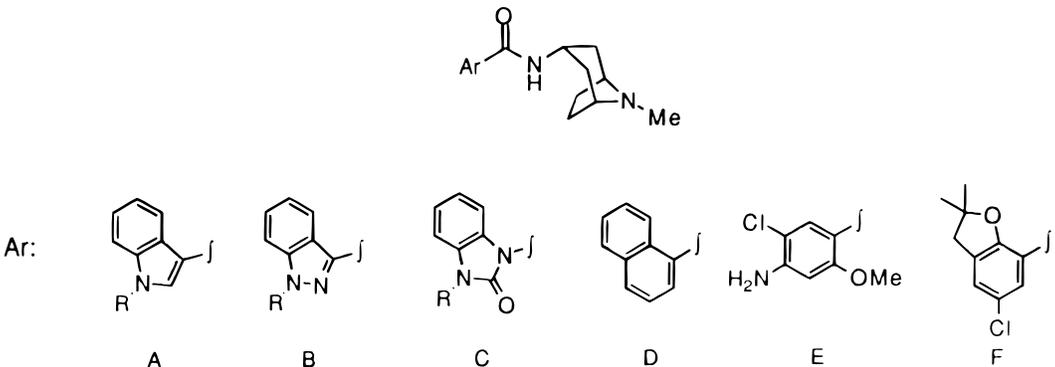
In the carbamylcholine-contracted rat esophagus preparation, serotonin produces a concentration-dependent relaxation mediated through activation of 5-HT<sub>4</sub> receptors.<sup>8,20</sup> The 5-HT<sub>4</sub> receptor antagonist affinity of compounds was evaluated by their ability to block serotonin-induced relaxation in carbamylcholine (10<sup>-6</sup> M)-contracted esophagus. Tissues were challenged with serotonin (1 μM), and relaxation of the tissue in the presence of a given concentration of the antagonist or vehicle was measured as reported in Tables 1–5. In the presence of vehicle, 1 μM serotonin produced 68.8 ± 4.3% (*N* = 10) relaxation of the carbamylcholine (10<sup>-6</sup> M)-contracted rat esophagus. Thus, the greater the antagonist affinity, the less relaxation produced by serotonin (1 μM).

**Modification of the Aromatic Acyl Group.** Using the amide analogue of tropisetron, **11a**, as our starting point, our initial goal was to study the effect of modification of the aromatic acyl group on 5-HT<sub>4</sub> receptor antagonist affinity. Our examination focused on a series of bicyclic aromatic amides. Replacement of the ester functionality in tropisetron by an amide linkage led to a significant decrease in 5-HT<sub>4</sub> receptor affinity (Table 1, **11a**). This finding is consistent with a study of metoclopramide analogues in which ester derivatives possessed an order of magnitude higher affinity than the corresponding amide analogues at the 5-HT<sub>4</sub> receptor.<sup>21</sup> Alkylation at N-1 of the indole-3-carboxamide ring (Table 1, **11b-f**) resulted in compounds which were slightly more potent than tropisetron in inhibiting serotonin-induced relaxation of the rat esophagus. The potency of these 1-alkylindole derivatives increased with the size of the N-1 substituent, the 1-isopropylindole derivative **11e** being the most potent of the 5-HT<sub>4</sub> receptor antagonists. Extending the alkyl group further as in the *sec*-butyl analogue **11f** led to a significant decrease in 5-HT<sub>4</sub> receptor affinity.

Replacement of the indole C-2 with a nitrogen atom (Table 1, **12a-f**) provided a series of indazole analogues that possessed higher 5-HT<sub>4</sub> receptor antagonist affinity than the corresponding indole counterparts. While the unsubstituted indazole **12a** was only weakly active as a 5-HT<sub>4</sub> receptor antagonist, methylation at N-1 led to a significant increase in antagonist potency. Similar to the indole derivatives, incorporation of larger N-1 alkyl substituents increased 5-HT<sub>4</sub> receptor antagonist affinity.

Since the *N*-ethyl (**13c**, BIMU-1) and *N*-isopropyl (**13e**, BIMU-8) derivatives were reported to possess affinity for the 5-HT<sub>4</sub> receptor,<sup>22</sup> we included the *N*-acylbenzimidazolone derivatives **13a-f** in our study. Interestingly, *N*-alkylbenzimidazolones possessed 5-HT<sub>4</sub> receptor antagonist activity in the rat esophagus similar to that of the indazole series (Table 1, **13a-f**). The

Table 1



compd no.	Ar	R	molecular formula	solvent of crystallization	mp (°C)	5-HT <sub>4</sub> receptor antagonist activity (% relaxation to 10 <sup>-6</sup> M 5-HT ± SEM (no. of determinations))				
						antagonist concentration				
						10 <sup>-8</sup> M	3 × 10 <sup>-8</sup> M	10 <sup>-7</sup> M	10 <sup>-6</sup> M	10 <sup>-5</sup> M
<b>1</b>								33.9 ± 3.9(3)	26.4 ± 5.4(13)	
<b>2</b>								7.0 ± 2.0 (3)	20.3 ± 3.3(3)	
<b>11a</b>	A	H	C <sub>17</sub> H <sub>21</sub> N <sub>3</sub> O·C <sub>4</sub> H <sub>4</sub> O <sub>4</sub>	EtOAc–MeOH	207				60.0 ± 4.0(4)	
<b>11b</b>	A	Me	C <sub>18</sub> H <sub>23</sub> N <sub>3</sub> O·C <sub>4</sub> H <sub>4</sub> O <sub>4</sub>	EtOAc–MeOH	205–208				47.3 ± 11.0(4)	
<b>11c</b>	A	Et	C <sub>19</sub> H <sub>25</sub> N <sub>3</sub> O·C <sub>2</sub> H <sub>2</sub> O <sub>4</sub>	EtOAc–MeOH	200–203				44.8 ± 10.1(4)	
<b>11d</b>	A	<i>n</i> -Pr	C <sub>20</sub> H <sub>27</sub> N <sub>3</sub> O·C <sub>2</sub> H <sub>2</sub> O <sub>4</sub>	EtOAc–MeOH	182–183				34.3 ± 9.1(3)	
<b>11e</b>	A	<i>i</i> -Pr	C <sub>20</sub> H <sub>27</sub> N <sub>3</sub> O·C <sub>2</sub> H <sub>2</sub> O <sub>4</sub>	EtOAc–MeOH	207–208				25.3 ± 4.1(4)	
<b>11f</b>	A	<i>s</i> -Bu	C <sub>21</sub> H <sub>29</sub> N <sub>3</sub> O·C <sub>2</sub> H <sub>2</sub> O <sub>4</sub>	EtOAc–MeOH	174–175				57.8 ± 7.8(4)	
<b>12a</b>	B	H	C <sub>17</sub> H <sub>20</sub> N <sub>4</sub> O	EtOAc	228–230				50.5 ± 15.9(4)	
<b>12b</b>	B	Me	C <sub>17</sub> H <sub>22</sub> N <sub>4</sub> O	Et <sub>2</sub> O	123–124				21.0 ± 7.1(3)	
<b>12c</b>	B	Et	C <sub>18</sub> H <sub>24</sub> N <sub>4</sub> O·C <sub>2</sub> H <sub>2</sub> O <sub>4</sub>	EtOAc–MeOH	232–233			6.7 ± 0.3(3)		
<b>12d</b>	B	<i>n</i> -Pr	C <sub>19</sub> H <sub>26</sub> N <sub>4</sub> O·C <sub>2</sub> H <sub>2</sub> O <sub>4</sub>	EtOAc–MeOH	214–215			2.5 ± 1.0(4)		
<b>12e</b>	B	<i>i</i> -Pr	C <sub>19</sub> H <sub>26</sub> N <sub>4</sub> O·C <sub>2</sub> H <sub>2</sub> O <sub>4</sub>	EtOAc–MeOH	185–187	43.2 ± 8.0(5)		9.0 ± 4.1(4)	8.0 ± 2.0(5)	
<b>12f</b>	B	<i>s</i> -Bu	C <sub>20</sub> H <sub>28</sub> N <sub>4</sub> O·C <sub>4</sub> H <sub>4</sub> O <sub>4</sub>	EtOAc–cyclohexane	130–132		46.3 ± 5.6(4)			
<b>13b</b>	C	Me	C <sub>17</sub> H <sub>22</sub> N <sub>4</sub> O <sub>2</sub> ·HCl	EtOAc–MeOH	>250			7.3 ± 4.3(4)		
<b>13c</b>	C	Et	C <sub>18</sub> H <sub>24</sub> N <sub>4</sub> O <sub>2</sub> ·HCl	acetone	227–230		37.5 ± 10.1(4)			
<b>13d</b>	C	<i>n</i> -Pr	C <sub>19</sub> H <sub>26</sub> N <sub>4</sub> O <sub>2</sub> ·HCl	MeOH	222–223		30.7 ± 15.1(3)		4.3 ± 2.0(4)	
<b>13e</b>	C	<i>i</i> -Pr	C <sub>19</sub> H <sub>26</sub> N <sub>4</sub> O <sub>2</sub> ·HCl	MeOH	138–140	55.0 ± 10.9(4)	13.8 ± 8.7(4)		2.7 ± 1.5(3)	
<b>13f</b>	C	<i>s</i> -Bu	C <sub>20</sub> H <sub>28</sub> N <sub>4</sub> O <sub>2</sub> ·C <sub>2</sub> H <sub>2</sub> O <sub>4</sub>	Et <sub>2</sub> O	205			8.8 ± 4.4(5)		
<b>14</b>	D		C <sub>19</sub> H <sub>22</sub> N <sub>2</sub> O·C <sub>7</sub> H <sub>8</sub> O <sub>3</sub> S	EtOH	238–239				76.5 ± 7.0(4)	
<b>15</b>	E		C <sub>16</sub> H <sub>22</sub> N <sub>3</sub> O <sub>2</sub> ·C <sub>4</sub> H <sub>4</sub> O <sub>4</sub>	EtOAc–MeOH	172			5.4 ± 2.5(5)	2.8 ± 1.5(4)	
<b>16</b>	F							37.0 ± 7.7(7)	22.2 ± 6.3(6)	

