

Novel Pyrrolopyrimidine-Based α -Helix Mimetics: Cell-Permeable Inhibitors of Protein–Protein InteractionsJi Hoon Lee,[†] Qi Zhang,[†] Sunhwan Jo,[‡] Sergio C. Chai,[†] Misook Oh,[†] Wonpil Im,[‡] Hua Lu,^{*,†} and Hyun-Suk Lim^{*,†}[†]Department of Biochemistry and Molecular Biology, and Indiana University Simon Cancer Center, Indiana University School of Medicine, Indianapolis, Indiana 46202, United States[‡]Department of Molecular Biosciences and Center for Bioinformatics, The University of Kansas, 2030 Becker Drive, Lawrence, Kansas 66047, United States

S Supporting Information

ABSTRACT: There is considerable interest in developing non-peptidic, small-molecule α -helix mimetics to disrupt α -helix-mediated protein–protein interactions. Herein, we report the design of a novel pyrrolopyrimidine-based scaffold for such α -helix mimetics with increased conformational rigidity. We also developed a facile solid-phase synthetic route that is amenable to divergent synthesis of a large library. Using a fluorescence polarization-based assay, we identified cell-permeable, dual MDMX/MDM2 inhibitors, demonstrating that the designed molecules can act as α -helix mimetics.

α -Helices represent one of the most common protein secondary structures and are involved in various protein–protein interactions (PPIs). In many PPIs, short helical peptides play an important role as a recognition motif, where side chains at i , $i+3$ or $i+4$, and $i+7$ positions often become a critical determinant for PPIs (Figure 1A).¹ Since α -helix-mediated PPIs are involved in a wide array of cellular signaling pathways, inhibitors of these interactions could be promising therapeutic targets. While peptide- and peptidomimetic-based approaches have shown successful applications toward such targets,² non-peptidic, small molecules have advantages in terms of desirable bioavailability and cell permeability. Thus, there is considerable interest in developing non-peptidic, small-molecule α -helix mimetics that can disrupt such PPIs. Hamilton and co-workers demonstrated that rationally designed terphenyl **1** and similar scaffolds can serve as α -helix mimetics (Figure 1B).^{1e} Following their pioneering work, a number of terphenyl-inspired structures have been reported.^{1e,3} While some of these compounds have been shown to effectively disrupt certain PPIs, there are several important drawbacks associated with terphenyl-related structures, such as low aqueous solubility, relatively flexible scaffold structure, and long synthetic routes. Herein, we report design of a novel class of small-molecule α -helix mimetics, development of a facile solid-phase synthetic pathway, and a subsequent high-throughput screen, which led to the identification of potent inhibitors that disrupt the interaction between p53 and MDMX/MDM2.

On the basis of Hamilton's terephthalamide **2** (Figure 1B),⁴ we designed a pyrrolopyrimidine-based scaffold **3** (Figure 1C)

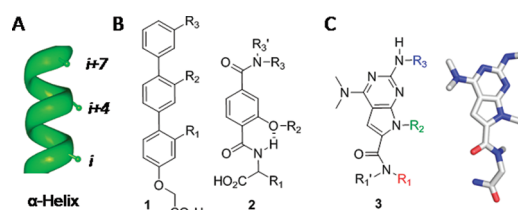


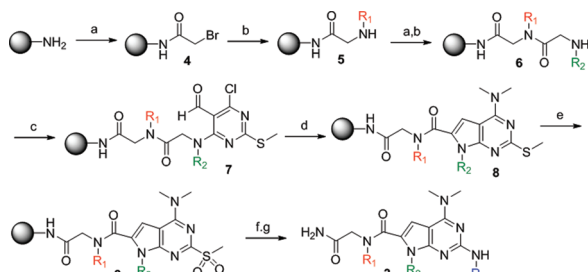
Figure 1. (A) α -Helix with i , $i+4$, and $i+7$ side-chain positions. (B) α -Helix mimetics with terphenyl **1** and terephthalamide **2** scaffolds. (C) Pyrrolopyrimidine-based scaffold **3** and its energy-minimized conformation (R_1 , R_2 , R_3 = CH_3).

