

# ANTIMICROBIAL AND TUBERCULOSTATIC ACTIVITY OF 5-ARYL(HETARYL)-1,3,4-OXADIAZOLE-2-THIONES AND THEIR DERIVATIVES

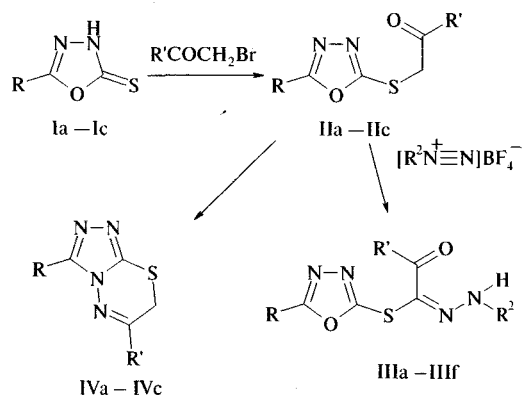
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As is known, some 5-R-1,3,4-oxadiazole-2-thione derivatives possess antimicrobial properties [1 – 3] and some 1,3,4-oxadiazole derivatives are used as antituberculous drugs [4 – 10].

In the search for new antimicrobial and tuberculostatic agents, we have employed the interaction of 5-aryl(hetaryl)-1,3,4-oxadiazole-2-thiones (Ia – Ic) [11] with substituted phenacyl bromides to obtain a series of new 5-aryl(hetaryl)-2-(phenacylthio)-1,3,4-oxadiazoles (IIa – IIc).



I: R = 3-BrC<sub>6</sub>H<sub>4</sub> (a), 4-pyridyl (b), C<sub>6</sub>H<sub>5</sub>CH<sub>2</sub> (c). II: R = 3-BrC<sub>6</sub>H<sub>4</sub> (a, b), 4-pyridyl (c); R' = C<sub>6</sub>H<sub>5</sub> (a), 4-O<sub>2</sub>NC<sub>6</sub>H<sub>4</sub> (b), 4-ClC<sub>6</sub>H<sub>4</sub> (c). III: R = 3-BrC<sub>6</sub>H<sub>4</sub> (a, b), 4-pyridyl (c, d, e), C<sub>6</sub>H<sub>5</sub>CH<sub>2</sub> (f); R' = C<sub>6</sub>H<sub>5</sub> (a), 4-O<sub>2</sub>NC<sub>6</sub>H<sub>4</sub> (b), 4-ClC<sub>6</sub>H<sub>4</sub> (c, d, e, f); R<sup>2</sup> = 4-O<sub>2</sub>NC<sub>6</sub>H<sub>4</sub> (a, d), 4-CH<sub>3</sub>OC<sub>6</sub>H<sub>4</sub> (b, c, e, f), 4-BrC<sub>6</sub>H<sub>4</sub> (c). IV: R = 3-BrC<sub>6</sub>H<sub>4</sub> (a, b), 4-pyridyl (c); R' = C<sub>6</sub>H<sub>5</sub> (a), 4-O<sub>2</sub>NC<sub>6</sub>H<sub>4</sub> (b), 4-ClC<sub>6</sub>H<sub>4</sub> (c).

We have established for the first time that compounds IIa – IIc exhibit the properties of CH-acids and readily react

with active methylene groups in aryldiazonium borofluorides to form the corresponding arylhydrazones (IIIa – IIIf). At the same time, reactions of compounds IIa – IIc with a hydrazine hydrate solution according to [11] led to 7H-s-triazolo-[3,4-b][1,3,4]thiadiazines (IVa – IVc) [12, 13].

Purity and identity of the synthesized compounds were checked and the proposed structures were confirmed by data of elemental analyses and the results of IR and <sup>1</sup>H NMR spectroscopic measurements (Table 1).

## EXPERIMENTAL CHEMICAL PART

The IR spectra were measured on an UR-20 spectrophotometer (Germany) using samples pelletized with KBr. The <sup>1</sup>H NMR spectra were recorded on a Bruker-300 (300 MHz) spectrometer using DMSO-d<sub>6</sub> as the solvent and TMS as the internal standard.

The initial 5-aryl(hetaryl)-1,3,4-oxadiazole-2-thiones (Ia – Ic) were synthesized as described in [14, 15].

**5-Aryl(hetaryl)-2-(phenacylthio)-1,3,4-oxadiazoles (IIa – IIc).** To a solution of 0.01 mole of compound I(a – c) in 30 ml of a 80% aqueous ethanol solution with 0.6 g KOH was added 0.01 mole of the corresponding  $\alpha$ -halogenoketone in 30 ml of ethanol and the mixture was kept at 18 – 20°C for 15 – 18 h, after which the precipitated product was filtered.

**5-Aryl(hetaryl)-2-( $\alpha$ -aryldiazenophenacylthio)-1,3,4-oxadiazoles (IIIa – IIIf).** To a solution of 0.01 mole of compound II(a – c) and 1 g of anhydrous sodium acetate in a mixture of 30 ml of CH<sub>3</sub>COOH and 5 ml of acetic anhydride was added a suspension of 0.01 mole of aryldiazonium borofluorides in a mixture of 20 ml of CH<sub>3</sub>COOH and 5 ml of acetic anhydride. The reaction mixture was kept at 18 – 20°C for 20 – 24 h in the dark, after which the precipitated product was filtered.

