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A Convergent Synthesis of the *tris*-Oxazole Ring System in Ulapualide A and Related Marine Metabolites

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Abstract: A convergent synthesis of the fully functionalised *tris*-oxazole system 16, found in the ulapualide family of marine macrolides, involving elaboration of the substituted oxazoles 13 and 8, followed by their coupling to the amide 14 and manipulation of the third oxazole ring *via* the oxazoline intermediate 15, is described.

Natural product structures based on the presence of three contiguous 2,4-disubstituted oxazole rings are unique to the family of marine metabolites known as ulapualides, eg ulapualide A 1. The tris-oxazole unit 2 in these metabolites is most likely derived by cyclodehydration of an appropriately substituted tris-serine precursor, eg 3, Y=CO₂R leading to the corresponding tris-oxazoline, followed by enzymic oxidation. A related, bis-oxazole, unit is found in the natural product hennoxazole A isolated from Polyfibrospongia sp, and muscoride A found in the freshwater cyanobacterium Nostac muscorum shows a bis-oxazole core which is formally derived from two threonine residues.

The first synthesis of the novel tris-oxazole unit in ulapualide A was described by ourselves, 6 and used a biomimetic type approach from three molecules of serine and three sequential oxazoline cyclisationoxidation reactions. Other approaches to the same tris-oxazole unit have been described more recently, which highlight the scope for the Hantzsch oxazole synthesis⁷ and for [3+2] cycloaddition reactions of acylcarbenes to nitriles in the elaboration of oxazoles.8 Like our approach however, these alternative methods have been used in a linear, step-wise fashion. In relation to our studies of the total synthesis of ulapualide A⁹ we required a convergent approach to the *tris*-oxazole unit in the natural product, which would permit the elaboration of the central oxazole ring as a final step, ultimately in an intramolecular fashion (see Scheme 1). In this Letter we demonstrate the scope for this approach with a concise synthesis of the tris-oxazole 16 involving elaboration of the substituted mono-oxazoles 13 and 8, followed by their coupling to the amide 14 and manipulation of the third ring in 16 via 15.

Thus, treatment of the serine-derived oxazolidine acid 4^{10} with serine methyl ester HCl first gave the corresponding amide 5 which on reaction with Burgess reagent¹¹ led to the oxazoline 6 as a mixture of

diastereoisomers. Oxidation of this mixture with CuBr/t-BuOOH¹² next led to the oxazole 7 which on deprotection with TFA produced the salt 8. Using similar chemistry, the oxazole acid chloride 13 was derived from the benzyl ester 9 of serine¹³ following conversion to the amide 10, cyclisation to 11, oxidation of the oxazoline 11 to the oxazole 12, and finally manipulation of the acid ester to the corresponding acid chloride 13.¹⁴

A coupling reaction between 13 and 8 in the presence of triethylamine then led to the intermediate *bis*-oxazole amide 14 which on cyclodehydration using Burgess reagent¹¹ followed by oxidation of the resulting *bis*-oxazole oxazoline 15 provided the doubly-differentiated

BOC NH
$$O_2$$
C O_2 Me O_2 Me

Reagents and conditions: i, Serine OMe HCl,DCC / HOBt / Et₃N,0° →r.t,18h,79%; ii,Burgess'reagent,THF,90°,1.5h,68%; iii,CuBr / Cu(OAc)₂/t-BuOOH,Benzene,reflux,5h, 78%;iv,dil.TFA,rt,1h,50%

Scheme 2

Reagents and conditions: i,AcOCH $_2$ Cl,Et $_3$ N,0° \longrightarrow r.t.,18h,86%;ii,Burgess'reagent,THF,90°, lh,66%;iii,CuBr $_2$,DBU,HMTA,CH $_2$ Cl $_2$,r.t.,18h,58%;iv,H $_2$,Pd-C,EtOAc,16h,100%;v,SOCl $_2$, eflux,4h,quantitative

Scheme 3

$$13 + 8$$

$$0 \longrightarrow NH$$

$$HO \longrightarrow NCO_2Me$$

$$14$$

$$15$$

$$0 \longrightarrow N$$

$$CO_2Me$$

$$CO_2Me$$

$$16$$

Reagents and conditions: i,Et₃N,0° \longrightarrow rt,18h,69%; ii,Burgess' reagent,THF,90°,2h,77%; iii,NiO₂,benzene,reflux,4h,40%

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tris-oxazole 16 (Scheme 4). This convergent synthesis of the tris-oxazole unit in ulapualide A, which proceeds in an overall yield of 7% from 9 has many attractions over the linear approaches described earlier, not least its possible application as the final macrocyclisation step to natural ulapualides. The development of this design, alongside others, towards a total synthesis of ulapualide A are in progress in our laboratories.

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- (14) All new compounds showed satisfactory spectroscopic data together with microanalytical and/or accurate mass spectrometry data. Typical procedure: conversion of 14 into 15: A solution of Burgess reagent (50 mg, 0.21 mmol) in THF (1.5 ml) was added to a solution of 14 (65 mg, 0.18 mmol) in THF (1.5 ml) and the mixture was heated in a sealed tube at 90°C for 1.5 hour. The mixture was cooled to room temperature, and then concentrated in vacuo to leave an oil. Purification by chromatography on silica gel using ethyl acetate as eluent gave 15 (50 mg, 77%) as a colourless oil. conversion of 15 into 16: Freshly prepared NiO₂ (150 mg) was added in three portions to a refluxing solution of 15 (50 mg, 0.15 mmol) in dry benzene (3 ml) at one hour intervals. The mixture was heated under reflux for two more hours, and then filtered through celite. The filtrate was concentrated in vacuo to leave a viscous mass. Purification by chromatography on silica gel using ethyl acetate as eluent gave 16 (21 mg, 40%) as a colourless solid.