with a hypothesis that the bicyclic ring in **3** could replace a pseudo-bicyclic structure formed by an intramolecular hydrogen bond in the terephthalamide **2**. Thus, the planar heterocyclic framework could provide a pre-organized structure with increased conformational rigidity. An energy-minimization study suggests that three functional groups on R_1 , R_2 , and R_3 of this scaffold can mimic the spatial orientation of the side chains of i , $i+3$ or $i+4$, and $i+7$ amino acids in an α -helix (Figure 1A,C). In addition, the pyrrolopyrimidine template is expected to possess favorable physical properties, including water solubility and cell permeability, similar to its structurally related purine and indole scaffolds, which are frequently used as privileged scaffolds in biology and medicine.⁵

For the preparation of the designed compound **3**, we developed a straightforward solid-phase synthetic strategy by modifying a solution-phase synthesis of pyrrolopyrimidines (Scheme 1).⁶ First, we utilized peptoid synthesis conditions to introduce two functionalities (R_1 and R_2). This efficient sub-monomer route⁷ includes bromoacetylation of amine group on Rink amide MBHA resin, followed by displacement of the bromide with primary amines. After repeating the same procedure, the resulting dimeric peptoids **6** were then coupled with 4,6-dichloro-2-(methylthio)pyrimidine-5-carbaldehyde to afford aldehydes **7** in nearly quantitative yield in five steps (Figure S1). Treatment of the aldehydes **7** with 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU) in DMF and MeOH provided pyrrolopyrimidines **8** through concomitant cyclization and dimethylamination.⁸ The thioether group was oxidized by *m*-chloroperbenzoic acid (mCPBA) to give a sulfone,

Received: September 12, 2010

Published: December 20, 2010

Scheme 1. Solid-Phase Synthesis^a

^a Reagents and conditions: (a) $\text{BrCH}_2\text{CO}_2\text{H}$, DIC, DMF, rt; (b) RNH_2 , DMF, rt; (c) 4,6-dichloro-2-(methylthio)pyrimidine-5-carbaldehyde, DIEA/DMF, rt; (d) DBU/DMF-MeOH, 90 °C; (e) mCPBA, NaHCO_3 , DCM, rt; (f) R_3NH_2 , DIEA/NMP, 170 °C; (g) TFA/DCM, rt.

which was subsequently substituted with various amines. The resin-bound trisubstituted bicyclic product **3** was cleaved with trifluoroacetic acid (TFA). Crude products were analyzed for purity and identity by LC-MS. To investigate the efficiency of our synthetic pathway, a series of compounds **3** were synthesized, employing a variety of different amines. Synthesis with most tested amines resulted in >80% purity of the final products **3**, as determined by LC-MS (Figure S1), which is sufficiently pure for biological testing without further purification. The most common byproduct (<5%) in the synthesis was the 4-methoxy-substituted product (instead of the *N,N*-dimethylamino group) obtained during the cyclization step. Those amines that resulted in low purity (<70%) of final products were excluded in library synthesis. Compared to the relatively lengthy, linear synthesis of most terphenyl-related structures, our synthetic route is simple and divergent and thus amenable to library synthesis. Notably, since many structurally diverse primary amines are commercially available, a large combinatorial library is readily accessible.

The efficacy of this scaffold as α -helix mimetics was assessed by monitoring its ability to disrupt the p53-MDMX interaction. MDM2 and its homologue MDMX bind to the tumor suppressor p53 and regulate its stability and activity. The interactions are mediated mainly by three key residues (Phe19, Trp23, and Leu26) of p53 and the hydrophobic pocket in MDMX and MDM2 (Figure 2A).^{2a,b,9} Despite the similarity in the p53 recognition surface of MDMX and MDM2, it proved to be challenging to identify MDMX inhibitors, whereas a number of MDM2 inhibitors have been reported.¹⁰ Indeed, inhibitors developed to target MDM2 show significantly less affinity to MDMX,^{9,11} and there is only one report of an MDMX-specific small-molecule inhibitor.¹² Given that MDMX is overexpressed in many cancers and functions as a major regulator of p53 activity (both independently of and synergistically with MDM2), the development of MDMX inhibitors (either MDMX-specific or dual MDMX/MDM2 inhibitors) is highly desirable but still remains challenging.^{11,13}