effect of N-alkylation on the benzimidazolones was also similar to the indazole series with the *N*-isopropyl derivative **13e** providing potent inhibitors of serotonin-induced relaxation in the rat esophagus.

The effect of *N*-isopropyl substitution might have been anticipated, since substitution of a heteroaromatic ring with an isopropyl group has led to increases in antagonist potency at other 5-HT receptor subtypes, as well. In a series of ergoline derivatives, *N*-1 isopropyl derivatives had higher affinity for the rat 5-HT<sub>2A</sub> receptor and were more potent 5-HT<sub>2A</sub> receptor antagonists than their corresponding *N*-1 unsubstituted derivatives.<sup>23</sup> Further, isopropyl substitution of tryptamine derivatives served to convert agonist activity to antagonist activity at the 5-HT<sub>2B</sub> receptor in the rat stomach fundus and at the rat 5-HT<sub>2A</sub> receptor.<sup>23, 24</sup>

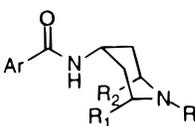
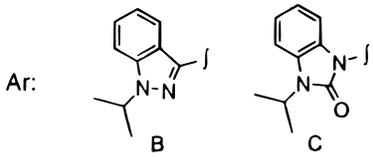
The 1-naphthalenecarboxamide **14** and the dihydrobenzofuran derivative zatosetron (**16**) are two bicyclic aromatic amides which displayed only weak 5-HT<sub>4</sub> receptor antagonism in the concentrations tested. The tropane analogue of cisapride, **15**, has been previously characterized as a potent 5-HT<sub>3</sub> receptor antagonist,<sup>18</sup> and we found that **15** was also a potent 5-HT<sub>4</sub> receptor antagonist.

The increased 5-HT<sub>4</sub> receptor affinity of **15** and the indazole and benzimidazolone derivatives over the acylindole and naphthalenecarboxylic acid derivatives

can be understood in terms of increased conformational rigidity of these systems. The amide N–H can participate in an intramolecular hydrogen bond with the methoxy group of **15**, the lone pair of N-2 in the indazole amides **12a–f**, and the carbonyl of the benzimidazolone amides **13a–f**. This hydrogen-bonding interaction enforces coplanarity of the aromatic ring with the atoms of the amide linkage. While the less active indole and naphthalene systems have similar steric requirements as the indazole and benzimidazolone systems, they cannot form a similar rigidifying intramolecular hydrogen bond. The importance of analogous intramolecular hydrogen bonding interactions has been demonstrated with a series of dopamine D2 receptor antagonists and serotonin 5-HT<sub>3</sub> receptor antagonists.<sup>18,25</sup> While zatosetron can theoretically form an intramolecular hydrogen bond, the *gem*-dimethyl substitution on the dihydrobenzofuran ring may provide sufficient steric bulk that the group can no longer participate in the required binding interactions with the 5-HT<sub>4</sub> receptor.

**Modification of the Tropane Group.** These studies identified 1-isopropylindazole and 1-isopropylbenzimidazolone as platforms producing compounds with relatively high affinity as 5-HT<sub>4</sub> receptor antagonists. To improve the affinity and selectivity of these derivatives, we incorporated structural elements of cisapride by replacing the tropane ring system with the piperidine

Table 2

compd no.	Ar	R <sub>1</sub>	R <sub>2</sub>	R <sup>a</sup>	molecular formula	solvent of crystallization	mp (°C)	5-HT <sub>4</sub> receptor antagonist activity (% relaxation to 10 <sup>-6</sup> M 5-HT ± SEM (no. of determinations))		
								antagonist concentration		
								3 × 10 <sup>-8</sup> M	10 <sup>-7</sup> M	10 <sup>-6</sup> M
<b>12g</b>	B	-CH <sub>2</sub> -CH <sub>2</sub> -	X		C <sub>27</sub> H <sub>33</sub> FN <sub>4</sub> O <sub>2</sub> ·HCl	2-PrOH	245–247			
<b>19b</b>	B	H	H	Me	C <sub>17</sub> H <sub>24</sub> N <sub>4</sub> O·C <sub>2</sub> H <sub>2</sub> O <sub>4</sub>	2-PrOH	185	37.3 ± 3.2(3)	22.2 ± 6.1(5)	17.7 ± 5.3(6)
<b>19l</b>	B	H	H	X	C <sub>25</sub> H <sub>31</sub> FN <sub>4</sub> O <sub>2</sub> ·HCl	EtOAc–MeOH	174–175	36.0 ± 6.2(6)	15.7 ± 5.5(6)	7.5 ± 3.5(8)
<b>13g</b>	C	-CH <sub>2</sub> -CH <sub>2</sub> -	X		C <sub>27</sub> H <sub>33</sub> FN <sub>4</sub> O <sub>3</sub> ·HCl	EtOAc–MeOH	235–236		29.5 ± 6.0(6)	2.1 ± 6.5(6)
<b>17a</b>	C	H	H	Me	C <sub>17</sub> H <sub>24</sub> N <sub>4</sub> O <sub>2</sub> ·HCl	EtOAc–MeOH	203–205		40.8 ± 11.4(4)	7.5 ± 2.7(4)
<b>17b</b>	C	H	H	X	C <sub>25</sub> H <sub>31</sub> FN <sub>4</sub> O <sub>3</sub> ·HCl	EtOAc	210–212		62.0 ± 13.9(6)	