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TABLE 1. Yields and Physicochemical Characteristics of the Synthesized Compounds

Compound	Yield, %	M.p., °C	Empirical formula	IR spectrum: $\lambda_{\max}$ , cm <sup>-1</sup>	Proton chemical shift $\delta$ , ppm		
					S-CH <sub>2</sub> - (s)	=N-NH- (s)	H <sub>arom</sub> (m)
IIa	84	118.5 – 119	C <sub>16</sub> H <sub>11</sub> BrN <sub>2</sub> O <sub>2</sub> S	1670	5.22	–	7.50 – 8.08
IIb	87	147 – 148.5	C <sub>16</sub> H <sub>10</sub> BrN <sub>3</sub> O <sub>4</sub> S	1672	5.26	–	7.53 – 8.42
IIc	85	150 – 151	C <sub>15</sub> H <sub>10</sub> ClN <sub>3</sub> O <sub>2</sub> S	1678	5.20	–	7.65 – 8.81
IIIa	73	267 – 268	C <sub>22</sub> H <sub>14</sub> BrN <sub>3</sub> O <sub>4</sub> S	1650, 3126, 2985	–	11.55	7.49 – 8.41
IIIb*	67	210 – 210.5	C <sub>23</sub> H <sub>16</sub> BrN <sub>3</sub> O <sub>5</sub> S	1650, 3130, 2990	–	11.58	7.10 – 8.40
IIIc	74	245 – 245.5	C <sub>21</sub> H <sub>13</sub> BrClN <sub>3</sub> O <sub>2</sub> S	1650, 3210, 2980	–	11.65	7.68 – 8.77
IIId	76	262 – 262.5	C <sub>21</sub> H <sub>13</sub> ClN <sub>3</sub> O <sub>4</sub> S	1660, 3126, 2986	–	11.76	7.73 – 8.79
IIIe*	67	227 – 227.5	C <sub>22</sub> H <sub>16</sub> ClN <sub>3</sub> O <sub>3</sub> S	1660, 3168, 3000	–	11.56	7.11 – 8.75
IIIf*	65	204 – 205	C <sub>24</sub> H <sub>16</sub> ClN <sub>4</sub> O <sub>3</sub> S	1660, 3184, 3030	–	10.78	7.09 – 8.18
IVa	79	224 – 225	C <sub>16</sub> H <sub>11</sub> BrN <sub>4</sub> S		4.45	–	7.53 – 8.22
IVb	78	242 – 243	C <sub>16</sub> H <sub>10</sub> BrN <sub>3</sub> O <sub>2</sub> S		4.51	–	7.55 – 8.40
IVc	82	246 – 247	C <sub>15</sub> H <sub>10</sub> ClN <sub>3</sub> S		4.32	–	7.56 – 8.75

\* Signal of OCH<sub>3</sub> protons, 3.83 ppm (s).

\*\* Signal of OCH<sub>3</sub> protons, 3.81 ppm (s) and signal of CH<sub>2</sub> protons, 3.53 ppm (s).

#### 7H-s-Triazolo[3,4-b][1,3,4]thiadiazines (IVa – IVc).

To a solution of 0.01 mole of compound II(a – c) in 30 ml of acetic acid was added 0.02 mole of 98.9% hydrazine hydrate. The mixture was boiled for 4 h and cooled, after which the precipitated product was filtered.

Compounds IIa – IIc, IIIe – IIIf, and IVa – IVc appear as crystalline substances insoluble in water and soluble in most of the common organic solvents (acetone, acetic acid, DMF, etc.). Compounds IIa – IIc and IVa – IVc are colorless, while compound IIIb is red and compounds IIIa and IIIc – IIIf and IVa – IVc have a yellow-orange color.

## EXPERIMENTAL BIOLOGICAL PART

The antimicrobial activity of the synthesized compounds was determined by the conventional method of double serial dilutions in a meat-extract broth. The stock solutions, prepared by dissolving 100 µg of each compound in 1 ml of the plain broth, were diluted to an initial working concentration of 12.5 µg/ml. Then, sequential half-diluted solutions were placed into tubes and test microbe cultures were added to a load of  $2.5 \times 10^5$  CFU/ml. The tests were performed with clinical strains of *E. coli*, *P. vulgaris*, *Anthracooides*, *St. aureus*, *Citrobacter*, *Enterobacter*, and *Ps. aeruginosa*. The cultures were incubated for 72 h at 37°C and for 48 h at 25°C.

The antimicrobial activity was judged by inhibition of the test culture growth after incubation for 1, 2, 3, or 5 days. The bactericidal activity was determined by additional tests, whereby the media with no visible growth were inoculated onto dishes with a meat-extract agar and into tubes with the meat-extract broth. These samples were incubated over three days and the results were evaluated after 24, 48, or 72 h [16].

The tuberculostatic activity of compounds IIa, IIIc – IIIf, and IVc *in vitro* was studied by the method of serial dilutions using the Levenstein – Jensen dense egg culture medium. Prior to coagulation of the medium, the test compounds were added to a concentration of 25, 5, 1, and 0.2 µg/ml [17]. The tests were performed with the following cultures: *M. tuberculosis* (strain 192), *M. bovinus* (strain Vallee), and *M. Avium* (strain 14141). The initial culture suspensions were referenced to the bacterial turbidity standard corresponding to  $500 \times 10^6$  BCG microbial bodies per ml, diluted 1 : 10 with a physiological solution, and introduced (0.2 ml) into each test tube containing the culture medium with (or without, for the control) substances studied. The samples were incubated at 37°C for a time period corresponding to the optimum incubation time for each culture studied (7 to 30 days).

It was found that the compounds studied exhibited no significant antibacterial activity (MTD > 100 – 200 µg/ml). Some inhibition of the growth of tuberculosis mycobacteria was observed in the presence of compound IVc.

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