To search for such inhibitors, we constructed a 900-member library of compound **3** following Scheme 1. This was accomplished by a manual parallel synthesis (see Supporting Information for details). Primary amines containing hydrophobic groups (aromatic or aliphatic, shown in Figure S2) were selected with the intent of mimicking the side chains of the three amino acids of p53 (Phe, Trp, and Leu). For quality control, 90 compounds (10%) were randomly chosen from the library, and their purity and identity were verified by LC-MS. As shown in Table S1, the

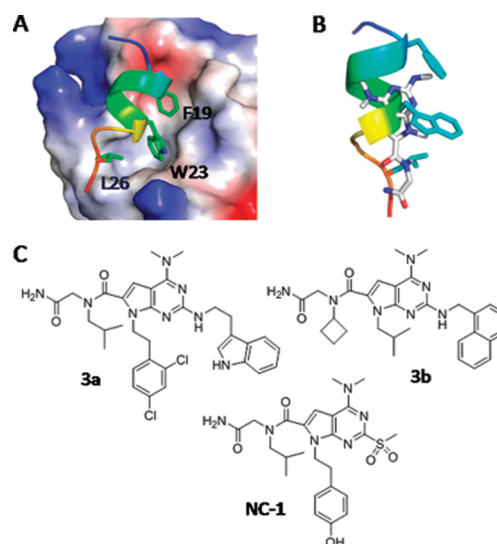


Figure 2. (A) Crystal structure of MDMX in complex with a p53 peptide (PDB entry: 2Z5S). (B) Overlay of compound **3** ($\text{R}_1, \text{R}_2, \text{R}_3 = \text{CH}_3$) and an MDMX-bound p53 peptide. (C) Structures of the active compounds (**3a** and **3b**) and an inactive compound (NC-1).

average purity of the crude final products was 83%, indicating the robustness of the synthetic route. The library was used for screening without further purification. A computational modeling showed that the scaffold could mimic MDMX-bound p53 peptide (Figure 2B and Figure S3). The library molecules were screened at $\sim 40 \mu\text{M}$ concentration for the ability to displace a Rhodamine-labeled 15-mer p53 peptide from MDMX protein by a fluorescence polarization (FP) assay (Figure S4; see Supporting Information for details). From the screen, we identified the two most active compounds, **3a** and **3b** (Figure 2C), which were resynthesized and purified by HPLC for further studies. The overall yields for the synthesis of **3a** and **3b** were 26% and 40%, respectively, based on the loading level of the resin (Figure S5). As expected, these compounds are highly soluble in aqueous solution.^{14,15}

We assessed the binding affinities of the selected compounds using the same FP-based competitive assays. A 15-mer peptide derived from p53 and a known MDM2 inhibitor, MI-63,¹⁶ were used as controls. As shown in Figure 3A, **3a** and **3b** effectively inhibited the p53-MDMX binding with $K_i = 0.62$ and $0.45 \mu\text{M}$, respectively, which were comparable with that of a 15-mer p53 peptide ($K_i = 0.8 \mu\text{M}$). We then tested whether they could bind to MDM2. Interestingly, they were found to inhibit the p53-MDM2 interaction as well, with $K_i = 0.62$ and $0.84 \mu\text{M}$, respectively, similar to the binding affinities for MDMX (Figure 3B,C). These results suggest that the compounds act as dual inhibitors of MDMX- and MDM2-p53 interactions.

Next we investigated the cellular activity of the inhibitors, since the compounds were expected to be cell-permeable, as indicated above. To test this, we examined the effect of the inhibitors on cellular levels of p53 and the cyclin-dependent kinase inhibitor p21, a major transcriptional target of p53. Human lung cancer H460 cells expressing wild-type p53 were incubated with **3a**, **3b**, DMSO, or NC-1 (as a negative control), and cell lysates were analyzed by Western blot to monitor p53 and p21 levels. Indeed, treatment of **3a** and **3b** led to the increase of p53 and p21 levels in a dose-dependent manner, whereas DMSO and NC-1 had no effects (Figure 3D and Figure S6). This result clearly