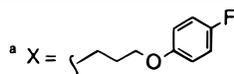
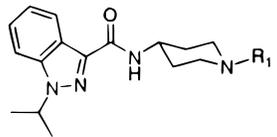


Table 3



compd no.	R <sub>1</sub> <sup>a</sup>	molecular formula	solvent of crystallization	mp (°C)	5-HT <sub>4</sub> receptor antagonist activity (% relaxation to 10 <sup>-6</sup> M 5-HT ± SEM (no. of determinations))			
					antagonist concentration			
					10 <sup>-8</sup> M	3 × 10 <sup>-8</sup> M	10 <sup>-7</sup> M	10 <sup>-6</sup> M
<b>19a</b>	H	C <sub>16</sub> H <sub>22</sub> N <sub>4</sub> O·HCl	EtOAc–MeOH	145				32.0 ± 11.3(4)
<b>19c</b>	CH <sub>2</sub> Ph	C <sub>23</sub> H <sub>28</sub> N <sub>4</sub> O·HCl	EtOAc–MeOH	209–212	38.5 ± 9.0(4)	23.3 ± 8.9(4)	24.5 ± 7.9(6)	3.85 ± 2.8(4)
<b>19d</b>	(CH <sub>2</sub> ) <sub>2</sub> Ph	C <sub>24</sub> H <sub>30</sub> N <sub>4</sub> O·C <sub>2</sub> H <sub>2</sub> O <sub>4</sub>	EtOH	188				1.3 ± 1.3(3)
<b>19e</b>	(CH <sub>2</sub> ) <sub>3</sub> Ph	C <sub>25</sub> H <sub>32</sub> N <sub>4</sub> O·C <sub>2</sub> H <sub>2</sub> O <sub>4</sub>	EtOH	184				3.7 ± 2.7(3)
<b>19f</b>	(CH <sub>2</sub> ) <sub>4</sub> Ph	C <sub>26</sub> H <sub>34</sub> N <sub>4</sub> O·C <sub>2</sub> H <sub>2</sub> O <sub>4</sub>	EtOH	167				4.0 ± 0.6(3)
<b>19g</b>	(CH <sub>2</sub> ) <sub>5</sub> Ph	C <sub>27</sub> H <sub>36</sub> N <sub>4</sub> O·HCl	EtOAc	133				43.7 ± 8.5(6)
<b>19h</b>	(CH <sub>2</sub> ) <sub>2</sub> Phth	C <sub>26</sub> H <sub>29</sub> N <sub>5</sub> O <sub>3</sub> ·C <sub>2</sub> H <sub>2</sub> O <sub>4</sub>	EtOAc–MeOH	148–150				7.5 ± 3.9(4)
<b>19i</b>	(CH <sub>2</sub> ) <sub>2</sub> NHMs	C <sub>19</sub> H <sub>29</sub> N <sub>5</sub> O <sub>3</sub> ·C <sub>4</sub> H <sub>4</sub> O	EtOH	180			16.0 ± 4.8(4)	6.8 ± 5.4(4)
<b>19j</b>	(CH <sub>2</sub> ) <sub>2</sub> NHCOPh	C <sub>25</sub> H <sub>31</sub> N <sub>5</sub> O <sub>2</sub> ·C <sub>2</sub> H <sub>2</sub> O <sub>4</sub>	EtOAc–MeOH	115			28.0 ± 8.7(4)	6.0 ± 2.7(5)
<b>19k</b>	(CH <sub>2</sub> ) <sub>2</sub> NHCOAd	C <sub>29</sub> H <sub>41</sub> N <sub>5</sub> O <sub>2</sub> ·C <sub>2</sub> H <sub>2</sub> O <sub>4</sub>	EtOAc–MeOH	124			38.5 ± 11.8(4)	1.0 ± 1.0(4)
<b>19b</b>	Me	C <sub>17</sub> H <sub>24</sub> N <sub>4</sub> O·C <sub>2</sub> H <sub>2</sub> O <sub>4</sub>	2-PrOH	185		37.3 ± 3.2(3)		
<b>19l</b>	X	C <sub>25</sub> H <sub>31</sub> FN <sub>4</sub> O <sub>2</sub> ·HCl	EtOAc–MeOH	174–175		36.0 ± 6.2(6)	15.7 ± 5.5(6)	7.5 ± 3.5(8)



ring system and by incorporating the *p*-fluorophenoxypropyl side chain into the compounds.

In the indazole series, incorporation of a *p*-fluorophenoxypropyl side chain (**12g**) led to a modest decrease in 5-HT<sub>4</sub> receptor antagonist activity. However, the piperidine analogue of this compound, **19l**, retained most of the activity of the original tropane derivative and was equipotent with the *N*-methylpiperidine derivative **19b** (Table 2). In contrast, modifying the benzimidazolone analogue **13e** by either replacing the tropane nucleus with the piperidine system (**17a**) or incorporating the *p*-fluorophenoxypropyl side chain (**13g**) decreased activity. When both of these modifications were made, the resulting compound, **17b**, was inactive as a 5-HT<sub>4</sub> receptor antagonist at 10<sup>-7</sup> M. Since the 1-isopropylindazole series was more tolerant to these modifications, we focused subsequent efforts on this series.

Table 3 outlines the effect of modification of the substituent on the piperidine nitrogen of **19b**. *N*-Methyl

substitution of the secondary amine increased 5-HT<sub>4</sub> receptor antagonist affinity by greater than 10-fold relative to the unsubstituted amine **19a**. Likewise, incorporation of a benzyl, phenethyl, phenylpropyl, or phenylbutyl group led to compounds with improved 5-HT<sub>4</sub> receptor antagonist activity (**19c–f**). Further lengthening of the side chain to the phenylpentyl group (**19g**) reduced 5-HT<sub>4</sub> receptor antagonism. Thus, relatively large groups were well-tolerated at the piperidine nitrogen. Since the mesylamidoethyl side chain of the known 5-HT<sub>4</sub> receptor antagonist GR113808 (**4**) is available to participate in a hydrogen bond, we wished to incorporate that and similar functionality into the series. As anticipated, the mesylamidoethyl derivative **19i** and its corresponding phthalimido (**19h**), benzamido (**19j**), and adamantylamide (**19k**) analogues were all potent 5-HT<sub>4</sub> receptor antagonists.

