

CONTROLLING BENZYLIC FUNCTIONALITY AND STEREOCHEMISTRY: 1. SYNTHESIS OF THE SECOPSEUDOPTEROSIN AGLYCONE

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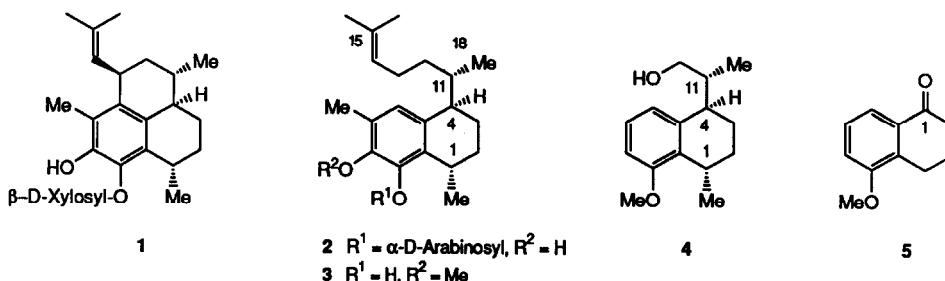
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Abstract: Directed, homogeneous hydrogenation of 1-(1-hydroxymethylethyl)-5-methoxy-3,4-dihydronaphthalene (7), followed by protection and selective benzylic oxidation gave the 1-oxo-(4*R**, 11*R**) compound (13). After addition of MeCeCl₂, the natural C-1 stereochemistry was established by intramolecular hydride delivery from the di-*t*-butylsilyl ether. Final elaboration of the sidechain and the Ar ring substituents gave the secopseudopterosin aglycone ether (3).

Pseudopterosin A (1)¹ and secopseudopterosin A (2)² are members of a family of diterpenes isolated from *pseudopterosorgia* sp. by Fenical and co-workers. The potent anti-inflammatory activity of these substances³ gives good reason to regard them as targets for flexible, stereocontrolled syntheses. Two routes to the tricyclic system have appeared,^{4,5} having in common the annelation of an aromatic ring onto a terpene-derived unit in which three stereocenters were already established. In this paper, we outline a conceptually different, tetralone-based route to the racemic secopseudopterosin aglycone ether (3), with >20:1 relative stereocontrol at each stereocenter.

Our route to key intermediate (4) from 5-methoxytetralone (5) utilises a directed hydrogenation in conjunction with a selective functionalisation of a benzylic methylene group. The final stereocenter is established by a different type of directed reduction, using *intramolecular ionic hydrogenation* as a new method for controlling benzylic stereochemistry. In the accompanying paper, we describe the conversion of (4) to the tricyclic series, and other aspects of chemo- and regioselectivity in benzylic carbonium ion chemistry.



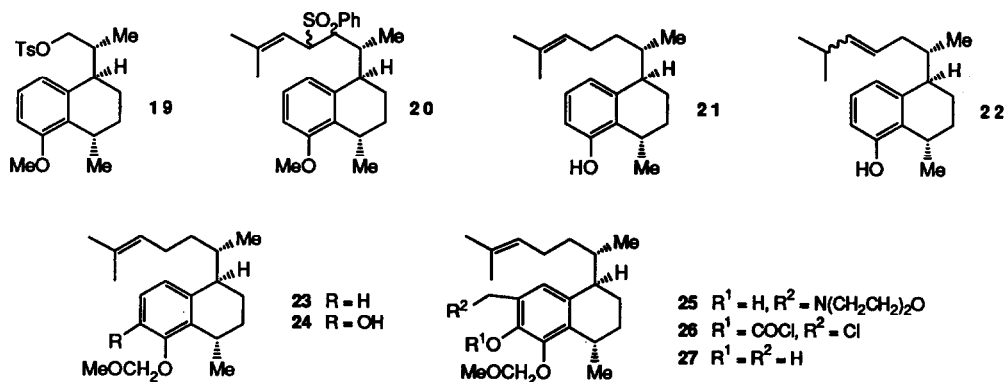
The conversion of (5) to (4) is shown in Scheme 1. Reformatsky reaction⁶ [Zn, MeCHBrCO₂Et, Me₃SiCl activation, THF, 65°], followed by dehydration [MsOH, 1,2-C₂H₄Cl₂, 80°] afforded racemic⁷ ester (6), which was reduced [NaH₂Al(OCH₂CH₂OMe)₂, Et₂O] to homoallylic alcohol (7) [mp 63-65°; 75-85%]. Dihydronaphthalenes (6) and (7) have a strongly preferred solution conformation,⁸ depicted for (7), which minimises interactions with the *peri*-ArH and differentiates the diastereotopic faces of the olefin, permitting stereoselective reduction.