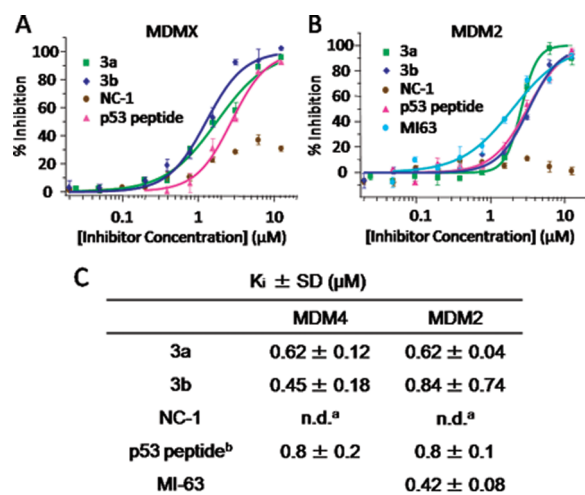


Figure 3. Inhibition curves for Rhodamine-labeled p53 peptide (SQETFSDLWKLLEN-NH-Rhodamine) binding to human MDMX (amino acids 1–137) (A) and human MDM2 (amino acids 1–138) (B) by fluorescence polarization. (C) K_i values of inhibitors. ^aNot determined. ^bUnlabeled 15-mer p53 peptide (SQETFSDLWKLLEN).

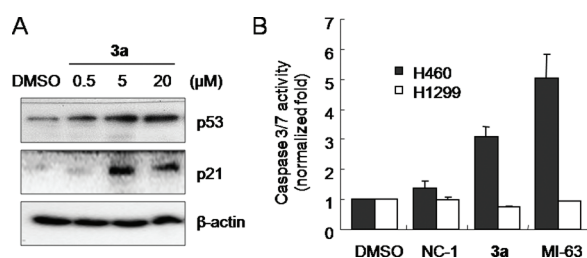


Figure 4. p53 activation by 3a. (A) Human lung cancer H460 cells expressing wild-type p53 were treated with DMSO or the indicated concentrations of 3a for 12 h. Lysates were subjected to Western blot analysis. (B) H460 cells with wild-type p53 and human lung cancer H1299 cells with deleted p53 were exposed to DMSO, NC-1 (20 μM), 3a (20 μM), or MI-63 (20 μM) for 24 h. Caspase activity was analyzed by Caspase-Glo 3/7 assay kit (Promega).

demonstrates that the compounds are not only cell-permeable but also able to induce p53 level and activity in cells.

Activation of p53 by the MDMX/MDM2 inhibitors would result in cell cycle arrest and apoptosis. We examined the ability of the inhibitors to induce apoptosis by monitoring the effect of 3a on caspase activity. H460 cells with wild-type p53 and human lung cancer H1299 cells with deleted p53 were exposed to DMSO, NC-1, 3a, or MI-63 (a positive control) for 24 h, and caspase activity was measured by Caspase-Glo 3/7 assay kit (see Supporting Information). As shown in Figure 4, 3a and MI-63 induced caspase 3/7 activity (about 3- and 5-fold, respectively) in H460 cells, while they had no effect on H1299 cells. This result indicates that 3a triggers apoptosis through a p53-dependent pathway by binding to MDMX/MDM2 and inhibiting their function toward p53.

While unnatural oligomer-based dual MDMX/MDM2 inhibitors such as β -peptides¹⁷ and *N*-acylpolymamines¹⁸ have been reported, our compounds would have advantages as cell-permeable, small “drug-like” molecules. These dual-specific inhibitors should have a unique, but overlapping, mechanism with current MDM2 inhibitors and thus could provide a promising starting point for the development of a novel class of anti-cancer agents. Because the inhibitors are

initial hit compounds, further studies are needed. These include biological evaluation of the inhibitors to elucidate the mode of mechanism as dual inhibitors, investigation of the selectivity of the compounds among α -helix-recognizing proteins, and X-ray and computational studies to better understand the molecular basis for the binding. Experiments along these lines are underway.