**Piperidine Replacement Strategy.** In general, we had found that modification of the rigid tropane skeleton

Table 4

		5HT4 Receptor Antagonist Activity						Ex vivo <sup>a</sup>		
Compound number	n	NR2	Molecular formula	solvent of crystallization	melting point (°C)	In vitro				(% relaxation to 10 <sup>-6</sup> M 5HT ± SEM (number of determinations))
						Antagonist concentration				
						10 <sup>-7</sup> M	3•10 <sup>-7</sup> M	10 <sup>-6</sup> M	10 <sup>-5</sup> M	
<b>21a</b>	2	NEt <sub>2</sub>	C <sub>17</sub> H <sub>26</sub> N <sub>4</sub> O•C <sub>2</sub> H <sub>4</sub> O <sub>4</sub>	EIOAc	128-130		26.8 ± 9.2 (4)			
<b>21b</b>	2		C <sub>17</sub> H <sub>24</sub> N <sub>4</sub> O•HCl	EIOAc-MeOH	210-212		4.3 ± 1.3 (3)			51 ± 4 (3)
<b>21c</b>	2		C <sub>18</sub> H <sub>26</sub> N <sub>4</sub> O•HCl	EIOAc-MeOH	>250		22.0 ± 9.9 (5)		6.6 ± 1.5 (5)	38 ± 4 (5)
<b>21d</b>	2		C <sub>19</sub> H <sub>28</sub> N <sub>4</sub> O•HCl	EIOAc-MeOH	216-218		4.8 ± 1.9 (4)			52 ± 2 (3)
<b>21e</b>	3		C <sub>19</sub> H <sub>28</sub> N <sub>4</sub> O•HCl	EIOAc-MeOH	156-158		14.3 ± 4.8 (6)		6.3 ± 3.7 (4)	15 ± 4 (3)
<b>21f</b>	4		C <sub>20</sub> H <sub>30</sub> N <sub>4</sub> O•C <sub>2</sub> H <sub>4</sub> O <sub>4</sub>	2-PrOH	143		43.3 ± 21.5 (2)		3.3 ± 0.6 (4)	49 ± 6 (4)
<b>21g</b>	2		C <sub>17</sub> H <sub>24</sub> N <sub>4</sub> O <sub>2</sub> •HCl	EIOAc-MeOH	201-203		42.3 ± 8.3 (3)		19.8 ± 7.7 (5)	
<b>21h</b>	2		C <sub>24</sub> H <sub>31</sub> N <sub>5</sub> O <sub>2</sub> •2HCl	2-PrOH	242				9.1 ± 2.3 (7)	
<b>21i</b>	2		C <sub>24</sub> H <sub>29</sub> N <sub>5</sub> O <sub>2</sub> •C <sub>2</sub> H <sub>4</sub> O <sub>4</sub>	EIOAc-MeOH	105				19.8 ± 4.2 (4)	
<b>21j</b>	2		C <sub>22</sub> H <sub>26</sub> N <sub>4</sub> O•C <sub>2</sub> H <sub>4</sub> O <sub>4</sub>	EIOH	191		20.3 ± 6.9 (4)			41 ± 3 (3)

<sup>a</sup> Compounds tested at 0.1 mg/kg po.



to the less rigid piperidine system increased 5-HT<sub>4</sub> receptor antagonist activity. To explore further increases in the flexibility of the side chain, replacement of the piperidine ring with the acyclic diethylaminoethyl group (**21a**) resulted in a compound with only moderate 5-HT<sub>4</sub> receptor antagonist activity (Table 4). However, replacement of the diethylamino group with a cyclic amine (**21b–d**) significantly increased activity, similar to previous observations with a series of metoclopramide ester analogues.<sup>26</sup> Increasing the length of the linking methylene chain to three and four carbons (**21e,f**) did not further increase 5-HT<sub>4</sub> receptor antagonist activity. Substitution with a morpholine (**21g**) or a substituted piperazine (**21h,i**) group slightly decreased 5-HT<sub>4</sub> receptor affinity, whereas the benzo-fused piperidine system **21j** was well-tolerated and equipotent with the parent piperidine system.

Pursuing our hypothesis that hydrogen bonding increased 5-HT<sub>4</sub> receptor antagonist activity, we proposed that incorporation of groups capable of hydrogen bond formation into **21c** would increase 5-HT<sub>4</sub> receptor affinity. We chose to attach these groups through C-4 of the piperidine of **21c** since the resulting derivatives would, because of the symmetry of the system, consist of a single stereoisomer. The 5-HT<sub>4</sub> receptor antagonist activity of this series is outlined in Table 5. Although the 4-aminopiperidine analogue **23a** had lower affinity than the parent piperidine derivative as a 5-HT<sub>4</sub> receptor antagonist, a series of aliphatic amide derivatives of the 4-aminopiperidine (**23b–g, 8**) were potent 5-HT<sub>4</sub> receptor antagonists *in vitro*. We propose that the amide groups interact with the 5-HT<sub>4</sub> receptor in an area of steric tolerance since substituents as large as *tert*-butyl and 1-adamantyl gave rise to potent 5-HT<sub>4</sub> receptor antagonists. The benzamide (**23i**) and phenacyl (**23j**) derivatives were also 5-HT<sub>4</sub> receptor antagonists as were the carbamate **21k**, the urea derivatives **23k–m**, and the methanesulfonamide **23n**. Because the 4-benzylpiperidine analogue **21l** was less potent than the other 4-substituted analogues, the C-4 substituent likely interacts with the 5-HT<sub>4</sub> receptor in a hydrogen-bonding interaction rather than strictly a hydrophobic interaction.

**5-HT<sub>3</sub>, 5-HT<sub>2A</sub>, and  $\alpha_1$  Adrenergic Receptor Selectivity.** Throughout the development of this SAR directed toward obtaining high affinity 5-HT<sub>4</sub> receptor antagonists, careful attention was paid to the specificity of the 5-HT<sub>4</sub> receptor interactions. We had chosen tropisetron and cisapride as the leads for these studies. Since tropisetron is a potent 5-HT<sub>3</sub> receptor antagonist<sup>27</sup> and cisapride has high affinity for 5-HT<sub>2A</sub> and  $\alpha_1$  receptors,<sup>6</sup> we initially focused our attention on interactions with these three receptors.

As the SAR progressed, selected compounds were tested as 5-HT<sub>3</sub> receptor antagonists in the guinea pig ileum. Serotonin induces a contraction in this tissue which is blocked by 5-HT<sub>3</sub> receptor antagonists.<sup>27</sup> The affinities of selected compounds are found in Table 6. The tropane derivatives tropisetron (**1**) and zatosetron (**16**) potently block 5-HT<sub>3</sub> receptors in the guinea pig ileum with pK<sub>b</sub>'s of 8.06 and 8.78, respectively.<sup>28</sup> Potent 5-HT<sub>3</sub> receptor antagonist activity was also seen for **12e** and **13e**, the tropane analogues in the indazole and benzimidazolone series. Since many high-affinity 5-HT<sub>3</sub>

**Table 6.** 5-HT<sub>3</sub> Receptor Affinity of Selected Compounds

compd no.	5-HT <sub>3</sub> receptor affinity <sup>a</sup> (pK <sub>b</sub> )
<b>1</b>	8.06 ± 0.05(16) <sup>b</sup>
<b>16</b>	8.78 ± 0.10(10) <sup>b</sup>
<b>12e</b>	7.4 ± 0.59(4)
<b>13e</b>	7.5 ± 0.29(3)
<b>19b</b>	6.3 ± 0.80(3)
<b>21c</b>	5.8 ± 0.3(4)

<sup>a</sup> Determined from compound's ability to block serotonin-induced contraction of the guinea pig ileum.<sup>28</sup> <sup>b</sup> Data from ref 28.