Reagents (see text for details): **a**: Zn, MeCHBrCO₂Et; MsOH. **b**: Red-Al. **c**: C₆Rh(PPh₃)₃, H₂, *t*-BuOK or H₂, Pd-C; *p*-NO₂C₆H₄COCl, py. **d**: K₂S₂O₈, CuSO₄. **e**: PCC; NaOMe. **f**: MeCeCl₂; TsOH; *t*-Bu₂SiHCl, imidazole. **g**: CF₃CO₂H (high dilution); *n*-Bu₄NF. **h**: H₂, Pd-C.

Although normal reduction [H_2 , Pd-C] of (7) gave a 3:2 mixture of (8) and epimer (9)⁹ [δ values (CDCl_3) for CHCH_3 in (8) and (9) were 0.77 and 1.04, respectively], directed homogeneous reduction¹⁰ by the method of Thompson¹¹ [0.05 eq. $\text{CIRh}(\text{PPh}_3)_3$, 0.1 eq. $t\text{-BuOK}$, 60 psi H_2 , THF, 23°, 50 h] afforded (8)/(9) in >95:5 ratio. Complete purity was secured upon one recrystallization of the derived *p*-nitrobenzoate (10) [mp 90-92°; 86% from (7)]. Selective oxidation was obtained by modifying a known procedure:¹² treatment of (10) with $\text{K}_2\text{S}_2\text{O}_8$ (2 eq.), CuSO_4 (0.2 eq.) and *sym*-collidine¹³ (2 eq.) in $\text{MeCN-H}_2\text{O}$ [1:1, 80°, 1.5 h] gave a mixture of alcohols (11)/(12) and ketone (13). Oxidation of this mixture [PCC ,¹⁴ celite, CH_2Cl_2] provided (13) [63% from (10)]. Following hydrolysis, treatment with MeCeCl_2 ¹⁵ [THF, -70 to 23°] and subsequent dehydration [TsOH] gave olefin (15) in 72% yield from (13).

Olefin (15) was inert to $\text{H}_2/\text{t-BuOK}/\text{CIRh}(\text{PPh}_3)_3$, but was reduced [H_2 , Pd-C] from the less hindered face to isomer (16), epimeric¹⁶ at C-1 with the natural series. With $\text{Et}_3\text{SiH}-\text{CF}_3\text{CO}_2\text{H}$, a 2:3 mixture of (16) and the desired (4) was produced [δ values (CDCl_3) for CHCH_3 : for (4), 0.87 and 1.16, and for (16), 0.68 and 1.18]. The rapid, clean nature of this "ionic hydrogenation"¹⁷ suggested the application of an *intramolecular* version¹⁸ to secure the needed stereochemistry. Accordingly, (15) was converted [2 eq. $\text{t-Bu}_2\text{SiHCl}$, 3 eq. imidazole, DMF, 23°] to the ether¹⁹ (18). Syringe pump addition²⁰ [16-20 h] of (18) in CH_2Cl_2 to $\text{CF}_3\text{CO}_2\text{H}$ [5 eq., 0.1M in CH_2Cl_2] followed by desilylation [$n\text{-Bu}_4\text{NF}$, THF, 23°] gave (4) of >95% isomeric purity in 65-75% yield from (15), 19-24% overall from 5-methoxytetralone.

Alcohol (4) was converted to the aglycone ether (3) as follows: tosylate (19), treated with $\text{Me}_2\text{C}=\text{CHCH}(\text{Li})\text{SO}_2\text{Ph}$ [THF, -70 to 23°], afforded diastereoisomer mixture (20).²¹ Desulfonation of (20) [Li , EtNH_2]²² was accompanied by demethylation to yield (21) containing 8-13% of (22). After protection [MeOCH_2Cl , $\text{t-Pr}_2\text{NEt}$] to afford (23), metalation²³ was effected [t-BuLi , Et_2O , 0°]. Workup with $\text{B}(\text{OMe})_3$ followed by $\text{H}_2\text{O}_2\text{-H}_2\text{O-K}_2\text{CO}_3$ ²⁴ gave catechol ether (24), and Mannich reaction [aq. CH_2O , morpholine, EtOH , 80°] produced (25).



Mannich base (25) was resistant to both hydride reagents and Na-NH_3 , but was efficiently converted to the chloromethyl compound (26) [3 eq. CCl_3OCOCI , 4 eq. *sym*-collidine or $\text{t-Pr}_2\text{NEt}$, CH_2Cl_2 , $0-23^\circ$, 6 h]. Reduction [NaBH_4 , DMSO] followed by hydrolysis²⁵ [NaOH , aq. EtOH] gave C6-methyl compound (27), which was O-methylated [MeI , K_2CO_3 , acetone]. Final hydrolysis of the MOM ether [TsOH , MeOH] then provided (3), identical (TLC, MS, 400MHz PMR) with a sample prepared² from secopseudopterosin.