In summary, we have described the design of a pyrrolopyrimidine-based scaffold 3 as a novel class of α -helix mimetics and the development of a facile solid-phase synthetic route (Scheme 1). In addition, we have identified potent, dual inhibitors of MDMX- and MDM2-p53 interactions through high-throughput screening, highlighting that they can serve as α -helix mimetics (Figures 2–4). Of note, our compounds have several important advantages compared to most terphenyl-related structures, which include increased conformational rigidity, improved aqueous solubility, excellent cell permeability, and ease of synthesis. Importantly, our divergent solid-phase synthesis is amenable to the construction of large libraries and subsequent high-throughput screening. Taken together, we believe that our pyrrolopyrimidine-based scaffold, along with the efficient solid-phase synthesis, will serve as a useful tool to discover inhibitors of many α -helix-mediated PPIs.

■ ASSOCIATED CONTENT

S Supporting Information. Detailed experimental procedures, LC-MS spectra, and complete refs 6 and 11. This material is available free of charge via the Internet at <http://pubs.acs.org>.

■ AUTHOR INFORMATION

Corresponding Author

limhyun@iupui.edu; hualu@iupui.edu

■ ACKNOWLEDGMENT

We are grateful to Prof. Zhong-Yin Zhang (Indiana University) for critical reading of the manuscript, Prof. Qizhuang Ye (Indiana University) for providing access to a liquid handling system, and Dr. Jaeki Min (St. Jude Children's Research Hospital) for technical assistance. H.L. was supported by NIH-NCI grants CA127724, CA095441, and CA129828. This research was also supported in part by NSF OCI-0503992 through TeraGrid resources provided by Purdue University.

■ REFERENCES

- (1) (a) Davis, J. M.; Tsou, L. K.; Hamilton, A. D. *Chem. Soc. Rev.* **2007**, 36, 326. (b) Hershsberger, S. J.; Lee, S. G.; Chmielewski, J. *Curr. Top. Med. Chem.* **2007**, 7, 928. (c) Wilson, A. J. *Chem. Soc. Rev.* **2009**, 38, 3289. (d) Jochim, A. L.; Arora, P. S. *Mol. Biosyst.* **2009**, 5, 924. (e) Cummings, C. G.; Hamilton, A. D. *Curr. Opin. Chem. Biol.* **2010**, 14, 341.
- (2) (a) Pazgier, M.; Liu, M.; Zou, G.; Yuan, W.; Li, C.; Li, C.; Li, J.; Monbo, J.; Zella, D.; Tarasov, S. G.; Lu, W. *Proc. Natl. Acad. Sci. U.S.A.* **2009**, 106, 4665. (b) Phan, J.; Li, Z.; Kasprzak, A.; Li, B.; Sebt, S.; Guida, W.; Schönbrunn, E.; Chen, J. *J. Biol. Chem.* **2010**, 285, 2174. (c) Walensky, L. D.; Kung, A. L.; Escher, I.; Malia, T. J.; Barbuto, S.; Wright, R. D.; Wagner, G.; Verdine, G. L.; Korsmeyer, S. J. *Science* **2004**, 305, 1466. (d) Fasan, R.; Dias, R. L.; Moehle, K.; Zerbe, O.; Vrijbloed, J. W.; Obrecht, D.; Robinson, J. A. *Angew. Chem., Int. Ed.* **2004**, 43, 2109. (e) Wang, D.; Liao, W.; Arora, P. S. *Angew. Chem., Int. Ed.* **2005**, 44, 6525. (f) Hara, T.; Durell, S. R.; Myers, M. C.; Appella, D. H. *J. Am. Chem. Soc.* **2006**, 128, 1995. (g) Sadowsky, J. D.; Fairlie, W. D.; Hadley, E. B.; Lee, H. S.; Umezawa, N.; Nikolovska-Coleska, Z.; Wang, S.;

Huang, D. C.; Tomita, Y.; Gellman, S. H. *J. Am. Chem. Soc.* **2007**, *129*, 139. (h) Bautista, A. D.; Appelbaum, J. S.; Craig, C. J.; Michel, J.; Schepartz, A. *J. Am. Chem. Soc.* **2010**, *132*, 2904.