**Table 7.**  $\alpha_1$  Adrenergic Receptor and 5-HT<sub>2</sub> Receptor Affinity of Selected Compounds

compd no.	$\alpha_1$ receptor affinity		5-HT <sub>2</sub> receptor affinity <sup>c</sup>
	pK <sub>b</sub> (nM) <sup>a</sup>	pK <sub>i</sub> (nM) <sup>b</sup>	pK <sub>b</sub> (nM)
<b>2</b>	6.7		7.89
<b>19l</b>	6.7	7.69	6.8
<b>19c</b>	<5.0	<4.0	5.5
<b>19d</b>	7.68		
<b>19e</b>	5.8		
<b>19f</b>	6.5		
<b>19g</b>	5.6		
<b>19j</b>	5.8	6.62	
<b>19i</b>	<5.0	<4.0	
<b>19h</b>			<5.0
<b>21b</b>		<4.0	
<b>21c</b>	4.9	4.8	<5.0
<b>21d</b>		5.0	
<b>21e</b>		<4.5	
<b>21f</b>		5.2	
<b>23b</b>		<4.0	
<b>8</b>		4.3	
<b>21k</b>	<5.0	5.9	

<sup>a</sup> As determined by block of norepinephrine-induced contraction of the rat aorta.<sup>31</sup> <sup>b</sup> As determined by displacement of [<sup>3</sup>H]prazosin from rat whole brain.<sup>30</sup> <sup>c</sup> As determined by block of serotonin-induced contraction of the rat aorta.<sup>31</sup>

receptor antagonists have a basic nitrogen constrained within a polycyclic framework,<sup>29</sup> we proposed that conformationally labile derivatives would have lower affinity at the 5-HT<sub>3</sub> receptor. Indeed, the piperidine **19b** had 10-fold lower 5-HT<sub>3</sub> receptor affinity than its tropane analogue, **12e**. In **21c**, the flexibility of the system is further increased by replacing the piperidine ring with the acyclic ethylene linker, and this compound showed a further decrease in 5-HT<sub>3</sub> receptor affinity with a pK<sub>b</sub> of 5.8. By replacement of the tropane system of **12e** with the more flexible piperidine or acyclic systems, we significantly minimized the interactions of these compounds with the 5-HT<sub>3</sub> receptor.

We then explored the interactions of compounds containing the piperidine and acyclic linkers with  $\alpha_1$  adrenergic receptors. The affinity for the  $\alpha_1$  adrenergic receptor was determined by displacement of [<sup>3</sup>H]prazosin from rat whole brain tissue<sup>30</sup> or by blockade of norepinephrine-induced contractions of rat aorta.<sup>31</sup> The results are reported in Table 7. Cisapride (**2**) blocked  $\alpha_1$  receptors in the rat aorta with a pK<sub>b</sub> of 6.7. Compound **12g**, the piperidine derivative containing the *p*-fluorophenoxypropyl side chain, was a similarly potent  $\alpha_1$  receptor blocker. Compound **19d**, in which the *p*-fluorophenoxypropyl group is replaced by a phenethyl group, possessed 10-fold higher affinity than cisapride as an  $\alpha_1$  receptor blocker. Increasing the length of the phenylalkylene side chain decreased  $\alpha_1$  receptor affinity (**19e–g**), but none of these compounds had pK<sub>b</sub>'s lower than 5.6. On the other hand, the *N*-benzylpiperidine

analogue **19c** had virtually no affinity for the  $\alpha_1$  receptor. While the (mesylamidoethyl)piperidine derivative **19i** was not an  $\alpha_1$  receptor antagonist, the analogous benzamidoethyl derivative **19j** did block  $\alpha_1$  receptors. In the acyclic-linked series, none of the compounds studied (**8**, **21b–f,k**, **23b**) had significant affinity for the  $\alpha_1$  adrenergic receptor.

The 5-HT<sub>2A</sub> receptor antagonist activity of selected compounds was determined by their ability to block the contractile effect of serotonin in the rat aorta (Table 7).<sup>31</sup> Cisapride (**2**) was an antagonist in this assay with a  $pK_b$  of 7.89. The *N*-(*p*-fluorophenoxypropyl)piperidine derivative **19l** was also a 5-HT<sub>2A</sub> receptor antagonist but with 10-fold lower affinity than cisapride. Replacing the *p*-fluorophenoxypropyl side chain with a benzyl (**19c**) or phthalimidoethyl (**19h**) group led to further decreases in 5-HT<sub>2A</sub> receptor affinity. In the acyclic series, **21c** was inactive as a 5-HT<sub>2A</sub> receptor blocker in the concentrations tested.

The trend from these selectivity studies is clear. In progressing from the highly constrained tropane skeleton through the less rigid piperidine ring system and onto the flexible acyclic systems, we were able to progressively increase the selectivity of these compounds for the 5-HT<sub>4</sub> receptor versus 5-HT<sub>3</sub>,  $\alpha_1$  adrenergic, and 5-HT<sub>2A</sub> receptors. Next, we focused our attention on the oral in vivo activity of these acyclic derivatives and, to a lesser extent, on selected substituted piperidines that possessed the highest 5-HT<sub>4</sub> receptor affinity and selectivity.

**In Vivo 5-HT<sub>4</sub> Receptor Antagonist Activity following Oral Administration.** During these SAR studies, our interest centered upon obtaining a potent and orally active 5-HT<sub>4</sub> antagonist with a long duration of in vivo activity for clinical studies. However, convenient and well-documented functional consequences of 5-HT<sub>4</sub> receptor activation were not readily available in vivo. Although 5-HT<sub>4</sub> receptor antagonist activities can be estimated by several complex gastrointestinal assays in dogs<sup>12,32</sup> and by tachycardia in pigs,<sup>33</sup> we utilized ex vivo rat esophageal relaxation to serotonin as a simple, efficient comparative estimate of the in vivo 5-HT<sub>4</sub> receptor antagonist activity of these compounds. We selected 3 h after oral administration of antagonist to rats to remove the esophagus and challenge with serotonin ex vivo. The time (3 h) was selected to minimize any mechanical effects of the gavage on the esophagus and to measure activity at a time that would support a relatively long duration of action necessary for clinical utility.<sup>34</sup>

Of the cyclic amines studied (Table 4) several were evaluated for ex vivo activity. Of these, **21e** showed excellent activity to inhibit ex vivo serotonin-induced relaxation of the rat esophagus after oral administration of 0.1 mg/kg. However, **21e** did possess some modest affinity at both  $\alpha_2$  and muscarinic receptors which precluded our further interest (data not shown).

A number of the 4-substituted piperidinoethyl derivatives, in particular **23a,g**, **8**, **21k,l**, and **23m,n**, were effective in blocking 5-HT<sub>4</sub> receptors ex vivo following oral administration. The selectivity profile and duration of action of compound **8** dictated further investigation and development of this analogue.<sup>35</sup> Thus, compound **8** has been identified as a potent and selective 5-HT<sub>4</sub>

receptor antagonist with pharmacodynamic properties suitable for clinical development.