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References and Notes:

1. Look, S. A.; Fenical, W.; Matsumoto, G. K.; Clardy, J. *J. Org. Chem.*, **1986**, *51*, 5140-5145
2. Look, S. A.; Fenical, W. *Tetrahedron*, **1987**, *43*, 3363-3370.
3. Look, S. A.; Fenical, W.; Jacobs, R. S.; Clardy, J. *Proc. Nat. Acad. Sci. USA*, **1986**, *83*, 6238-6240.
4. Broka, C. A.; Chan, S.; Peterson, B. *J. Org. Chem.*, **1988**, *53*, 1584-1586.

5. Corey, E. J.; Carpino, P. J. *Am. Chem. Soc.*, **1989**, *111*, 5472-5474.
6. Review: Furstner, A. *Synthesis*, **1989**, 571-590.
7. The derived acid is an obvious point for chiral intervention; these studies are in progress.
8. The *peri*-ArH in (6) showed a strong NOE with the allylic methine, but very weak effects with other protons.
9. Over Pd-C, (6) was reduced with 94:6 selectivity for the *epimeric* series. This provides potentially useful intermediates for the serrulatane series of diterpenes (see: Croft, K. D.; Ghisalberti, E. L.; Jefferies, P. R.; Proudfoot, G. M. *Aust. J. Chem.*, **1981**, *34*, 1954-1957.). For a very recent synthesis of dihydroxy serrulatic acid *via* Tetralin-Cr(CO)₃ complexes, see; Uemura, M.; Nishimura, H.; Hayashi, Y. *Tetrahedron Letts.*, **1990**, *31*, 2319-2322.
10. Review: Brown, J. M. *Angew. Chem. Int. Edit. Engl.*, **1987**, *26*, 190-203.
11. Thompson, H. W.; McPherson, E. *J. Am. Chem. Soc.*, **1974**, *96*, 6232-6233. This process, which utilises the anionic species to ensure entry into the coordination sphere of the metal complex, permits the use of the air-stable ClRh(PPh₃)₃, and also allowed highly diastereoselective reduction of the corresponding *carboxylate* [H₂, ClRh(PPh₃)₃ in EtOH-H₂O-NaHCO₃]. LiAlH₄ reduction of the dihydroacid afforded pure (8).
12. Bhatt, M. V.; Perumal, P. T. *Tetrahedron Letts.*, **1981**, *22*, 2605-2608.
13. This more hindered buffer was tried on the assumption that pyridine (which gave ca. 30% of the alcohol/ketone mixture) was intercepting radical and/or cationic intermediates.
14. Corey, E. J.; Suggs, J. W. *Tetrahedron Letts.*, **1975**, 2647-2650.
15. Imamoto, T.; Sugiura, Y.; Takiyama, N. *Tetrahedron Letts.*, **1984**, *25*, 4233-4236. Other organometallics (MeMgBr, Ph₃P=CH₂ and Me₃SiCH₂MgCl) caused extensive enolate formation from (14).
16. Confirmed by an X-ray structure determination on the *p*-nitrobenzoate (17). Details may be obtained from A. T. McP.
17. Review: Kursanov, D. N.; Parnes, Z.; Loim, N. M. *Synthesis*, **1974**, 633-651.
18. One reaction of this type, involving reduction of a ketone through a 6-membered transition state, has been described: Anwar, S.; Davis, A. P. *J. C. S. Chem. Commun.* **1986**, 831-832.
19. RCH₂OSiHt-Bu₂ showed acid stability similar to RCH₂OSiMe₂t-Bu; the less hindered RCH₂OSiHMe₂ and RCH₂OSiHPh₂ were labile to weak acids and to silica gel.
20. From rapid addition of excess CF₃CO₂H to solutions of (18), the ratios of (3):(16) were 3:2 at 5x10⁻²M and 5:1 at 5x10⁻³ M. High dilution ensured dominance of the intramolecular transfer, through the 9-membered transition state depicted in Scheme 1. We believe that this type of process will be of general use for stereocontrol, and we are currently examining the scope and limitations with respect to tether, ring size and functionality in systems involving benzylic, oxocarbenium and iminium ions.
21. A variety of cuprate-based, direct prenylations of (19) or the corresponding chloride gave regioisomer mixtures.
22. Greico, P. A.; Masaki, Y. *J. Org. Chem.*, **1974**, *39*, 2135-2136.
23. The MeO group was not an effective director of *ortho*-metalation in 5-methoxytetralins.
24. Compare: Kidwell, R. L.; Murphy, M.; Darling, S. D. *Org. Syn. Coll. Vol. V*, Wiley, New York, **1973**, 918-921.
25. The major product from the NaBH₄ step was the symmetrical carbonate of phenol (27).
26. All new compounds gave satisfactory elemental analyses and/or high resolution mass spectra.

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