(3) (a) Lu, F.; Chi, S. W.; Kim, D. H.; Han, K. H.; Kuntz, I. D.; Guy, R. K. *J. Comb. Chem.* **2006**, *8*, 315. (b) Volonterio, A.; Moisan, L.; Rebek, J., Jr. *Org. Lett.* **2007**, *9*, 3733. (c) Maity, P.; Konig, B. *Org. Lett.* **2008**, *10*, 1473. (d) Plante, J. P.; Burnley, T.; Malkova, B.; Webb, M. E.; Warriner, S. L.; Edwards, T. A.; Wilson, A. J. *Chem. Commun.* **2009**, 5091. (e) Shaginian, A.; Whitby, L. R.; Hong, S.; Hwang, I.; Farooqi, B.; Searcey, M.; Chen, J.; Vogt, P. K.; Boger, D. *J. Am. Chem. Soc.* **2009**, *131*, 5564. (f) Maringanti, S.; Cheemala, M. N.; Ahn, J.-M. *Org. Lett.* **2009**, *11*, 4418. (g) Tosovska, P.; Arora, P. S. *Org. Lett.* **2010**, *12*, 1588. (h) Bourne, G. T.; Kuster, D. J.; Marshall, G. R. *Chemistry* **2010**, *16*, 8439.

(4) Yin, H.; Lee, G. I.; Sedey, K. A.; Rodriguez, J. M.; Wang, H. G.; Sebt, S. M.; Hamilton, A. D. *J. Am. Chem. Soc.* **2005**, *127*, 5463.

(5) Welsch, M. E.; Snyder, S. A.; Stockwell, B. R. *Curr. Opin. Chem. Biol.* **2010**, *14*, 347.

(6) Clark, M. P.; et al. *Bioorg. Med. Chem. Lett.* **2007**, *17*, 1250.

(7) Figliozzi, G. M.; Goldsmith, R.; Ng, S. C.; Banville, S. C.; Zuckermann, R. N. *Methods Enzymol.* **1996**, *267*, 437.

(8) Agarwal, A.; Chauhan, P. M. S. *Synth. Commun.* **2004**, *34*, 2925.

(9) Czarna, A.; Popowicz, G. M.; Pecak, A.; Wolf, S.; Dubin, G.; Holak, T. A. *Cell Cycle* **2009**, *8*, 1176.

(10) (a) Murray, J. K.; Gellman, S. H. *Biopolymers* **2007**, *88*, 657. (b) Dömling, A. *Curr. Opin. Chem. Biol.* **2008**, *12*, 281. (c) Shangary, S.; Wang, S. *Annu. Rev. Pharmacol. Toxicol.* **2009**, *49*, 223.

(11) Laurie, N. A.; et al. *Nature* **2006**, *444*, 61.

(12) Reed, D.; Shen, Y.; Shelat, A. A.; Arnold, L. A.; Ferreira, A. M.; Zhu, F.; Mills, N.; Smithson, D. C.; Regni, C. A.; Bashford, D.; Cicero, S. A.; Schulman, B. A.; Jochemsen, A. G.; Guy, R. K.; Dyer, M. A. *J. Biol. Chem.* **2010**, *285*, 10786.

(13) (a) Toledo, F.; Wahl, G. M. *Nat. Rev. Cancer* **2006**, *6*, 909. (b) Marine, J. C.; Dyer, M. A.; Jochemsen, A. G. *J. Cell Sci.* **2006**, *120*, 371.

(14) Solubilities of **3a** and **3b** in water and phosphate buffer saline (pH 7.4) were >150 $\mu\text{g/mL}$, which is above the desirable water solubility (100 $\mu\text{g/mL}$) for favorable oral absorption of drugs.¹⁵

(15) Muller, C. E. *Chem. Biodiv.* **2009**, *6*, 2071.

(16) Ding, K.; Lu, Y.; Nikolovska-Coleska, Z.; Wang, G.; Qiu, S.; Shangary, S.; Gao, W.; Qin, D.; Stuckey, J.; Krajewski, K.; Roller, P. P.; Wang, S. *J. Med. Chem.* **2006**, *49*, 3432.

(17) Michel, J.; Harker, E. A.; Tirado-Rives, J.; Jorgensen, W. L.; Schepartz, A. *J. Am. Chem. Soc.* **2009**, *131*, 6356.

(18) Hayashi, R.; Wang, D.; Hara, T.; Iera, J. A.; Durell, S. R.; Appella, D. H. *Bioorg. Med. Chem.* **2009**, *17*, 7884.