## Experimental Section

**General Experimental.** All solvents and reagents were used as commercially available. Moisture- and/or air-sensitive reactions were run under a positive pressure of nitrogen. The solvent system used for flash chromatographic purification is reported in parentheses. "(NH<sub>4</sub>OH)" indicates that ca. 0.1% ammonium hydroxide solution was added to the eluting solvent. Analytical thin-layer chromatography (TLC) was performed on 0.25-mm Merck silica gel 60 F254 plates. *R<sub>f</sub>*'s are reported using the same solvent system that was used for the flash column. Melting points were determined on a Hoover-Thomas Uni-Melt capillary melting point apparatus and are uncorrected. NMRs were recorded at 300 MHz. Chemical shifts are reported in parts per million downfield ( $\delta$ ) from tetramethylsilane in the form: chemical shift (multiplicity, coupling constant, number of protons). *J* values are reported in hertz (Hz). Mass spectra were recorded using field desorption (FD) ionization on a Varian MAT 731 mass spectrometer. Mass spectral data are reported in the form: *m/e* of parent ion (relative intensity). Elemental analyses were performed by the Physical Chemistry Research department of the Lilly Research Laboratories.

**N-(1-Benzylpiperidin-4-yl)-1*H*-indazole-3-carboxamide (**18**).** To a solution of 1*H*-indazole-3-carboxylic acid (8.11 g, 50 mmol) in DMF (140 mL) under a nitrogen atmosphere was added in one portion 1,1'-carbonyldiimidazole (8.92 g, 55 mmol). The resulting solution was warmed at 60 °C for 2 h and then cooled to room temperature before adding a solution of 4-amino-1-benzylpiperidine (9.51 g, 50 mmol) in DMF (20 mL). The resulting solution was heated at 60 °C for 2 h. The DMF was evaporated under reduced pressure and the residue dissolved in CH<sub>2</sub>Cl<sub>2</sub> (ca. 250 mL). This solution was washed sequentially with water, 1 N NaOH solution, water, and brine, dried (Na<sub>2</sub>SO<sub>4</sub>), filtered, and evaporated under reduced pressure. The residue was recrystallized twice from EtOH to give the product as light-yellow crystals (10.63 g, 76%); mp 198–200 °C. <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>): 1.60–1.84 (m, 4H), 1.97–2.11 (m, 2H), 2.77–2.90 (m, 2H), 3.48 (s, 2H), 3.77–3.94 (m, 1H), 7.19–7.45 (m, 7H), 7.6 (d, 1H, *J* = 9.0 Hz), 8.10–8.20 (m, 2H), 13.55 (s, 1H). MS: 334 (100). Anal. (C<sub>20</sub>H<sub>22</sub>N<sub>4</sub>O) C, H, N.

**N-(1-Benzylpiperidin-4-yl)-1-isopropylindazole-3-carboxamide Hydrochloride (**19c**).** To a solution of *N*-(1-benzylpiperidin-4-yl)-1*H*-indazole-3-carboxamide (9.94 g, 29.7 mmol) in DMF (200 mL) under a nitrogen atmosphere was added sodium hydride (60% dispersion in mineral oil, 1.2 g, 30 mmol), and the resulting mixture was allowed to stir at room temperature for 3 h. The reaction mixture was cooled to about 20 °C, and 2-iodopropane (3.3 mL, 33 mmol) was added. The resulting solution was allowed to stir at room temperature for about 18 h. The DMF was evaporated under reduced pressure, and the residue was dissolved in EtOAc (300 mL). This solution was washed sequentially with 10% aqueous sodium carbonate solution, water, and brine, dried (Na<sub>2</sub>SO<sub>4</sub>), filtered, and evaporated under reduced pressure to give an oil which crystallized. The crude product was purified via preparative HPLC (silica gel, CH<sub>2</sub>Cl<sub>2</sub>, 0–2% MeOH, 0.5% NH<sub>4</sub>OH) to give 8.77 g (78%) of product. A sample of this product (753 mg, 2 mmol) was dissolved in EtOH (30 mL) and treated with ethanolic HCl solution. The mixture was evaporated under reduced pressure, and the residue was stirred in about 40 mL of Et<sub>2</sub>O for 2 h. The resulting precipitate was collected by filtration and recrystallized from MeOH/EtOAc to give the hydrochloride salt (478 mg, 58%), mp 209–212 °C. <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>): 1.56 (d, 6H, *J* = 8.0 Hz), 1.94–2.12 (m, 4H), 3.00–3.45 (m, 4H), 3.97–4.17 (m, 1H), 4.30 (d, 2H, *J* = 6.0 Hz), 5.08 (septet, 1H, *J* = 7.0 Hz), 7.27 (t, 1H, *J* = 8.0 Hz), 7.38–7.70 (m, 6H), 7.80 (d, 1H, *J* = 9.0 Hz), 8.15 (d, 1H, *J* = 9.0 Hz), 8.34 (d, 1H, *J* = 8.0 Hz), 10.36 (br s, 1H). MS: 376 (100). Anal. (C<sub>23</sub>H<sub>29</sub>N<sub>4</sub>ClO) C, H, N.

**N-(Piperidin-4-yl)-1-isopropylindazole-3-carboxamide Hydrochloride (**19a**).** 1-Chloroethyl chloroformate (2.86

g, 20 mmol, 2 equiv) was added dropwise to a solution of *N*-(1-benzylpiperidin-4-yl)-1-isopropylindazole-3-carboxamide hydrochloride (3.76 g, 10 mmol) in  $\text{CH}_2\text{Cl}_2$  (50 mL) at 0 °C. The reaction mixture was allowed to warm to room temperature and then heated to reflux for 1 h. The reaction mixture was concentrated under vacuum and the residue dissolved in MeOH (50 mL) and heated to reflux for 1 h. The reaction mixture was concentrated to give a white solid which was recrystallized from MeOH/EtOAc to give the product (2.92 g, 91%), mp 145 °C.  $^1\text{H NMR}$  (DMSO- $d_6$ ): 1.57 (d, 6H,  $J = 6.0$  Hz), 1.80–2.05 (m, 4H), 2.93–3.01 (m, 2H), 4.05–4.20 (m, 1H), 5.08 (septet, 1H,  $J = 8.0$  Hz), 7.25 (t, 1H,  $J = 8.0$  Hz), 7.44 (t, 1H,  $J = 8.0$  Hz), 7.80 (d, 1H,  $J = 9.0$  Hz), 8.30 (d, 1H,  $J = 9.0$  Hz), 8.85 (br s, 1H). MS: 286 (100). Anal. ( $\text{C}_{16}\text{H}_{23}\text{N}_4\text{ClO}$ ) C,H,N.

***N*-(1-(2-(Phthalimid-1-yl)ethyl)piperidin-4-yl)-1-isopropylbenzimidazole-3-carboxamide Oxalate (19h).** To a solution of *N*-(piperidin-4-yl)-1-isopropylindazole-3-carboxamide hydrochloride (2.24 g, 6.9 mmol) in DMF (35 mL) under a nitrogen atmosphere was added sodium carbonate (2.94 g, 28 mmol). *N*-(2-Bromoethyl)phthalimide (1.76 g, 6.9 mmol) in DMF (10 mL) was then added, and the resulting solution was heated at 100 °C for about 18 h. The DMF was evaporated under reduced pressure to provide a residue which was taken up in EtOAc (ca. 250 mL). The solution was washed with water and brine, dried ( $\text{Na}_2\text{SO}_4$ ), filtered, and evaporated under reduced pressure to give 3.08 g of crude product. The free base was purified via flash chromatography (silica gel, EtOAc/EtOH, 100:5) to give a viscous oil (2.14 g, 68%). A sample of the above oil was converted to the oxalate salt in warm EtOAc and recrystallized from MeOH/EtOAc to give the title product as colorless crystals, mp 148–150 °C.  $^1\text{H NMR}$  (DMSO- $d_6$ ): 1.55 (d, 6H,  $J = 7.0$  Hz), 1.68–2.06 (m, 4H), 2.63–2.88 (m, 2H), 2.97–3.20 (m, 2H), 3.25–3.55 (m, 2H), 3.78–4.15 (m, 4H), 5.08 (septet, 1H,  $J = 7.0$  Hz), 7.25 (t, 1H,  $J = 8.0$  Hz), 7.42 (t, 1H,  $J = 8.0$  Hz), 7.68–8.02 (m, 5H), 8.03–8.31 (m, 2H). MS: 459 (100). Anal. ( $\text{C}_{28}\text{H}_{31}\text{N}_5\text{O}_7$ ) C,H,N.

***N*-(1-(2-Aminoethyl)piperidin-4-yl)-1-isopropylindazole-3-carboxamide (19m).** To a solution of *N*-(1-(2-phthalimid-1-yl)ethyl)piperidin-4-yl)-1-isopropylindazole-3-carboxamide (1.86 g, 4.0 mmol) in EtOH (80 mL) at 60 °C was added hydrazine hydrate (2.0 mL), and the resulting solution was stirred at 60 °C for about 4 h. The reaction mixture was cooled to 0 °C and filtered. The filtrate was evaporated under reduced pressure, and to the residue was added 1 N NaOH solution. This mixture was extracted four times with  $\text{Et}_2\text{O}$ . The extracts were washed with brine, dried ( $\text{Na}_2\text{SO}_4$ ), and evaporated under reduced pressure to give a colorless solid (1.21 g, 92%) which was used in the next step without further purification.

***N*-(1-(2-((*N*-Methylsulfonyl)amino)ethyl)piperidin-4-yl)-1-isopropylindazole-3-carboxamide Maleate (19i).** To a solution of *N*-(1-(2-aminoethyl)piperidin-4-yl)-1-isopropylindazole-3-carboxamide (264 mg, 0.8 mmol) in THF (5 mL) under a nitrogen atmosphere at about 15 °C were added triethylamine (0.12 mL, 0.84 mmol) and methanesulfonyl chloride (0.06 mL, 0.8 mmol). The resulting mixture was allowed to stir for about 18 h at room temperature. The mixture was filtered and the filtrate evaporated under reduced pressure to give a cloudy oil. The maleate salt was made in warm EtOAc and was recrystallized from about 10 mL of EtOH to give the title product as colorless crystals (250 mg, 60%) mp 180 °C.  $^1\text{H NMR}$  (DMSO- $d_6$ ): 1.56 (d, 6H,  $J = 6.0$  Hz), 1.80–2.15 (m, 4H), 2.94–3.68 (m, 13H), 3.98–4.20 (m, 1H), 5.10 (septet, 1H,  $J = 7.0$  Hz), 6.05 (s, 2H), 7.22–7.38 (m, 2H), 7.45 (t, 1H,  $J = 7.0$  Hz), 7.80 (d, 1H,  $J = 9.0$  Hz), 8.17 (d, 1H,  $J = 9.0$  Hz), 8.22–8.40 (m, 1H). MS: 407 (100). Anal. ( $\text{C}_{23}\text{H}_{33}\text{N}_5\text{O}_7\text{S}$ ) C,H,N.

**1-Benzyl-4-((Carbobenzoyloxy)amido)piperidine.** A solution of benzyl chloroformate (2 equiv, 144 mmol, 24.4 g) in  $\text{CH}_2\text{Cl}_2$  (100 mL) was added to a mixture of 4-amino-1-benzylpiperidine (72 mmol, 13.7 g) in  $\text{CH}_2\text{Cl}_2$  (200 mL) and 1 N NaOH solution (100 mL) at room temperature. The reaction mixture was stirred vigorously for 20 min, poured into water, and extracted with  $\text{CH}_2\text{Cl}_2$ . The extract was dried ( $\text{Na}_2\text{SO}_4$ )

and concentrated to yield 36.3 g of crude product as an orange oil. Flash chromatography (2:1 ether/hexanes ( $\text{NH}_4\text{OH}$ )) provided the pure product ( $R_f$ : 0.21, 15.3 g, 65%) as a colorless oil which solidified upon standing. Recrystallization of a sample of this material ( $\text{Et}_2\text{O}$ /hexanes) gave the product as fine white needles, mp 78–80 °C.  $^1\text{H NMR}$  ( $\text{CDCl}_3$ ): 1.46 (br q, 2H,  $J = 11$ ), 1.92 (br d, 2H,  $J = 11$ ), 2.11 (br t, 2H,  $J = 11$ ), 2.79 (br d, 2H,  $J = 11$ ), 3.49 (s, 2H), 3.42–3.60 (m, 1H), 4.58–4.70 (m, 1H), 5.07 (s, 2H), 7.20–7.45 (m, 10H). MS: 324 (100). Anal. ( $\text{C}_{20}\text{H}_{24}\text{N}_2\text{O}_2$ ) C,H,N.

**4-((Carbobenzoyloxy)amido)piperidine Hydrochloride.** A solution of 1-chloroethyl chloroformate (2 equiv, 84.4 mmol, 12.5 g) in  $\text{CH}_2\text{Cl}_2$  (25 mL) was added to a solution of 1-benzyl-4-((carbobenzoyloxy)amido)piperidine (42.2 mmol, 13.7 g) in  $\text{CH}_2\text{Cl}_2$  (200 mL) at room temperature. After stirring at room temperature for 1 h, the reaction mixture was concentrated, dissolved in MeOH (200 mL), and heated to reflux for 1 h. The reaction mixture was concentrated to give a white solid. Recrystallization (MeOH/EtOAc) gave the title compound as a white solid (7.56 g, 67%), mp 210.0–210.5 °C.  $^1\text{H NMR}$  (DMSO- $d_6$ ): 1.61 (br q, 2H,  $J = 11$ ), 1.84 (dd, 2H,  $J = 3, 14$ ), 2.76–2.96 (m, 2H), 3.14 (br d, 2H,  $J = 13$ ), 3.47–3.60 (m, 1H), 5.47 (s, 2H), 7.22–7.34 (m, 5H), 7.51 (d, 1H,  $J = 7$ ), 8.97 (br s, 1H), 9.19 (br s, 1H). MS: 234 (100), 235 (40). Anal. ( $\text{C}_{13}\text{H}_{18}\text{N}_2\text{O}_2 \cdot \text{HCl}$ ) C,H,N.

**1-(2-*N*-Phthalimidoethyl)-4-((benzyloxycarbonyl)amino)piperidine.** Sodium carbonate (2.83 g, 26.6 mmol) was added to a solution of 4-((carbobenzoyloxy)amido)piperidine hydrochloride (2.07 g, 7.6 mmol) and *N*-(2-bromoethyl)phthalimide (1.94 g, 7.6 mmol) in DMF (40 mL) and the reaction mixture stirred at 100 °C for 18 h. The reaction mixture was concentrated, diluted with water, and extracted with EtOAc. The extracts were washed with water and brine, dried ( $\text{Na}_2\text{SO}_4$ ), and concentrated to give a solid. Crystallization from EtOH provided the desired product as colorless crystals (1.58 g, 51%), mp 159–161 °C.  $^1\text{H NMR}$  (DMSO- $d_6$ ): 1.20–1.38 (m, 2H), 1.60–1.77 (m, 2H), 1.93–2.10 (m, 2H), 2.76–2.94 (m, 2H), 3.62–3.78 (m, 2H), 5.00 (s, 2H), 7.12–7.24 (m, 1H), 7.24–7.45 (m, 5H), 7.77–7.97 (m, 4H). MS: 407 (100). Anal. ( $\text{C}_{23}\text{H}_{25}\text{N}_3\text{O}_4$ ) C,H,N.

***N*-(2-(4-((Benzyloxycarbonyl)amino)-1-piperidinyl)ethyl)-1-isopropylindazole-3-carboxamide Oxalate (21k).** The following steps were performed without purification of the intermediates.

**1-(2-*N*-Aminoethyl)-4-((benzyloxycarbonyl)amino)piperidine (22).** A mixture of 1-(2-*N*-phthalimidoethyl)-4-((benzyloxycarbonyl)amino)piperidine (11.9 g, 29.2 mmol), hydrazine hydrate (15.8 mL, 16.4 g, 32.5 mmol), and EtOH (600 mL) was heated to reflux for 4 h. The reaction mixture was cooled to room temperature and filtered. The filtrate was concentrated. The resulting residue was taken up in 1 N NaOH solution. This mixture was saturated with NaCl and extracted with  $\text{Et}_2\text{O}$ . The ether extracts were dried ( $\text{Na}_2\text{SO}_4$ ) and concentrated to provide the desired product as an oil (8.19 g, 100%) which was used directly in the subsequent reaction. MS: 278 (100).

***N*-(2-(4-((Benzyloxycarbonyl)amino)-1-piperidinyl)ethyl)-1*H*-indazole-3-carboxamide (20k).** 1,1'-Carbonyldiimidazole (4.70 g, 29 mmol) was added to a solution of 1*H*-indazole-3-carboxylic acid (4.70 g, 29 mmol) in DMF (120 mL) and the reaction mixture stirred at room temperature for 4 h. A solution of 1-(2-*N*-aminoethyl)-4-((benzyloxycarbonyl)amino)piperidine (8.1 g, 29 mmol) in DMF (18 mL) was added dropwise and the reaction mixture stirred at room temperature for 18 h. The reaction mixture was concentrated to remove the DMF and the residue taken up in water and extracted with EtOAc. The extract was washed with water and brine, dried ( $\text{Na}_2\text{SO}_4$ ), and concentrated to the desired product (11.17 g, 91%) which was sufficiently pure for use in the next reaction, mp 184–187 °C. MS: 422 (100).

***N*-(2-(4-((Benzyloxycarbonyl)amino)-1-piperidinyl)ethyl)-1-isopropylindazole-3-carboxamide Oxalate (21k).** Sodium hydride (60% dispersion in mineral oil, 1.05 g, 26.3 mmol) was added to a solution of *N*-(2-(4-((benzyloxycarbonyl)-

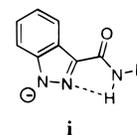
amino)-1-piperidinyl)ethyl)-1*H*-indazole-3-carboxamide (11.1 g, 26.3 mmol) in DMF (100 mL) and the reaction mixture stirred at room temperature for 4 h. The reaction was cooled to 15 °C, 2-iodopropane (2.9 mL, 5.13 g, 29 mmol) was added dropwise, and the reaction mixture was stirred at room temperature for 18 h. The DMF was removed in vacuo and the residue dissolved in EtOAc and water. The EtOAc layer was washed sequentially with 10% sodium carbonate solution, water and brine and evaporated to give the crude product. Flash chromatography (5% MeOH in CH<sub>2</sub>Cl<sub>2</sub>) provided the product as an oil (10.6 g, 87%). Crystallization of the product as the oxalate salt provided colorless crystals, mp 149–151 °C. <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>): 1.55 (d, 6H, *J* = 8.0 Hz), 1.55–1.75 (m, 1H), 1.85–2.02 (m, 2H), 2.60–4.40 (m, 12H), 5.03 (s, 2H), 5.10 (septet, 1H, *J* = 7.0 Hz), 7.23–7.55 (m, 7H), 7.82 (d, 1H, *J* = 9.0 Hz), 8.18 (d, 1H, *J* = 9.0 Hz), 8.40–8.50 (m, 1H). MS: 463 (100). Anal. (C<sub>28</sub>H<sub>35</sub>N<sub>5</sub>O<sub>7</sub>) C,H,N.

**N-[2-(4-Amino-1-piperidinyl)ethyl]-1-isopropylindazole-3-carboxamide dioxalate (23a).** A solution of *N*-(2-(4-(benzyloxycarbonyl)amino)-1-piperidinyl)ethyl)-1-isopropylindazole-3-carboxamide (8.71 g, 18.8 mmol) in EtOH (135 mL) was hydrogenated over 5% palladium on carbon (3.0 g) at 60 psi and room temperature for 18 h. The reaction mixture was filtered to remove the catalyst and concentrated to give an oil (5.0 g, 81%). Crystallization of this oil as the dioxalate salt gave colorless crystals, mp 232 °C. <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>): 1.55 (d, 6H, *J* = 7.0 Hz), 1.60–1.80 (m, 2H), 1.90–2.10 (m, 2H), 2.50–2.70 (m, 2H), 2.80–3.00 (m, 2H), 3.05–3.24 (m, 1H), 3.24–3.40 (m, 2H), 3.48–3.67 (m, 2H), 5.10 (septet, 1H, *J* = 7.0 Hz), 7.28 (t, 1H, *J* = 8.0 Hz), 7.82 (d, 1H, *J* = 9.0 Hz), 8.18 (d, 1H, *J* = 9.0 Hz), 8.30–8.42 (m, 1H). MS: 329 (100). Anal. (C<sub>22</sub>H<sub>31</sub>N<sub>5</sub>O<sub>9</sub>) C,H,N.

**N-(2-(4-(1-Adamantylcarboxamido)-1-piperidinyl)ethyl)-1-isopropylindazole-3-carboxamide Oxalate (8).** 1-Adamantanecarbonyl chloride (199 mg, 1 mmol) was added to a solution of *N*-(2-(4-amino-1-piperidinyl)ethyl)-1-isopropylindazole-3-carboxamide (330 mg, 1.0 mmol) and triethylamine (0.15 mL, 1.07 mmol) in THF (10 mL) at 0 °C. The reaction mixture was allowed to warm to room temperature, stirred for 18 h, and filtered. The filtrate was concentrated to give an oil. Crystallization of the oxalate salt (MeOH/EtOAc) gave the title compound as colorless crystals (267 mg, 46%), mp 228 °C. <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>): 1.55 (d, 6H, *J* = 8.0 Hz), 1.56–1.90 (m, 16H), 1.95 (br s, 3H), 2.80–3.05 (m, 2H), 3.05–3.20 (m, 2H), 3.30–3.55 (m, 2H), 3.55–3.70 (m, 2H), 3.70–3.90 (m, 1H), 5.10 (septet, 1H, *J* = 7.0 Hz), 7.23–7.32 (m, 1H), 7.27 (t, 1H, *J* = 8.0 Hz), 7.46 (t, 1H, *J* = 8.0 Hz), 7.80 (d, 1H, *J* = 9.0 Hz), 8.18 (t, 1H, *J* = 9.0 Hz), 8.40–8.55 (m, 1H). MS: 491 (100). Anal. (C<sub>31</sub>H<sub>43</sub>N<sub>5</sub>O<sub>6</sub>) C,H,N.

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- (35) See ref 6 for a detailed report on the pharmacology of **8**.

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